

6-28-2024

The discovery of first RET PROTAC with *in vivo* activity

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The Discovery of First RET PROTAC with *in vivo* Activity

by

Yafeng Wang

A dissertation submitted in partial fulfillment
of the requirements for the degree of
Doctor of Philosophy
Department of Chemistry
College of Arts and Sciences
University of South Florida

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Date of Approval:
June 28, 2024

Keywords: PROTAC, Sulfono- γ -AApeptide, SPPS, Michael receptor, HPLC purification,
Cleavage cocktail

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ACKNOWLEDGEMENTS

I would like to thank my major professor, Dr. Jianfeng Cai, for his guidance and patience during these five years. Whenever I was stuck with my project, he would offer help, was very patient with my slow progress, and encouraged me to learn to solve research problems. It is my great honor to work in this lab and I have learnt so much from him.

Moreover, I would like to thank all my committee members, Dr. David Merkler, Dr. Feng Cheng, and Dr. Wenqi Liu. They have encouraged me and given me suggestions since my first year in 2020. Also, I must thank my previous committee member, Dr. Wayne Guida. I attended his advanced organic class that helped me learn deeper about organic chemistry. As a committee member, he shared his great vast treasure of knowledge to help me pass each milestone.

In our lab, I must thank Dr. Lulu Wei, she taught me how to run reactions and do research. I also need to thank Dr. Yizhan Zhai, and previous senior graduate students, such as Dr. Lei Wang, Dr. Songyi Xue, Huaxuan Yu, Jun Ou, Yu Yu Win, Haohui Ye. They offered great help and convenience for lab research and life. Furthermore, I would like to thank the undergraduate student, Ting Chao, she helped me prepare important intermediates and taught me new generation fancy techniques.

I would like to thank all my collaborators, Dr. Jie Wu, Dr. Jiandong Chen, Dr. Danny Chou. I appreciate their sticking and biological experiment help with my research projects.

I am also grateful for other graduate students and faculty members in the chemistry department who shared their happy time and experience with me.

Finally, I really appreciate the help from my family and my high school friend, Jun Cao. They are the spark, the inspiration, the guide, and the candle in my life.

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LIST OF ABBREVIATIONS

Abbreviations	Definitions
AA	Amino acid
BB	Building block
Boc	<i>tert</i> -Butoxycarbonyl
Bn	Benzyl
DCM	Dichloromethane
Cbz	Benzyloxycarbonyl
DCC	1,3-Dicyclohexylcarbodiimide
DIC	<i>N, N</i> -Di(<i>propan-2-yl</i>) methanediimine
DMF	<i>N, N</i> -Dimethylformamide
DIPEA	Diisopropylethylamine
DMAP	4-Methylaminopyridine
DMSO	Dimethylsulfoxide
EDCI	1-ethyl-3-[3-(dimethylamino)propyl] carbodiimide
Pd ₂ (dba) ₃	Tris(dibenzylideneacetone)dipalladium (0)
Pd ₂ Cl ₂ (PPh ₃)	Bis(triphenylphosphine)palladium chloride
Fmoc	9-Fluorenylmethoxycarbonyl
HOBt	1 <i>H</i> -1,2,3-Benzotriazol-1-ol
HPLC	High-performance liquid chromatography
IC ₅₀	Half-maximal inhibitory concentration
MIC	Minimum inhibitory concentration
NMM	<i>N</i> -methylmorpholine
Pbf	2,2,4,6,7-pentamethyldihydrobenzofuran-5-sulfonyl
PG	Protecting group
PROTAC	Proteolysis targeting chimera
SNAr	Nucleophilic aromatic substitution
SPPS	Solid phase peptide synthesis
TEA	Triethylamine
THF	Tetrahydrofuran
T _{1/2}	Half life
Ts	Tosyl
Trt	Tryl
Xantphos	(9,9-Dimethyl-9 <i>H</i> -xanthene-4,5-diyl) Bis(diphenylphosphane)

ABSTRACT

This report discusses the development of RET Proteolysis targeting chimera PROTAC molecules based on Selpercatinib (LOXO-292) and Sulfonyl- γ -AApeptide PROTAC. A PROTAC molecule is a bifunctional molecule composed of two active domains and a linker. It works by inducing selective protein intracellular degradation. This technology was first discussed by Kathleen Sakamoto, Ray Deshaies and Craig Crews in 2001. Using various E3 Ligase, such as CRBN, pVHL, beta-TrCP1, Mdm2, DCAF15, DCAF16, RNF114, and c-IAP1, the PROTAC technology has been applied to several novel drug development. Yale University licensed the PROTAC technology to Arvinas in 2013–14. For my research, Chapter one summarizes our development of RET PROTAC in detail. Using Selpercatinib as warhead binding to RET protein, different methylene, PEG and heterocycle linkers, various E₃ ligases are also tried to develop RET proteac molecules. Chapter two discusses our novel design of covalent inhibitors that targets RET G810C mutation using different Michael receptors. Chapter three concludes our design of unnatural γ -AApeptide to overcome antibiotic resistance due to the abuse of current antibiotics. Chapter four starts with the basic knowledge of solid phase synthesis. After the brief review, our research on the SPPS of BCL-9 sulfonyl- γ -AApeptide PROTAC linked with VHL and PG E₃ ligase is discussed. Details of sulfonyl- γ -amino acid building block synthesis, procedure of SPPS, peptide cleavage and purification are also summarized in detail. Chapter five starts with the background unnatural amino acids and foldamers. Then our design based on natural peptide S597 with our sulfonyl- γ -amino acid to mimic the sequence of site 1 or replacing natural amino acids with our

sulfono- γ -amino acid are discussed. At the end of each chapter, NMR, Q-TOF, ESI-MS, and LC-MS data are attached.

CHAPTER ONE: FIRST RET DEGRADER WITH *IN VIVO* ACTIVITY

1.1 Introduction

The rearranged-during-transfection (*RET*) gene encodes the RET receptor protein tyrosine kinase (PTK)¹⁻³. Aberrantly activated RET kinase caused by germline or somatic *RET* mutations is one of the major oncogenic drivers frequently found in medullary thyroid cancer (MTC)^{2, 3}. Chromosomal rearrangements that result in aberrant expression and constitutive activation of the fusion RET kinases are associated with ~10% papillary thyroid cancer (PTC) and ~2% non-small cell lung cancer (NSCLC) and are proven driver oncogenes^{1, 4}. Furthermore, RET alterations were observed at a low rate in diverse cancer types that contribute to tumorigenesis and maintenance of these RET-positive cancers, which are approved targets for cancer therapy⁵⁻⁷. Although more than 100 different RET fusion genes have been detected³, the most frequently observed RET fusions are CCDC6-RET and KIF5B-RET fusions in PTC and NSCLC.

Two RET-selective protein tyrosine kinase inhibitors (TKIs), selpercatinib (LOXO-292)⁸ and pralsetinib (BLU-667)⁹, have been approved for treating RET-positive cancers^{1, 3}. These are ATP competitive RET inhibitors that inhibit the RET kinase activity¹⁰. Inhibition of RET kinase activity by these inhibitors does not reduce the RET protein level. Instead, inhibition of the RET kinase activity by selpercatinib or pralsetinib consistently increased the CCDC6-RET fusion protein level in human cancer cells¹¹ (see also Results below). A higher level of RET fusion oncogene expression has been implicated in the reduced effectiveness of RET kinase inhibitors, such as in the case of RET inhibitor RXDX-105^{1, 12}. While the effectiveness of selpercatinib and pralsetinib at their full prescription strength may not be impacted by a higher level of oncogenic

RET expression, dose reduction is required in patients who have treatment-related adverse events. In fact, 30% of thyroid cancer patients and 41% of NSCLC patients treated with selpercatinib had dose reduction^{13, 14}. Conceivably, the efficiency of these occupancy-based drugs may be reduced at a lower drug dose. This may contribute to the persistence of residual tumors that exist in over 90% of patients in selpercatinib- or pralsetinib treatment³. Thus, depletion of oncogenic RET protein is likely to have an added benefit in addition to inhibition of the oncogenic RET kinase activity in RET-targeted cancer therapy.

Proteolysis targeting chimera (PROTAC) is a developing modality to achieve targeted protein degradation with heterobifunctional small chemical molecules^{15, 16}. A PROTAC brings an E3 ligase into close proximity to the target protein of interest to form a ternary complex, resulting in ubiquitination and proteasomal degradation of the target protein. A PROTAC typically consists of three moieties: a ligand that engages the target protein of interest, a ligand of an E3 ubiquitin ligase that recruits a ubiquitin ligase, and a linker that tethers these two ligands¹⁵. For protein kinases, a kinase inhibitor may be used as the basis for the design of the target-binding ligand^{17, 18}. A ligand to the von Hippel-Lindau tumor suppressor (VHL) protein and immunomodulatory imide drugs (IMiDs) that bind cereblon (CRBN) are the most commonly used ligands for E3 ubiquitin ligases in PROTAC design. The VHL ligand recruits the CUL2-RBX1-ElongB/C-VHL (CRL2^{VHL}) ubiquitin ligase¹⁹. IMiDs, including thalidomide, lenalidomide, and pomalidomide, recruit the ubiquitously expressed cullin ring ligase 4 (CUL4)-RBX1-DDB1-CRBN (CRL4^{CRBN}) E3 ligase^{16, 20}. More recently, phenyl glutarimide (PG) analogues of CRBN binders were used to improve the chemical stability of CRBN ligands²¹. In addition to high-affinity ligands for binding the target protein and the E3 ligase, an appropriate linker that connects the two ligands is critical for the formation of a productive ternary complex to allow efficient degradation of the target

protein. An optimal linker is determined by its composition, length, and the position of coupling to the ligands^{22, 23}.

In this study, we designed, synthesized, and evaluated in cell cultures *in vitro* and in an animal model *in vivo* RET PROTACs based on selpercatinib (LOXO-292) as the RET ligand, IMiDs-Lenalidomide, and PG as the E3 ligase ligand, and various flexible and rigid linkers of different compositions and lengths coupling to different positions of the IMiDs or PG. Our investigation identified YW-N-7 as a RET PROTAC that could fully degrade RET in the cell lines tested and demonstrated potent *in vivo* activity in inhibiting oncogenic RET-driven tumor growth in animals.

1.2 Results

1.2.1 Design of PROTAC

Our design was based on LOXO-292 (selpercatinib), an FDA-approved RET kinase inhibitor to treat lung or thyroid cancer (Figure 1A). We recently published the co-crystal structure of LOXO-292/RET (PDB: 7JU6),¹⁰ which reveals that the terminal hydroxypropyl group of LOXO-292 has minimal impact on its binding to RET (Figure 1B and 1C). Unlike the rest of the structure buried in the RET hydrophobic cleft, this hydroxypropyl group (shaded in light blue) is outside the protein and exposed to the solvent. No interactions are identified between this group and the protein based on the crystal structure. We thereby postulated that this group could be replaced with certain linkers to link an ubiquitin E3 ligase binding ligand to make PROTAC compounds for degrading RET. We decided to choose lenalidomide which is known to target the E3 ligase cereblon (CRBN) (Figure 1C). In parallel, we also attempted to select phenyl glutarimide (PG) moiety as PG is also a CRBN ligand but is believed to have enhanced chemical

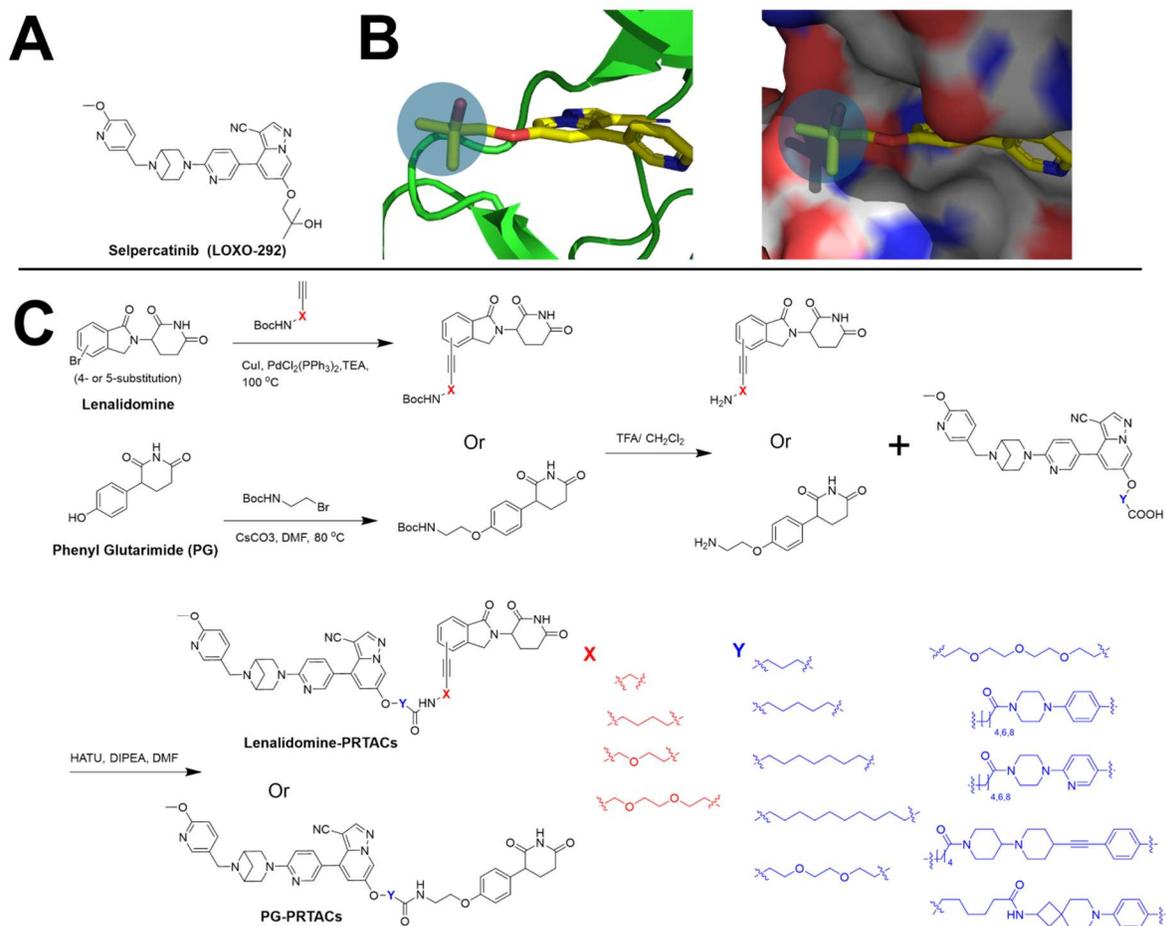


Figure 1.1. Design of PROTAC. A. The structure of LOXO-292. B. The co-crystal structure of LOXO-292/RET complex (PDB: 7JU6) shown in cartoon and surface representation, respectively. C. The synthetic routes of lenalidomine and phenyl glutarimide (PG) based PROTAC derivatives of LOXO-292.

stability (Figure 1C). To this end, a variety of PROTAC compounds bearing various flexible and rigid linkers of different compositions and lengths were synthesized and studied for their ability to inhibit RET activity and degrade RET protein (Table 1).

1.2.2 SAR study

Table 1.1. PROTAC compounds and their ability to degrade RET at 50 nM; The three most potent lead compounds are shown in red.

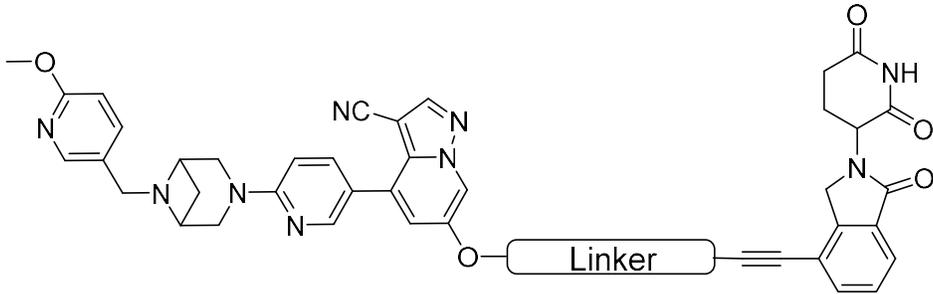
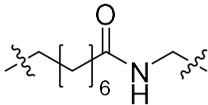
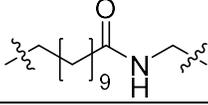
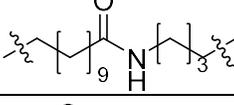
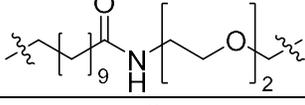
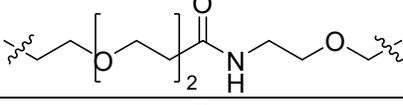
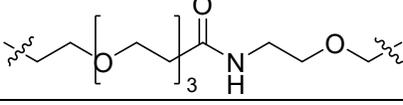
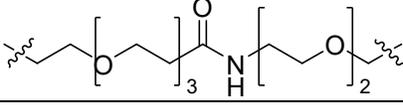
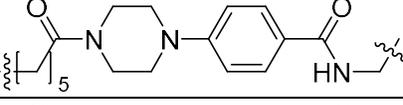
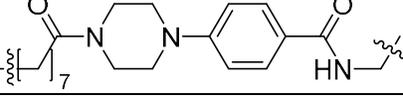
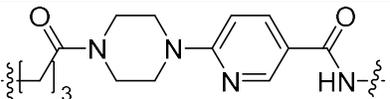
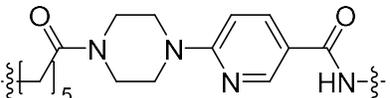
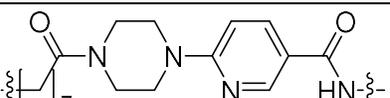
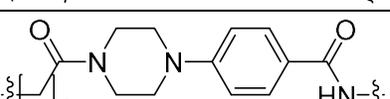
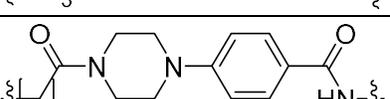
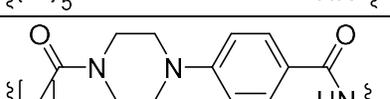
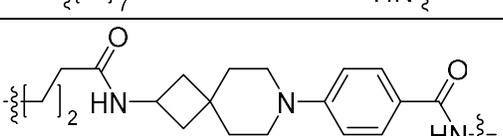
		
Manuscript Name	Linkers	Degradation rates at 50 nM (%)
YW-L-1		-73
YW-L-2		29.7
YW-L-3		43.1
YW-L-4		-4.8
YW-L-5		43.8
YW-L-6		-38.3
YW-L-7		-20.3
YW-L-8		36
YW-L-9		33

Table 1 1. Continued

YW-L-10		69
YW-L-11		76.2
YW-L-12		84
YW-L-13		96
YW-N-1		-105
YW-N-2		-75
YW-N-3		-133
YW-N-4		40

Table 1 1. Continued

YW-N-5		68
YW-N-6		37
YW-N-7		88
YW-N-8		60
YW-N-9		82
YW-N-10		57
YW-N-11		92

1.2.3 LOXO-292 and BLU-667 treatment

To assess the impact of RET kinase inhibitors on the RET protein, we first examined the RET protein level in the presence of two FDA-approved drugs Selpercatinib (LOXO-292) and Pralsetinib (BLU-667). Surprisingly, upon treatment of both compounds, the level of RET protein dramatically increased (Figure 2A and Figure 2B), in contrast to the intuition that LOXO-292 and BLU-667 should change the phosphorylation level but not alter the total RET protein level. This may be owing to the ability of cancer cells to express more RET protein due to the decreased level of RET kinase activity upon the inhibition of LOXO-292 and BLU-667, respectively. This phenomenon has been implicated in the reduced effectiveness of RET kinase inhibitors.

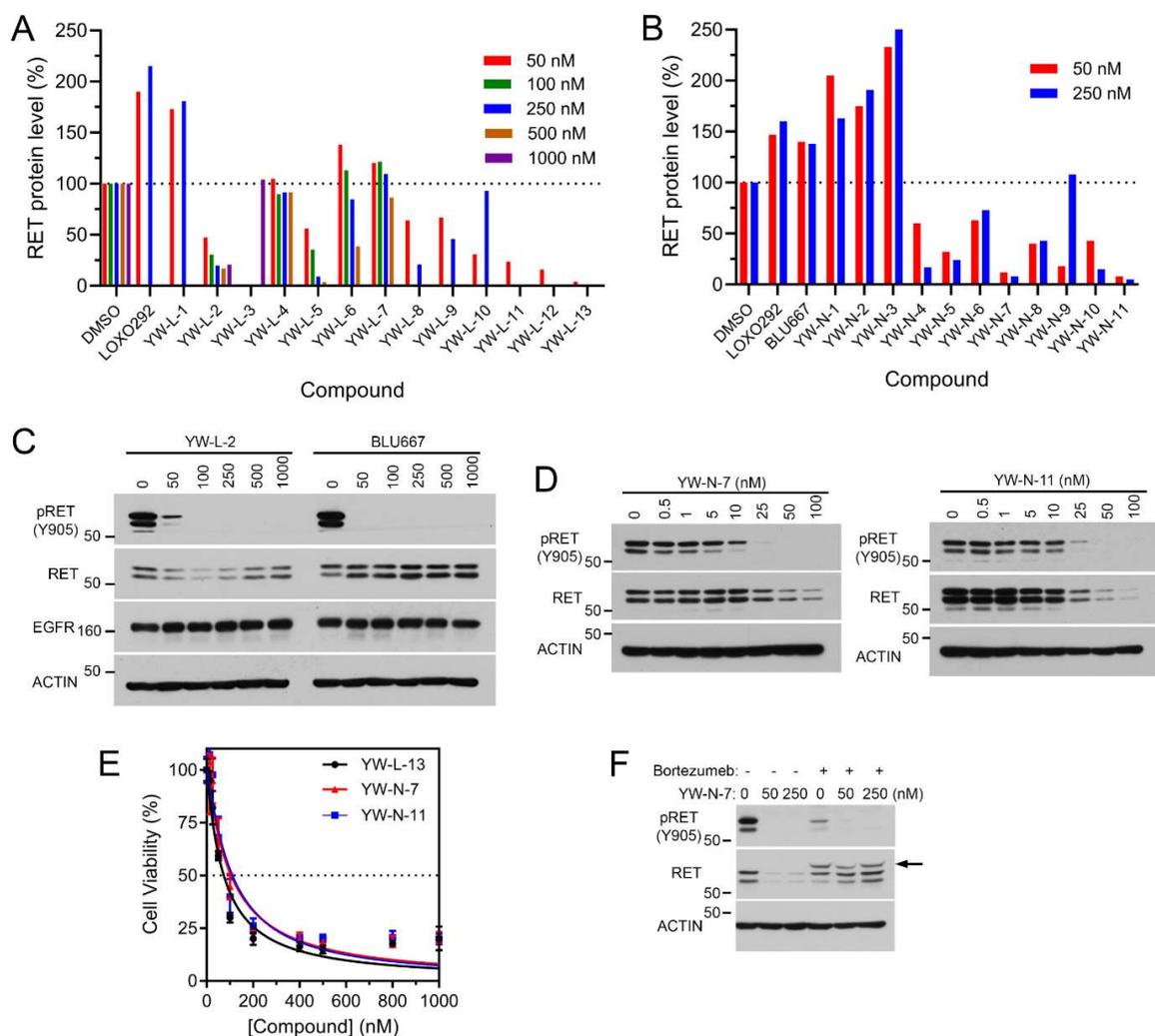


Figure 1.2. Evaluation of RET PROTACs. **A, B:** Degradation of the RET protein. **C:** RET kinase inhibition, RET and EGFR protein degradation by YW-L-2 and Blu 667 **D:** Concentration-dependent inhibition and degradation of RET by YW-N-7 and YW-N-11. **E:** cell cytotoxicity of YW-L-13, YW-N-7 and YW-N-11 **F:** Bortezumeb's effect on YW-N-7-induced CCDC6-RET degradation.

1.2.4 Lenalidomide-based PROTAC degradation result

To assess whether our design strategy could lead to RET protein degradation, we first synthesized a series of conjugates of lenalidomide and LOXO-292 bearing different linkers (Table 1 and Figure 2A). Our first attempt, **YW-L-1**, showed no capability to degrade RET. Instead, it seemed that the compound boosted the RET expression, similar to the effect of LOXO-292 (Figure 2A and 2B) and BLU-667 (Figure 2B). However, elongation of the linkers led to compounds **YW-**

L-2 and **YW-L-3** which started to exhibit activity to degrade RET. This initial success demonstrated the strategy of linking LOXO-29 and lenalidomide for targeted degradation of RET is possible. The distance between RET and the E3 ligase is crucial, as longer linkers seem to have stronger activity in degrading RET. However, none of the compounds were able to completely degrade RET at higher concentrations (Figure 2C). The western blot is consistent with the screening (Figure 2A) that BLU-667 treatment led to elevated expression level of RET (Figure 2C). We next examined whether a hydrophobic or hydrophilic linker was ideal for the degradation of RET. Interestingly, the replacement of hydrophobic linkers with hydrophilic linkers only gave promiscuous SAR. Except for **YW-L-5**, the other three compounds **YW-L-4**, **YW-L-6**, and **YW-L-7** did not exhibit any activity for RET degradation. Since longer hydrophobic linkers were favored, we speculated a rigid hydrophobic linker may lead to more potent RET degraders, as the distance between RET and CRBN would be more defined with the rigid spacer. Indeed, all the compounds from **YW-L-8** to **YW-L-12** showed good activity in degrading RET. In particular, **YW-L-12** could degrade 84% RET at 50 nM. Lastly, we also investigated the impact of the linkage position on the aromatic ring of the lenalidomide. Intriguingly, with the same linker used for **YW-L-11** in which the linkage point is at position 4, the linkage at position 5 led to an even more potent compound **YW-L-13** that could degrade 96% RET protein at 50 nM.

1.2.5 Phenyl glutarimide-based PROTAC for RET protein degradation

Since phenyl glutarimide (PG) is currently recognized as the alternative ligand to target CRBN E3 ligase but with enhanced stability in comparison to lenalidomide, we next studied the conjugates of PG and LOXO-292 for their potential as PROTAC compounds to degrade RET. Similar to lenalidomide-based PROTAC compounds, with shorter linkers **YW-N-1**, **YW-N-2** and **YW-N-3** did not show any inhibitory activity (Table 1), whereas **YW-N-4** bearing a longer linker

revealed 40% RET degradation at 50 nM. As anticipated, compounds from **YW-N-5** to **YW-N-11** all show significant degradation of RET protein due to the rigid linkers employed between PG and the parent compound LOXO-292, a trend already observed for lenalidomide PROTAC development. Among them, both **YW-N-7** and **YW-N-11** demonstrated the most potent activity in RET degradation (Table 1) on a dose-dependent manner (Figure 2D). As shown in Figure 2D, both **YW-N-7** and **YW-N-11** virtually completely degraded RET protein at 100 nM.

1.2.6 The cytotoxicity of YW-L-13, YW-N-7 and YW-N-11

Since RET is crucial for cancer cell growth, the inhibition of RET kinase activity or the eradication of RET protein itself are expected to induce cancer cell death. As shown in Figure 2E, the most potent compounds from both lenalidomide- and PG-based PROTACs **YW-L-13**, **YW-N-7** and **YW-N-11** all revealed potent anti-cancer activity by killing cancer cells effectively. The EC₅₀ ranged between 50-100 nMs, which are in good agreement with their ability to degrade RET protein.

1.2.7 YW-N-7 RET degradation with proteasome inhibitor

To ensure that our PROTAC compounds exerted their activity through proteasome-mediated protein degradation, we next chose to test the ability of **YW-N-7** to degrade RET in the presence of Bortezomib, an FDA-approved proteasome inhibitor. As shown in Figure 2, in the absence of Bortezomib, **YW-N-7** could completely inhibit RET phosphorylation and eradicate RET protein as low as 50 nM. In the presence of Bortezomib, **YW-N-7** could still function as a kinase inhibitor as LOXO-292 to block the phosphorylation of RET. However, since Bortezomib blocked the proteasomal enzymatic activity, the ability of **YW-N-7** to degrade RET was inhibited on a dose-dependent manner. Overall, this experiment demonstrated that our PROTAC compounds

exerted dual functional roles to inhibit RET by suppressing RET kinase activity and degrading RET protein.

1.2.8 Pharmacokinetics (PK) Test

To assess the potential of our PROTAC compounds as a research tool or potential RET-targeted therapy, we set out to carry out *in vivo* studies. We first evaluated the PK profile of the aforementioned lead compounds. As shown in Figure 3, among three compounds, **YW-L-13**, which exhibited the most potent *in vitro* activity, revealed a poor PK profile with a short half-life and ultrafast clearance. This is somehow in alignment with the reports that some CRBN-targeting ligands including lenalidomide are not stable. **YW-N-11** showed improved PK properties, whereas **YW-N-7** demonstrated the best PK profile with a half-life time of ~16 h. As such, we decided to study the efficacy of **YW-N-7** on a B/KR cell-derived xenografted mouse tumor model.

1.2.9 YW-N-7 inhibits B/KR CDX tumor growth in mice

After the pharmacokinetics (PK) test, we did a further *in vivo* cell viability test with the best PK data, **YW-N-7**. From Figure 4A, it was clear that the B/KR cell viability decreased notably after being treated with **YW-N-7** for three days. The IC_{50} of **YW-N-7** was 105.2 nM. After being treated with 500 nM for three days, the cell viability dropped to less than 20%. It suggested that **YW-N-7** could effectively inhibit B/KR cell growth by degrading RET protein level and inhibiting RET kinase activity. This could be further confirmed by immunoblotting data. After cells were treated with **YW-N-7** at different concentrations for 15 h, it showed both pRET (Y905) and KIF5B-RET could be degraded at as low as 50 nM (Figure 4B) To determine if the

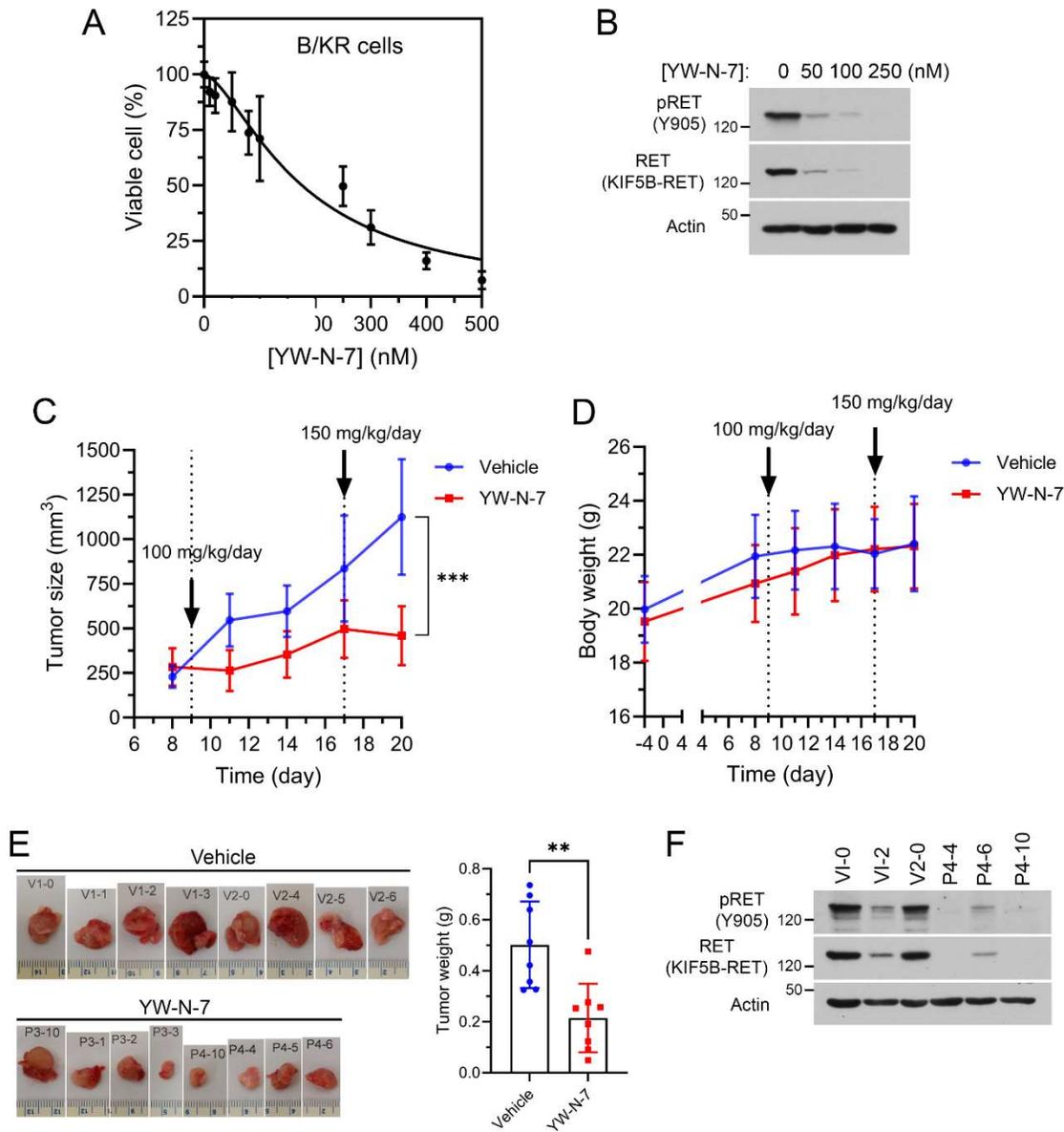


Figure 1.3. YW-N-7 inhibits cell-derived xenograft (CDX) tumors in animals. **A:** Cells treated with various concentrations of YW-N-7. **B:** Cells treated with indicated concentrations of YW-N-7. **(C-D):** CDX tumors were treated with YW-N-7 or mock-treated with vehicle. **C:** Body weights were measured with a scale. **D, E:** Images of tumors collected at the endpoint. **F:** Immunoblots of tumor tissue samples.

RET PROTAC YW-N-7 could inhibit RET oncogene-driven tumors in animals, we tested the effect of YW-N-7 on tumor growth of B/KR CDXs. After measurable s.c. tumors were established, mice bearing a similar size of tumors were treated with YW-N-7 (100 mg/kg/day) or the solvent (control) by intraperitoneal injection starting on Day 9. Tumors grew rapidly in the solvent-treated group. The YW-N-7 treatment significantly inhibited tumor growth (Figure 4C). Because the YW-N-7 treatment did not reduce the animal body weight (Figure 4D), the drug dose was increased to 150 mg/kg/day on Day 17, which remained tolerated by the animals (Figure 4D). Figure 4E showed that tumors from YW-N-7 -treated animals were significantly smaller than tumors from the solvent-treated animals. Immunoblotting analysis of tumor samples showed that the phospho-RET and KIF5B-RET protein were decreased in samples from the YW-N-7-treated group (Figure 4F).

1.3 Discussion

The rearranged-during-transfection (RET) protein is an attractive target for cancer therapy. Aberrantly activated RET kinase caused by germline or somatic *RET* mutations is one of the major oncogenic drivers frequently found in medullary thyroid cancer (MTC). However, the current drugs selpercatinib (LOXO-292) and pralsetinib (BLU-667) as RET kinase inhibitors do not reduce RET protein level. Instead, consistently increased the RET fusion protein level in human cancer cells. The overexpression of RET could reduce the effectiveness of the kinase inhibitors and contribute to protein mutation and drug resistance. Thus, depletion of oncogenic RET protein is likely to have an added benefit in addition to inhibition of the oncogenic RET kinase activity.

Proteolysis targeting chimera (PROTAC) is a developing modality to achieve targeted protein degradation with heterobifunctional small chemical molecules. A PROTAC, bearing an E₃ binding ligand, a linker and a ligand that engages the target protein of interest, brings an E₃ ligase

into close proximity to the target protein of interest to form a ternary complex, resulting in ubiquitination and proteasomal degradation of the target protein. Immunomodulatory imide drugs (IMiDs), including thalidomide, lenalidomide, and pomalidomide, bind ubiquitously expressed cereblon (CRBN) E₃ ligase, are commonly used ligands for E₃ ubiquitin ligases in PROTAC design. More recently, phenyl glutarimide (PG) analogues of CRBN binders were used to improve the chemical stability of CRBN ligands.

To deplete RET protein, we decided to design PROTAC compounds using LOXO-292 (selpercatinib) as the RET-targeting ligand. This is because the co-crystal structure of LOXO-292/RET (PDB: 7JU6), suggested that the terminal hydroxypropyl group of LOXO-292 has minimal impact on its binding to RET (Figure 1B and 1C). We postulated that this group could be replaced with certain linkers to link an ubiquitin E₃ ligase binding ligand to degrade RET. For targeting E₃, we chose lenalidomide, a common ligand of the E₃ ligase cereblon (CRBN). We also made bifunctional molecules containing LOXO-292 scaffold and PG as a comparison.

Our subsequent SAR studies suggested that a rigid and hydrophobic linker is preferable to join the LOXO-292 moiety and the CRBN binding ligand to make PROTAC compounds that could degrade RET effectively. Indeed, both lenalidomide- and PG-based PROTAC compounds could lead to the depletion of RET. The three lead compounds, lenalidomide-based **YW-L-13**, and PG-based **YW-N-7** and **YW-N-11**, demonstrated remarkable activity that completely depleted RET in cancer cells at a concentration of 100 nM or below. Consistent with their ability to degrade RET, they exhibited potent cytotoxicity toward cancer cells with EC₅₀s less than 100 nM. In contrast, we observed treatment of either selpercatinib (LOXO-292) or pralsetinib (BLU-667) increased RET protein level dramatically, which is consistent with the previous research findings. Also, in the presence of high concentration of the proteasome inhibitor Bortezomib, the ability of **YW-N-**

7 to degrade RET was impaired, however, it still exhibited potent inhibitory activity for RET phosphorylation, suggesting that **YW-N-7** has synergistic dual roles in inhibiting RET function: acting as RET kinase inhibitor as well as RET protein degrader.

To select the lead compound for in vivo animal study, the lead compounds were first evaluated by PK studies. Consistent with the previous reports, lenalidomide-based **YW-L-13** revealed a poor PK profile, whereas **YW-N-7** showed a good half-life time of 15.6 h. Therefore **YW-N-7** was selected for further investigation. Next, we demonstrated that **YW-N-7** could potentially kill RET-positive B/KR cancer cells, by depleting RET protein and inhibiting RET kinase activity (Figure 4A and 4B). To determine if the **YW-N-7** could inhibit RET oncogene-driven tumor growth in animals, we tested the effect of **YW-N-7** on tumor growth on a B/KR cell-derived xenograft mouse model. The treatment of **YW-N-7** led to significant tumor regression in mice. Immunoblotting analysis of tumor samples showed that the phospho-RET and KIF5B-RET protein were decreased in samples from the **YW-N-7**-treated group, demonstrating the consistent mechanism of in vitro and in vivo. As **YW-N-7** revealed synergistic dual roles of RET inhibition by inhibiting RET kinase activity and degrading RET, it could be used as a powerful research tool to gain insight into RET functional mechanism by selectively depleting RET protein. Its potent activity also enables it a lead compound for development of RET-targeted cancer therapy.

1.4 Conclusion

In summary, we have developed a series of bifunctional molecules as novel PROTAC compounds for the targeted degradation of RET. These molecules are based on a LOXO-292 scaffold, a CRBN E3 ligand using either lenalidomide or phenyl glutarimide (PG), and a linker of varied lengths, flexibility, and hydrophobicity. Among them, **YW-N-7** as our lead tool compound exhibited synergistic dual functional roles by selectively depleting RET while inhibiting the kinase

activity of RET simultaneously in vitro and in vivo. The compound demonstrates significant antitumor efficacy on a B/KR cell-derived xenografted mouse model. This study exemplifies the feasibility of **YW-N-7** as both a powerful tool compound to study RET biomechanism and a promising lead compound for RET-targeted cancer therapy.

1.5 Synthesis and biological test

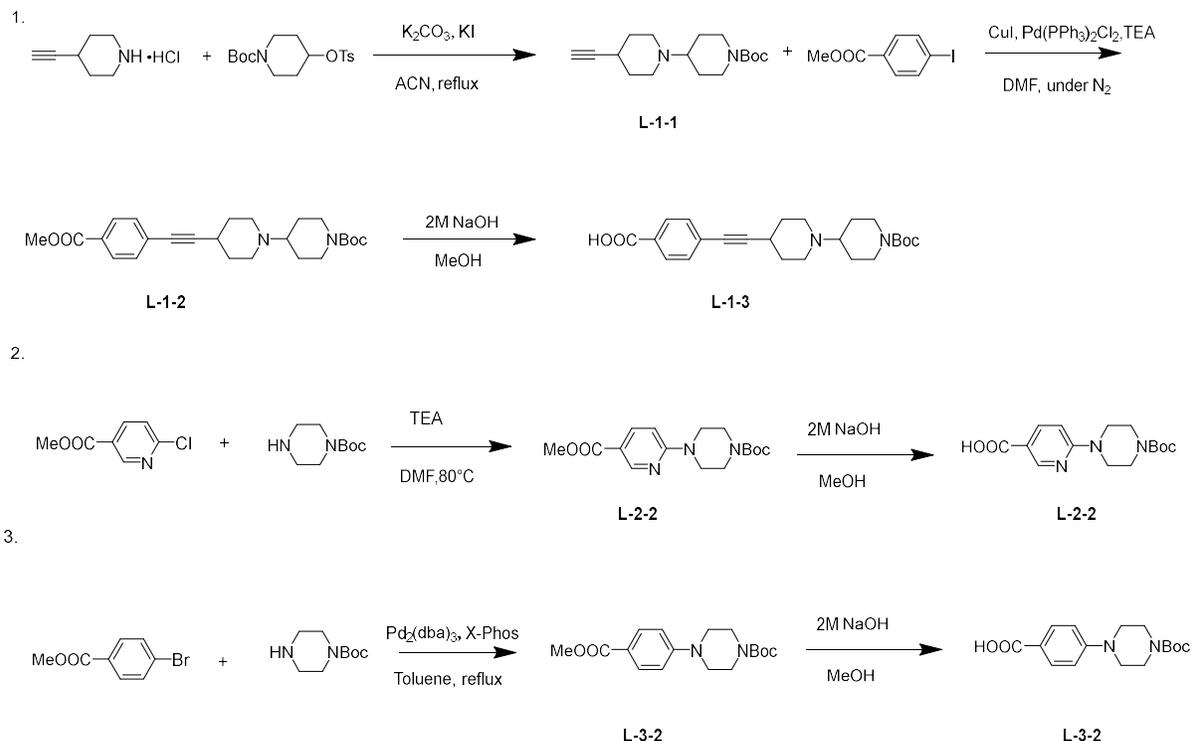


Figure 1.4. Synthesis route of L-1-1, L-2-2, L-3-2.

1.5.1 Synthesis of L-1-3

Step 1: Tert-butyl 4-(Tosyloxy) piperidine-1-carboxylate and 4-ethynylpiperidine HCl salt were dissolved in ACN, then 10eq of K_2CO_3 and 0.2eq of KI were added. The reaction solution was stirred for 48h. The reaction was diluted with water and extracted with EA 3 times. Combine the EA phase, the organic phase was washed with brine, dried with anhydrous Na_2SO_4 , then evaporated under reduced pressure. The left residue was purified by column chromatography to get **L-1-1**, using EA to get product. $[M+H^+]:293.2$

Step 2: Methyl 4-iodobenzoate(1eq) and tert-butyl 4-ethynyl-[1,4'-bipiperidine]-1'-carboxylate(1eq) were dissolved in anhydrous DMF, then 0.2eq of CuI and Pd (PPh₃)₂Cl₂ were added. The mixture was purged under N₂ for 5min. Then 3eq of TEA was added under N₂ environment. The mixture was stirred under 100°C for 4h. Then the reaction was cooled down to R.T., DMF was evaporated under reduced pressure. The left residue was purified by column chromatography, using EA: MeOH=30:1 to get product, **L-1-2**. [M+H⁺]:427.6

Step 3: Tert-butyl 4-((4-(methoxycarbonyl) phenyl) ethynyl)-[1,4'-bipiperidine]-1'-carboxylate was dissolved in MeOH, 2eq of 2M NaOH was added. The reaction solution was stirred under 100°C for 2h. Next, MeOH was evaporated under reduced pressure, the left residue was diluted with H₂O and acidified to pH around 4 using 1M HCl. The H₂O solution was extracted by DCM for two times, combine the organic layer, washed by Brine and dried by anhydrous Na₂SO₄. EA was removed under reduced pressure, the left residue was purified by column chromatography, using DCM: MeOH=20:1 to get product, **L-1-3**. [M+H⁺]:412.5

1.5.2 Synthesis of L-2-2:

Step 1: Methyl 4-bromobenzoate, tert-butyl piperazine-1-carboxylate, Pd₂(dba)₃, and X-phos were dissolved in toluene. The reaction solution was purged with nitrogen and sealed. The mixture was refluxED 8 hours. Next, toluene was evaporated under reduced pressure, water was added and extracted with 300ml EA three times. EA was evaporated and the left oil was purified by column to get product tert-butyl 4-(4-(methoxycarbonyl) phenyl) piperazine-1-carboxylate.

Step 2: tert-butyl 4-(4-(methoxycarbonyl) phenyl) piperazine-1-carboxylate was dissolved in MeOH, then 2M NaOH 3eq was added. The resulting mixture was refluxed under 70°C for 8 hours. After the reaction was done, cool down to R.T., followed by evaporating MeOH and acidifying the solution to pH around 4, then extracted by EA 3 times. Combine the EA phase and washed by

Brine, then dried by Na₂SO₄, EA was evaporated under reduced pressure and purified by column to get product 4-(4-(tert-butoxycarbonyl) piperazin-1-yl) benzoic acid, using Hex: EA=3:1 to get product. ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.42 (s, 1H), 7.77 (s, 2H), 6.96 (s, 2H), 3.45 (s, 4H), 3.29 (s, 4H), 1.42 (s, 9H). ¹³C NMR (600 MHz, DMSO-*d*₆) δ 167.69, 157.20, 153.58, 130.92, 121.56, 115.62, 81.28, 48.87, 29.38.

1.5.3 Synthesis of L-3-2:

Methyl 4-bromobenzoate, tert-butyl piperazine-1-carboxylate, Pd₂(dba)₃, and X-phos were dissolved in toluene. The reaction solution was purged with nitrogen and sealed. The mixture was refluxed for 8 hours. Next, toluene was evaporated under reduced pressure, water was added and extracted with 300ml EA three times. EA was evaporated and the left oil was purified by column to get product **L-3-1**.

Tert-butyl 4-(4-(methoxycarbonyl) phenyl) piperazine-1-carboxylate was dissolved in MeOH, then 2M NaOH 3eq was added. The resulting mixture was refluxed under 70°C for 8 hours. After the reaction was done, cool down to R.T., followed by evaporating MeOH and acidifying the solution to pH around 4, then extracted by DCM 3 times. Combine the DCM phase and washed by Brine, then dried by Na₂SO₄, DCM was evaporated under reduced pressure and purified by column to get product **L-3-2**. ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.42 (s, 1H), 7.77 (s, 2H), 6.96 (s, 2H), 3.45 (s, 4H), 3.29 (s, 4H), 1.42 (s, 9H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.69, 157.20, 153.58, 130.92, 121.56, 115.62, 81.28, 48.87, 29.38.

1.5.4 Synthesis of W-1-4

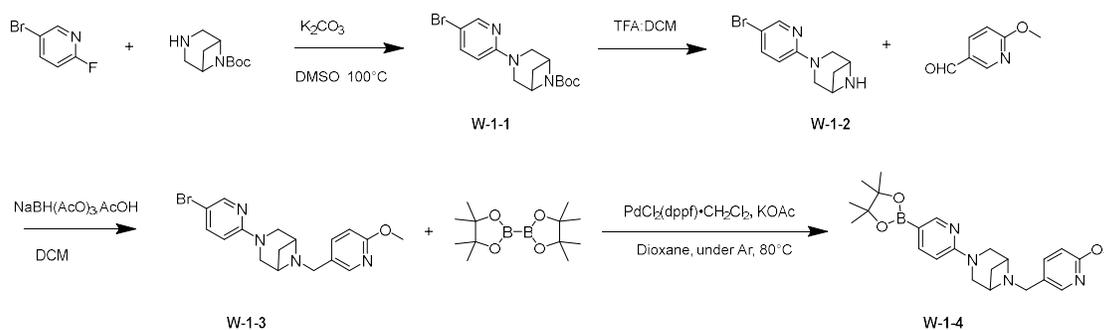
Step 1: 5-bromo-2-fluoropyridine and tert-butyl 3,6-diazabicyclo[3.1.1]heptane-6-carboxylate were mixed dissolved in DMSO, then 10 eq K₂CO₃ was added. The solution was stirred under 100°C overnight. On the second day, the reaction solution was diluted water and extracted with

EA, then organic phase was washed with Brine and dried with Na₂SO₄, followed by concentrated under reduced pressure. The residue was purified by chromatography to get **W-1-1**. (Hexane:EA=1:1, get product).

Step 2: **W-1-1** was dissolved in 4 M HCl in dioxane and stirred for 2. Then, the solvent was evaporated under reduced pressure. The left residue, W-1-2, was used for the next step without any further purification.

Warhead synthesis

1.



2.

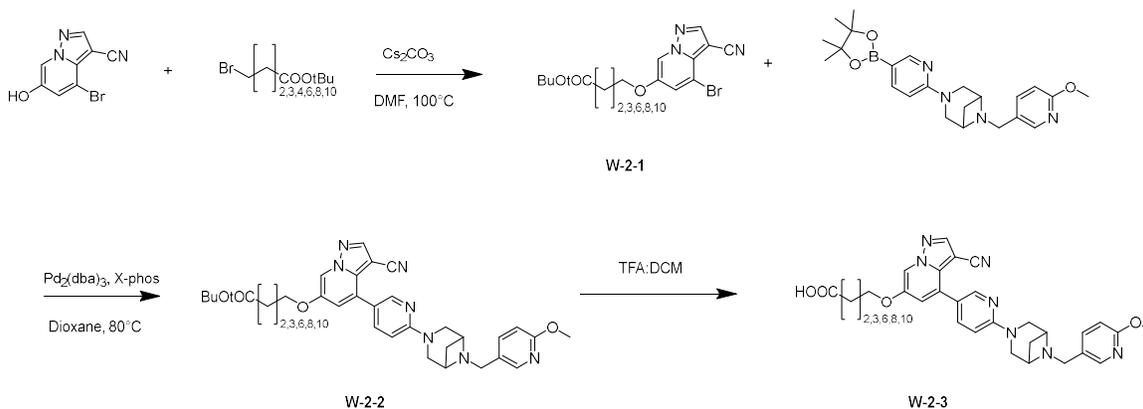


Figure 1.5. Synthesis of different methylene linkers of W-2-3.

Step 3: **W-1-2** was mixed with 6-methoxynicotinaldehyde in 1,2-Dichloroethane (DCE), then 3 eq of Sodium triacetoxyborohydride [NaBH(AcO)₃] and 3 eq of Triethylamine (TEA) were added. The resulting solution was stirred under room temperature (RT) overnight. The second day, the

reaction solution was diluted with water and extracted with EA. The organic phase was washed with Brine and dried by Na₂SO₄, then evaporated under reduced pressure. The left residue was purified by chromatography (pure EA get product **W-1-3** with 1% TEA). ¹H NMR (600 MHz, Chloroform-*d*) δ 8.25 (d, *J* = 2.5 Hz, 1H), 8.09 – 8.00 (m, 1H), 7.59 (ddd, *J* = 10.1, 8.7, 2.5 Hz, 2H), 6.71 (d, *J* = 8.5 Hz, 1H), 6.45 (d, *J* = 9.0 Hz, 1H), 3.91 (s, 3H), 3.76 – 3.68 (m, 4H), 3.51 (s, 2H), 3.46 (d, *J* = 11.9 Hz, 2H), 2.66 (q, *J* = 6.8 Hz, 1H), 1.60 (d, *J* = 8.6 Hz, 1H).

Step 4: The 3-(5-bromopyridin-2-yl)-6-((6-methoxypyridin-3-yl)methyl)-3,6-diazabicyclo[3.1.1]heptane (1eq), Bis(pinacolato) diboron (2eq), Acetic potassium(3eq) and PdCl₂(dppf)·CH₂Cl₂(0.1eq) were mixed together in dioxane and sparged under N₂ for 10min, the reaction solution was sealed and stirred under 90°C overnight. The second day, the reaction solution was filtered through filter paper and evaporated under reduced pressure. The left residue was purified by chromatography (EA:MeOH=10:1, get product).

[M+H]⁺:341.2 in form of boron acid in LC-MS.

1.5.5 Synthesis of **W-2-3**

Step 1: 4-bromo-6-hydroxypyrazolo[1,5-a] pyridine-3-carbonitrile (1.0 eq) was stirred with tert-butyl 4-bromobutanoate (1.1eq) in DMF under 70°C for 3h. Three hours later, the reaction solution was diluted with H₂O and extracted with EA, then washed with Brine and concentrated under reduced pressure. The left residue was purified by chromatography to get product **W-2-1**. (Hexane:EA=3:1 get product), [M+H]⁺:366.0

Step 2: tert-butyl 4-((4-bromo-3-cyanopyrazolo[1,5-a]pyridin-6-yl)oxy)butanoate (1.0 eq), 6-((6-methoxypyridin-3-yl)methyl)-3-(5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)pyridin-2-yl)-3,6-diazabicyclo[3.1.1]heptane (1.5 eq), X-phos (0.2 eq), Pd₂(dba)₃ (0.05eq) and 2M K₃PO₄ were mixed together in dioxane and sparged under N₂ for 10min. The resulting mixture was stirred

overnight. Next, the dioxane was evaporated under reduced pressure and the residue was directly purified by column chromatography using EA:MeOH=10:1, product **W-2-2**. $[M+H]^+$:596.6

Step 3: The previous step product was dissolved in DCM:TFA=1:1 to react for 2h. Then, the solution was evaporated under reduced pressure and co-evaporated with hex and DCM to totally remove TFA. The residue was purified by column chromatography, using DCM:MeOH=10:1 to get **W-2-3**. $[M+H]^+$:540.1

The rest of warheads with different length of linkers were synthesized using the same procedure.

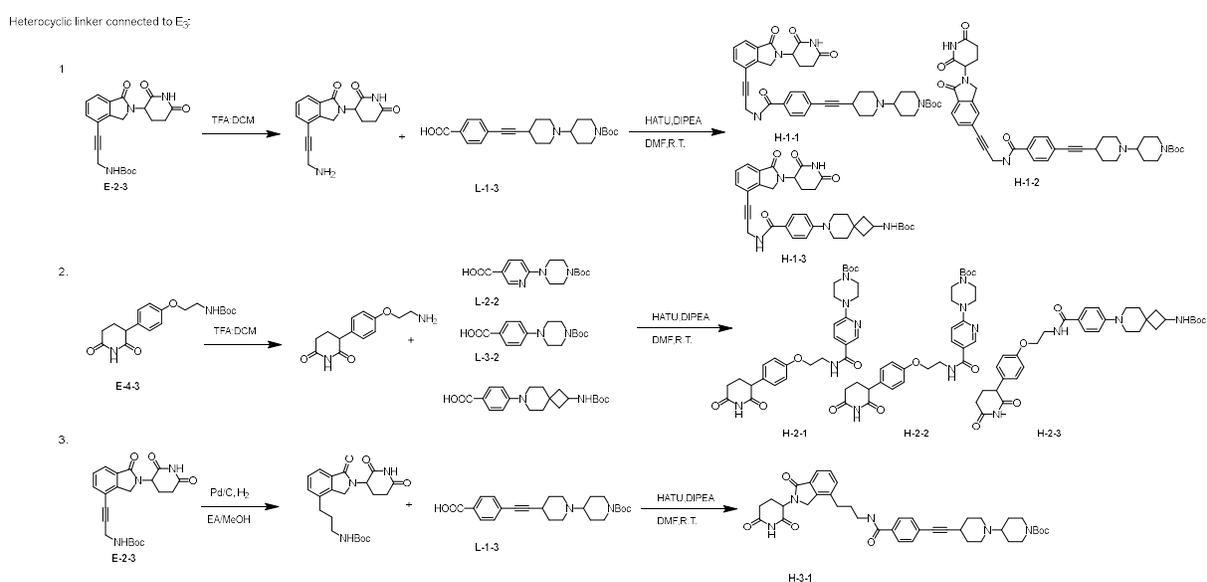


Figure 1.6. Synthesis of various E₃ ligands with different heterocycle linkers.

1.5.6 Synthesis of H-1-1 to H-1-3

Compound **E-2-3** was dissolved in TFA:DCM=1:1 solution and stirred for 2 hours. Then, solvent was removed under reduced pressure, and the left residue was co-evaporated with hexane for 5 times to totally remove TFA. The left oil residue was dissolved with **L-1-3** in DMF, 1.2 eq of HATU and 3 eq of DIPEA were added. The reaction solution was stirred under R.T. overnight. Next, DMF was removed under reduced pressure, the left solid was purified by column

chromatography to get product, using EA:MeOH=10:1 to get product H-1-1. $[M+H]^+$: 691.8 With the same strategy, H-1-1 to H-3-1 were prepared.

1.6 Cell and animal test material

1.6.1 Antibodies

Antibodies to phospho-RET (Y905) (#3221), RET (#14698), and EGFR (#2232) were from Cell Signaling Technology (Danvers, MA, USA). anti- β -actin antibody (#A5316) was from Sigma Aldrich (St. Louis, MO, USA).

1.6.2 Cell lines and cell culture

The CCDC6-RET fusion-positive human TPC1 thyroid carcinoma cells were from the European Collection of Authenticated Cells Culture (ECACC). Mouse BaF3 cells expressing KIF5B-RET (B/KR) were generated using a lentiviral vector as described previously⁴. Cells were free of mycoplasma. TPC1 cells were cultured in RPMI-1640/5% fetal bovine serum (FBS) plus penicillin-streptomycin (100 U/ml) at 37 °C/5% CO₂. B/KR cells were cultured in RPMI-1640/10% FBS plus penicillin-streptomycin (100 U/ml) at 37 °C/5% CO₂.

1.6.3 Cell-based assays and Immunoblotting analysis

Cell viability assay was performed using CellTiter-glo reagent (Promega, Madison, WI, USA) in 96-well plates as described previously. For experiments with TPC1 cells, cell viability was determined after 5 days of drug treatment^{4, 24-26}. For experiments with B/KR cells, cell viability was determined after 3 days of drug treatment. For immunoblotting analysis of CCDC6-RET protein degradation, TPC1 cells cultured in 60-mm plates (50% confluent) were either treated with the test compound for 24 h or for the length of time indicated in the Figure **x** legend. For immunoblotting analysis of KIF5B-RET protein degradation in B/KR cells, 2×10^6 cells in 6-well plate were treated with the test compound for 15 h.

Cell lysates were prepared by lysis in cold Lysis Buffer A (50 mM Tris-HCl, pH7.5, 150 mM NaCl, 1 mM EDTA, 1 mM EGTA, 25 mM NaF, 5 mM sodium pyrophosphate, 1 mM Na₃VO₄, 2 µg/ml aprotinin, 2 µg/ml leupeptin, 1 mM dithiothreitol, 20 mM p-nitrophenyl phosphate, 1% Triton X-100). Cell lysate supernatants were obtained by microcentrifugation twice at 16000 rpm for 10 min at 4 °C. Equal amounts of soluble cell lysate proteins were separated on 10% SDS-polyacrylamide gels, transferred to nitrocellulose filters and analyzed by immunoblotting.

1.6.4 PK study

In two separate experiments, the compound YW-N-7 was administered either p.o. or i.p. to mice at the dose of 50 mg/kg (1 mg/20g, volume 150 µL). After administration, 75 µL blood samples are collected at 10 min, 20 min, 30 min, 1h, 2h, 4h, 8h, 16h, 24h, 36h, 48h (n=3 per time point, and each mouse was used for three time points, thus 12 mice were used for either p.o. or i.p. to make a total of 24 mice) after drug administration. Blood samples were collected into 1.5-mL Eppendorf tubes containing 30 µL disodium EDTA (0.5 M, pH 8.0), kept on ice, then centrifuged at 4000 rpm/min for 10 min under 4 °C. The supernatants (serum) were collected and stored under –80 °C for future analysis. 100 µL of the serum samples are added to 9:1 ratio of acetonitrile and glacial acetic acid. The samples are cooled in ice for 10 min, followed by centrifuging at 10,000 rpm and 4 °C for 10 min. The clear supernatants are transferred to vials and analyzed by LC/MS/MS.

1.6.5 Cell-derived xenograft (CDX) tumor study

The CDX tumor growth experiment in animals was approved by the IACUC of the University of Oklahoma Health Sciences Center. B/KR cells used for the animal experiment were verified for free of mouse pathogens by IDEXX BioAnalytics. Cells were prepared in serum-free RPMI-1640 medium and inoculated s.c. (5×10^6 cells/0.1 ml/each) into the right flanks of 6-week old female

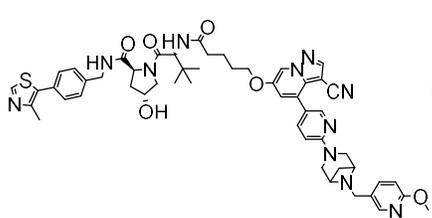
SHO mice (Charles River). After measurable tumors were established, mice were treated either with YW-N-7 at the dose specific in the figure legend or with the same volume of vehicle by i.p. injection. Each group had 8 mice bearing tumors of the similar size. The compound was dissolved in a vehicle that contained 5% 1-methyl-2-pyrrolidinone, 40% polyethylene glycol 400, 5% solutol HS-15, and 49.5% saline. Tumor sizes and animal body weights were measured similar to that described²⁴. Statistical analysis was performed using unpaired t-test with Welch's correction. $p < 0.05$ was considered statistically significant.

1.7. Additional Compounds

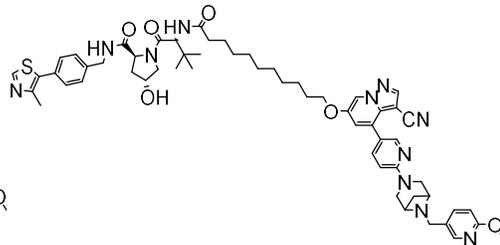
In addition, we also synthesize VHL, pomalidomide, and thalidomide as E₃ ligands PROTAC molecules with different length of methylene linkers (Figure 1.7.). However, none of them have comparable degradation rate compared with lenalidomide or PG E₃ ligands. We didn't continue to do further research about these three ligands.

Table 1.2. HRMS data

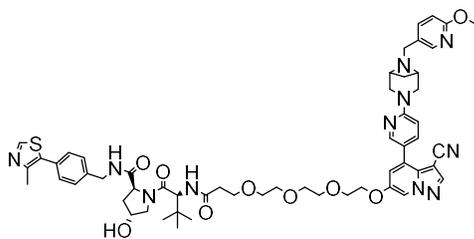
Compound	Degradation rates at 50 nM (%)
YW-V-1	-120.7
YW-V-2	54.4
YW-V-3	15
YW-V-4	-7.8
YW-T-1	4.1
YW-T-2	1.9
YW-T-3	ND
YW-T-4	ND
YW-T-5	25.3
YW-T-6	43.5



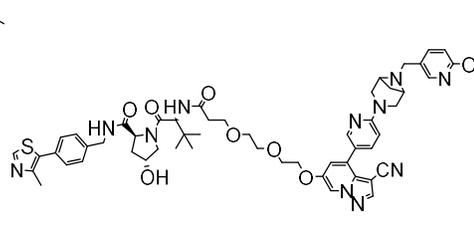
YW-V-1



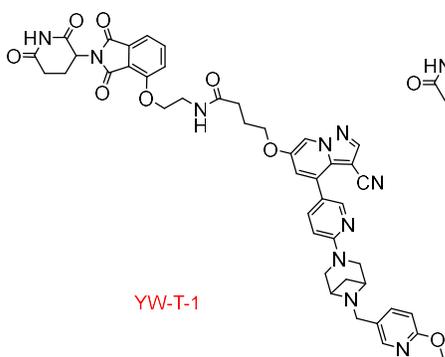
YW-V-2



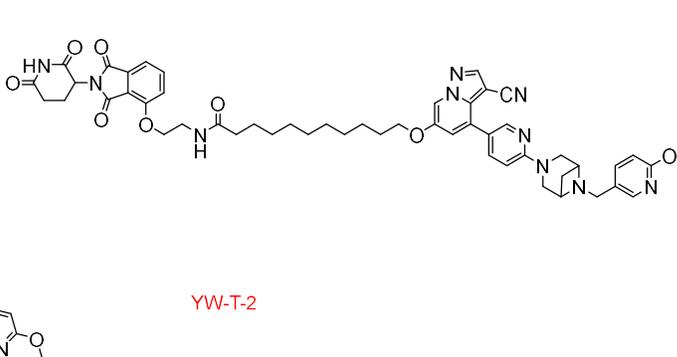
YW-V-3



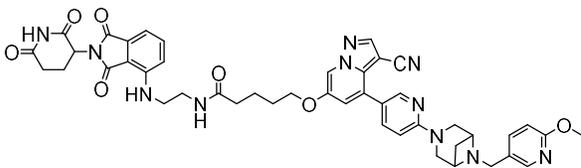
YW-V-4



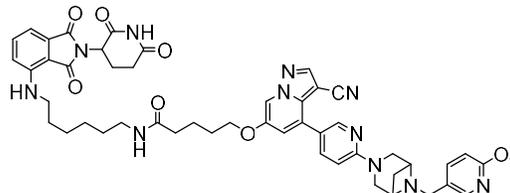
YW-T-1



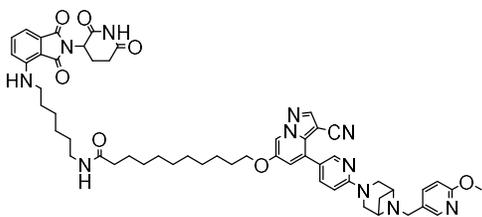
YW-T-2



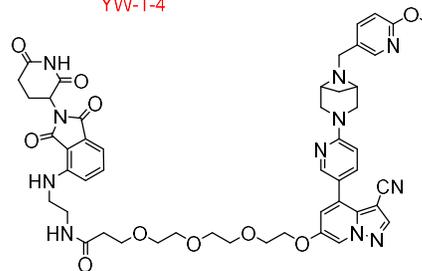
YW-T-3



YW-T-4



YW-T-5



YW-T-6

Figure 1.7. Additional compounds

CHAPTER TWO: COVALENT INHIBITOR TO TARGET RET G810C MUTATION

2.1. Introduction

RET receptor tyrosine kinase is encoded by RET proto-oncogene, and RET tumors are usually caused by chromosome rearrangements or point mutations. About 1%-2% NSCLC originated from RET chimeric proteins, including fusions with KIF5B, CCDC6, NCOA4, and TRIM33. Aberrant RET signaling also can result in thyroid cancer. For example, MTC either arise sporadically (75%) or as a consequence of RET germline mutation.²⁷ In addition, RET chromosomal rearrangements, such as, CCDC6-RET, PRKAR1A-RET, and TRIM24-RET, are related to PTC.

Previously, first generation inhibitors are multi-kinase inhibitors, Cabozantinib, Vandetanib, and Lenvatinib, which were used to treat RET malignancy.²⁷ Vandetanib showed a potency with 100nM IC₅₀. It can effectively inhibit the phosphorylation and signaling transduction of RET/PTC3, RET/MEN2B, and EGF/RET chimeric receptor.²⁸ In clinical trial, it revealed an 11-month prolongation of median progression-free survival (from 19.3 months to an estimated 30.5 months).²⁹ For Cabozantinib, it inhibited TT cell RET phosphorylation bearing C634W mutation, which is usually related to familial MTC (FMTC) and MEN2A.³⁰ However, the overall survival (OS) has no big difference between Cabozantinib and Placebo (26.6 months and 21.1 months). In selective phase 3 III trial of Lenvatinib, it showed significantly improved median PFS and ORR, reaching 18.3 months/64.8% compared to 3.6 months/1.5% in 392 patients with differentiated thyroid carcinoma (DTC).³¹ Moreover, in a phase II trial against 25 NSCLC patients, it revealed promising clinical activity with an ORR of 16% and a median PFS of 7.3 months.³²

In 2020, the US Food and Drug administration granted two RET selective inhibitors, Selpercatinib and Pralsetinib. It demonstrated good potency *in vitro* and *in vivo* towards cells harboring RET alternations, especially gatekeeper V804M/L mutants. The crystal structures unveil that they bind to RET kinase domain very different from multi-kinase inhibitors (MKI). They occupy both the front and back pockets of the catalytic cleft, in contrast to MKI, they don't pass through the channel formed by V804 and K758. Instead, they wrap around the lysine, thus avoiding steric hinderance with V804M/L gatekeeper mutations. Especially, Selpercatinib displays exceptional efficacy towards RET wild type (WT) and RET gatekeeper V804M/L mutations with impressive IC₅₀ value of 0.4 nM and 0.8/0.4 nM.³³ Among 105 participants clinical trial who has been treated with platinum-based chemotherapy, it achieved 64% ORR with a median PFS of 16.5 months. In the previously untreated 39 patients, the ORR was 85%, with 90% of the response last for 6 months.³⁴ However, in recent research, solvent front mutations, particularly G810S/C/R, have been identified as the main drivers to long-term administration of acquired resistance to selective RET inhibitors, such as Selpercatinib and Pralsetinib. The first case of solvent front mutations patients was observed in a patient with KIF5B-RET fusion NSCLC which had good clinical result at first, but later developed acquired drug resistance. In circular tumor DNA sequencing test, it turned out to correlate with G810S/C/R mutations. The potency of selpercatinib towards these three mutations decrease several folds compared with WT.³⁵ Among G810S/C/R mutations, the G810C is the main solvent front mutation. Now, it is urgent to develop next generation RET inhibitor to solve the problem of G810C mutation.

Targeted covalent inhibitors are rationally designed inhibitors that bind to protein first, then react rapidly to a proximate nucleophile, usually cysteine, to form a bond. Compared with conventional inhibitors, it offers several advantages, such as improved potency, selectivity,

pharmacodynamics, and efficacy.³⁶ At the same time, due to its very reactive warheads, it has shortcomings of indiscriminate reactivity that could form covalent adduct with unexpected targets and idiosyncratic toxicities. Even though most pharmaceutical companies don't put that much resource on covalent drug pipelines, still there are several examples of covalent drug that have been approved for therapeutics or at late-stage clinical trials. For example, Afatinib, Mobocertinib, Dacomitinib are approved for EGFR driven lung cancer; Neratinib is used for HR2 positive breast cancer; Ibrutinib is a BTK inhibitor for leukemia, lymphoma; Penicillin has been used as broad-spectrum antibiotics since last century.

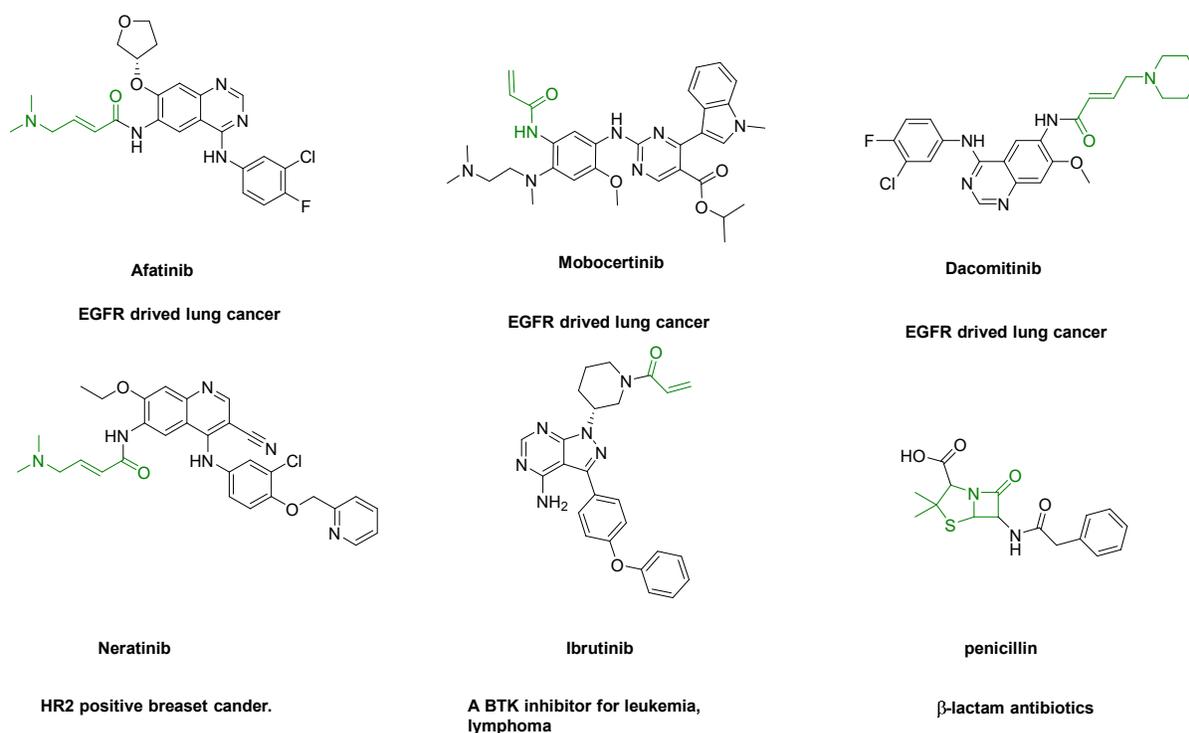


Figure 2.1. Examples of Covalent Inhibitor.

Table 2.1. SAR of different Michael Receptors.

Compound	R	IC ₅₀ B/KR(G810C)
YW-D-50-2		26.10
YW-D-50-3		360.5
YW-D-50-4		81.19
YW-D-50-5		93.85
YW-D-50-6		14.19
YW-D-50-7		11.53

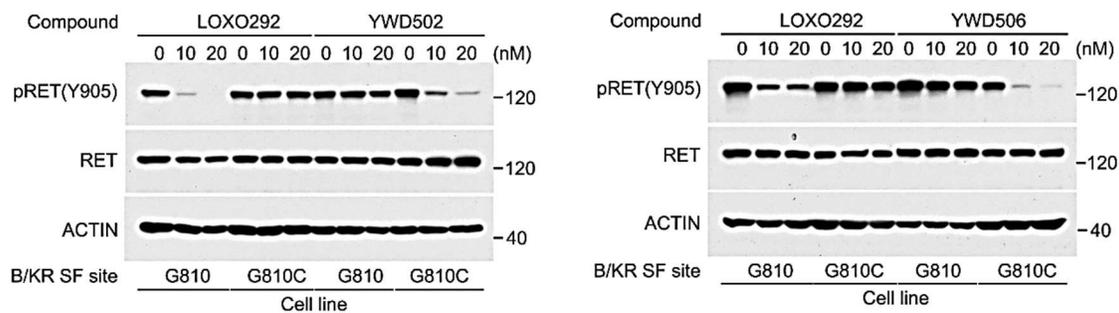


Figure 2.3. WB data of YW-D-50-2, YW-D-50-6.

2.4. Chemistry Synthesis

Synthesis route:

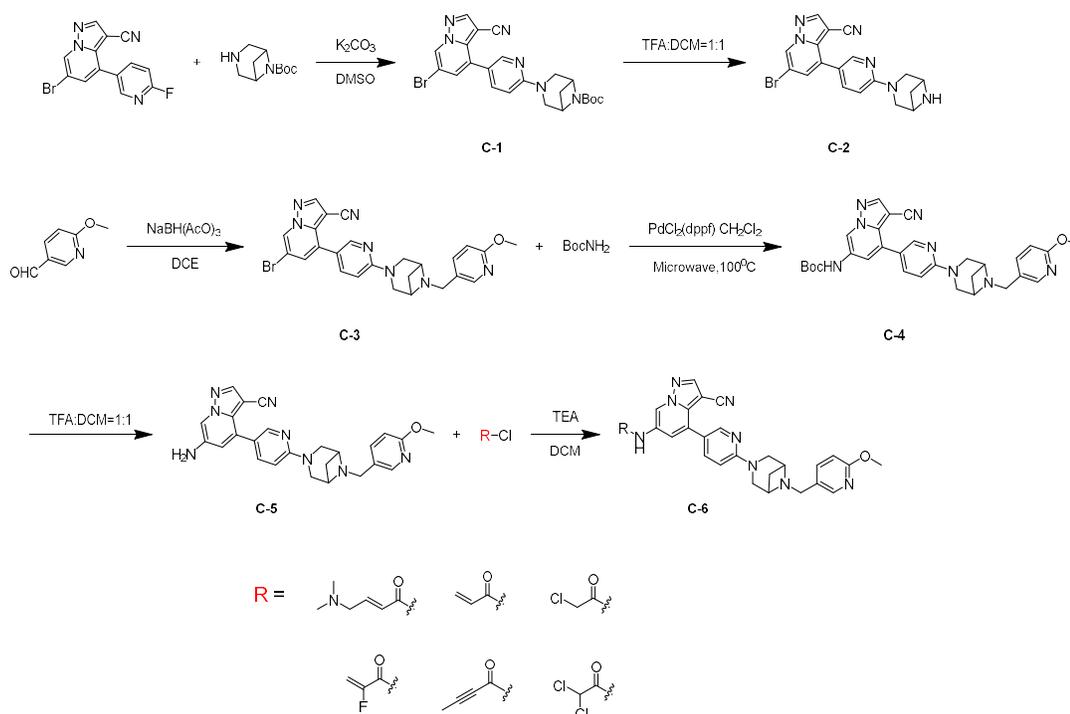


Figure 2.4. Covalent Inhibitor Synthesis Route.

C-1

6-bromo-4-(6-fluoropyridin-3-yl) pyrazolo[1,5-a] pyridine-3-carbonitrile(1eq) and tert-butyl 3,6-diazabicyclo [3.1.1] heptane-6-carboxylate (1.1eq) were dissolved in DMSO with 5eq of K_2CO_3 . The reaction solution was stirred for 3 days under 110°C. After, the reaction solution was cooled down to R.T, extracted with EA for 3 times. Combine the organic layers, washed with Brine, dried by anhydrous Na_2SO_4 . Then, EA was removed under reduced pressure, the left oil run column chromatography using Hex:EA=1:1 to get product, Rf=0.54.

C-2

C-1 was dissolved in a mixture of DCM:TFA=1:1 for 2h, then acid was removed under reduced pressure. The left oil was purified by column chromatography, using DCM:MEOH=10:1 to get product, Rf=0.4.

C-3

Compound C-2 (1eq) was dissolved in DCM under R.T.. Then, 2eq of 6-methoxynicotinaldehyde was added, followed by 1.5eq of sodium triacetoxyhydride and 1eq of AcOH. The reaction mixture was stirred overnight. Next day, reaction was quenched with NaHCO₃ and extracted with DCM for two times. Combine the organic layers, washed with Brine, and dried with anhydrous Na₂SO₄. DCM was removed under reduced pressure, the left residue was purified by column chromatography, using DCM:MeOH=10:1 to get product, Rf=0.2

C-4

A mixture of compound C-3 (1eq) and tert-butyl carbamate (1.1eq) were dissolved in toluene, then 0.05 eq of Pd₂(dba)₃, 4eq of Cs₂CO₃, and 0.05 eq of Xantphos were added. This Hartwig-Buchwald reaction runs under microwave at 110°C for 2h. Then, the solvent was removed under reduced pressure. The left residue was dissolved in acetonitrile and water, filtered, and purified by HPLC.

C-6

Compound C-4 was dissolved in TFA: DCM=1:1 and stirred for 2 h. After, solvent was removed under reduced pressure, the left oil was co-evaporated with hex for 5 times to totally removed TFA to get C-5. C-5 was used for the next step without any further purification. C-5 was dissolved in DCM in ice bath, then 2 eq of TEA was added, followed by 2 eq of different Michael receptor warheads. The reaction solution was stirred overnight. Next, DCM was removed, the left residue was dissolved in ACN and water, filtered, purified by HPLC at last.

CHAPTER THREE: DENDRIMERIC PEPTIDOMIMETIC AS ANTIMICROBIAL

3.1. Introduction

It is estimated that 700,000 to several million deaths result from bacterial infections per year, and 2.8 million people are infected by bacteria resistant to current antibiotics in the USA, with at least 35,000 people dying from that. Due to the scarcity of new antibiotics to combat antibacterial resistance (AMR), over 50 million people could die by 2050, with the yearly death toll being 10 million under the current predicted model.³⁷ As a result, it is extremely urgent to develop new antibacterial agents that can mitigate emerging antibiotic resistance.

Host defense peptides (HDP), which are produced by organisms as the first-line agents to defend against a wide range of bacteria, have gained much attention from scientists.³⁸ Usually, HDPs share two common features: one is the cationic charges, and the other one is a proper ratio of hydrophobic residues, which enable HDPs to adopt amphipathic structures and exhibit significant selectivity toward bacteria over mammalian cells. This is because the outer leaflet of the mammalian cell membrane is composed of zwitterionic lipids, while the outer leaflet of the bacteria membrane mainly consists of negatively charged phospholipids. As a result, the negatively charged bacterial membrane tends to interact with cationic HDPs preferentially, leading to bacterial cell death while maintaining low hemolysis risks and cytotoxicity.³⁹ In human, the two main classes of HDP are defensin and cathelicidin. Defensins usually have a β -sheet core and it can be subdivided into α , β , and θ defensins.⁴⁰ Cathelicidins are produced as prepropeptides that commonly have cathelin-like domain before cleavage by serine proteases. A majority of cathelicidins have secondary helical structure. One side of the helix is hydrophilic cationic charged

amino acids, the other side is hydrophobic. Thus, cathelicidins can interact and perturb with anionic surface of bacteria.⁴⁰ Nevertheless, there are still drawbacks associated with HDPs, such as proteolytic degradation susceptibility, poor selectivity, and moderate activity. To address these problems, several classes of peptidomimetics have been developed to overcome the drawbacks of HDP, including peptoids,⁴¹ β -peptides,⁴² γ -AApeptides,⁴³ and oligourea (Figure 3.1.).⁴⁴ These new unnatural peptidomimetic sequences can mimic HDP function against pathogens and retain resistance to proteolytic hydrolysis (Figure 3.2.).⁴⁵

Lipidated peptides have been presented as antibiotics for years. For instance, polymyxin B⁴⁶ and daptomycin⁴⁷ are two FDA-approved lipo-cyclic peptides. While daptomycin only displays activity against Gram-positive bacteria, polymyxin B is only active for Gram-negative bacteria. Despite distinct antibacterial mechanisms, it has been shown that lipid tails are critical for their activity, which facilitate bacterial membrane interaction.⁴⁸ Recently, our group designed a new class of antimicrobial peptidomimetic compounds composed of γ -AA amino acid (Figure 1). To further explore the antimicrobial potential of γ -AApeptides, we herein report a new class of short, lipidated dendrimeric γ -AApeptides as potential antibacterial agents.

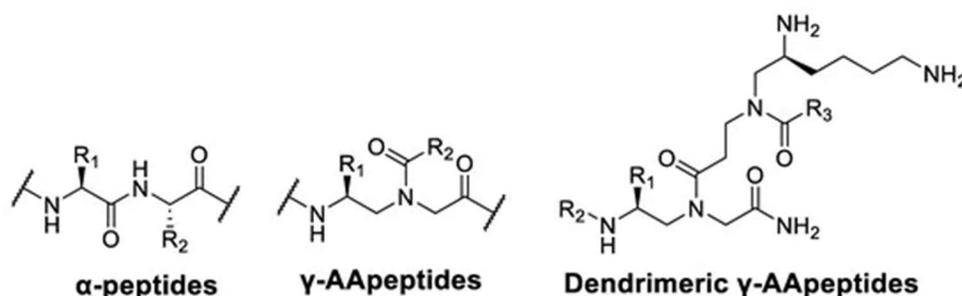


Figure 3.1. Structure of α -peptides, γ -AA peptides, and dendrimeric γ -AA peptides.

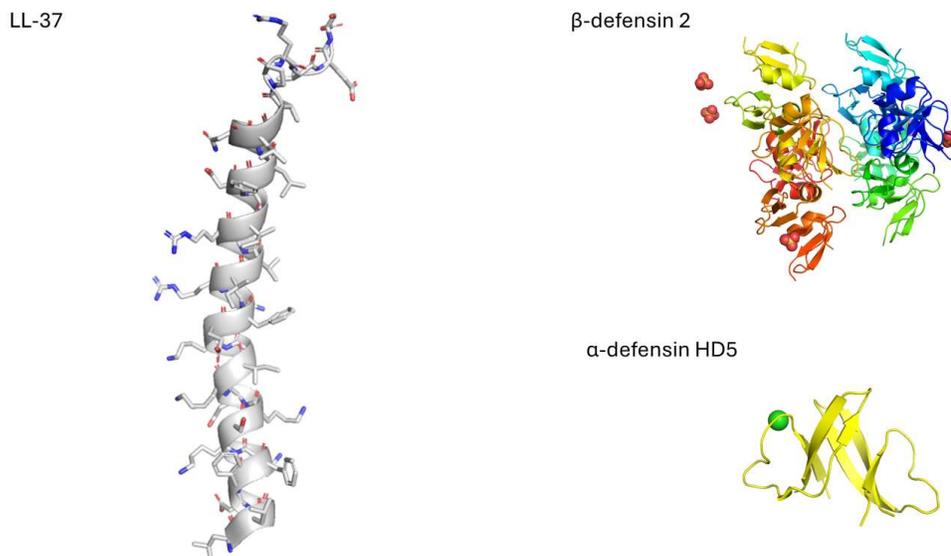


Figure 3.2. Examples of HDPS

3.2. Examples of HDP and mimics

In recent years, our group developed several γ -AA peptides and lipidated γ -AA peptides, such as compounds, **26**, **29**, and **31**, based on the structure of HDP. For the lipidated γ -AA peptides, the lipidated tail would be inserted into bacterial membrane, while the positive lysine's function as positive charged group to attach to the surface of the bacterial membrane. However, lipidated peptides don't guarantee good selectivity. The cytotoxicity and *in vivo* activity were not evaluated.

To achieve the goal of developing small-molecule antimicrobial peptide derivatives with drug-like properties, a new series of compounds with adamantaneacetyl groups was developed, they showed broad-spectrum activities. Moreover, the lead compound **28** didn't show significant activity loss even under Na^+ , K^+ , and Ca^{2+} cations. But we didn't run extensive structure activity

relationships research about compound **28**. In the future, more research can be done with **28** to explore more enhanced activities and therapeutic index (Figure 3.3).⁴⁹

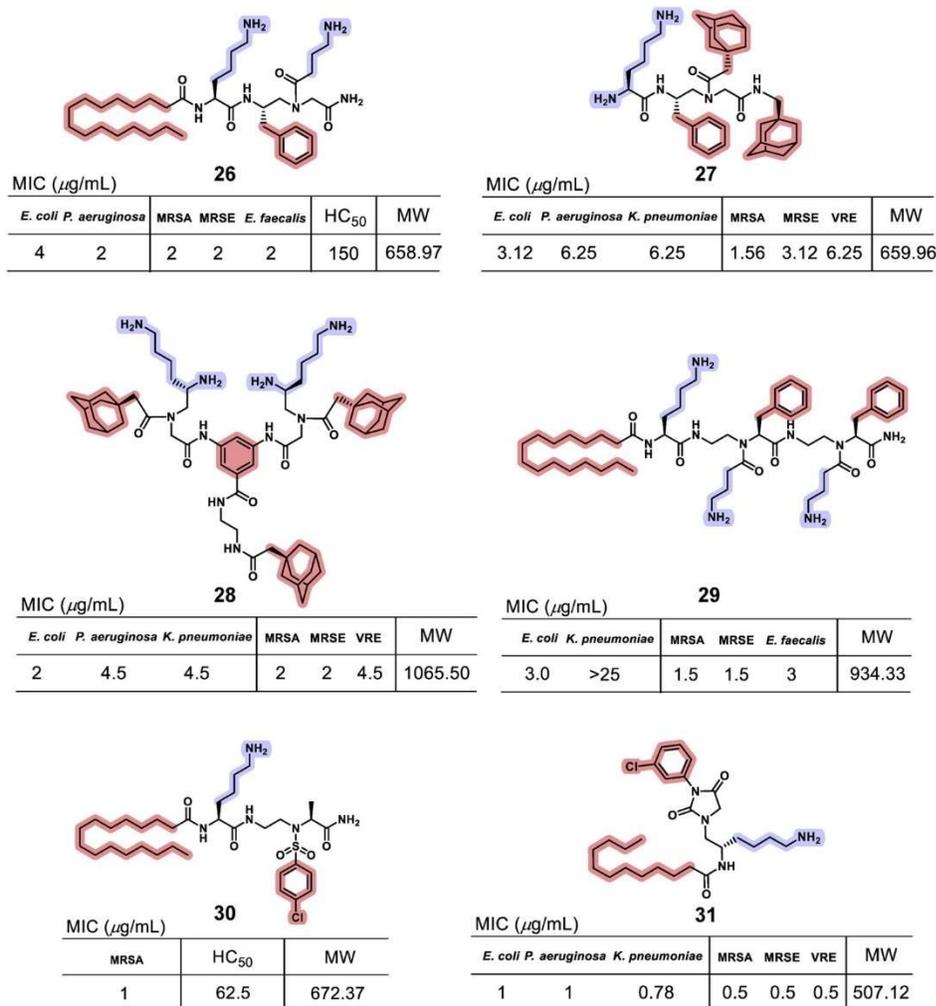


Figure 3.3. Examples of recently developed γ -AA peptides and lipitated γ -AA peptides to mimic HDPs in our group.

3.3. Results and Discussion

In our current design, a positively charged γ -AApeptide building block was attached to the secondary amine of another γ -AApeptide building block, to which different lengths of lipid tails were introduced to make a series of miniature lipo-dendrimeric γ -AApeptides (Figure 1). As such, these amphipathic structures were expected to mimic the antibacterial function of HDPs. The

positively charged side chains would form electrostatic attraction with bacterial membranes, whereas the lipid tail would facilitate the insertion of the compounds into bacterial membranes, leading to membrane disruption. To this end, the compounds were tested for the ability to kill a panel of Gram-positive and Gram-negative bacteria (Table 1). For the building blocks conjugating to the C16 lipid tail, such as Leu (YW-1), Phe (YW-2), Ala (YW-3), Tyr (YW-4), Ser (YW-5), Lys (YW-6), and Arg (YW-7), different γ -AA amino acids were synthesized and solid phase was used to do final synthesis, the structure of newly synthesized compounds listed in Scheme 1. It shows these compounds generally exhibited effective antimicrobial activity and did not show hemolytic activity up to 125 $\mu\text{g}/\text{mL}$. While YW-1 displayed the most broad-spectrum antimicrobial activity against a panel of Gram-positive and Gram-negative bacteria, certain compounds, such as YW-3, 4, 6, and 7, demonstrated highly selective activity toward MRSA (MIC: 0.75–1.5 $\mu\text{g}/\text{mL}$) (Table 3.1.). The findings suggested that functional groups on the building blocks conjugated to the C16 lipid tail could play an important role in antibacterial activity and selectivity, which could direct the future design and optimization of this class of compounds (Figure 3.4).

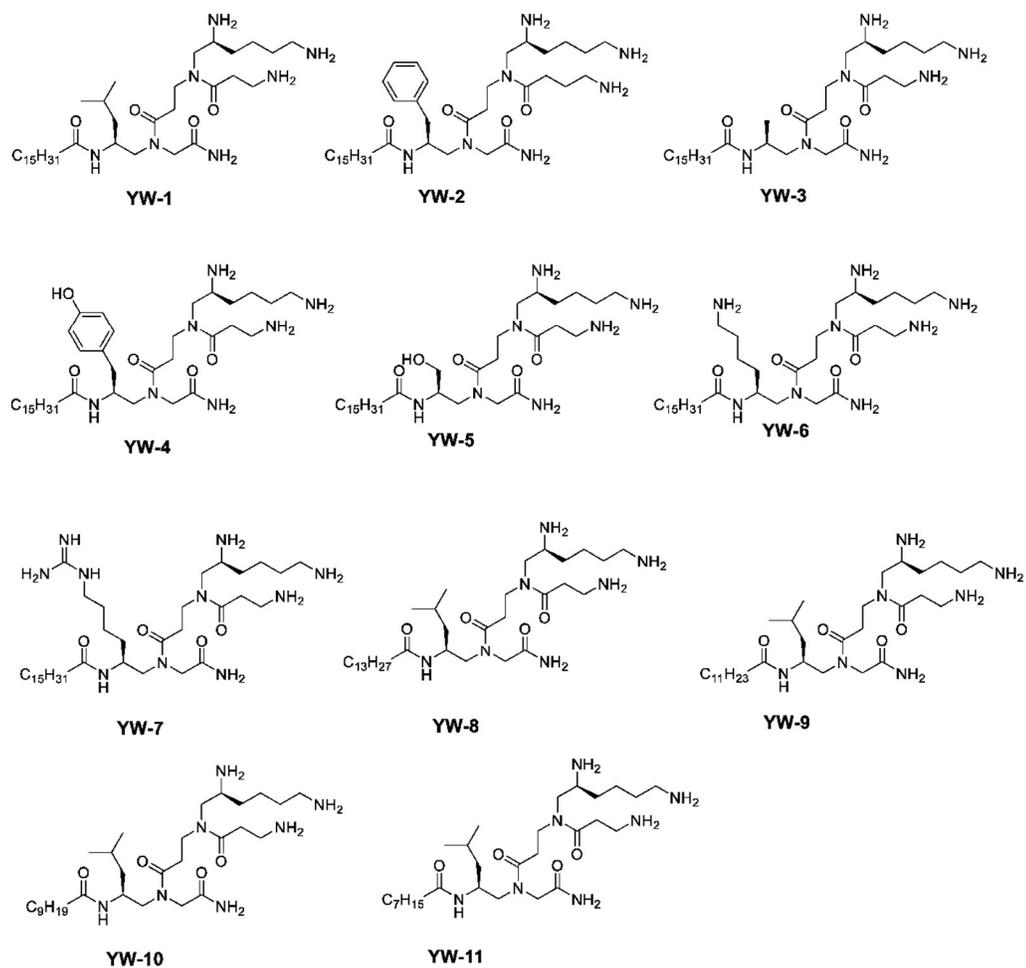


Figure 3.4. The structures of lipitated dendrimeric γ -AApeptides.

Table 3.1. Antibacterial activity, hemolytic activity, and selectivity of dendrimeric γ -Aapeptides. “---” means not tested.

	MIC ($\mu\text{g}/\text{mL}$)						HC ₅₀ ($\mu\text{g}/\text{mL}$)	Selectivity Index (HC ₅₀ /MICM _{MRSA})
	Gram-Positive (+)			Gram-Negative (-)				
	MRSA	E.F.	MRS E	P.A.	K.P.	E. coli		
YW-1	1.5-3	1.5-3	1.5-3	6-12.5	12.5-25	3-6	>125	>41.7
YW-2	1.5-3	1.5-3	1.5-3	12.5-25	6-12.5	3-6	>125	>41.7
YW-3	0.75-1.5	3-6	6-12.5	6-12.5	>25	3-6	>125	>83.3
YW-4	0.75-1.5	3-6	3-6	>25	>25	3-6	>125	>83.3
YW-5	1.5-3					6-12.5	>125	>41.7
YW-6	1.5-3					6-12.5	>125	>83.3
YW-7	0.75-1.5	3-6	3-6	12.5-25	>25	3-6	>125	>83.3
YW-8	12.5-25					>25		
YW-9	>25					>25		
YW-10	>25					>25		
YW-11	>25					>25		

3.4. Materials and Methods

3.4.1. Membrane Depolarization Study

To probe the mechanism of antibacterial activity, we first carried out a membrane depolarization study with the membrane potential-sensitive dye 3,3'-dipropylthiadicarbocyanine iodide (diSC35) (Figure 2A,B).⁵⁰ DiSC35 usually accumulates in living bacterial membranes, and due to self-quenching, it shows low fluorescence intensity. However, if the membrane is disrupted, and the fluorophore is released from the membrane, fluorescence will improve dramatically. As shown in Figure 2, when the bacteria were treated with different concentrations of the lead compound YW-1, both E. coli and MRSA exhibited dose-dependent increases in fluorescence intensity. It was intriguing that YW-1, at a two-fold of MIC or above, caused even more intensive fluorescence than the positive control Triton, which is well known for causing membrane damage. This experiment result demonstrated that the bacterial membranes were disrupted by YW-1, indicating a potential mechanism similar to that of HDPs (Figure 3.5.).

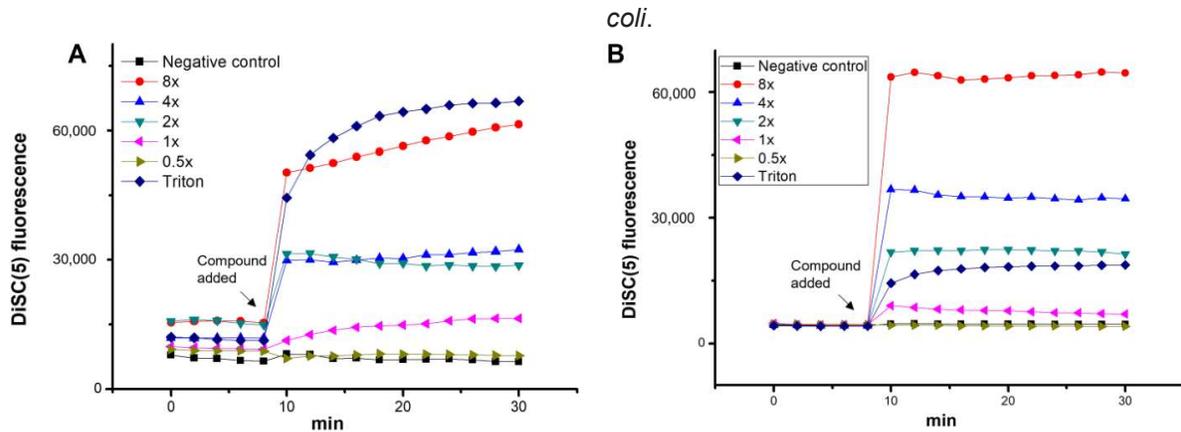


Figure 3.5. Membrane depolarization study of YW-1 against Gram-negative bacteria *E. coli* (A) and Gram-positive bacteria MRSA (B). Triton x-100 as a positive control.

3.4.2 Outer Membrane (OM) Permeabilization

Next, using *E. coli* as a microorganism, we evaluated the OM permeabilization of YW-1. In aqueous conditions, 1-N-Phenyl naphthylamine (NPN) is blocked by the cell wall. However, if the OM is permeabilized, and NPN is taken up as a result, the fluorescence intensity will increase, compared to non-treated OM.⁵¹ As shown in Figure 3A, with 1% Triton as a control, the permeabilization capability of the OM was determined by the absorbance of NPN in a concentration-dependent manner. YW-1 exhibited a good potency in outer membrane permeability with 1 × MIC of 98%, compared with that of 1% Triton (Figure 3.6.).

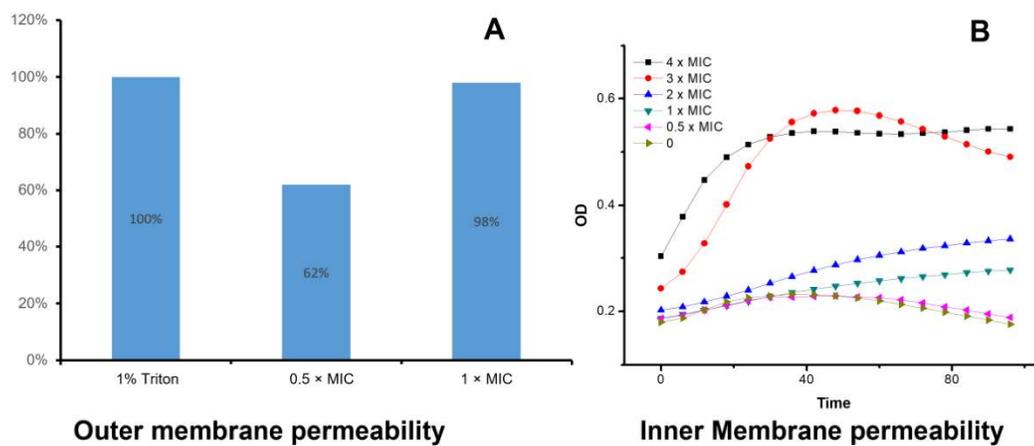


Figure 3.6. Outer membrane (OM) permeabilization (A) and inner membrane permeabilization (B) against *E. coli*.

3.4.3. Inner Membrane (IM) Permeabilization

We next used the o-nitrophenyl- β -d-galactopyranoside (ONPG) hydrolysis assay to test the ability of the lead compound YW-1 to permeabilize the inner membrane of Gram-negative bacteria (Figure 3B). If the lead compound YW-1 compromised the bacterial inner membrane, ONPG would interact with the cytoplasmic enzyme β -galactosidase to form o-nitrophenol, which can be measured under OD 420 nm.⁵² As shown in Figure 3B, when *E. coli* were treated with YW-1 with $8 \times \text{MIC}$, $4 \times \text{MIC}$, $2 \times \text{MIC}$, and $1 \times \text{MIC}$, the OD 420 nm intensity was enhanced to a great extent.

3.4.4. Fluorescence Microscopy

To further assess that the bacteria were killed with YW-1 by comprising the cell membrane, we conducted a fluorescence microscopy experiment (Figure 3.7). Two dyes, 4',6-diamidino-2-phenylindole (DAPI) and propidium iodide (PI), were employed in this experiment. PI can only stain dead or injured cells since it does not have cell membrane permeability. Instead, DAPI can be used to stain both dead and live cells because it is cell permeable. After MRSA and *E. coli* were treated with YW-1, red fluorescence was observed under the PI channel (Figure 3.7.), suggesting that bacterial membranes were compromised by YW-1.

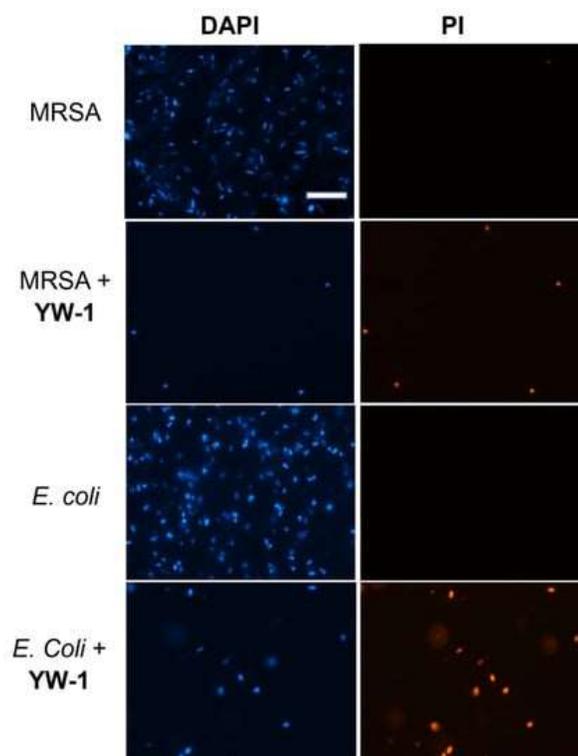


Figure 3.7. Fluorescence microscopy of the bacteria MRSA and *E. coli* treated with $2 \times$ MIC of YW-1, scale bar 10 μm .

3.4.5. Transmission Electron Microscopy (TEM)

TEM microscopy offers convenient access to check the intactness of cell membranes. The cell membranes of MRSA and *E. coli* were intact. However, after treatment with YW-1, the membranes clearly demonstrated damages, and bacterial cells lost their spherical (MRSA) or rod (*E. coli*) shapes (Figure 5), indicating that the membranes of these bacteria were disrupted (Figure 3.8).

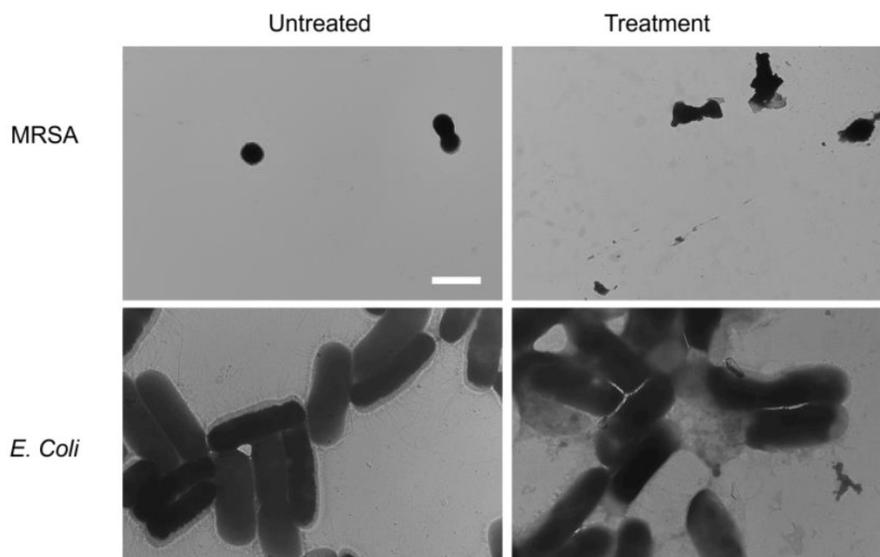


Figure 3.8. TEM graphs of MRSA and *E. coli* in the presence and absence of YW-1 at $2 \times$ MIC, scale bar $2 \mu\text{m}$.

3.4.6. Bacterial Killing Efficiency

3.4.6.1. Time-Kill Kinetics Study

To evaluate how rapidly bacteria could be eradicated (Figure 3.9), the lead compound YW-1 was examined for its ability to eradicate MRSA and *E. coli* by studying the time-kill kinetics. As shown in Figure 6, YW-1 could remove MRSA and *E. coli* thoroughly with $4 \times$ MIC and $8 \times$ MIC within 2 h.

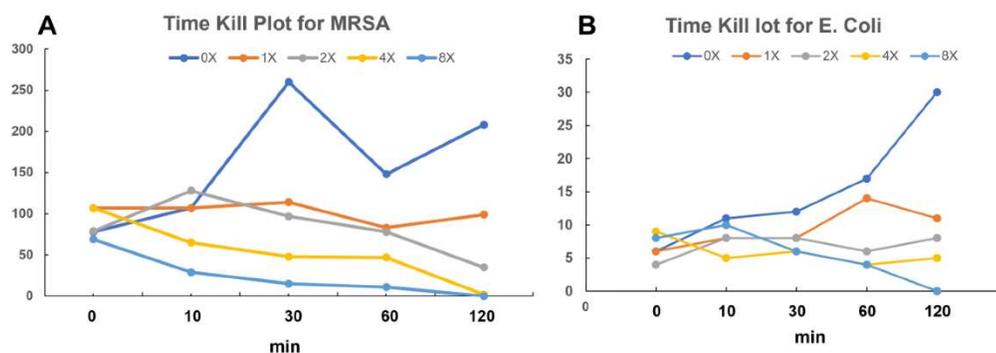


Figure 3.9. Time-kill kinetics of YW-1 toward MRSA (A) and *E. coli* (B).

3.4.6.2. Drug Resistance Test

Since the abuse of antibiotics, antibacterial drug resistance has become an increasingly serious threat. Thus, it is of great importance to assess the drug resistance of newly developed lead compounds. As shown in figure 3.10, the control, ciprofloxacin, developed drug resistance after 14 generations, since the MIC increased about 100-fold. In comparison, YW-1 did not have an obvious change in MIC, which suggests that YW-1 has a low probability of developing antibiotic resistance. Our compound had a lower probability of developing drug resistance.

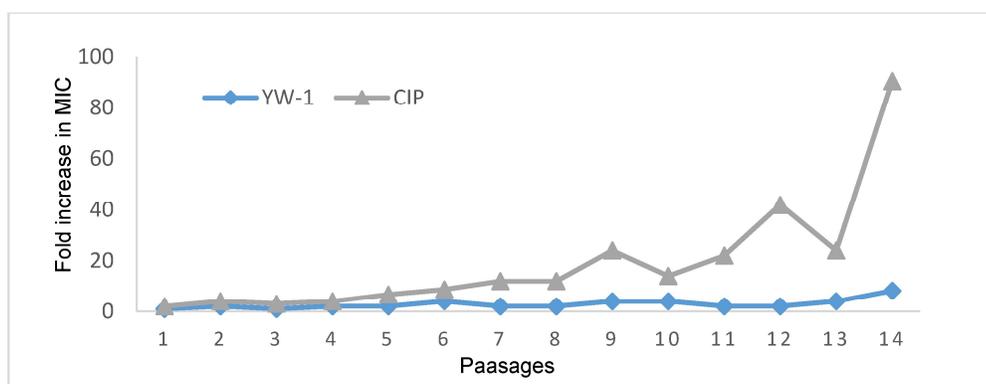


Figure 3.10. Drug resistance study of YW-1 toward MRSA.

3.5. Experiment Procedure

3.5.1. General Experiment Methods

Chemical reagents and solvents were bought from Sigma Aldrich, Oakwood, TCI, and Chem-Impex. Final products were purified with Waters Breeze 2 HPLC and lyophilized using Labconco lyophilizer. HPLC traces of the final product were collected using 5–95% acetonitrile in water with 0.1% HPLC-grade TFA for at least 40 min. The nuclear magnetic resonance (NMR) data were collected using the Agilent 600 MHz NEO instrument. High-resolution mass spectra of compounds (Table S1) were collected using an Agilent Technologies 6540 UHD accurate-mass Q-TOF LC/MS spectrometer. Antibacterial assays and mechanism of action studies were performed using a Biotec multimode microplate reader synergy H4. Six different species of

bacteria were used for bacteria assay tests, such as MRSA (ATCC 33591), MRSE (RP62A), VREF (ATCC 700802), E. coli (ATCC 25922), K. pneumoniae (ATCC 13383), and P. aeruginosa (ATCC 27853).

3.5.2. MIC (Minimum Inhibitory Concentration)

Concisely, E. coli, K. pneumoniae, P. aeruginosa, MRSA, MRSE, and VREF were cultivated in 37 °C TSB medium for 16 h. Subsequently, 4 ml of new TSB medium was added to 100 µL of cultivated bacterial solution and incubated for another 6 h to reach the mid-log phase. Next, 96-well plates were injected with 50 µL of bacterial solution that reached mid-log phase and 50 µL of compounds with different concentrations, which ranged from 0.75–25 µg/mL. Then, the 96-wells plates were incubated for another 16 h at 37 °C. After 16 h, MICs were determined by a multimode microplate reader.

3.5.3. Hemolytic Activity

Human red blood cells were washed with 1 × PBS buffer, then centrifuged at 700 g for 10 min. After discarding the top clear solution, the bottom cells were diluted to obtain 5% solution using 1 × PBS buffer.

Then, 50 µL of cell solution was injected into 96-well plates. Following this, 50 µL of synthesized compounds with different concentrations from 250 to 1.95 µg/mL were added into 96-well plates. Then, the mixture was incubated for 1 h at 37 °C. After centrifuging at 3500× g rpm for 10 min, 30 µL of the supernatant was transferred into a new 96-well plate with 100 µL of 1 × PBS buffer. With the same microplate reader, the data of absorbance at 540 nm were compared. Positive control: 2% Triton-100. Calculation formula: percentage of hemolysis = [(Absorbance of sample-Absorbance of PBS)/(Absorbance of Triton-Absorbance of PBS)] × 100.

3.5.4. Drug Resistance Study

The first generation of MIC data of YW-1 against MRSA and E. coli was already obtained in an MIC study. Then, the MRSA of the first generation in the well next to the last clear well was diluted to the mid-log phase, and MICs were tested at 37 °C. This step was repeated for 14 passages. The data of drug resistance for E. coli were obtained using the same method with bacterial E. coli.

3.5.5. TEM

An amount of 30 µL of mid-log phase MRSA and E. coli was diluted to 3 mL in TSB medium with 2 × MIC of YW-1, and the mixture was incubated for 2 h. The bacterial pellets were centrifuged at 3000× g rpm for 10 min. For the next step, PBS buffer was used to wash three times. Then, the suspended bacterial samples were dropped on grids and dried in a vacuum oven at 45 °C. TEM images were obtained using a FEI Morgagni 268D TEM, with an Olympus MegaView III CAMERA on the microscope, at 60 kV.

3.5.6. Inner Membrane Permeability

Mid-log phase E. coli was obtained in Mueller Hinton Broth with 2% lactose at 37 °C; then, it was centrifuged at 3000× g rpm for 10 min at 4 °C and washed with 20 mM glucose and 1.5 mM ONPG in 5 nM HEPES buffer one time. Next, the bacterial solution was diluted until OD600 = 0.1 using the same buffer. Following this, 50 µL of diluted bacterial solution was injected into a 96-well plate, and 50 µL of YW-1 in different concentrations and melittin were injected into the bacterial solution, respectively. The OD420 was read at 37 °C every 6 min until the fluorescence reached the highest plateau.

3.5.7. Fluorescence Microscopy

A total of 30 µL of mid-log phase bacterial solution was diluted to 3 mL in TSB medium with 2 × MIC of YW-1. The mixture was incubated for 2 h at 37 °C. Following this, the bacterial

solution was centrifuged at 3000 rpm for 10 min. The top solution was thrown away, the bottom bacterial pellets were washed with PBS buffer, and PI (5 µg/mL) and DAPI (10 µg/mL) were added sequentially on ice under a dark environment. After dyeing with PI and DAPI, the bacterial cells were washed with PBS buffer. Immediately after 100 µL of PBS was added to suspend the bacterial cell, 10 µL of the suspended solution was dropped onto the slide, and data were obtained using a Zeiss Axiovert 200 inverted microscope.

3.5.8. Time-Kill Kinetics Study

Different concentrations of YW-1 and ciprofloxacin were mixed with 300 µL of mid-log phase bacterial solution in TSB medium. The mixture was incubated for 0, 10, and 30 min and 1 and 2 h, respectively. At the time, *E. coli* were diluted 100-fold, and the MRSA was diluted 100-fold. A total of 100 µL of each was transferred on TSB agar plates. After 16 h at 37 °C, CFUs were read using the Biotec multimode microplate reader.

3.6. Conclusions

In summary, we developed a new series of antibacterial compounds. The lead compound, YW-1, showed potency towards both Gram-positive and Gram-negative bacteria. Further, YW-1 killed bacteria by disrupting cell membranes, confirmed with TEM, OM permeabilization, and membrane depolarization though without apparent drug-developed resistance. Meanwhile, it displayed good selectivity with low hemolytic toxicity. Taken together, YW-1 showed good therapeutic potential, and it could be a candidate to solve the problem of AMR.

3.7. General Synthesis of Compound

The Fmoc protected amino acids were bought from Chemimpex, reagents used in reaction are bought from Fisher Scientific and Sigma Aldrich. The building blocks were synthesized

following previously published paper from our group. Next, the building block was loaded on Rink-Amide beads with DIC and HOBt in DMF reacting for 4 hours. After deprotecting Fmoc protecting, palmitic acid was coupled DIC and HOBt. The Alloc protecting group of the secondary amine was removed with Dimethylamine Boron ((CH₃)₂NH.BH₃) and Tetrakis(triphenylphosphine) Palladium, the second building was coupled like the first building block. Deprotect the alloc protecting group again, then couple Fmoc protected beta alaine. Subsequently, the Fmoc was removed, and the beads were cleaved with 50% TFA in DCM. The solvent was removed and solid was precipitated with cold ether. Finally, the oil was purified by HPLC and got the last product.

Table 3.2. HRMS data

Peptide	HRMS (ESI) [Exact Mass+H ⁺] calculated	HRMS (ESI) [Exact Mass+H ⁺] found
YW-1	667.5742	668.5747
YW-2	701.5568	702.5604
YW-3	625.5255	626.5286
YW-4	717.5517	718.5532
YW-5	641.5204	642.5214
YW-6	682.5833	683.5852
YW-7	724.6051	725.6050
YW-8	673.5255	674.5298
YW-9	645.4942	646.4979
YW-10	617.4629	618.4668
YW-11	589.4316	590.4355

CHAPTER FOUR: BCL-9 P-PROTAC TO DEGRADE B-CATENIN

4.1. Fmoc SPPS

4.1.1. Background of Fmoc SPPS

Pioneered by Robert Bruce Merrifield, the well-established peptide synthesis in lab is known as solid phase peptide synthesis. On a solid-support resin, it allows scientists to synthesize peptide chain via successive coupling AA.⁵³ In contrast to solution phase synthesis that time-consuming isolation of product from reaction solution, the nascent peptide chain attached covalently to the solid supportive resin that functionalized with different reactive groups, such as hydroxy or amine groups. Usually, vessels for SPPS have silica pads that help researchers remove excessive starting material and side products easily. Depending on the side chain and protecting strategy, nowadays, the most often used two SPPS are Fmoc-SPPS and Boc-SPPS. The general procedure of Fmoc-SPPS is to repeat several cycles of coupling N-terminal Fmoc protected AA. This cycle is repeated until the desired peptide sequence is done. During coupling steps, capping methods, such as a mixture of pyridine and acetic anhydride, are often used to block the unreacted amine to reduce side products. Compared with Boc/Bzl SPPS using HF as N-terminal deprotecting strategy, Fmoc-SPPS deprotection method is much more milder using 20-50% piperidine in DMF (Figure 4.1).⁵⁴ Moreover, the new free amine is neutral, not like Boc SPPS that free amine is protonated under strong acidic condition, that is directly used for next step coupling reaction without any neutralization. It is also easy to confirm whether the deprotection is done since chromophores are produced. Because chromophores show strong

color under UV, it is very convenient to check the reaction. Usually, after two times deprotection with 20% piperidine for 10 min, the deprotection is done.

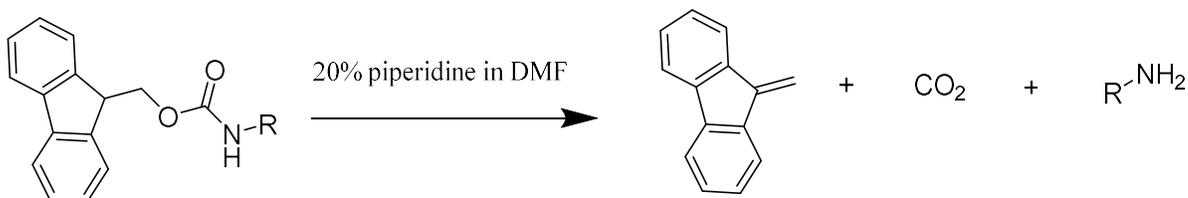


Figure 4.1. Deprotection Mechanism of Fmoc-SPPS.

On Fmoc SPPS, the side chains of AA are usually acid sensitive protecting groups, such as Boc used for Lys, tBu used for Glu and Asp, Trt used for Gln and Asn. The cleavage cocktails can remove all the acid sensitive protecting groups at cleavage step. Scavengers including triisopropylsilane (TIPS) and water are most added with a small ratio during cleavage step to prevent side reactions. Nevertheless, some other scavenger reagents could also be use.^{55, 56}

4.1.2. Coupling Reagents and Limitations for Fmoc SPPS

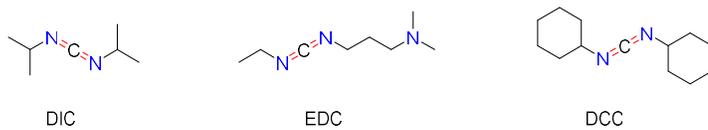
For the coupling reagent, diisopropylcarbodiimide (DIC) is frequently used SPPS since it is liquid, and the urea byproduct is easily washed away by DMF. Another carbodiimide coupling reagent, 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC), is usually used for solution phase coupling as it has a tertiary amine, as a result, the byproduct is easily washed away during aqueous work-up. To circumvent racemization problem, different “racemization suppressing” chemicals are added to reaction, such as the triazoles 1-hydroxy-benzotriazole (HOBt), and 1-hydroxy-7-aza-benzotriazole (HOAt). After carboxylic acid forms O-acylisourea intermediate with carbodiimide, these two chemicals will attack O-acylisourea to form an active ester, which will react with a free amine to form an amide bond.⁵⁷

The other two often used coupling reagents are aminium/uronium and phosphonium salts (Figure 4.2). This type coupling reagent completely omit carbodiimide. Instead, it incorporates

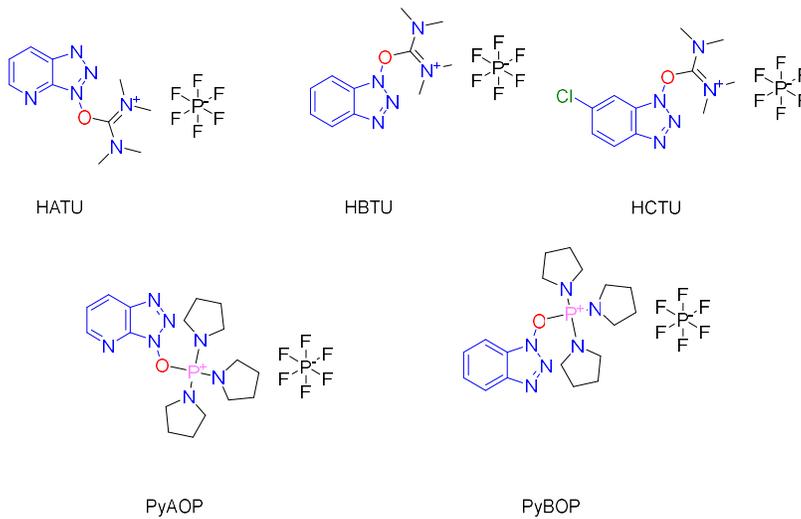
HOAt/HOBt moiety as an aminium/uronium or phosphonium as an anion (tetrafluoroborate or hexafluorophosphate). Aminium/uronium coupling reagents include HATU (HOAt), HBTU/TBTU (HOBt), HCTU (6-Cl HOBt). The only difference between HBTU and TBTU is the choice of anion. Phosphonium coupling reagents include PyBOP (HOBt) and PyAOP (HOAt) .⁵⁸

However, this step-by-step coupling SPPS has limitations. Even though it is ideal for small peptides containing AA from 2 to 100, it is difficult to synthesize even longer peptide sequence. For longer peptide sequences, the methods developed are chemical ligations, such as native chemical ligation (NCL), Ser/Thr ligation (STL), α -Ketoacid-Hydroxylamine (KAHA) Amide-Forming Ligation. In solution phase, unprotected peptide chain reacts chemo selectively with each other. For example, in NCL, a peptide thioester would selectively react with an N-terminal cysteine peptide sequence.

A Carbodiimide Coupling Reagent:



Aminium/uronimu and Phosphonium Coupling Reagent:



B

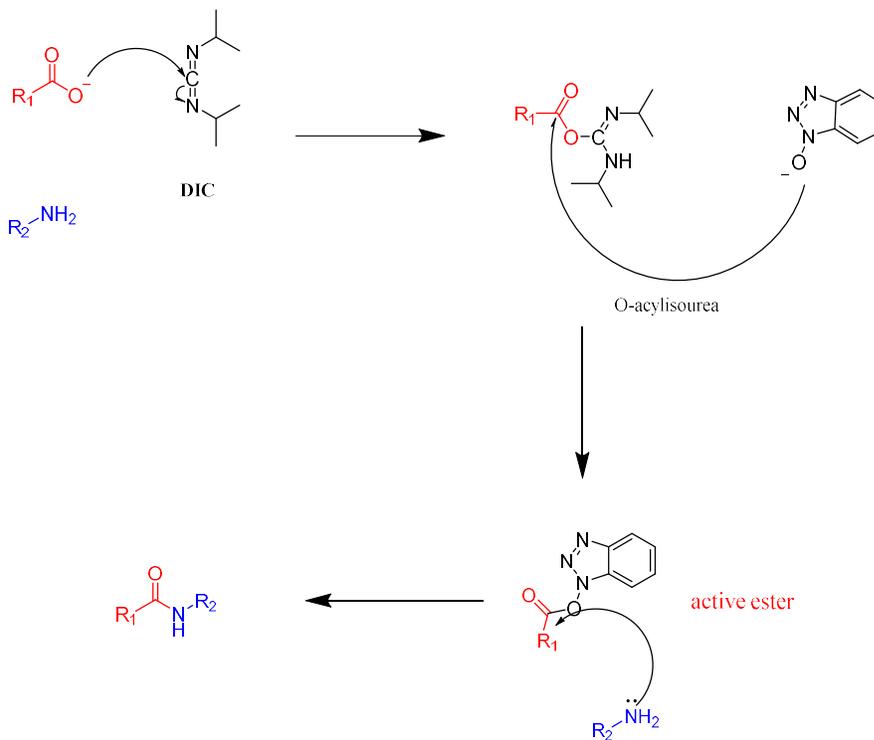
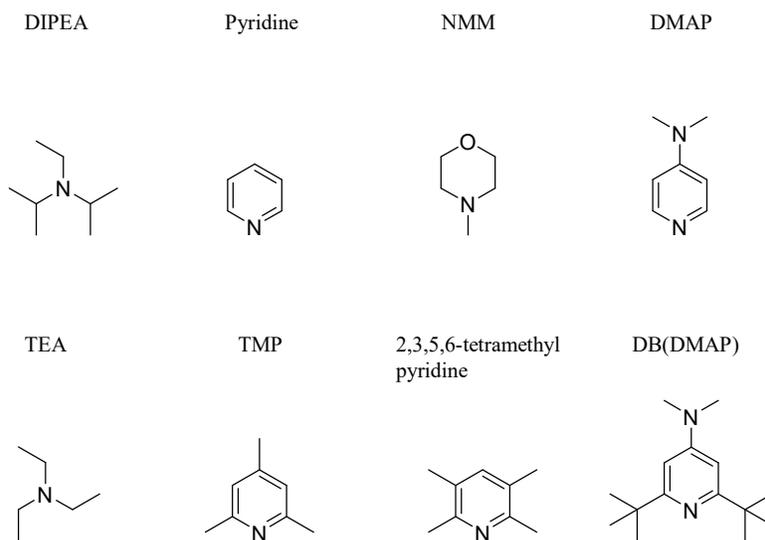


Figure 4.2. Coupling reagents and mechanism. A: Different Types of Coupling Reagents. B: Carbodiimide Reagent Coupling Mechanism.

Bases for SPPS:



Additives for Coupling

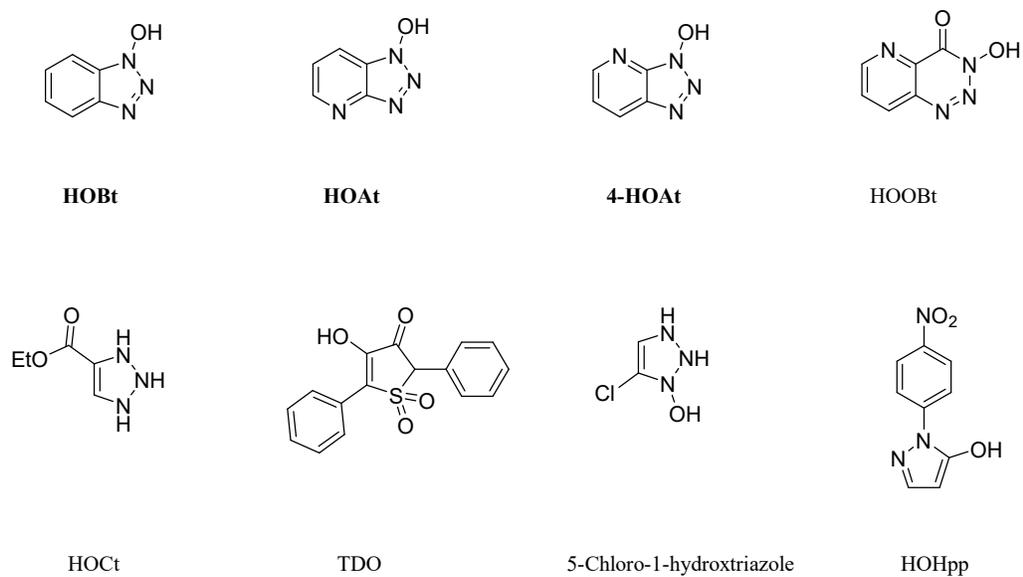


Figure 4.3. Base and additives for coupling.

4.1.3. Resins in Fmoc-SPPS

Resin Core: Polystyrene is most often used resin core in SPPS, but other core matrices are also used in special synthesis, including polyacrylate, polyacrylamide, polyethylene glycol.⁵⁹ These other core resins are utilized in some special peptide synthesis, such as preparing some peptides that are prone to aggregate.

Linear polystyrene tends to dissolve in hydrophobic solvents, but precipitate in protic solvents. Usually, polystyrene supports used in SPPS contain 1-2% DVB as crosslinking agent. Crosslinked polystyrene is insoluble in common solvents; thus, it is normally prepared and used as small, spherical beads.

Polystyrene

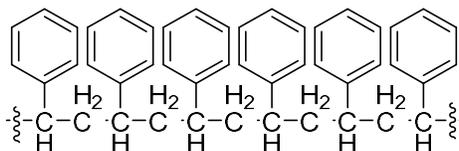
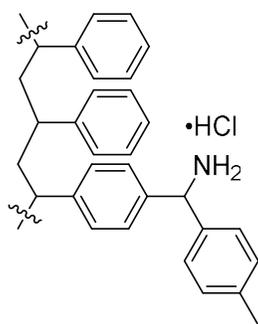
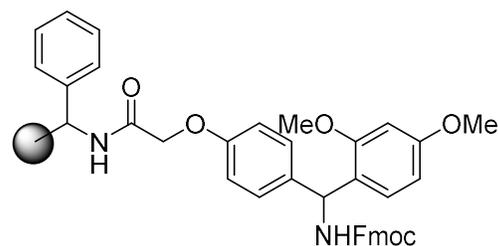


Figure 4.4. Structure of Polystyrene

Amino Core Resins: Aminomethyl (AM) has long been used as core resin in SPPS.⁶⁰ At first, 4-Methylbenzhydryl amine resins (MBHA) were developed for the formation of peptide bond in Boc-N protection/TFA deprotection. Generally, carboxylic acids and electrophilic alkyl substrates would form very stable amide or amine bonds. To cleave product, strong acid is required, such as high concentration of TFA. At the same time, it can also be used as base resin to anchor acid labile linkers such as Rink Amide linker.⁶¹ These resins are generated via electrophilic aromatic substitution (EAS), and the efficiency of EAS is hard to control. Consequently, despite optimized protocols, the qualities of AM resins vary in different batches.



MBHA resin HCl



Fmoc-Rink Amide MBHA Resin

Figure 4.5. Structure of AM resin

Merrifield Resin: Merrifield resin was named after the Noble Laureate who first used it in peptide synthesis, the structure is chloromethyl polystyrene.⁵³ Linkers are attached to Merrifield Resin by nucleophilic attack replacing the original chloride. The resulting new bond is usually acid stable and requires strong acid for cleavage. However, carboxylic acid is not easily freed from Merrifield resin using acid cleavage cocktail. Instead, other cleavage methods including saponification, transesterification, and cyclization-release have proven effective.⁶²⁻⁶⁵ Merrifield resin is usually generated by two methods: direct incorporation via EAS or substituted monomer co-polymerized with styrene. Overall, substituted resins are generated by substrate direct incorporation which leads to isomers. For instance, 70% of Merrifield resin are para-chloromethyl substituted, the other 30% resin are meta-substituted. Conversely, by using pure monomer, co-polymerization can prepare 98% para substituted resin. In addition, it can control the degree of resin substitution by changing the ratio of styrene and substituted monomer (Figure 4.6).

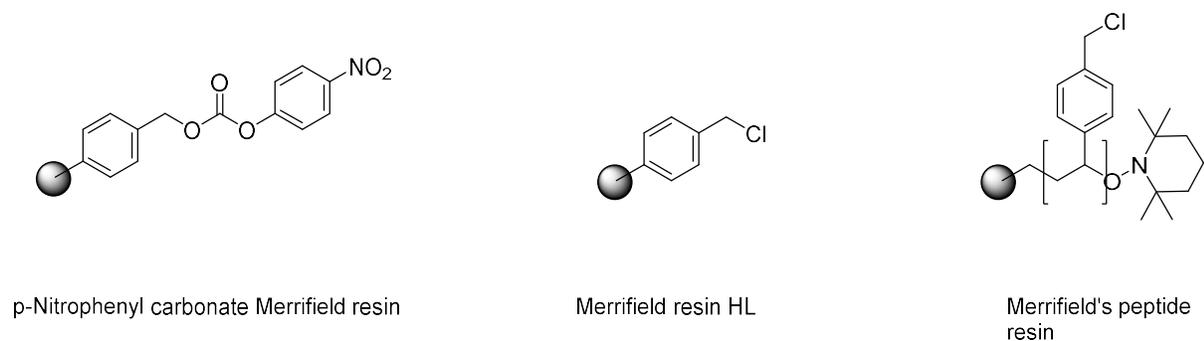


Figure 4.6. Different kinds of Merrifield resin

Swelling Factor of Resin: Even though resins are insoluble in organic solvents, they can be swollen by aprotic solvents such as toluene, dimethylformamide (DMF), and dichloromethane (DCM) (Table 4.1.). For example, one gram of 1% of divinylbenzene (DVB) cross-linked resin swells 4-6 times as its original volume in DCM. In comparison, one gram of 2% DVB cross-linked resin swells 2-4 times in DCM. Since the reaction kinetics depends on diffusion, the swelling factor matters in SPPS. Consequently, well-swollen beads have a shorter reaction time and better product conversion since it has a higher rate of reagents diffusing into the core part of matrix.

Table 4.1. Swelling factor of 1% cross-linked polystyrene resins in different solvents.

THF	5.5	Acetonitrile	3.5
Toluene	5.3	Et ₂ O	3.2
DCM	5.2	EtOH	2.0
Dioxane	4.9	MeOH	1.8
DMF	4.7	H ₂ O	1.0 (no swelling)

4.1.4. Cleavage Cocktails

The linker is where the C-terminal amino acid attaches, and there is a wide range of commercially available linkers for different polymeric supports. Actually, the selected cleavage cocktail depends on the linker.

Low concentrations of TFA: With a XAL (“Seiber”) or HAL linkers, synthesized peptides are cleaved in low concentration of TFA to get fully protected peptide amides or peptide acid respectively.

High concentration of TFA: With a PAC (“Wang”), PAL, AM (“Rink Amide”), or BAL linker, synthesized peptides are cleaved in high concentration of TFA to get fully-deprotected product (Table 4.2.).

Table 4.2. High Concentration of TFA Fmoc-SPPS Cleavage Cocktail.

	Recipe	Time (h)	Comments
B	TFA/water/phenol/TIPS (88/5/5/2)	1-4	All peptides
K	TFA/phenol/water/thioanisole/EDC (82.5/5/5/5/2.5)	1-4	All peptides
K'	TFA/phenol/water/thioanisole/1-dodecanethiol (82.5/5/5/5/2.5)	1-4	All peptides
L	TFA/DTT/Water/TIPS (88/5/5/2)	1-4	All peptides
P	TFA/phenol (95/5)	1-4	tBu group. Do not use with Trp, Met or Cys.
P'	TFA/phenol/Methanesulfonic acid (95/2.5/2.5)	15 min	All peptides
R	TFA/thioanisole/EDT/Anisole (90/5/3/2)	1-8	All peptides
T	TFA/TES (95/5)	1-4	Boc, tBu, Trt. Do not used with Arg or Trp
	TFA/water (95/5)	1-4	Boc, tBu, Trt, Pbf. Do not used with Trp, Met or Cys
	TFA/DCM/indole (70/28/2)	1-4	Do not use with Arg.

Both cleavage cocktails should be prepared fresh before using. Scavengers should also be fresh. Try to buy TFA, and scavengers in small quantities, and discard any long unused scavengers.

4.1.5. HPLC Analysis and Purification

HPLC is the most powerful and convenient tool to analyze and purify peptides. Following is the suggested guidelines:

1. With an unknown peptide, usually, a C-18 reversed phase column is recommended for medium-sized peptides that are moderately hydrophilic.

Use the following protocol for a standard analytical column:

Buffer A: 0.1% TFA in H₂O.

Buffer B: 0.1% TFA in ACN.

Flow rate: 1-1.5ml/min

Gradient: 0-90% B in 90min.

2. For peptides with 20-40 amino acids, medium-sized peptides, C-8, C-4 or a polymeric reversed phase column are recommended. These peptides are expected to elute at a high percentage of CAN, and it is safe to start with a high percentage of 10-20% B.
3. Long peptides, or those having many similarly charged groups, may best be purified by either polymeric reversed phase or aqueous ion exchange columns.
4. Peptides containing aromatic side chain, such as Tyr, Phe, or Trp, can be monitored at 240-260 nm, due to characteristic absorbance of aromatic ring. Otherwise, monitor at 210-214 nm, this is close to the wavelength of peptide bond.
5. If DMF is used to help dissolve the peptide, you will see a large peak at the beginning. If peptide is small and eluted early, use low percentage of B to separate peptide from DMF.

4.2. Introduction

With Wnt ligand, the canonical Wnt pathway causes accumulation of β -catenin in cell cytoplasm, next, it is translocated to nucleus and form a complex with BCL 9, PYGO, TCF/LEF to activate Wnt target gene transcription. Previous research demonstrated that BCL 9 amplification was related to several human cancers and other diseases, such as breast cancer, prostate cancer, type 2 diabetes and others.⁶⁶ It is also associated with tumor progression, low survival rate and not ideal clinical outcomes. Furthermore, an enhanced level of β catenin can increase cell proliferation, migration invasion, and tumor metastasis.⁶⁷ Without Wnt ligand binding, adenomatous polyposis coli (APC) and glycogen synthase kinase 3 β (GSK3 β) bind to β -catenin, after several cycles of phosphorylation and ubiquitination, it is degraded by proteasome.⁶⁸

Proteolysis Targeting Chimera is a strong modality tool to degrade intracellular or nuclear protein of interest. It provides a new way to solve the drug resistance problem of small molecular inhibitors. The PROTAC molecules consists of three parts: E₃ ligand, linker, and targeting protein bind ligand. After recruiting E₃ ligase to targeting protein, they form a tertiary complex and E₃ ligase ubiquitinates the target protein several times, then recognized and degraded by proteosome.⁶⁹ PROTAC molecules can be used to target undruggable protein, as only a tight binder is required for the target protein.⁷⁰ Furthermore, PROTAC molecules is also a great tool to validate the targets and gather more information about protein functions and signaling pathways. Compared with nucleotide-based methods like oligonucleotides, RNAi or genome editing methods that have limited applications due to their stability *in vivo* and *in vitro*. Most PROTAC molecules are developed from small molecule bind ligand, usually their molecular weights are not more than 1500 Da. As a result, they possess the advantages of small molecules, such as, good membrane permeability, *in vivo* stability, target specific and economic synthesis.⁷¹ These make PROTAC

molecules can enter cell and rapidly delete target proteins efficiently. It has been 20 years since the first research paper about PROTAC molecule been published. Now, this technology is translocated from academic to industry. As a new area of drug discovery, PROTAC strategies are successfully applied to degrade different proteins *in vitro* and *in vivo*, including estrogen receptor (ER),⁷² androgen receptor (AR), Kirsten rat sarcoma virus (KRAS) G12C mutation,⁷³ Bromodomain-containing protein 9 (BRD 9),⁷⁴ Signal transducer and activator of transcription 5 (STAT5),⁷⁵ Cyclin-dependent kinase 9 (CDK 9)⁷⁶, etc. Many biotech startups and companies build their own PROTAC pipelines and several PROTAC molecules have entered clinical trials. For example, ARV-110 and ARV-471 entered clinical trials in 2019, they provided the first clinical proof-of-concept for well-established tumor modality with PROTAC molecules.⁷⁷ In contrast to small molecule PROTAC, there are not so researches about peptidyl PROTAC (p-PROTAC) (Table 4.3.). The main challenges p-PROTAC face are poor cell permeability and low stability. However, p-PROTAC also has several advantages. First, it can be used to target some undruggable proteins with large shallow surfaces. Second, it has fewer side effects and is safer.⁷⁸ For instance, previously published protein binding ligand and E₃ recruiting ligand were endogenous peptides with high specificity and affinity. Those endogenous ligands compared with exogenous small molecular inhibitors and antibodies are ideal choices for drug development.⁷⁸

Table 4.3. Comparison of p-PROTAC and small molecule PROTAC.

	Peptide-PROTAC	Small molecule PROTAC
Targeting warhead	Peptides	Small molecules
Advantages	Specific targeting “undruggable” POI with specificity Resistance to target mutation Not labor intensive Low toxicity and high safety <i>in vivo</i>	Higher cellular permeability Better stability Cost effective
Disadvantages	Poor cell membrane permeability Lower stability Few research on their efficacy	Limitation of degradation of “undruggable” protein Inability to target “undruggable” shallow surface protein Toxic side effects
Clinical trials	NA	ARV-110, ARV-471, NX-2127

In 2019, Dr. Cai’s lab reported several sulfonyl γ -AA peptides that could effectively inhibit the protein-protein interaction (PPI) between BCL 9 and β -catenin.⁷⁹ Sulfonyl γ -AA peptides can mimic the secondary structure of natural helix. Owing to its unnatural amino acids and secondary helix structure, it shows better cell permeability and is resistant to enzyme degradation. We had a question: Can we degrade β -catenin using our sulfonyl γ -AA peptides?

4.3. Design

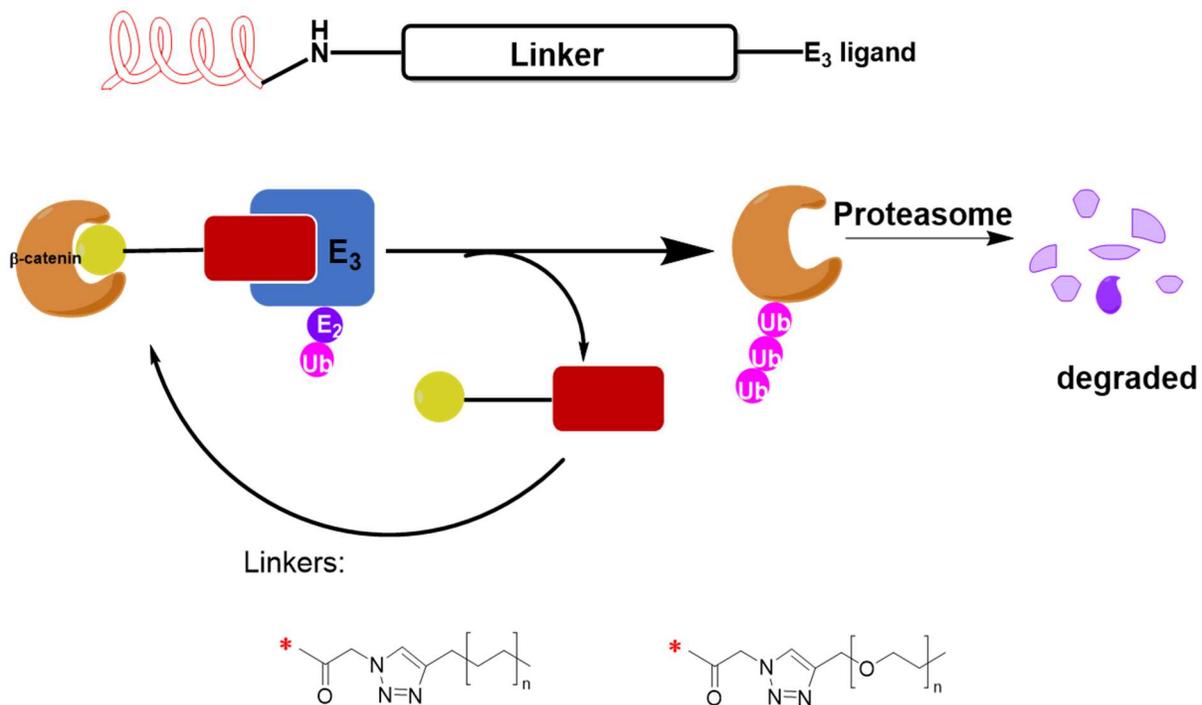


Figure 4.7. Design of Sulfonyl- γ -AApeptide PROTAC.

For our sulfonyl γ -AA p-PROTAC design, at one end, we put our sulfonyl γ -AA peptide helix and at the other end, put E₃ ligand linked by click reaction. At first, we chose sequence 2 from the paper since it showed a good binding affinity. For E₃ ligands, we tried VHL and phenyl glutarimide (PG).

4.4. Synthesis

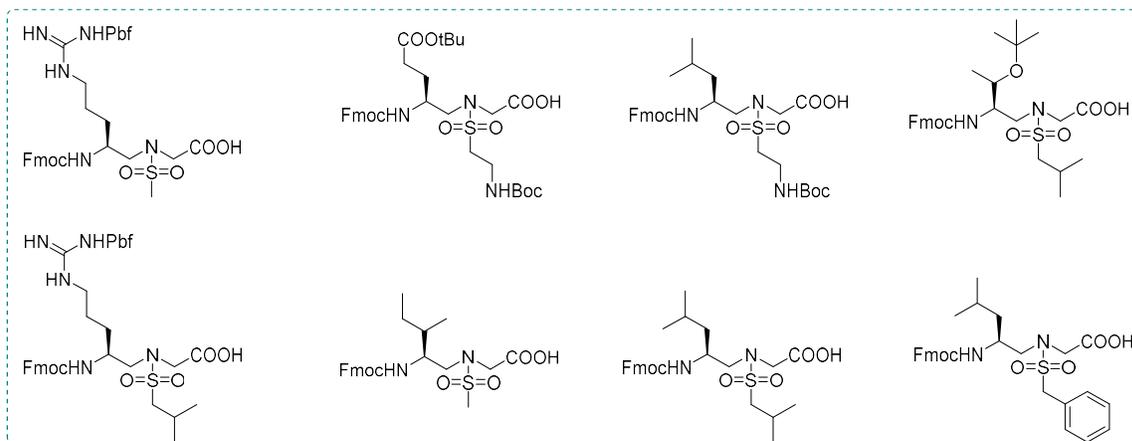
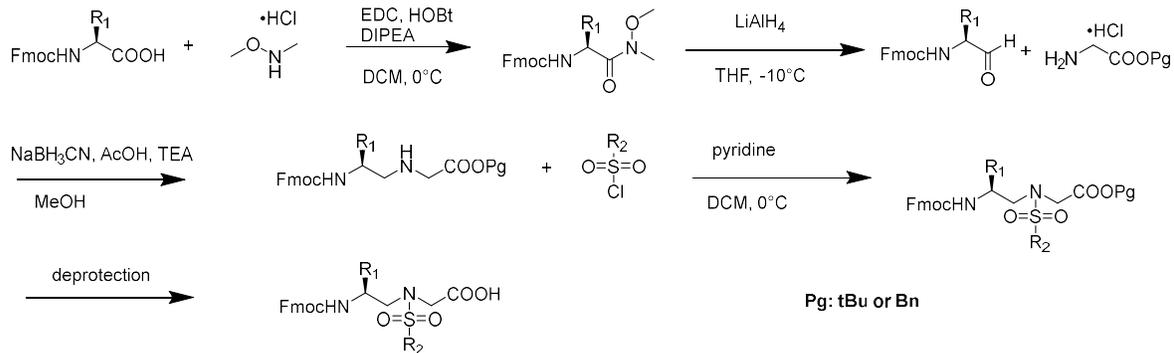
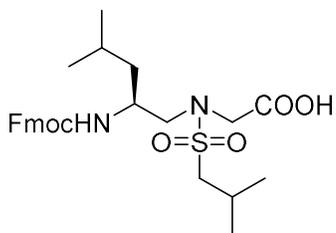


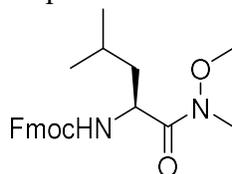
Figure 4.8. Building preparation.

4.4.1. Building Preparation

BB₁ Preparation:



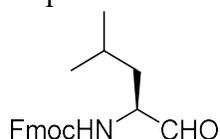
Step 1:



10g Fmoc-Leu-OH (28.3 mmol, 1eq), EDC (34.0 mmol, 1.2eq), HOBt (34.0 mmol, 1.2eq) and N, N-diisopropylethylamine (DIPEA, 62.3 mmol, 2.2eq) were dissolved in DCM under 0°C. After 5 min, 3.3g of N, O-dimethylhydroxyamine (34.0 mmol, 1.2eq) was added. The reaction solution was stirred for 2 h.

Work-up: After 2 h, DCM was removed under reduced pressure. Then 1M HCl solution was added to remove unreacted starting material and excessive EDC. The aqueous was extracted with EA two times, combined organic layers and washed with Brine, dried with Na₂SO₄. Then, EA was filtered with cotton, and removed under reduced pressure. The left oil is used for the next step without any further purification.

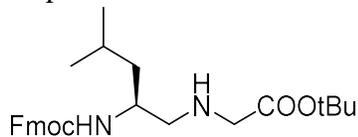
Step 2:



The first step product was dissolved in THF under sodium chloride and ice mixture (temperature between -15°C - -10°C). After cooling down for 10 min, 1.3g of Lithiumaluminumhydride (34 mmol, 1.2eq) was added in three portions. Then, the reaction was stirred for 30 min.

Work-up: Excessive 1M HCl was poured to quench reaction. Next, the aqueous layer was extracted with EA times. Combine organic layer and washed with Brine, then dried by Na₂SO₄. Then, EA was evaporated under reduced pressure. The left residue is used for the next step without any further purification.

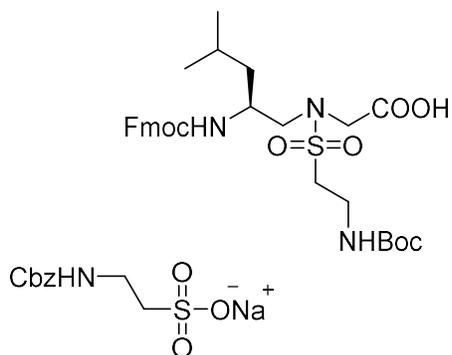
Step 3:



The previous step product was dissolved in MeOH under 0°C, then 5.1g of tert-butyl glycinate (28.3 mmol, 1eq) was added, followed by 2.8g of TEA (28.3 mmol, 1eq). The reaction mixture

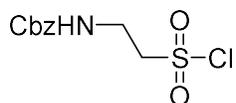
Work-up: The reaction solvent was evaporated under reduced pressure, then co-evaporated with hex 4 times to totally remove TFA. The left oil was purified by column chromatography, using HEX: EA=1:1 to get final product.

BB₂ Preparation:



Taurine (1 eq) was dissolved in a mixture of dioxane and H₂O, followed by 1 eq of CbzCl and 2 eq of NaOH. The reaction mixture was stirred overnight.

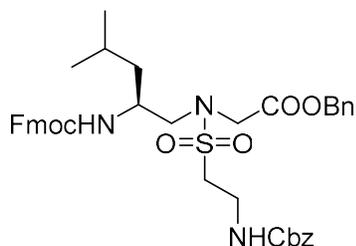
Work-up: Dioxane was evaporated under reduced pressure, then the aqueous layer was extracted with EA two times to remove unreacted CbzCl. Keep the aqueous layer, and co-evaporated with toluene to remove water several times. When the white solid precipitated, it can be used for the next step without any further purification, sodium 2-((benzyloxy)carbonyl) amino) ethane-1-sulfonate.



The previous step product, sodium 2-((benzyloxy)carbonyl) amino) ethane-1-sulfonate, was added with SO₂Cl₂ slowly. Then the reaction mixture was refluxed under 85°C for 4 h.

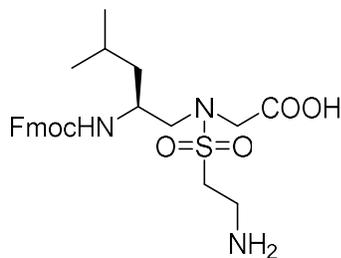
Work-up: The reaction mixture was cooled down to R.T., then SO₂Cl₂ was removed under reduced pressure. The left residue was co-evaporated with DCM for 4 time to totally remove SO₂Cl₂, then

purified by column chromatography, using HEX: EA=2:1 to get product, benzyl (2-(chlorosulfonyl)ethyl) carbamate.



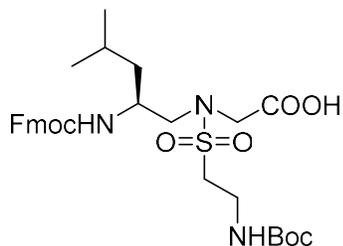
Benzyl (S)-(2-(((9H-fluoren-9-yl) methoxy) carbonyl) amino)-4-methylpentyl) glycinate was prepared using the same method as BB₁, then it was dissolved in DCM and cooled down to 0°C. Next, 10 eq of pyridine and 1.2 eq of benzyl (2-(chlorosulfonyl)ethyl) carbamate to reaction solution. The reaction mixture was stirred under R.T. overnight.

Work-up: DCM was removed under reduced pressure, then EA and 1M HCl were added. Then, the aqueous layer was extracted with EA for another time. Combine EA, washed with Brine and dried with Na₂SO₄. Next, EA was removed under reduced pressure. The left residue was purified by column chromatography, using HEX: EA=4:1 to get product, benzyl (S)-N-(2-(((9H-fluoren-9-yl) methoxy) carbonyl)amino)-4-methylpentyl)-N-((2-(((benzyloxy)carbonyl)amino)ethyl)sulfonyl)glycinate.



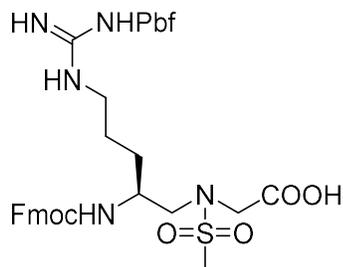
The previous step product was dissolved in a mixture of EA: MeOH=1:1, then catalytic amount (5%-10%) of Pd/C was added. The reaction ran under H₂ environment overnight. Next, the reaction was monitored by TCL. Usually, it takes three times for the reaction to be completely done.

Work-up: The reaction solution was filtered through celite, evaporated under reduced pressure. The left oil is used for the next step without any further purification, (S)-N-(2-(((9H-fluoren-9-yl) methoxy) carbonyl) amino)-4-methylpentyl)-N-((2-aminoethyl) sulfonyl) glycine.



The previous step product, (S)-N-(2-(((9H-fluoren-9-yl) methoxy) carbonyl) amino)-4-methylpentyl)-N-((2-aminoethyl) sulfonyl) glycine, was dissolved in THF and H₂O. Then, 3 eq of NaHCO₃ was added to adjust pH to basic, checked by pH paper around 9, followed by adding 1.5 eq of (Boc)₂O. The reaction mixture was stirred under 0°C overnight. Next day, the reaction was monitored by TLC, if the reaction is not done, add another 1 eq of (Boc)₂O and 1 eq of NaHCO₃. Work-up: THF was removed under reduced pressure, then 1M HCl was added to turn the pH of the aqueous to acidic. Then, the aqueous was extracted with EA for two times. Combine the EA, washed with Brine, and dried with Na₂SO₄. Next, EA was removed under reduced pressure. The left residue was purified by column chromatography, using HEX: EA=1:1 to get final BB₂.

BB₃ Preparation:



The preparation of BB₃ uses the same strategy as previous BB₁ and BB₂ preparation. However, we need to pay attention to the second step. For other natural AA, it is fine under LiAlH₄,

arginine was easy to be reduced under this strong reducing reagent. In arginine BB preparation, the LiAlH_4 reduction step should be less than 12min, if exceeding 20min, the final yield is very low.

4.4.2. Fmoc-SPPS synthesis

The Rink amide beads were sink in DCM to swell for 10 min. Then, 20% percent of piperidine in DMF was used to deprotect Fmoc protecting group. Next, the first building block was coupled with HOBt (4eq), DIC (2eq) in DMF for 4 h. Check reaction by ninhydrin test. When the reaction is done, continue coupling the next one building block until the sequence is finished. Lastly, couple the azidoacetic acid for a later click reaction or put a different length of linkers, then couple the PG E₃ ligand. After finishing all the coupling reaction, a cleavage cocktail containing 88% TFA, 5% phenol, 5% water, and 2% TIPS was used to cleave the product. Precipitated by cold ether, centrifuged, then purified by HPLC (Figure 4.9).

Sulfonyl p-PROTAC Synthesis

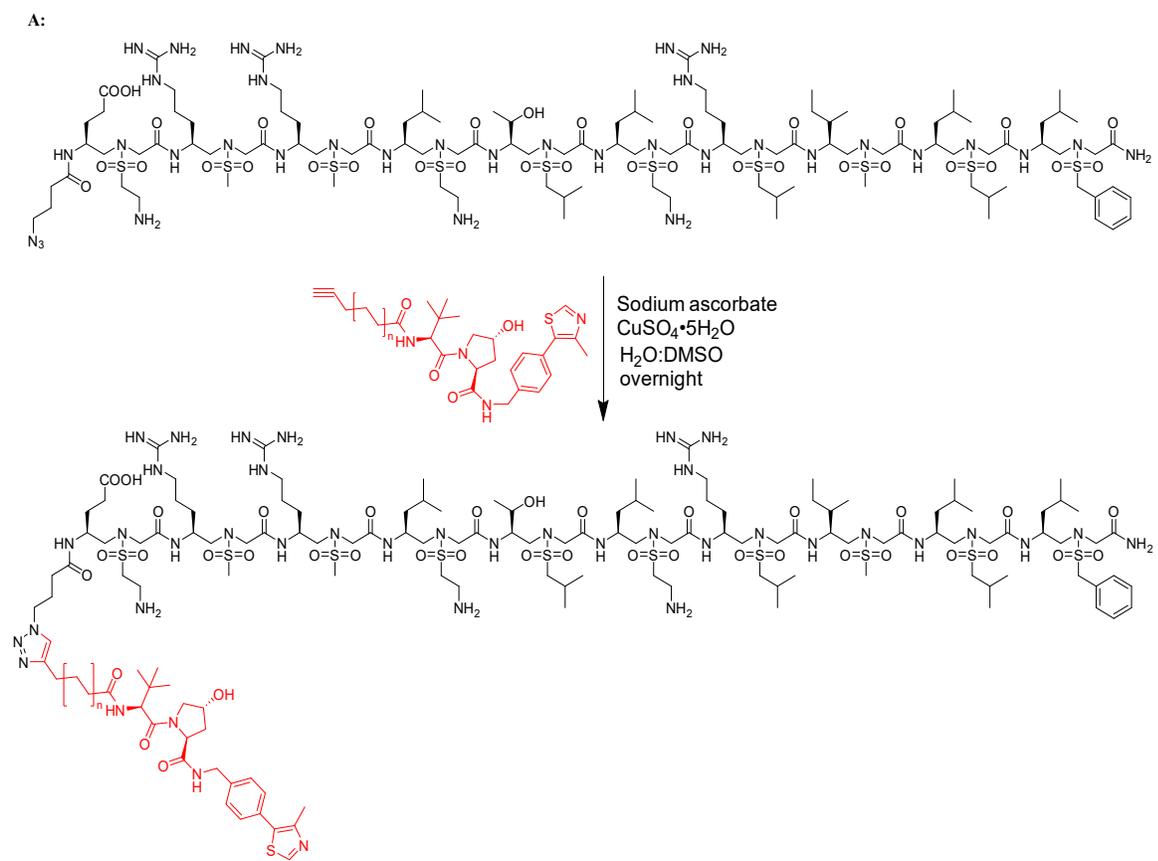
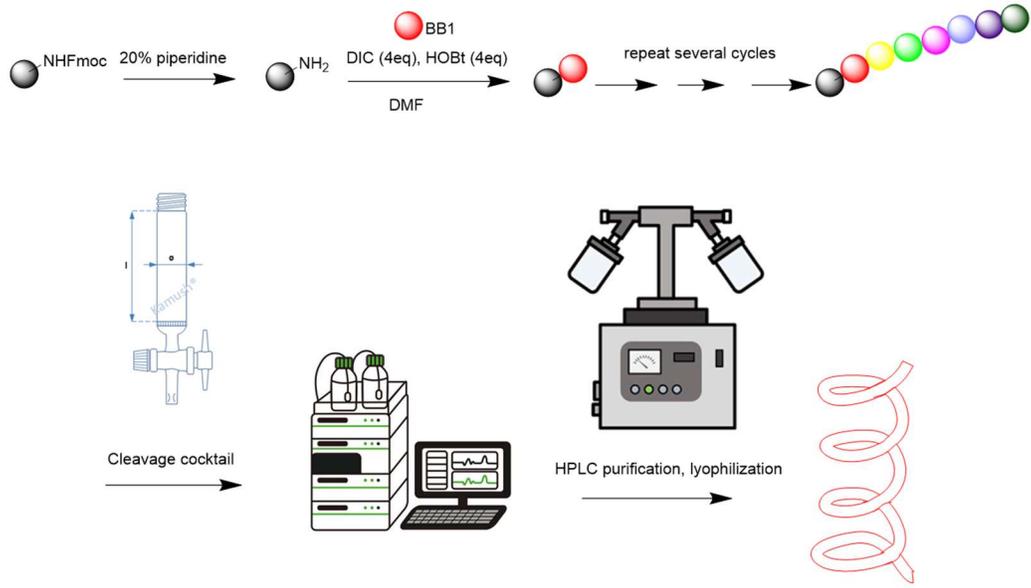


Figure 4.9. Synthesis procedure of Sulfonyl- γ -AApeptide PROTAC

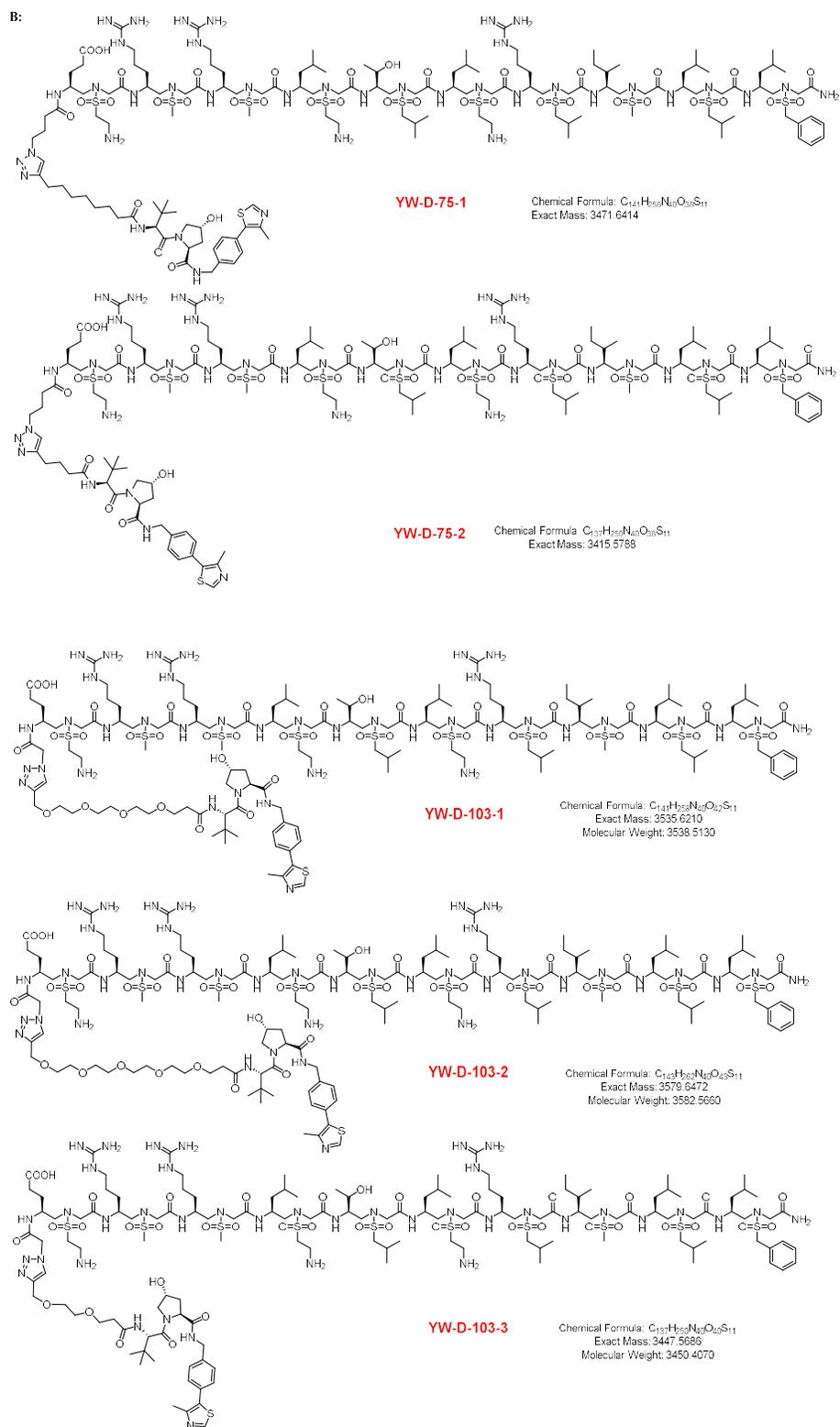


Figure 4.10. Synthesis design. A: Solution phase click reaction. B: Synthesis of different lengths of p-PROTAC.

Until now, five VHL sulfonyl p-PROTAC sequences are synthesized, our collaborator Dr. Chen at Moffit Cancer Center test their abilities to degrade β -catenin against U2OS and SW480 cells (Figure 4.11.).

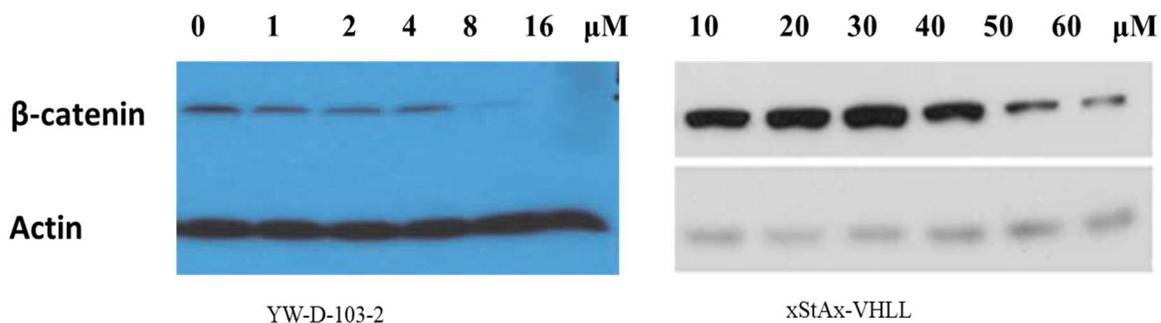
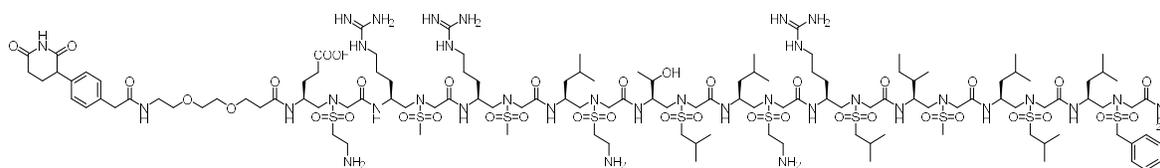


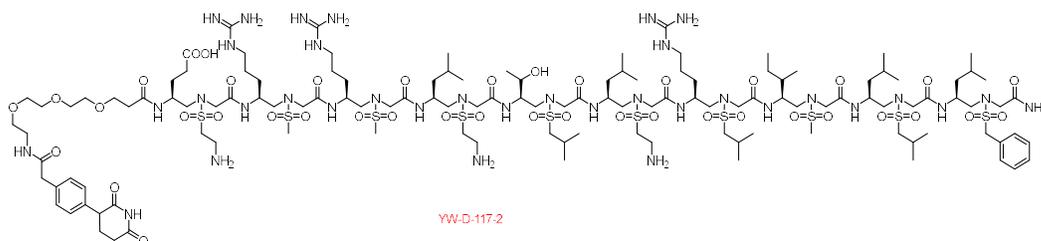
Figure 4.11. WB data of YW-D-103-2 and previously reported xStAx-VHLL

At the first western blot experiment, we found that YW-D-103-2 could effectively degrade β -catenin at 8 μ M (Figure). However, the repeat experiment was not consistent with the first time. Moreover, we found that p-PROTAC with VHL ligand showed cell toxicity, it not only degraded β -catenin but also degraded actin. As a result, we are trying to replace VHL with PG E₃ ligand since they recruit different E₃ Ligase.⁷⁷ By changing the E₃ ligand, we want to know whether the cell toxicity would disappear. The p-PROTAC with PG plan is undergoing, we finished the synthesis, and are waiting for our collaborator to test.



YW-D-117-1

Chemical Formula: $C_{127}H_{233}N_{39}O_{36}S_{10}$
 Exact Mass: 3168.4532
 Molecular Weight: 3171.0450



YW-D-117-2

Chemical Formula: $C_{127}H_{233}N_{39}O_{40}S_{10}$
 Exact Mass: 3212.4794
 Molecular Weight: 3215.0980

Figure 4.12. Structure of PG Sulfonyl- γ -AApeptide PROTAC.

CHAPTER FIVE: S597 PEPTIDOMIMETIC TO ACTIVATE INSULIN RECEPTOR

5.1. Background

Peptides are short sequences of amino acids linked by peptide bond. A polypeptide is a longer, continuous, unbranched peptide chain. Peptide sequences of not more than twenty amino acids are called oligopeptides, such as dipeptides, tripeptides, and tetrapeptides.⁸⁰ If the molecular weight of polypeptides exceeds 10000, they are called proteins. Proteins consist of one or more polypeptides, in biological functional pathway, they are bound to ligands such as coenzymes or cofactors. For example, proteins bind to DNA, RNA or another protein to form complex functional macromolecular assemblies.

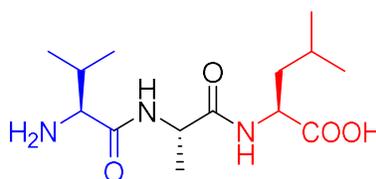


Figure 5.1. A tripeptide consists of Val, Ala, and Leu AA

Among proteins, an alpha helix is a sequence of amino acid in protein that twists into a helix. It is the most common secondary structure in proteins. It is also the most often appeared secondary structure in proteins that can be easily predicted with a sequence of amino acids. Naturally, the alpha helix has a right-handed conformation in which the N-H group of one amino acid in backbone forms a hydrogen bond to the C=O group that is four residues earlier, ($i+4-i$). Amino-acid propensities: Different amino acids have different propensities to form an α -helical structure. Alanine, leucine, methionine, lysine, and glutamic acid all have especially high helix-forming propensities; however, proline and glycine have poor helix-forming propensities. Proline

either breaks or kinks a helix since it has tertiary amine instead of the N-H bond as a hydrogen bond acceptor. Furthermore, its side chain interfere sterically with the preceding backbone-inside a helix, this forces a 30° bend of the backbone.⁸¹ For glycine, it has high conformation flexibility that entropy cost would be expensive to form α -helix structure.

Table 5.1. Table of L amino acids propensities to form α -helix: Difference in free energy change each amino acid.⁸²

Amino acid	1-letter	Helical penalty	
		Kcal/mol	kJ/mol
Alanine	A	0.00	0.00
Arginine	R	0.21	0.88
Asparagine	N	0.65	2.72
Aspartate	D	0.69	2.89
Cysteine	C	0.68	2.85
Glutamate	E	0.40	1.67
Glutamine	Q	0.39	1.63
Glycine	G	1.00	4.18
Histidine	H	0.61	2.55
Isoleucine	I	0.41	1.72
Leucine	L	0.21	0.88
Lysine	K	0.26	1.09
Methionine	M	0.24	1.00
Phenylalanine	F	0.54	2.26
Proline	P	3.16	13.22
Serine	S	0.50	2.09
Threonine	T	0.66	2.76
Tryptophan	W	0.49	2.05
Tyrosine	Y	0.53	2.22
Valine	V	0.61	2.55

Estimated free energy change in kcal/mol within an α helical conformation. Relative to alanine set as zero. More positive numbers are not favored, it means more free energy changes. Depending on the detailed chemical environment, significant deviations from these standard numbers may happen.

5.2. Functional roles of α -helix

5.2.1. DNA binding

α -helices have particular importance in DNA binding motifs, including zinc-finger, leucine zipper, and helix-turn-helix motifs. Since the conventional structure diameter of an α -helix is about 12 Å (1.2nm), including the average length of AA side chains, about the same width of the major in B-form DNA. Furthermore, the coil-coil dimers of helices can position interaction surfaces to the symmetrical repeat double-helical DNA.⁸³ For example, the double-helical DNA of transcription factor Max use a helical coil-coil to dimerize, position both helices to interact with two successive turns of DNA groove, (PDB: 1HLO).

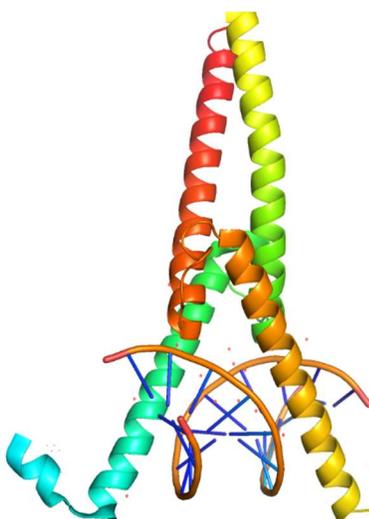


Figure 5.2. Transcriptional factor Max DNA binding with leucine zipper coil-coil(PDB:1HLO).

5.2.2. Membrane spanning

For transmembrane proteins, α -helices are the most common structures. Researchers presume that the helical structure can hide all hydrogen bonds internally, leaving no polar groups to membrane if the side chains of AA are nonpolar. Proteins are anchored to membrane by single

transmembrane helix, sometimes by coil-coil helices, or by a helix bundle, in most occasions containing seven helices arranged in a ring, such as rhodopsins, (PDB: 1GZM).⁸³

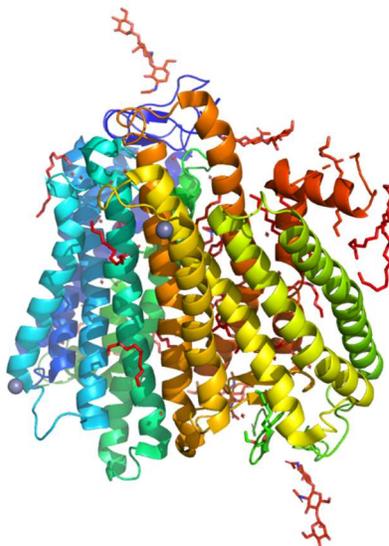


Figure 5.3. Structure of Bovine Rhodopsin in a Trigonal Crystal Form.

5.3. Unnatural helical peptide foldamers

Compared with unconventional peptide helical foldamers, conventional peptide helical foldamers have the disadvantages of easy loss of secondary helical structure, susceptibility to proteolytic degradation, and cell membrane penetration problem.^{81, 82, 84} Since Seebach and Gellman independently reported helical structure of β -peptides, several classes of foldamers have been developed.⁸⁴⁻⁸⁶ The advantages of unnatural peptide helical foldamers are that it can mimic the functions of natural peptide helices at same time overcome its limitations.

Recently, researchers developed several new types of helical peptide foldamers, such as β -peptides,⁸⁷ azapeptides,^{88, 89} oligoureas, peptoids,⁸⁹ α -aminoxy-peptides,⁹⁰ γ -peptides,⁹¹ and sulfono- γ -AApeptides.⁸⁴ Since the unnatural peptide helical foldamers are composed of unnatural AA, thus, they have unique advantages over natural α -peptide helices, such as good *in-vivo* stability and bioavailability, improved selectivity and diversity.

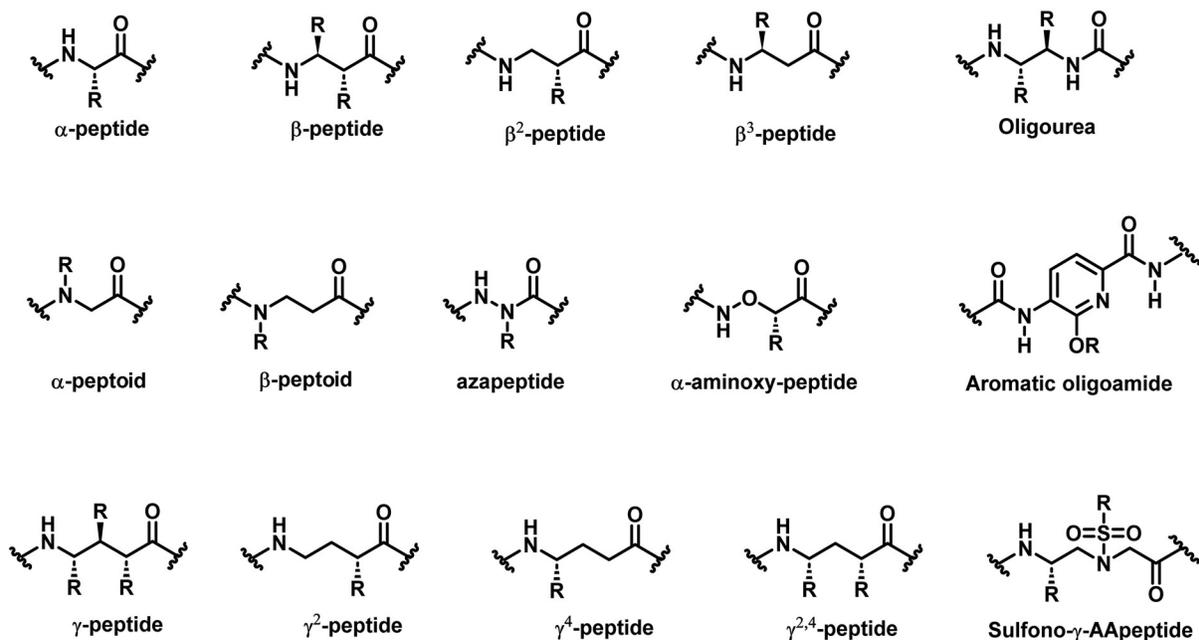


Figure 5.4. structure of α -peptide helical foldamer residues and unnatural helical foldamer residues.

5.4. Sulfonyl- γ -AApeptide

In 2015, our group developed a new type of helix using sulfonyl- γ -AA.⁸⁴ Sulfonyl- γ -AA are derivatives of natural AA, it is comparable to two natural AA in length, and have two side chains as dipeptide, one is from the side chain of canonical twenty natural amino acids, another is from the introduced natural or unnatural side chain of sulfonyl chloride. As a result, it provides great diversities of side chain since different sulfonyl chlorides are commercially available. Recently, we solved the crystal structure of homogeneous L-sulfonyl- γ -AApeptide foldamers (Figure 5.5), it turned to be an unexpected left-handed 14-helix hydrogen bond pattern; furthermore, it has a helical pitch of 5.1 Å (versus 5.4 Å of α -helix), and the side chain aligned perfectly from top to bottom along the helical backbone. It showed both the intramolecular hydrogen bond and the curvature of sulfonyl- γ -AA helped stabilize the helical backbone.

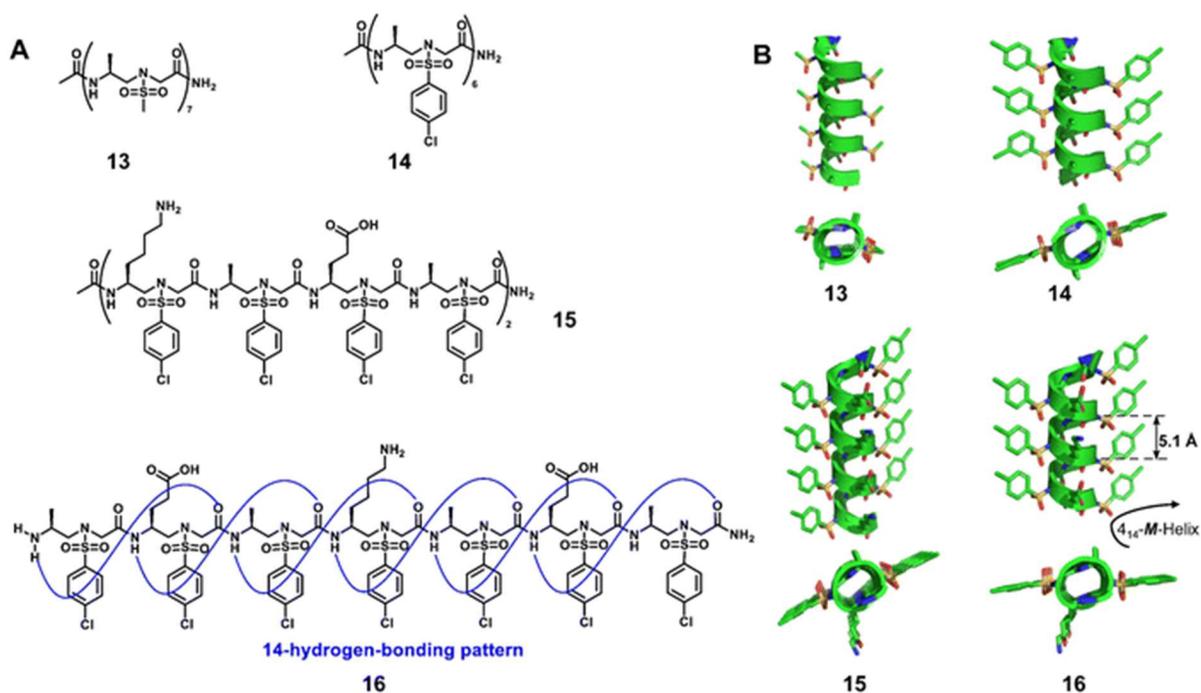


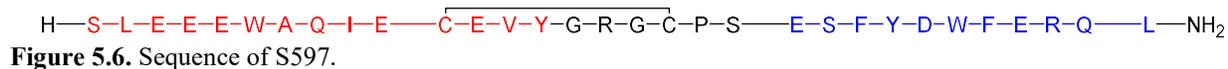
Figure: 5.5. Chemical structure of 13, 14, 15, 16. B: Crystal structure of 13, 14, 15, 16.

5.5. S597 insulin peptidomimetic

In 2007, Maja Jensen reported that S597 peptide leads to comparable level activation of protein kinase B (PKB) and glycogen synthesis to activation by insulin, even though the phosphorylation level is lower. However, Src homology 2/ α collagen-related (Shc) and extracellular signal-regulated kinase (ERK) 2 were absent from activation by S597. It resulted in cell proliferation is only slightly stimulated by S597. It is the first time to possible design insulin receptor binding mimetics with metabolic equipotency but low mitogenicity.⁹² In 2022, Junhee solved the crystal structure of S597 binding with insulin receptor.⁹³ Even though S597 activate insulin receptor (IR) and has comparable potency on glycemic control, however, the binding mechanisms are different. IR is fully activated by four insulins to two distinct sites forming a compact T-shape, in contrast, two S597 molecules bind and form an extended T-shaped IR by simultaneously binding to both the L1 domain of one protomer and the FnIII-1 domain of another.

Moreover, S597 could activate IR mutants that were resistant to insulin and elicit insulin-like signaling.

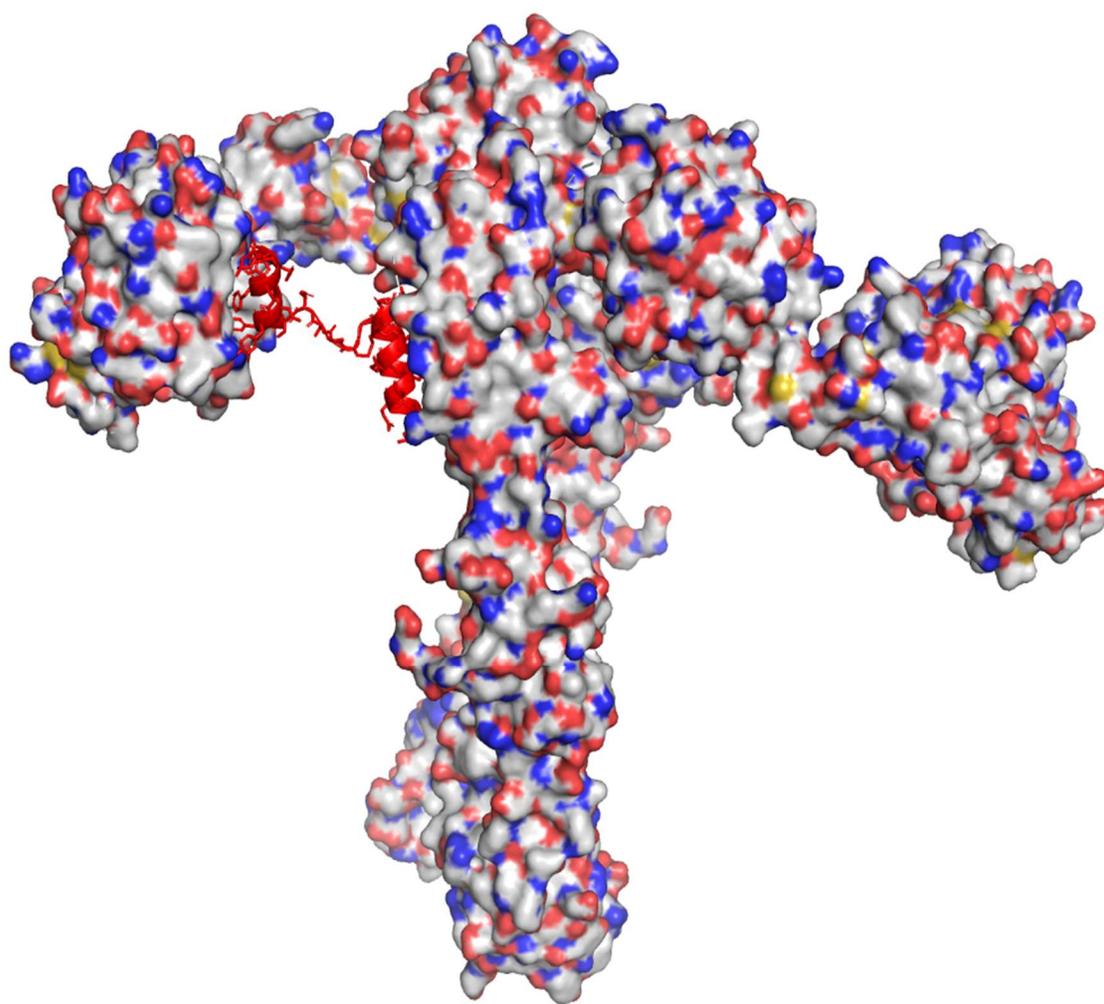
S597 Peptide sequence:



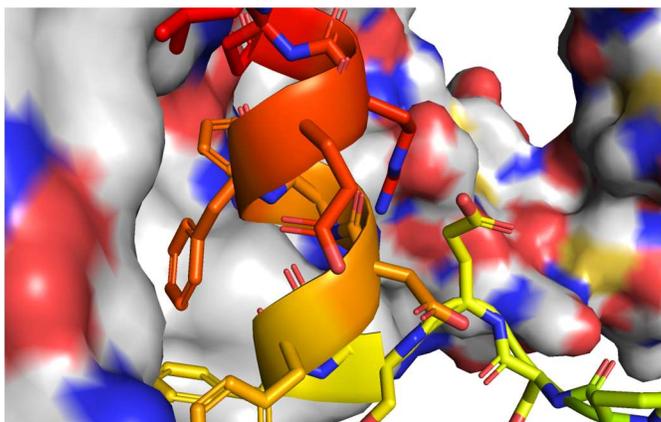
In the future, researchers are going to evaluate whether insulin in combination with S597 could lower the insulin requirements in diabetes patients.

5.5.1. Design of S597 Sulfonyl-γ-AApeptide

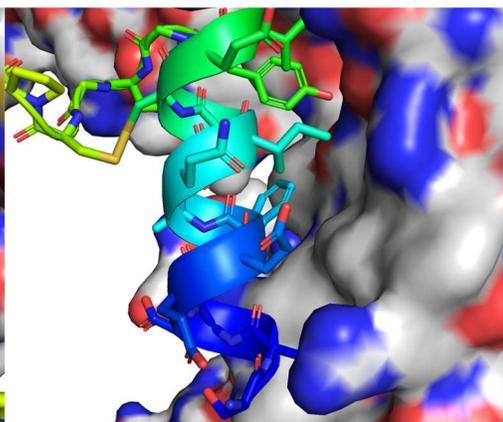
In our first design (Figure 5.8.), we tried to replace the two helices of S597 with our sulfonyl-γ-AA peptide. Our sulfonyl-γ-AA are unnatural AA while main the same length of dipeptides, it is resistant to proteolytic degradation at same time show better cell permeability. By introducing a part of unnatural helix or replacing natural AA with unnatural AA, we guessed that it would improve analogues' stability and cell permeability, thus, might improve its potency and pharmacokinetics.



A



B



C

Figure 5.7. Crystal structure of S597 bind with IR (A), (PDB:8DTL), site 1(B) binding and site 2(C) binding.

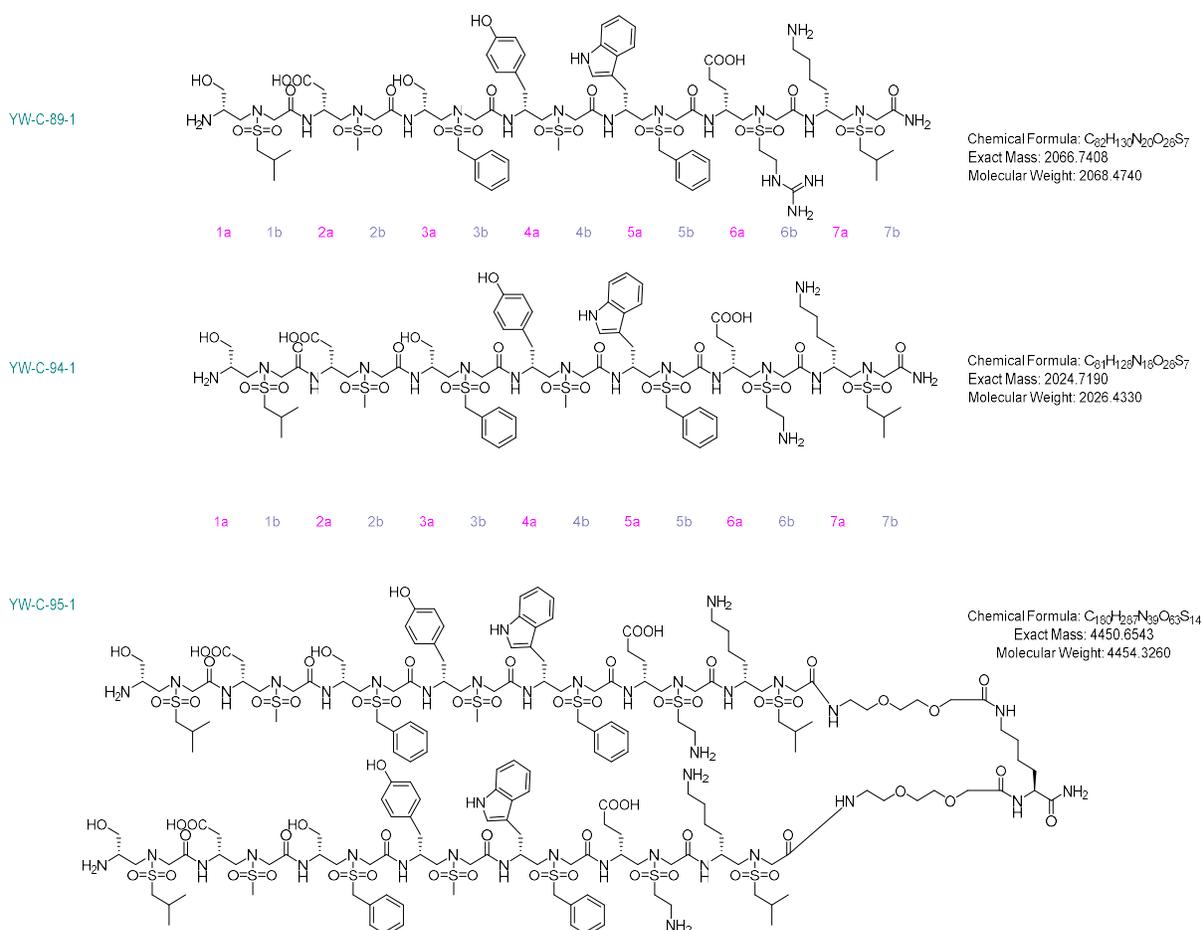


Figure 5.8. Structure of Site 1 sulfono- γ -AApeptide mimic, monomer and dimer.

First, we prepare several S597 sulfono- γ -AApeptides. YW-C-89-1 and YW-C-94-1 are the site 1 mimetic of S597 peptide, YW-C-95-1 is the dimer of site 1 mimetic. Our collaborator tested whether it could activate insulin receptor, and the result was negative. Then, we proposed another strategy. Now, we are trying to replace several AA of S597 that don't show apparent interaction with insulin receptor. Using this strategy (Figure 5.9.), we got YW-D-109-1 and YW-D-109-2.

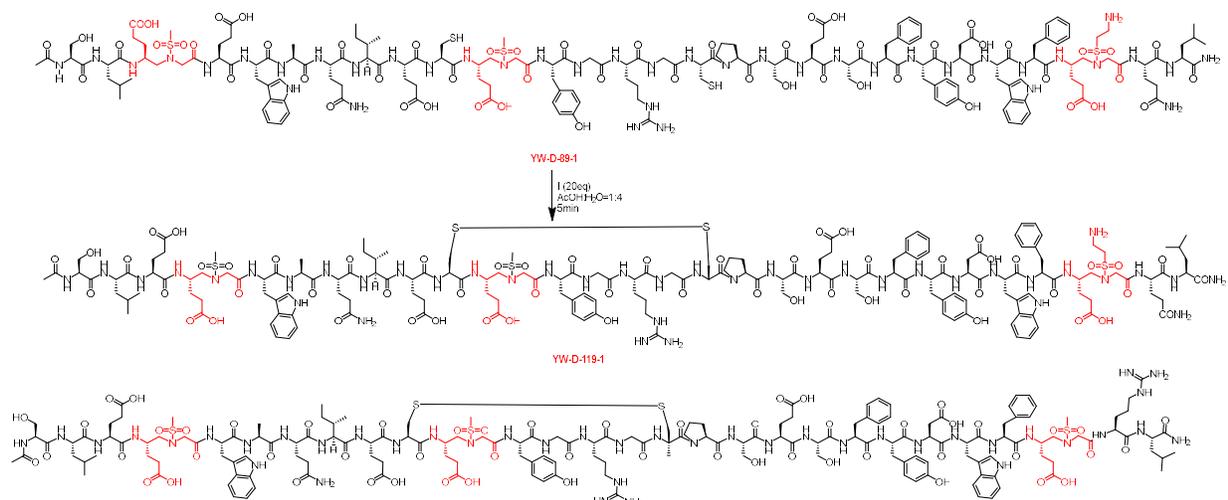


Figure 5.9. Structure of site modification with Sulfonyl- γ -AA.

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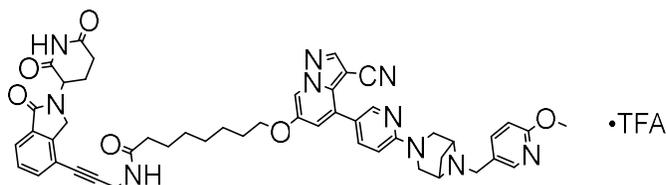
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APPENDICES

Appendix A: S1 ¹H-NMR, ¹³C-NMR, and HRMS

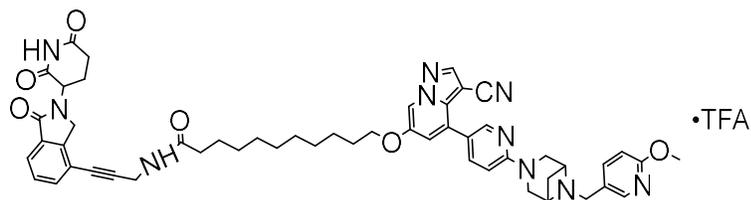
YW-L-1



¹H NMR (400 MHz, DMSO-*d*₆) δ 11.05 (s, 1H), 9.39 (q, *J* = 5.6 Hz, 1H), 8.73 – 8.69 (m, 1H), 8.68 (d, *J* = 2.1 Hz, 1H), 8.59 (d, *J* = 0.9 Hz, 1H), 8.42 (t, *J* = 2.7 Hz, 1H), 8.41 – 8.36 (m, 2H), 7.93 – 7.86 (m, 2H), 7.84 (dd, *J* = 8.6, 2.5 Hz, 1H), 7.76 – 7.62 (m, 2H), 7.53 (t, *J* = 7.6 Hz, 1H), 7.28 (dd, *J* = 3.2, 2.1 Hz, 1H), 7.26 (d, *J* = 2.1 Hz, 1H), 6.94 (d, *J* = 8.6 Hz, 2H), 6.82 (dd, *J* = 9.0, 1.6 Hz, 1H), 5.16 (ddd, *J* = 13.3, 5.1, 1.9 Hz, 3H), 4.63 (t, *J* = 6.3 Hz, 5H), 4.47 (d, *J* = 6.1 Hz, 4H), 4.16 (d, *J* = 5.4 Hz, 3H), 4.13 – 4.01 (m, 8H), 3.88 (d, *J* = 6.6 Hz, 8H), 3.52 (dt, *J* = 11.9, 6.5 Hz, 2H), 2.97 – 2.88 (m, 1H), 2.88 – 2.83 (m, 1H), 2.61 (d, *J* = 1.7 Hz, 1H), 2.41 – 2.36 (m, 2H), 2.21 (td, *J* = 7.3, 2.2 Hz, 2H), 2.13 (td, *J* = 7.4, 2.1 Hz, 3H), 2.10 – 1.98 (m, 4H), 1.75 (tq, *J* = 13.7, 6.7 Hz, 4H), 1.52 (ttt, *J* = 14.6, 7.6, 2.7 Hz, 4H), 1.42 (h, *J* = 7.1 Hz, 4H), 1.37 – 1.22 (m, 8H).

¹³C NMR (400 MHz, DMSO-*d*₆) δ 173.69, 172.51, 171.37, 168.22, 167.80, 158.17, 157.72, 150.06, 149.15, 147.67, 146.41, 141.87, 141.55, 138.59, 136.04, 129.50, 123.68, 123.03, 115.37, 111.24, 105.23, 93.08, 81.33, 69.58, 62.74, 53.89, 52.03, 49.99, 48.70, 47.30, 43.70, 35.52, 34.09, 31.64, 29.06, 28.87, 26.58, 25.77, 25.55, 24.89, 22.92.

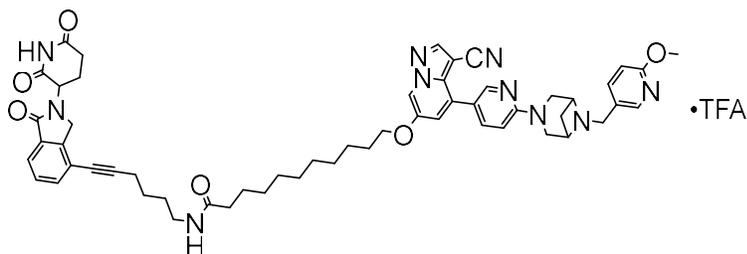
YW-L-2



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 10.40 (s, 1H), 9.31 (q, $J = 5.9, 5.4$ Hz, 1H), 8.61 (dd, $J = 6.3, 2.1$ Hz, 1H), 8.51 (s, 1H), 8.33 (dd, $J = 13.6, 2.4$ Hz, 1H), 8.28 (t, $J = 5.5$ Hz, 1H), 7.93 – 7.72 (m, 2H), 7.67 (d, $J = 7.5$ Hz, 1H), 7.58 (d, $J = 7.6$ Hz, 1H), 7.47 (t, $J = 7.6$ Hz, 1H), 7.19 (d, $J = 2.0$ Hz, 1H), 6.86 (d, $J = 8.3$ Hz, 1H), 6.74 (d, $J = 8.8$ Hz, 1H), 5.09 (dd, $J = 13.3, 5.1$ Hz, 1H), 4.57 (d, $J = 5.7$ Hz, 4H), 4.40 (t, $J = 3.7$ Hz, 3H), 4.36 (s, 1H), 4.26 (s, 1H), 4.22 (s, 1H), 4.14 (s, 1H), 4.08 (d, $J = 5.3$ Hz, 2H), 4.06 – 3.94 (m, 5H), 3.84 (s, 1H), 3.80 (s, 1H), 3.45 (dt, $J = 12.1, 6.5$ Hz, 1H), 2.85 (ddd, $J = 18.2, 13.6, 5.4$ Hz, 1H), 2.55 (s, 2H), 2.04 (t, $J = 7.4$ Hz, 2H), 2.00 – 1.90 (m, 2H), 1.66 (p, $J = 6.6$ Hz, 2H), 1.44 (q, $J = 7.0$ Hz, 2H), 1.33 (t, $J = 7.5$ Hz, 2H), 1.18 (s, 10H).

^{13}C NMR (400 MHz, $\text{DMSO-}d_6$) δ 173.29, 172.52, 171.44, 150.06, 149.76, 147.99, 147.04, 144.44, 141.84, 138.76, 134.60, 129.16, 123.68, 123.01, 113.13, 111.23, 105.22, 93.18, 81.34, 77.75, 69.62, 62.79, 53.89, 52.06, 50.03, 48.71, 47.34, 43.70, 35.57, 31.66, 29.37, 29.16, 29.06, 26.60, 25.88, 25.63, 22.93.

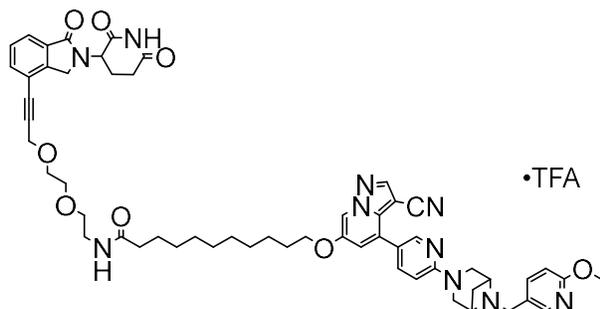
YW-L-3



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.37 (q, $J = 5.5$ Hz, 1H), 8.68 (d, $J = 6.2$ Hz, 1H), 8.59 (d, $J = 10.6$ Hz, 1H), 8.52 – 8.24 (m, 2H), 7.88 (dd, $J = 8.6, 2.4$ Hz, 1H), 7.81 (q, $J = 5.7, 4.3$ Hz, 1H), 7.73 (dd, $J = 21.0, 7.6$ Hz, 2H), 7.55 (t, $J = 7.6$ Hz, 1H), 7.31 – 7.22 (m, 1H), 6.93 (d, $J = 8.2$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 5.15 (dd, $J = 13.3, 5.0$ Hz, 1H), 4.63 (d, $J = 5.9$ Hz, 3H), 4.46 (d, $J = 4.7$ Hz, 5H), 4.37 (s, 2H), 4.32 (s, 1H), 3.91 (s, 1H), 3.88 (d, $J = 4.1$ Hz, 4H), 3.65 (t, $J = 4.5$ Hz, 3H), 3.54 (dt, $J = 19.3, 5.0$ Hz, 3H), 3.41 (t, $J = 5.9$ Hz, 2H), 3.18 (q, $J = 5.9$ Hz, 2H), 2.91 (ddd, $J = 18.1, 13.6, 5.5$ Hz, 2H), 2.04 (h, $J = 8.0, 6.7$ Hz, 4H), 1.74 (p, $J = 6.8, 6.1$ Hz, 2H), 1.43 (dp, $J = 15.3, 7.2$ Hz, 4H), 1.35 – 1.10 (m, 11H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 173.31, 172.69, 171.42, 167.95, 164.62, 157.75, 150.06, 149.75, 147.15, 144.43, 141.84, 141.35, 138.76, 136.21, 134.85, 132.57, 129.51, 129.18, 123.97, 123.00, 120.59, 120.37, 117.97, 114.93, 113.12, 111.23, 105.22, 91.79, 81.78, 81.34, 69.83, 69.63, 69.20, 62.79, 58.59, 53.89, 52.12, 50.04, 48.71, 47.40, 38.89, 35.78, 31.66, 29.70 – 28.60 (m), 25.90, 25.72, 22.83.

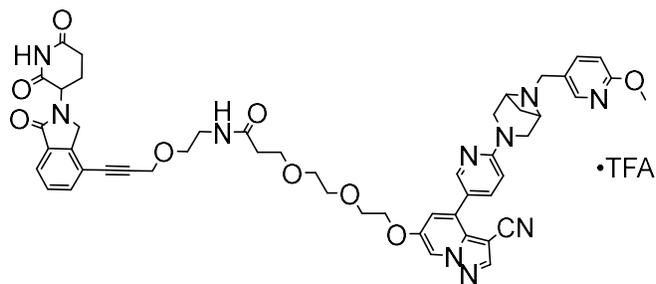
YW-L-4



¹H NMR (400 MHz, DMSO-*d*₆) δ 9.37 (q, *J* = 5.5, 5.0 Hz, 1H), 8.68 (t, *J* = 3.7 Hz, 1H), 8.59 (d, *J* = 10.5 Hz, 1H), 8.40 (dd, *J* = 13.1, 2.4 Hz, 1H), 7.88 (dd, *J* = 8.8, 2.4 Hz, 1H), 7.77 (t, *J* = 5.7 Hz, 1H), 7.70 (d, *J* = 7.5 Hz, 1H), 7.61 (d, *J* = 7.5 Hz, 1H), 7.50 (t, *J* = 7.6 Hz, 1H), 7.27 (t, *J* = 4.3 Hz, 1H), 6.92 (t, *J* = 7.2 Hz, 1H), 6.81 (d, *J* = 8.8 Hz, 1H), 5.14 (dd, *J* = 13.2, 5.1 Hz, 1H), 4.63 (d, *J* = 5.7 Hz, 2H), 4.47 (d, *J* = 5.7 Hz, 2H), 4.42 (s, 1H), 4.33 (s, 1H), 4.28 (s, 1H), 4.21 (s, 1H), 4.10 (d, *J* = 5.8 Hz, 4H), 4.07 (d, *J* = 5.6 Hz, 4H), 4.04 (s, 3H), 3.52 (dt, *J* = 12.1, 6.6 Hz, 1H), 3.07 (q, *J* = 6.2 Hz, 2H), 2.91 (ddd, *J* = 17.9, 13.4, 5.5 Hz, 2H), 2.61 (s, 2H), 2.03 (h, *J* = 6.1 Hz, 4H), 1.73 (p, *J* = 6.8, 6.4 Hz, 2H), 1.56 (q, *J* = 3.4 Hz, 4H), 1.43 (dp, *J* = 21.3, 7.4 Hz, 4H), 1.24 (d, *J* = 13.0 Hz, 11H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 173.30, 172.46, 171.44, 168.13, 164.62, 157.77, 149.76, 148.01, 147.04, 144.20, 141.84, 141.35, 138.74, 136.21, 134.51, 132.45, 129.28, 120.38, 119.28, 113.12, 111.23, 105.20, 96.61, 81.34, 76.98, 69.62, 62.80, 53.89, 52.11, 50.04, 48.71, 43.69, 38.27, 35.92, 31.69, 29.38, 28.87, 25.94, 25.90, 22.80, 18.92.

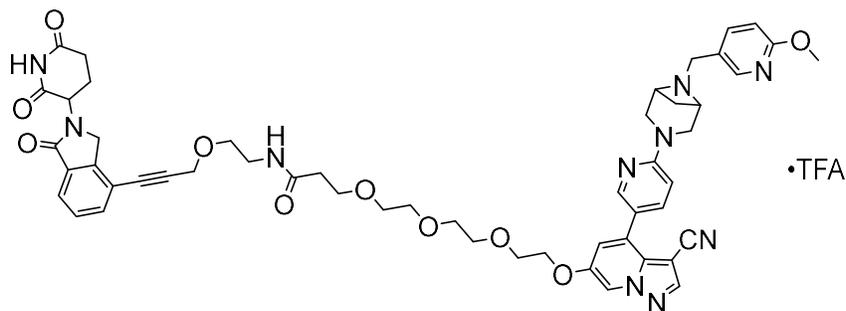
YW-L-5



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.38 (q, $J = 5.7, 4.7$ Hz, 1H), 8.71 (d, $J = 6.5$ Hz, 1H), 8.58 (s, 1H), 8.40 (dd, $J = 12.7, 2.4$ Hz, 1H), 7.97 (q, $J = 5.7$ Hz, 1H), 7.92 – 7.86 (m, 1H), 7.72 (dd, $J = 26.5, 7.6$ Hz, 2H), 7.54 (t, $J = 7.6$ Hz, 1H), 7.31 (d, $J = 7.5$ Hz, 1H), 6.93 (t, $J = 7.4$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 5.15 (dd, $J = 13.3, 5.1$ Hz, 1H), 4.64 (d, $J = 5.8$ Hz, 3H), 4.52 – 4.42 (m, 8H), 4.36 (s, 2H), 4.23 (q, $J = 4.9, 3.9$ Hz, 4H), 4.14 – 4.01 (m, 3H), 3.94 – 3.84 (m, 4H), 3.77 (q, $J = 4.3$ Hz, 2H), 3.57 (dt, $J = 16.3, 6.4$ Hz, 6H), 3.52 – 3.45 (m, 2H), 3.26 (q, $J = 5.7$ Hz, 2H), 2.94 – 2.84 (m, 1H), 2.31 (d, $J = 6.4$ Hz, 2H), 2.05 (ddd, $J = 21.8, 11.2, 4.9$ Hz, 2H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 173.33, 171.42, 170.70, 167.94, 164.62, 158.44, 149.90, 149.75, 147.98, 147.08, 144.44, 141.84, 138.75, 134.76, 132.56, 129.53, 129.16, 123.95, 122.96, 120.37, 117.92, 114.91, 113.28, 111.22, 105.22, 91.72, 81.89, 69.99, 69.12, 68.62, 62.78, 58.42, 53.88, 52.11, 50.03, 48.71, 47.39, 43.69, 38.86, 36.49, 31.67, 26.59, 22.79.

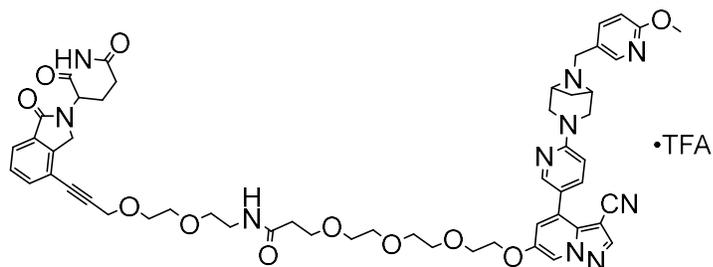
YW-L-6



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.43 (s, 1H), 8.73 (d, $J = 6.6$ Hz, 1H), 8.64 – 8.55 (m, 1H), 8.41 (d, $J = 12.8$ Hz, 1H), 7.97 (t, $J = 5.5$ Hz, 1H), 7.88 (t, $J = 12.0$ Hz, 2H), 7.74 (dd, $J = 24.9, 7.6$ Hz, 2H), 7.56 (t, $J = 7.8$ Hz, 1H), 7.32 (d, $J = 7.5$ Hz, 1H), 6.94 (d, $J = 8.3$ Hz, 1H), 6.83 (d, $J = 8.7$ Hz, 1H), 5.16 (dd, $J = 13.4, 5.0$ Hz, 1H), 4.65 (d, $J = 5.6$ Hz, 4H), 4.47 (d, $J = 8.9$ Hz, 5H), 4.25 (d, $J = 5.8$ Hz, 3H), 4.15 – 4.01 (m, 3H), 3.97 – 3.85 (m, 4H), 3.80 (t, $J = 4.3$ Hz, 2H), 3.65 – 3.39 (m, 15H), 3.27 (d, $J = 6.4$ Hz, 2H), 2.93 (td, $J = 16.1, 14.6, 5.2$ Hz, 1H), 2.61 (d, $J = 17.3$ Hz, 1H), 2.31 (t, $J = 6.4$ Hz, 2H), 2.14 – 1.94 (m, 2H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 173.33, 171.43, 170.71, 167.96, 164.63, 157.76, 149.77, 147.97, 149.06, 147.10, 144.46, 141.86, 141.37, 138.77, 136.29, 134.79, 132.58, 129.54, 123.98, 122.99, 120.57, 120.39, 117.95, 114.92, 113.35, 111.23, 105.71, 105.25, 91.75, 81.90, 81.40, 70.37, 70.26, 70.12, 69.97, 69.29, 69.16, 68.63, 67.25, 62.78, 58.43, 53.89, 52.13, 50.03, 48.71, 47.40, 43.71, 38.86, 36.50, 31.68, 26.61, 22.81.

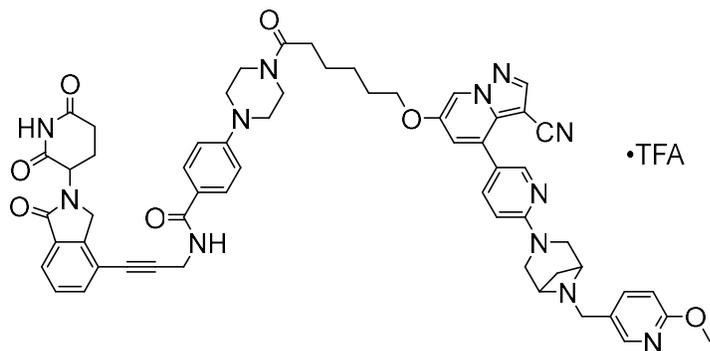
YW-L-7



¹H NMR (400 MHz, DMSO-*d*₆) δ 9.37 (q, *J* = 5.4 Hz, 1H), 8.72 (dd, *J* = 6.2, 2.1 Hz, 1H), 8.60 (d, *J* = 10.4 Hz, 1H), 8.53 – 8.24 (m, 2H), 8.00 – 7.80 (m, 3H), 7.74 (dd, *J* = 21.5, 7.6 Hz, 2H), 7.56 (t, *J* = 7.6 Hz, 1H), 7.31 (dd, *J* = 7.5, 2.1 Hz, 1H), 6.93 (dd, *J* = 10.2, 4.2 Hz, 1H), 6.82 (d, *J* = 8.8 Hz, 1H), 5.15 (dd, *J* = 13.3, 5.1 Hz, 1H), 4.64 (d, *J* = 5.6 Hz, 2H), 4.29 – 4.18 (m, 3H), 4.16 – 3.99 (m, 3H), 3.88 (d, *J* = 4.3 Hz, 3H), 3.79 (dd, *J* = 5.6, 3.3 Hz, 4H), 3.49 (p, *J* = 2.7, 2.3 Hz, 5H), 3.46 (q, *J* = 3.7, 3.1 Hz, 4H), 3.41 (t, *J* = 5.9 Hz, 3H), 3.20 (q, *J* = 5.8 Hz, 3H), 2.90 (dtd, *J* = 16.3, 12.2, 10.9, 5.7 Hz, 2H), 2.60 (dt, *J* = 15.1, 3.0 Hz, 2H), 2.42 (dd, *J* = 13.3, 4.5 Hz, 1H), 2.30 (t, *J* = 6.4 Hz, 2H), 2.16 – 1.91 (m, 2H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 173.32, 171.42, 170.61, 167.94, 164.62, 157.79, 149.91, 149.75, 148.03, 147.09, 144.43, 141.84, 141.33, 138.71, 134.86, 132.56, 129.19, 123.97, 122.97, 120.56, 120.37, 117.95, 114.91, 113.33, 111.22, 105.19, 91.79, 81.78, 81.39, 70.36, 70.26, 70.12, 69.97, 69.84, 69.58, 69.20, 67.24, 62.79, 58.58, 53.88, 52.12, 50.03, 48.71, 47.39, 38.97, 36.48, 31.66, 26.60, 22.82.

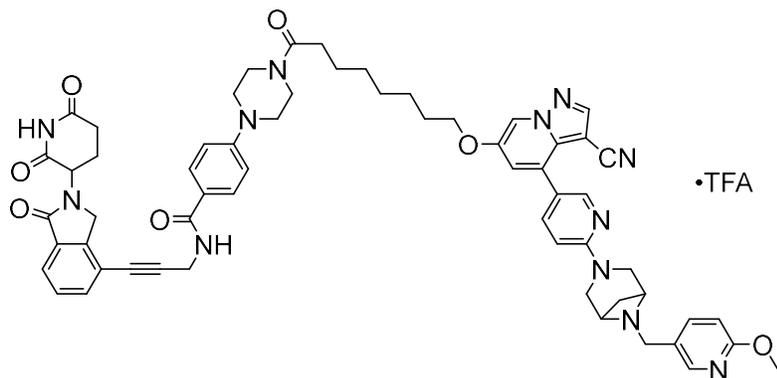
YW-L-8



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.37 (q, $J = 5.4$ Hz, 1H), 8.77 (t, $J = 5.5$ Hz, 1H), 8.69 (dd, $J = 6.4, 2.1$ Hz, 1H), 8.59 (d, $J = 10.6$ Hz, 1H), 8.40 (dd, $J = 14.4, 2.5$ Hz, 1H), 7.88 (dt, $J = 8.7, 2.1$ Hz, 1H), 7.81 – 7.71 (m, 3H), 7.67 (d, $J = 7.6$ Hz, 1H), 7.54 (t, $J = 7.6$ Hz, 1H), 7.28 (dd, $J = 7.3, 2.1$ Hz, 1H), 6.94 (ddd, $J = 17.1, 9.0, 4.2$ Hz, 3H), 6.81 (d, $J = 8.8$ Hz, 1H), 5.14 (dd, $J = 13.3, 5.1$ Hz, 1H), 4.63 (d, $J = 5.8$ Hz, 3H), 4.47 (d, $J = 6.6$ Hz, 2H), 4.36 – 4.31 (m, 2H), 4.11 (d, $J = 5.3$ Hz, 2H), 3.89 (d, $J = 15.2$ Hz, 5H), 3.59 (t, $J = 5.1$ Hz, 3H), 3.27 (dt, $J = 21.3, 5.1$ Hz, 3H), 2.34 (q, $J = 12.9, 10.1$ Hz, 4H), 2.11 – 1.92 (m, 4H), 1.76 (h, $J = 6.1$ Hz, 2H), 1.61 – 1.11 (m, 12H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 173.30, 171.45, 171.24, 168.03, 168.03, 166.14, 164.62, 157.77, 150.07, 149.76, 149.05, 148.01, 147.05, 144.57, 141.85, 141.36, 138.76, 134.47, 132.49, 131.34, 129.54, 129.17, 123.63, 120.33, 118.49, 114.94, 114.17, 113.13, 111.23, 105.21, 93.75, 77.71, 69.62, 62.79, 53.89, 52.03, 50.03, 48.71, 47.75, 47.32, 44.89, 43.69, 41.04, 32.66, 31.65, 29.73, 29.49, 29.17, 28.99, 28.84, 25.79, 25.14, 22.89.

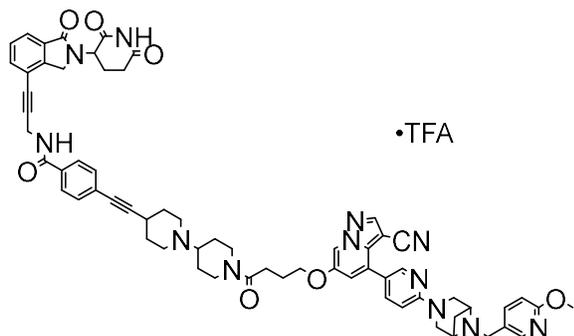
YW-L-9



^1H NMR (400 MHz, DMSO- d_6) δ 9.39 (q, J = 5.4 Hz, 1H), 8.77 (t, J = 5.4 Hz, 1H), 8.69 (dd, J = 6.6, 2.1 Hz, 1H), 8.59 (d, J = 10.7 Hz, 1H), 8.40 (dd, J = 14.6, 2.4 Hz, 2H), 7.89 (dt, J = 8.6, 2.8 Hz, 2H), 7.79 (d, J = 8.5 Hz, 2H), 7.74 (d, J = 7.5 Hz, 1H), 7.67 (d, J = 7.6 Hz, 1H), 7.54 (t, J = 7.6 Hz, 1H), 7.28 (dd, J = 7.5, 2.1 Hz, 1H), 7.03 – 6.87 (m, 4H), 6.82 (d, J = 8.8 Hz, 1H), 5.15 (dd, J = 13.3, 5.1 Hz, 1H), 4.64 (d, J = 5.6 Hz, 3H), 4.42 (s, 1H), 4.35 (s, 1H), 4.33 (d, J = 3.1 Hz, 1H), 4.25 (d, J = 26.4 Hz, 1H), 4.11 (d, J = 5.2 Hz, 3H), 4.08 (d, J = 4.3 Hz, 1H), 4.04 (s, 1H), 3.96 – 3.82 (m, 5H), 3.59 (t, J = 4.9 Hz, 4H), 3.51 (q, J = 6.6, 6.2 Hz, 1H), 3.27 (dt, J = 21.9, 5.2 Hz, 4H), 2.93 – 2.82 (m, 1H), 2.63 – 2.54 (m, 1H), 2.42 – 2.26 (m, 3H), 2.11 – 2.04 (m, 1H), 2.04 – 1.94 (m, 1H), 1.76 (dq, J = 10.9, 6.5, 5.5 Hz, 2H), 1.62 – 1.19 (m, 10H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 173.30, 171.44, 171.24, 168.03, 166.14, 164.62, 158.88, 158.52, 157.71, 153.07, 150.07, 149.75, 149.04, 147.91, 147.04, 144.57, 141.84, 141.34, 138.81, 136.21, 134.46, 132.48, 129.16, 123.62, 123.02, 120.59, 120.37, 118.49, 114.17, 113.14, 111.22, 105.26, 93.75, 81.35, 77.71, 69.62, 62.78, 53.88, 52.03, 50.03, 48.72, 47.76, 47.32, 44.90, 43.72, 32.66, 31.65, 29.73, 29.17, 28.99, 26.60, 25.79, 25.14, 22.89.

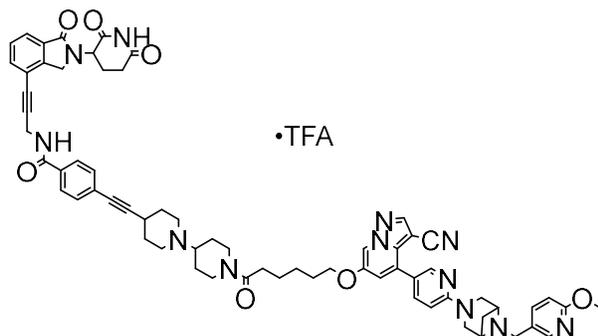
YW-L-11



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.62 (s, 1H), 9.42 (s, 1H), 9.15 (q, $J = 5.4$ Hz, 1H), 8.79 – 8.67 (m, 1H), 8.60 (dd, $J = 10.6, 3.4$ Hz, 1H), 8.40 (dd, $J = 15.0, 2.4$ Hz, 1H), 7.89 (dd, $J = 8.4, 6.2$ Hz, 3H), 7.75 (d, $J = 7.5$ Hz, 1H), 7.68 (dd, $J = 7.5, 2.3$ Hz, 1H), 7.54 (dt, $J = 22.9, 8.3$ Hz, 3H), 7.28 (d, $J = 2.0$ Hz, 1H), 6.93 (dd, $J = 9.2, 3.0$ Hz, 1H), 6.82 (dd, $J = 8.9, 2.7$ Hz, 1H), 5.16 (dd, $J = 13.3, 5.1$ Hz, 2H), 4.63 (t, $J = 5.3$ Hz, 2H), 4.57 (d, $J = 12.9$ Hz, 1H), 4.47 (d, $J = 5.9$ Hz, 2H), 4.38 (td, $J = 19.5, 18.8, 5.9$ Hz, 4H), 4.21 (s, 1H), 4.18 – 3.99 (m, 6H), 3.96 – 3.81 (m, 4H), 3.63 – 3.36 (m, 4H), 3.35 – 3.14 (m, 1H), 3.04 (t, $J = 11.0$ Hz, 2H), 2.91 (tq, $J = 14.4, 8.7, 7.0$ Hz, 2H), 2.64 – 2.53 (m, 4H), 2.43 – 2.25 (m, 1H), 2.20 (d, $J = 13.6$ Hz, 1H), 2.16 – 2.10 (m, 1H), 2.10 – 1.93 (m, 7H), 1.86 (q, $J = 12.5$ Hz, 1H), 1.73 – 1.36 (m, 2H), 1.24 (d, $J = 3.3$ Hz, 1H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 173.30, 171.46, 170.39, 168.00, 165.73, 158.83, 158.349, 157.79, 149.75, 149.02, 147.99, 147.09, 144.56, 141.84, 141.32, 138.74, 136.26, 134.59, 133.67, 132.50, 132.01, 131.84, 129.59, 129.20, 128.10, 125.98, 123.74, 122.99, 120.71, 120.56, 120.38, 118.35, 114.91, 113.24, 111.22, 105.65, 105.21, 94.25, 93.01, 83.19, 81.41, 77.96, 69.10, 62.78, 53.89, 52.04, 50.03, 48.70, 47.34, 46.15, 43.72, 31.64, 29.95, 29.60, 28.79, 27.81, 26.82, 26.61, 26.00, 24.55, 24.11, 22.91.

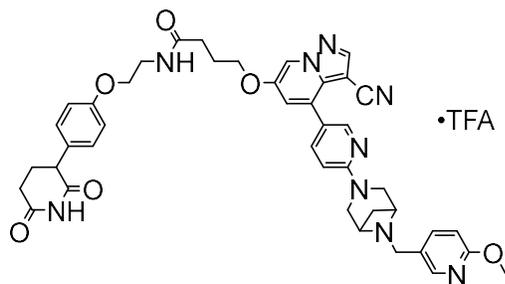
YW-L-12



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.91 – 9.58 (m, 1H), 9.44 (q, $J = 5.9, 4.6$ Hz, 1H), 9.16 (q, $J = 5.8$ Hz, 1H), 8.70 (dt, $J = 5.1, 2.5$ Hz, 1H), 8.59 (d, $J = 2.0$ Hz, 1H), 8.43 (d, $J = 2.3$ Hz, 1H), 8.39 (d, $J = 2.4$ Hz, 1H), 8.01 – 7.80 (m, 4H), 7.75 (d, $J = 7.5$ Hz, 1H), 7.68 (dd, $J = 7.5, 2.7$ Hz, 1H), 7.61 – 7.54 (m, 2H), 7.54 – 7.47 (m, 1H), 7.28 (dt, $J = 8.3, 2.2$ Hz, 1H), 6.92 (t, $J = 7.9$ Hz, 2H), 6.82 (d, $J = 9.0$ Hz, 1H), 5.16 (dd, $J = 13.3, 5.1$ Hz, 1H), 4.64 (t, $J = 5.4$ Hz, 2H), 4.55 (d, $J = 12.8$ Hz, 1H), 4.48 (d, $J = 6.0$ Hz, 2H), 4.38 (td, $J = 19.5, 18.9, 6.5$ Hz, 3H), 4.22 (s, 1H), 4.17 – 3.98 (m, 6H), 3.96 – 3.84 (m, 4H), 3.50 (dd, $J = 23.9, 8.7$ Hz, 4H), 3.34 – 3.14 (m, 1H), 3.03 (dt, $J = 17.2, 8.0$ Hz, 2H), 2.91 (dp, $J = 20.4, 6.8, 6.1$ Hz, 2H), 2.61 (d, $J = 3.6$ Hz, 1H), 2.44 – 2.28 (m, 3H), 2.25 – 2.16 (m, 1H), 2.16 – 1.94 (m, 6H), 1.88 (t, $J = 13.4$ Hz, 1H), 1.80 (q, $J = 7.1, 6.5$ Hz, 2H), 1.54 (ddt, $J = 40.0, 15.6, 9.6$ Hz, 6H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 173.30, 171.46, 170.94, 168.01, 165.74, 164.37, 158.93, 157.75, 157.49, 150.05, 149.75, 149.02, 147.94, 147.06, 144.56, 141.84, 141.32, 138.78, 136.22, 134.60, 133.66, 132.50, 132.01, 131.84, 129.54, 129.19, 128.10, 125.99, 123.73, 122.96, 120.57, 120.38, 118.36, 117.94, 115.03, 114.92, 113.17, 111.22, 105.24, 94.27, 93.01, 92.83, 81.38, 81.06, 77.95, 69.61, 62.87, 62.77, 53.88, 52.05, 50.03, 48.70, 48.49, 48.31, 47.34, 46.13, 44.61, 43.82, 43.72, 32.56, 31.64, 29.95, 29.58, 28.79, 27.80, 26.94, 26.60, 26.14, 26.01, 25.72, 24.93, 24.11, 22.91.

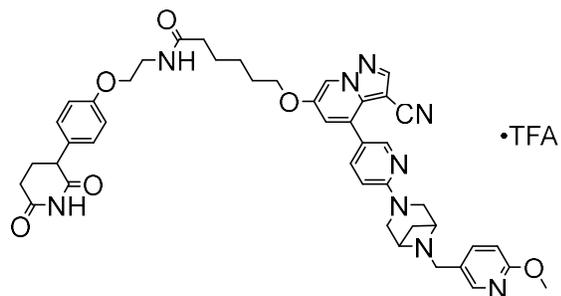
YW-N-1



^1H NMR (400 MHz, DMSO- d_6) δ 10.79 (s, 1H), 9.39 (s, 1H), 8.67 (d, $J = 7.3$ Hz, 1H), 8.59 (d, $J = 10.4$ Hz, 1H), 8.41 (d, $J = 12.4$ Hz, 1H), 8.07 (q, $J = 8.7, 7.4$ Hz, 1H), 7.89 (d, $J = 8.7$ Hz, 1H), 7.28 (d, $J = 7.7$ Hz, 1H), 7.11 (d, $J = 8.5$ Hz, 2H), 6.97 – 6.86 (m, 3H), 6.83 (d, $J = 9.0$ Hz, 1H), 4.64 (d, $J = 6.0$ Hz, 2H), 4.48 (d, $J = 6.3$ Hz, 1H), 4.08 (dd, $J = 15.9, 10.4$ Hz, 4H), 3.97 (s, 2H), 3.89 (s, 4H), 3.41 (s, 4H), 2.70 (s, 1H), 2.23 – 1.93 (m, 5H), 1.76 (t, $J = 7.2$ Hz, 2H), 1.59 (t, $J = 7.7$ Hz, 2H), 1.43 (q, $J = 7.7$ Hz, 2H), 1.24 (dd, $J = 17.0, 8.1$ Hz, 1H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 174.93, 173.96, 172.95, 164.63, 158.56, 157.80, 150.05, 149.74, 147.86, 147.04, 141.85, 138.82, 136.21, 131.66, 130.03, 123.01, 120.58, 120.34, 114.78, 113.14, 111.23, 105.31, 81.34, 69.57, 66.78, 62.79, 53.90, 50.04, 48.71, 46.95, 43.72, 38.63, 35.63, 31.81, 28.65, 26.45, 25.52, 25.38.

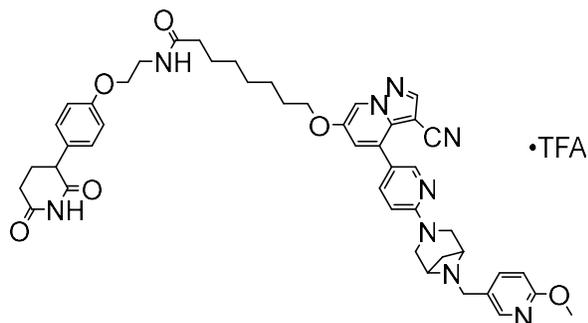
YW-N-2



^1H NMR (600 MHz, $\text{DMSO-}d_6$) δ 10.32 (s, 1H), 9.30 (d, $J = 5.1$ Hz, 1H), 8.63 (d, $J = 2.1$ Hz, 1H), 8.53 (s, 1H), 8.33 (dd, $J = 18.0, 2.5$ Hz, 2H), 8.16 – 8.05 (m, 1H), 7.81 (dd, $J = 8.6, 2.5$ Hz, 1H), 7.20 (d, $J = 2.1$ Hz, 1H), 7.03 (d, $J = 8.5$ Hz, 2H), 6.86 (dd, $J = 8.7, 2.9$ Hz, 1H), 6.83 – 6.78 (m, 2H), 6.75 (d, $J = 8.8$ Hz, 1H), 4.56 (t, $J = 6.4$ Hz, 3H), 4.40 (d, $J = 6.5$ Hz, 2H), 4.15 (d, $J = 3.7$ Hz, 1H), 4.08 – 4.00 (m, 3H), 3.98 (d, $J = 12.7$ Hz, 2H), 3.90 (td, $J = 5.7, 2.8$ Hz, 2H), 3.70 (dd, $J = 11.5, 4.9$ Hz, 1H), 3.45 (dt, $J = 12.3, 6.6$ Hz, 1H), 3.40 – 3.30 (m, 2H), 2.61 – 2.56 (m, 1H), 2.33 (s, 1H), 2.24 (td, $J = 7.4, 4.6$ Hz, 2H), 2.07 (dtd, $J = 13.0, 11.7, 4.4$ Hz, 1H), 2.02 – 1.96 (m, 1H), 1.96 – 1.87 (m, 3H).

^{13}C NMR (151 MHz, $\text{DMSO-}d_6$) δ 178.31, 174.95, 173.98, 172.04, 164.66, 158.76, 158.51, 157.57, 149.96, 147.05, 141.90, 138.92, 136.01, 132.06, 130.06, 129.51, 123.03, 117.14, 114.74, 111.24, 105.23, 81.64, 69.09, 66.77, 62.75, 53.90, 50.06, 49.99, 48.71, 46.94, 43.74, 38.70, 31.83, 26.43, 24.92.

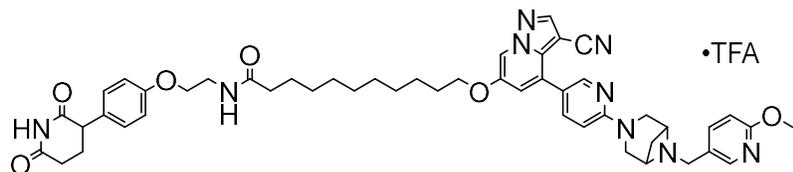
YW-N-3



^1H NMR (600 MHz, DMSO- d_6) δ 9.32 (q, $J = 5.4$ Hz, 1H), 8.63 (dd, $J = 10.2, 2.1$ Hz, 1H), 8.53 (d, $J = 16.4$ Hz, 1H), 8.33 (dd, $J = 20.3, 2.5$ Hz, 1H), 7.99 (dd, $J = 6.5, 4.8$ Hz, 1H), 7.82 (ddd, $J = 8.7, 4.9, 2.5$ Hz, 1H), 7.21 (dd, $J = 11.0, 2.1$ Hz, 1H), 7.09 – 7.01 (m, 2H), 6.87 (d, $J = 8.5$ Hz, 1H), 6.83 – 6.79 (m, 2H), 6.75 (d, $J = 8.8$ Hz, 1H), 4.56 (t, $J = 6.6$ Hz, 3H), 4.40 (d, $J = 6.5$ Hz, 2H), 4.15 (s, 1H), 4.08 – 3.94 (m, 5H), 3.88 (t, $J = 5.7$ Hz, 2H), 3.81 (d, $J = 6.4$ Hz, 4H), 3.70 (dd, $J = 11.5, 4.9$ Hz, 1H), 3.45 (dt, $J = 12.1, 6.7$ Hz, 1H), 3.33 (q, $J = 5.7$ Hz, 2H), 2.61 – 2.53 (m, 2H), 2.42 – 2.29 (m, 1H), 2.12 – 1.96 (m, 4H), 1.91 (dq, $J = 13.6, 4.8$ Hz, 1H), 1.72 – 1.62 (m, 2H), 1.48 – 1.40 (m, 2H), 1.34 (qd, $J = 9.2, 7.9, 4.6$ Hz, 2H), 1.26 (dt, $J = 13.9, 9.7, 5.5$ Hz, 2H), 1.22 – 1.14 (m, 2H).

^{13}C NMR (151 MHz, DMSO- d_6) δ 174.94, 173.93, 172.93, 164.60, 158.77, 158.53, 157.80, 148.87, 147.76, 141.87, 138.80, 136.19, 131.68, 130.05, 129.50, 123.02, 117.27, 115.33, 114.74, 113.16, 111.24, 105.23, 81.33, 69.60, 67.26, 62.74, 53.89, 49.67, 48.71, 46.95, 38.59, 35.68, 31.83, 28.98, 28.88, 28.82, 26.45, 25.77, 25.63.

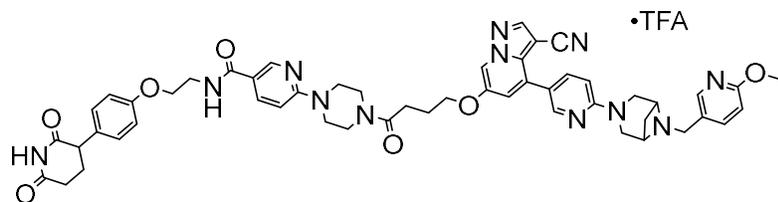
YW-N-4



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.37 (s, 1H), 8.69 (d, $J = 7.3$ Hz, 1H), 8.60 (d, $J = 10.2$ Hz, 1H), 8.41 (d, $J = 13.8$ Hz, 1H), 8.02 (d, $J = 5.8$ Hz, 1H), 7.89 (d, $J = 8.7$ Hz, 1H), 7.28 (d, $J = 8.2$ Hz, 1H), 7.13 (d, $J = 8.1$ Hz, 1H), 6.97 – 6.86 (m, 2H), 6.83 (d, $J = 8.9$ Hz, 1H), 4.64 (d, $J = 6.0$ Hz, 1H), 4.47 (s, 1H), 4.12 (s, 3H), 3.89 (s, 2H), 3.77 (s, 2H), 2.95 (s, 1H), 2.80 (s, 1H), 2.69 (s, 1H), 2.35 (s, 1H), 2.08 (t, $J = 7.5$ Hz, 3H), 1.76 (t, $J = 6.8$ Hz, 1H), 1.46 (dt, $J = 22.3, 7.5$ Hz, 3H), 1.38 – 1.08 (m, 8H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 174.94, 173.96, 173.05, 164.79, 157.81, 150.08, 149.39, 147.99, 147.04, 141.86, 138.73, 131.67, 130.04, 114.79, 111.23, 81.33, 69.65, 66.80, 62.81, 53.90, 46.95, 38.62, 35.75, 31.80, 29.40, 29.23, 29.14, 29.07, 28.89, 26.47, 25.89, 25.69.

YW-N-5

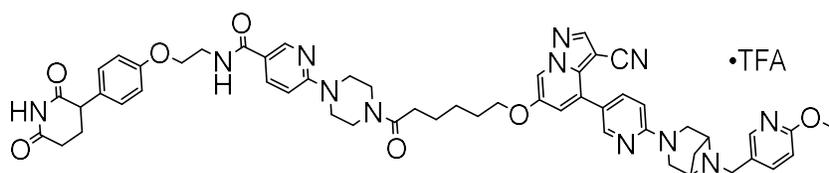


^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.44 (q, $J = 5.2$ Hz, 1H), 8.75 (dd, $J = 7.8, 2.0$ Hz, 1H), 8.68 – 8.54 (m, 3H), 8.50 (d, $J = 2.4$ Hz, 1H), 8.41 (dd, $J = 14.6, 2.4$ Hz, 2H), 8.29 (d, $J = 2.4$ Hz, 1H), 8.04 (dd, $J = 9.0, 2.4$ Hz, 1H), 7.98 (dd, $J = 8.7, 2.5$ Hz, 1H), 7.90 (ddd, $J = 8.0, 4.8, 2.5$ Hz, 2H), 7.85 (dd, $J = 8.6, 2.5$ Hz, 1H), 7.32 (dd, $J = 8.0, 2.0$ Hz, 1H), 7.13 (d, $J = 8.3$ Hz, 2H), 6.93 (dd, $J = 8.6, 4.4$ Hz, 4H), 6.83 (d, $J = 8.9$ Hz, 1H), 4.64 (t, $J = 5.4$ Hz, 2H), 4.48 (d, $J = 6.4$ Hz, 1H), 4.23 (s, 1H), 4.11 (ddd, $J = 27.3, 14.7, 8.9$ Hz, 7H), 3.92 (s, 1H), 3.88 (d, $J = 4.6$ Hz, 4H), 3.79 (dd, $J =$

11.5, 4.9 Hz, 1H), 3.73 – 3.65 (m, 2H), 3.55 – 3.46 (m, 1H), 2.71 – 2.60 (m, 1H), 2.57 (t, $J = 6.9$ Hz, 2H), 2.45 (t, $J = 4.2$ Hz, 1H), 2.16 (qd, $J = 12.1, 4.3$ Hz, 1H), 2.02 (tp, $J = 13.6, 5.1, 4.1$ Hz, 4H).

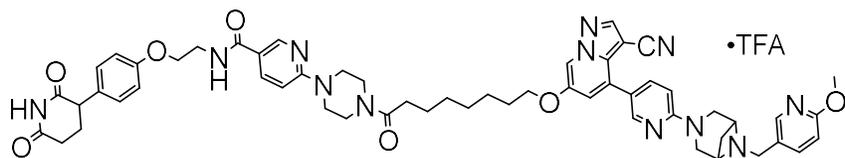
^{13}C NMR (101 MHz, DMSO- d_6) δ 175.00, 173.99, 170.74, 165.35, 159.24, 157.78, 149.91, 149.76, 147.80, 147.36, 147.10, 141.89, 138.89, 137.48, 131.73, 130.11, 129.46, 123.06, 120.55, 120.40, 119.02, 114.75, 113.22, 111.24, 106.77, 105.33, 81.34, 69.06, 66.52, 62.68, 53.89, 49.97, 48.72, 46.96, 44.64, 41.02, 31.86, 28.90, 26.47, 24.46.

YW-N-6



^1H NMR (400 MHz, DMSO- d_6) δ 10.83 (s, 1H), 9.40 (q, $J = 5.2$ Hz, 1H), 8.77 – 8.67 (m, 2H), 8.66 – 8.53 (m, 4H), 8.49 (d, $J = 2.3$ Hz, 1H), 8.46 – 8.36 (m, 3H), 8.29 (d, $J = 2.4$ Hz, 1H), 8.02 (dd, $J = 9.0, 2.5$ Hz, 1H), 7.97 (dt, $J = 8.8, 3.1$ Hz, 1H), 7.89 (dt, $J = 8.7, 3.0$ Hz, 3H), 7.84 (dd, $J = 8.6, 2.5$ Hz, 1H), 7.28 (dt, $J = 7.1, 2.5$ Hz, 2H), 7.13 (d, $J = 8.4$ Hz, 3H), 6.93 (dd, $J = 8.5, 5.0$ Hz, 5H), 6.82 (dd, $J = 8.9, 2.8$ Hz, 2H), 4.63 (t, $J = 5.4$ Hz, 4H), 4.47 (d, $J = 6.4$ Hz, 3H), 4.22 (s, 1H), 4.18 – 4.00 (m, 11H), 3.87 (s, 8H), 3.78 (dd, $J = 11.5, 4.9$ Hz, 1H), 3.72 – 3.46 (m, 11H), 2.69 (s, 1H), 2.63 (dd, $J = 11.7, 5.3$ Hz, 1H), 2.25 (td, $J = 7.2, 2.2$ Hz, 2H), 2.16 (dd, $J = 12.7, 4.2$ Hz, 1H), 2.12 – 2.03 (m, 2H), 2.03 – 1.93 (m, 2H), 1.78 (q, $J = 8.4$ Hz, 4H), 1.58 (p, $J = 7.7$ Hz, 4H), 1.53 – 1.38 (m, 4H), 1.36 – 1.15 (m, 4H).

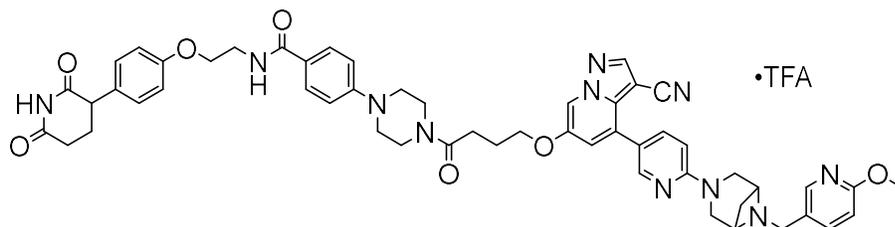
YW-N-7



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.35 (q, $J = 5.4$ Hz, 1H), 8.62 (t, $J = 4.0$ Hz, 1H), 8.57 – 8.45 (m, 2H), 8.34 (dd, $J = 14.5, 2.4$ Hz, 1H), 7.97 (dd, $J = 9.0, 2.4$ Hz, 1H), 7.82 (ddd, $J = 7.8, 4.5, 2.3$ Hz, 1H), 7.77 (dd, $J = 8.6, 2.4$ Hz, 1H), 7.22 (d, $J = 7.9$ Hz, 2H), 7.06 (d, $J = 8.2$ Hz, 2H), 6.86 (d, $J = 8.2$ Hz, 3H), 6.76 (d, $J = 8.8$ Hz, 1H), 4.57 (d, $J = 5.5$ Hz, 2H), 4.41 (d, $J = 6.3$ Hz, 1H), 4.15 (s, 1H), 4.10 – 3.91 (m, 5H), 3.85 (s, 1H), 3.82 (s, 2H), 3.80 (s, 1H), 3.71 (dd, $J = 11.3, 4.9$ Hz, 1H), 3.65 – 3.35 (m, 8H), 2.65 – 2.51 (m, 1H), 2.38 (s, 1H), 2.29 (t, $J = 7.5$ Hz, 2H), 2.15 – 1.87 (m, 3H), 1.70 (p, $J = 6.2$ Hz, 2H), 1.46 (p, $J = 7.3$ Hz, 2H), 1.38 (d, $J = 9.2$ Hz, 2H), 1.26 (dd, $J = 10.3, 4.9$ Hz, 2H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 174.92, 173.91, 171.39, 165.33, 164.62, 157.81, 150.06, 149.75, 147.76, 147.15, 141.84, 138.90, 137.55, 131.71, 130.07, 123.03, 119.08, 114.81, 111.22, 111.22, 106.87, 81.34, 69.62, 66.57, 62.76, 53.87, 50.03, 48.73, 46.96, 44.95, 44.71, 43.75, 32.69, 31.81, 29.16, 28.99, 28.84, 26.47, 25.79, 25.10.

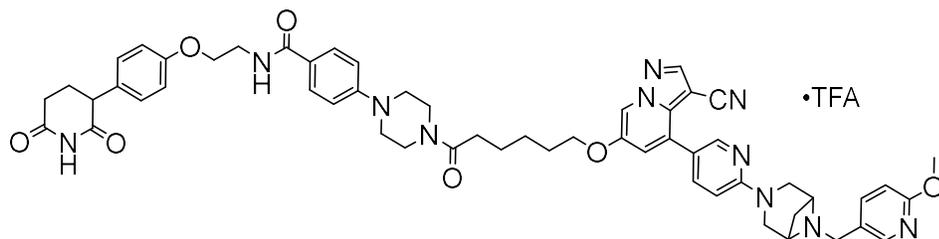
YW-N-8



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.43 (q, $J = 5.1$ Hz, 1H), 8.73 (dd, $J = 7.5, 2.1$ Hz, 1H), 8.59 (d, $J = 10.8$ Hz, 1H), 8.47 – 8.34 (m, 3H), 8.29 (d, $J = 2.5$ Hz, 1H), 7.89 (dd, $J = 8.7, 2.5$ Hz, 1H), 7.77 (d, $J = 8.6$ Hz, 2H), 7.31 (dd, $J = 8.4, 2.1$ Hz, 1H), 7.13 (d, $J = 8.4$ Hz, 2H), 7.04 – 6.87 (m, 5H), 6.83 (d, $J = 8.9$ Hz, 1H), 4.64 (t, $J = 5.1$ Hz, 2H), 4.48 (d, $J = 6.4$ Hz, 1H), 4.23 (s, 1H), 4.11 (ddd, $J = 26.8, 13.2, 7.1$ Hz, 6H), 3.90 (d, $J = 19.9$ Hz, 4H), 3.79 (dd, $J = 11.3, 4.9$ Hz, 1H), 3.61 (q, $J = 6.0$ Hz, 6H), 3.52 (q, $J = 6.4, 6.0$ Hz, 1H), 3.28 (dt, $J = 20.3, 5.0$ Hz, 4H), 2.71 – 2.54 (m, 3H), 2.24 – 1.91 (m, 5H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 174.92, 173.91, 170.60, 166.61, 164.62, 158.97, 158.61, 157.83, 152.90, 149.92, 149.75, 149.03, 147.79, 147.06, 141.84, 138.86, 136.25, 131.67, 130.06, 129.47, 129.02, 124.27, 123.05, 120.58, 120.37, 119.82, 114.79, 114.24, 113.30, 111.22, 105.86, 105.34, 81.37, 69.13, 66.59, 62.76, 53.87, 50.03, 48.72, 47.79, 47.50, 46.96, 44.83, 43.75, 31.81, 28.84, 26.47, 24.52.

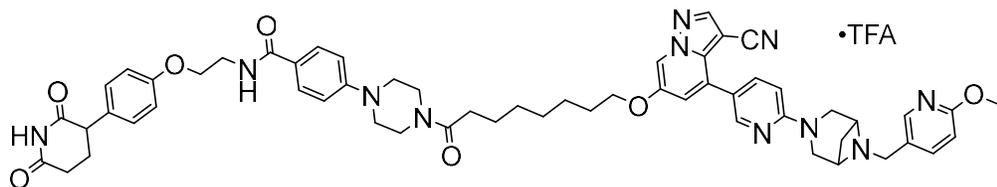
YW-N-9



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.42 (q, $J = 5.7, 4.7$ Hz, 1H), 8.70 (dd, $J = 6.1, 2.1$ Hz, 1H), 8.59 (d, $J = 10.5$ Hz, 1H), 8.51 – 8.24 (m, 3H), 8.00 – 7.81 (m, 2H), 7.76 (d, $J = 8.4$ Hz, 2H), 7.27 (dd, $J = 6.9, 2.0$ Hz, 1H), 7.13 (d, $J = 8.3$ Hz, 2H), 7.01 – 6.87 (m, 5H), 6.81 (d, $J = 8.9$ Hz, 1H), 4.64 (d, $J = 5.8$ Hz, 2H), 4.47 (d, $J = 6.3$ Hz, 2H), 4.26 – 3.99 (m, 7H), 3.90 (d, $J = 17.5$ Hz, 4H), 3.78 (dd, $J = 11.3, 4.9$ Hz, 1H), 3.69 – 3.45 (m, 6H), 3.26 (dt, $J = 22.3, 5.2$ Hz, 4H), 2.65 (ddd, $J = 17.0, 11.5, 5.4$ Hz, 1H), 2.39 (t, $J = 7.2$ Hz, 2H), 2.08 (dddd, $J = 43.0, 18.0, 9.2, 6.0$ Hz, 3H), 1.79 (p, $J = 6.2$ Hz, 2H), 1.60 (p, $J = 7.4$ Hz, 2H), 1.48 (tt, $J = 9.0, 5.8$ Hz, 2H), 1.24 (d, $J = 3.5$ Hz, 1H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 174.92, 173.91, 171.13, 166.60, 164.61, 158.79, 157.83, 152.92, 150.05, 149.75, 149.03, 147.97, 147.05, 141.84, 141.32, 138.75, 136.21, 131.67, 130.06, 129.53, 129.00, 124.26, 122.98, 120.57, 120.38, 114.79, 114.24, 113.18, 111.22, 105.67, 105.22, 81.36, 69.60, 66.59, 62.78, 53.88, 50.03, 48.70, 47.91, 47.55, 46.96, 44.93, 43.70, 32.61, 31.81, 28.76, 26.47, 25.71, 24.88.

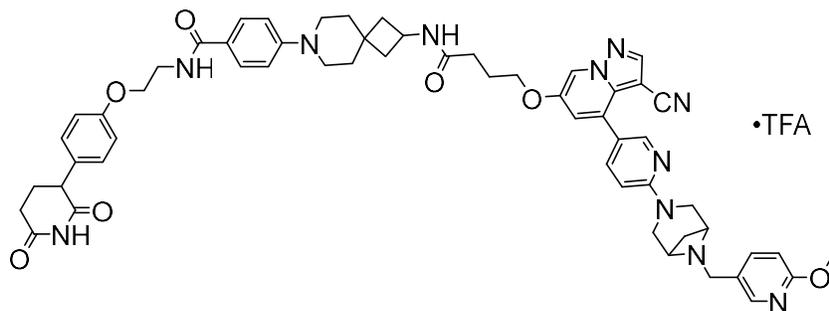
YW-N-10



^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.37 (q, $J = 5.3$ Hz, 1H), 8.69 (dd, $J = 6.2, 2.1$ Hz, 1H), 8.58 (s, 1H), 8.46 – 8.33 (m, 2H), 7.88 (dd, $J = 8.7, 2.5$ Hz, 1H), 7.76 (d, $J = 8.5$ Hz, 2H), 7.28 (dd, $J = 7.2, 2.1$ Hz, 1H), 7.12 (d, $J = 8.4$ Hz, 2H), 6.99 – 6.87 (m, 5H), 6.81 (d, $J = 8.8$ Hz, 1H), 4.62 (t, $J = 6.2$ Hz, 2H), 4.47 (d, $J = 6.3$ Hz, 2H), 4.21 (s, 1H), 4.07 (dd, $J = 15.4, 9.6$ Hz, 6H), 3.88 (d, $J = 4.4$ Hz, 4H), 3.78 (dd, $J = 11.3, 4.9$ Hz, 2H), 3.25 (dt, $J = 22.2, 5.2$ Hz, 4H), 2.92 – 2.80 (m, 1H), 2.72 – 2.58 (m, 1H), 2.35 (t, $J = 7.4$ Hz, 2H), 2.23 – 2.10 (m, 1H), 2.03 (ddt, $J = 22.3, 8.7, 4.2$ Hz, 2H), 1.76 (p, $J = 6.5$ Hz, 2H), 1.53 (p, $J = 7.4$ Hz, 2H), 1.48 – 1.40 (m, 2H), 1.40 – 1.28 (m, 4H), 1.25 (d, $J = 9.4$ Hz, 1H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 174.92, 173.92, 171.22, 166.60, 157.83, 152.93, 150.07, 149.75, 148.04, 147.04, 141.84, 138.73, 131.67, 130.07, 129.01, 124.25, 123.00, 120.58, 114.79, 114.25, 111.23, 105.18, 81.35, 69.62, 66.60, 62.79, 53.89, 50.03, 48.70, 47.91, 47.54, 46.96, 44.92, 32.65, 31.82, 29.16, 28.98, 28.84, 26.47, 25.79, 25.14.

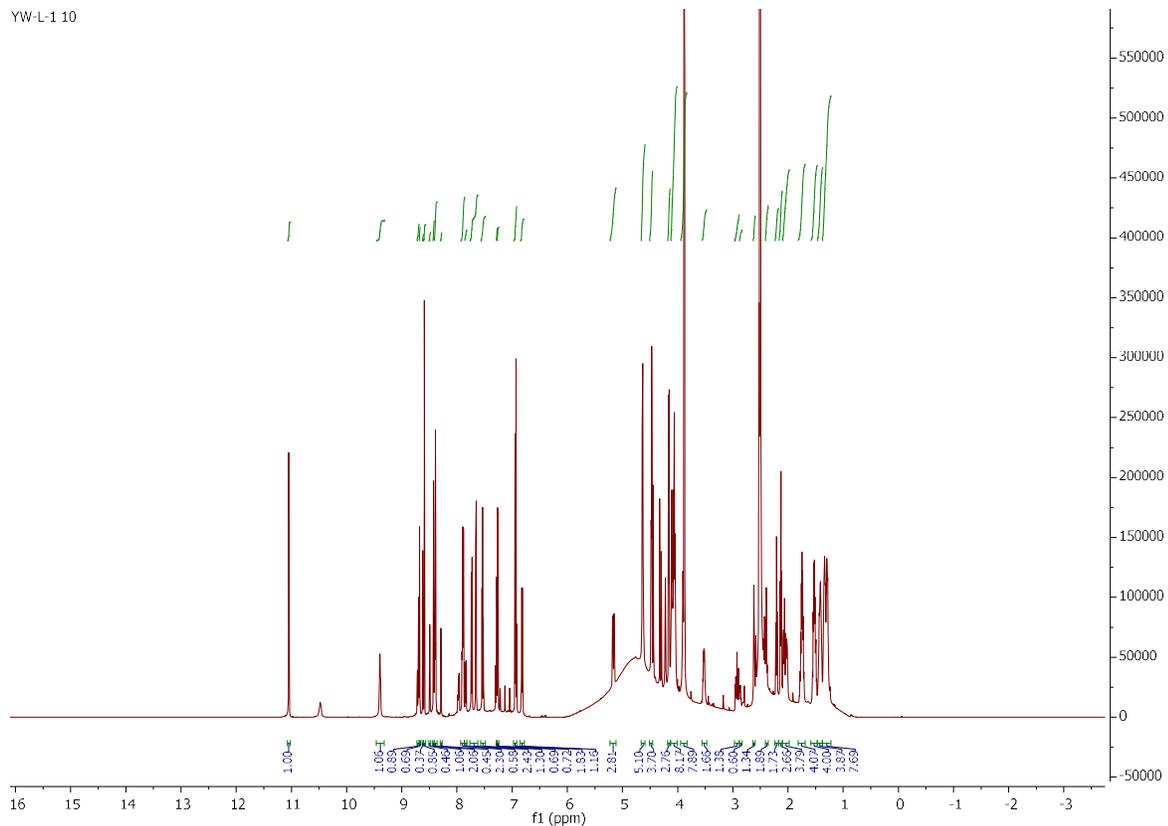
YW-N-11



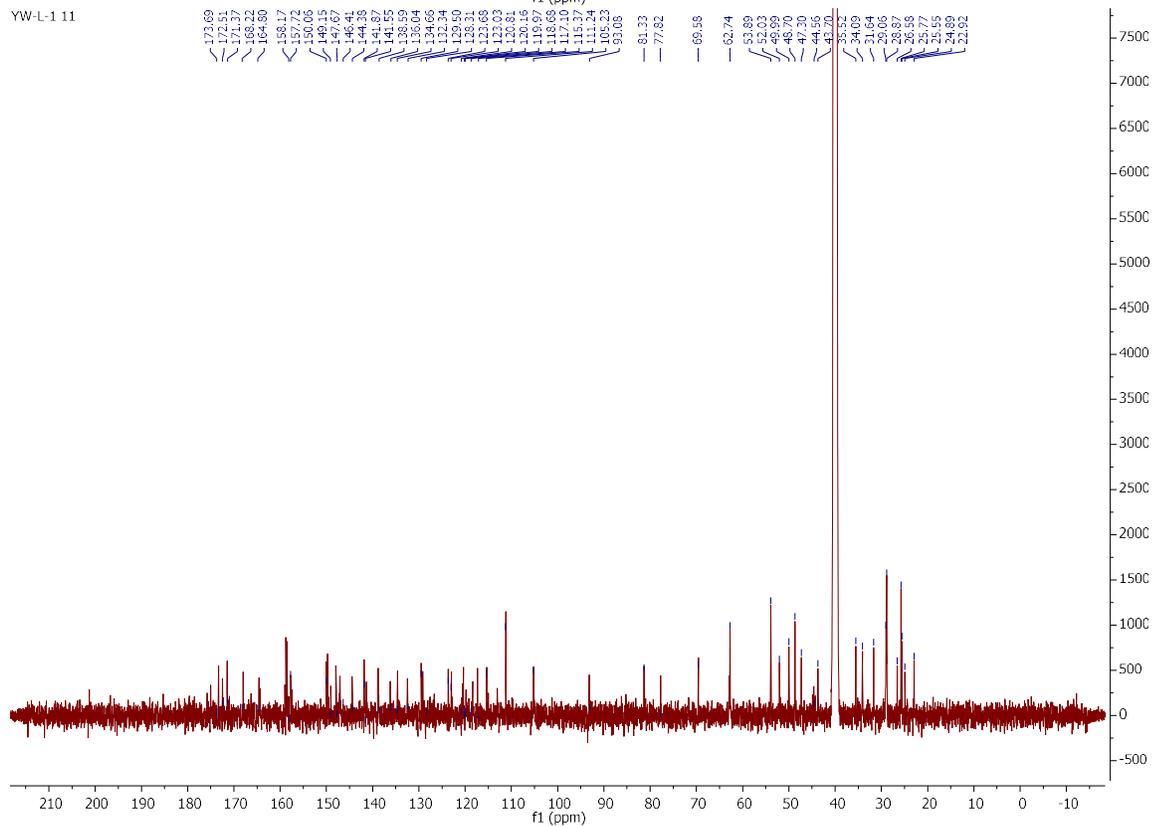
^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ 9.38 (q, $J = 5.3$ Hz, 1H), 8.71 (dd, $J = 6.8, 2.1$ Hz, 1H), 8.60 (d, $J = 10.9$ Hz, 1H), 8.52 – 8.24 (m, 3H), 8.12 (dd, $J = 7.7, 2.9$ Hz, 1H), 8.01 – 7.79 (m, 2H), 7.73 (d, $J = 8.5$ Hz, 2H), 7.27 (dd, $J = 9.4, 2.1$ Hz, 1H), 7.19 – 7.11 (m, 2H), 6.93 (td, $J = 7.1, 3.7$ Hz, 5H), 6.81 (d, $J = 8.8$ Hz, 1H), 4.63 (d, $J = 5.8$ Hz, 2H), 4.46 (d, $J = 6.3$ Hz, 2H), 4.22 (h, $J = 7.7$ Hz, 2H), 4.16 – 3.96 (m, 6H), 3.88 (d, $J = 4.4$ Hz, 4H), 3.78 (dd, $J = 11.3, 4.9$ Hz, 1H), 3.54 (dq, $J = 34.8, 6.1$ Hz, 3H), 3.19 (dt, $J = 39.5, 5.3$ Hz, 4H), 2.62 (dd, $J = 11.6, 5.4$ Hz, 1H), 2.45 (t, $J = 4.3$ Hz, 1H), 2.30 – 2.21 (m, 2H), 2.20 – 2.09 (m, 3H), 2.01 (ddt, $J = 14.9, 11.9, 8.0$ Hz, 4H), 1.69 – 1.57 (m, 4H), 1.57 – 1.47 (m, 2H), 1.34 – 1.06 (m, 1H).

^{13}C NMR (101 MHz, $\text{DMSO-}d_6$) δ 174.92, 173.91, 170.94, 166.65, 157.83, 152.98, 149.75, 149.04, 147.93, 147.08, 141.84, 138.76, 136.25, 131.66, 130.06, 129.02, 114.78, 114.29, 111.22, 105.25, 81.37, 69.24, 66.61, 62.77, 53.89, 50.03, 48.70, 46.96, 45.05, 43.70, 35.35, 31.82, 26.47, 24.89.

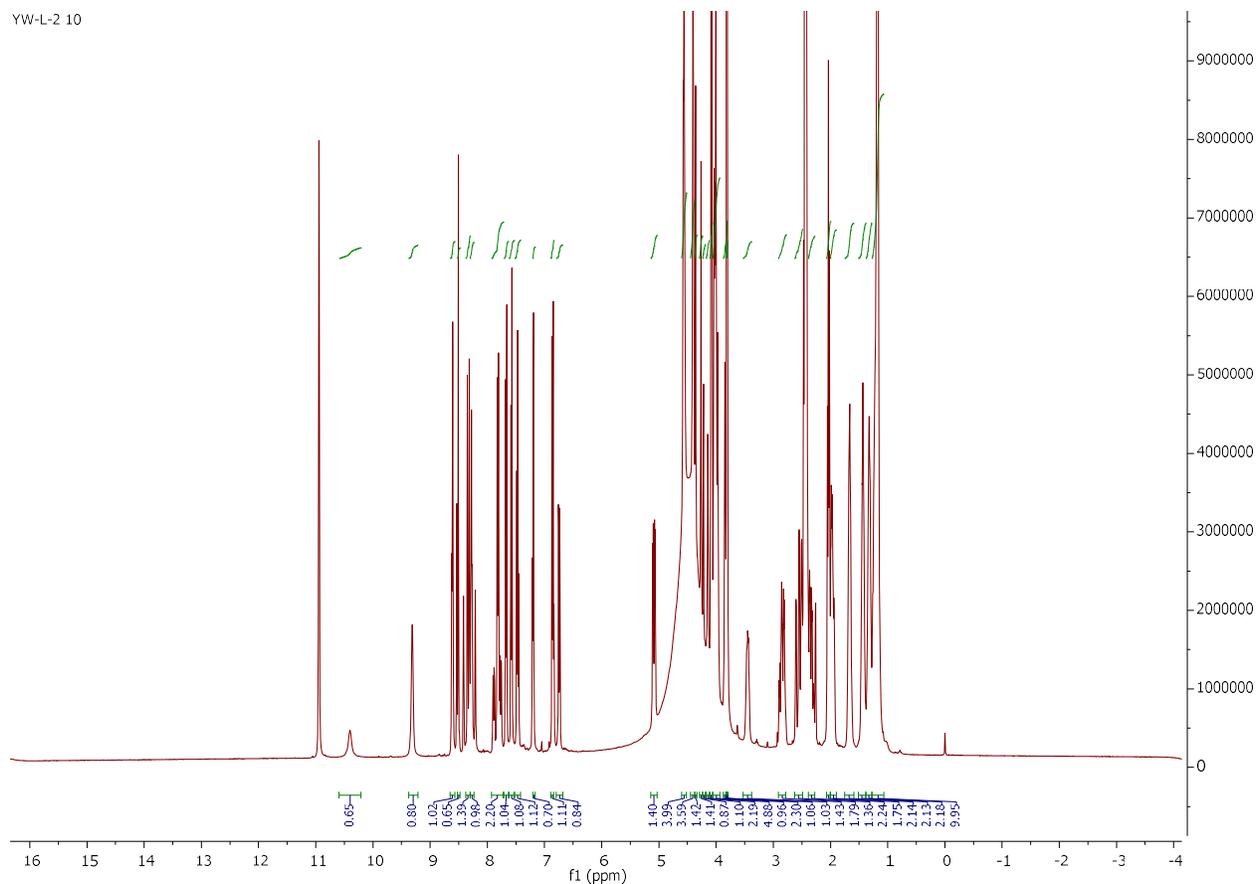
YW-L-1.10



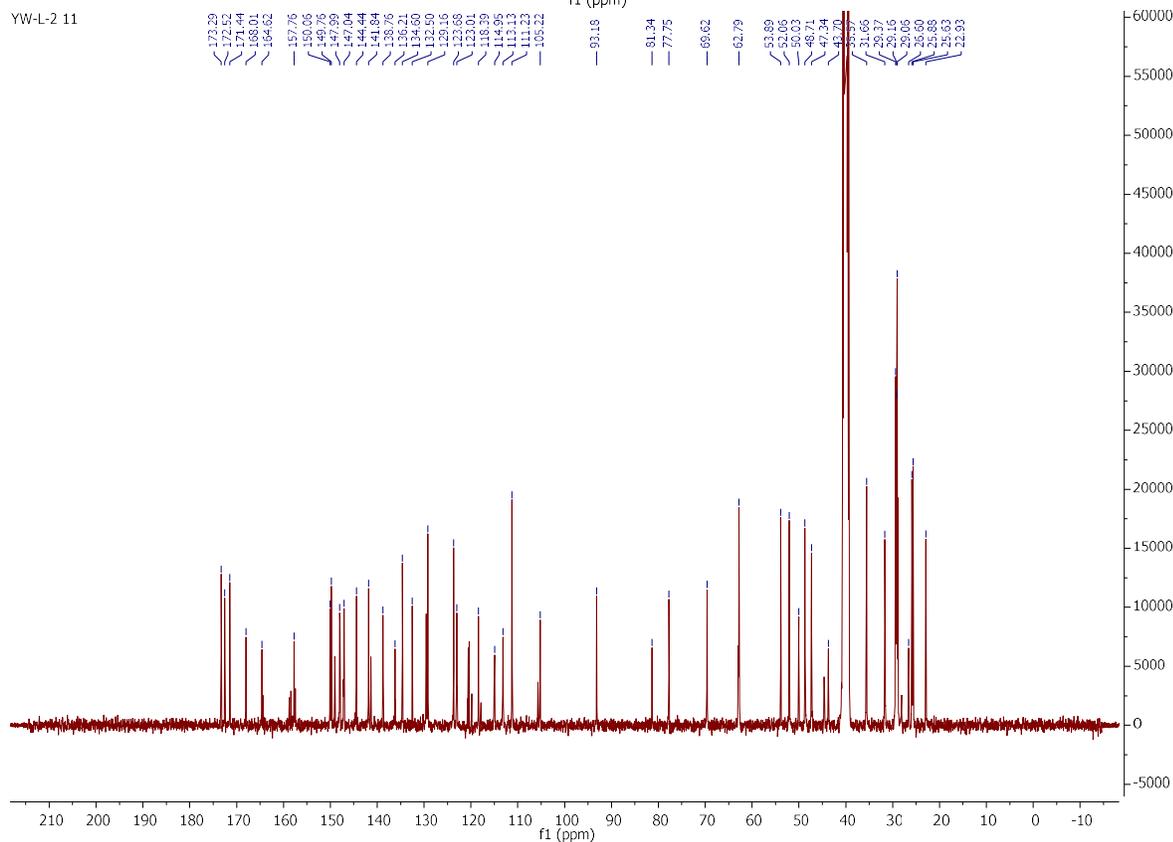
YW-L-1.11

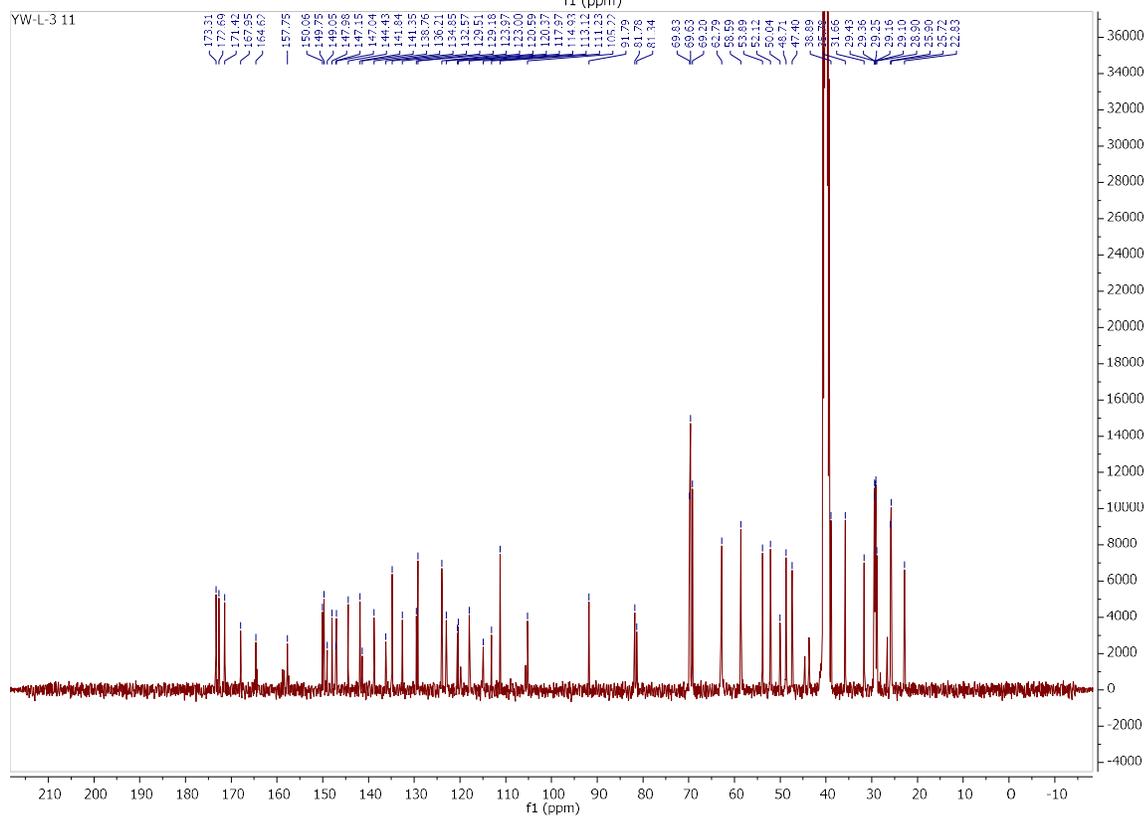
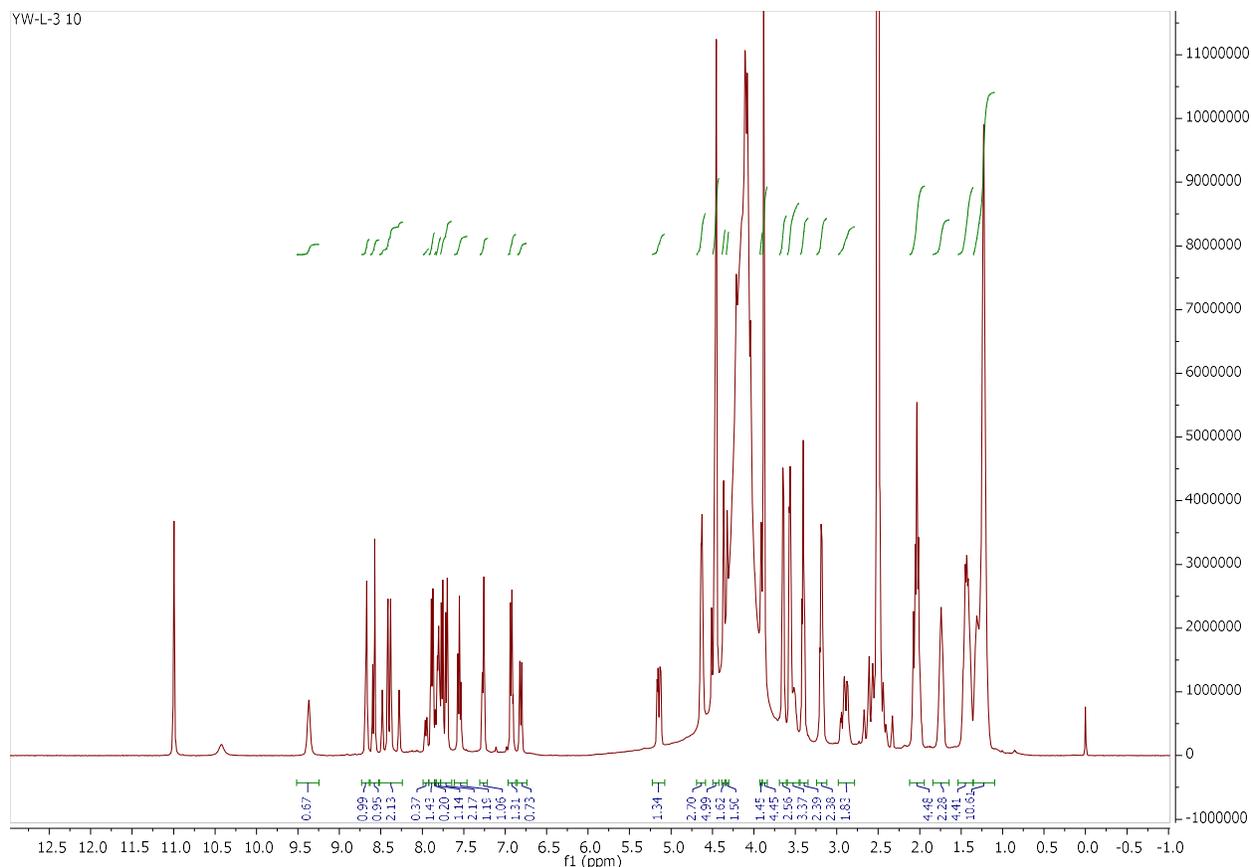


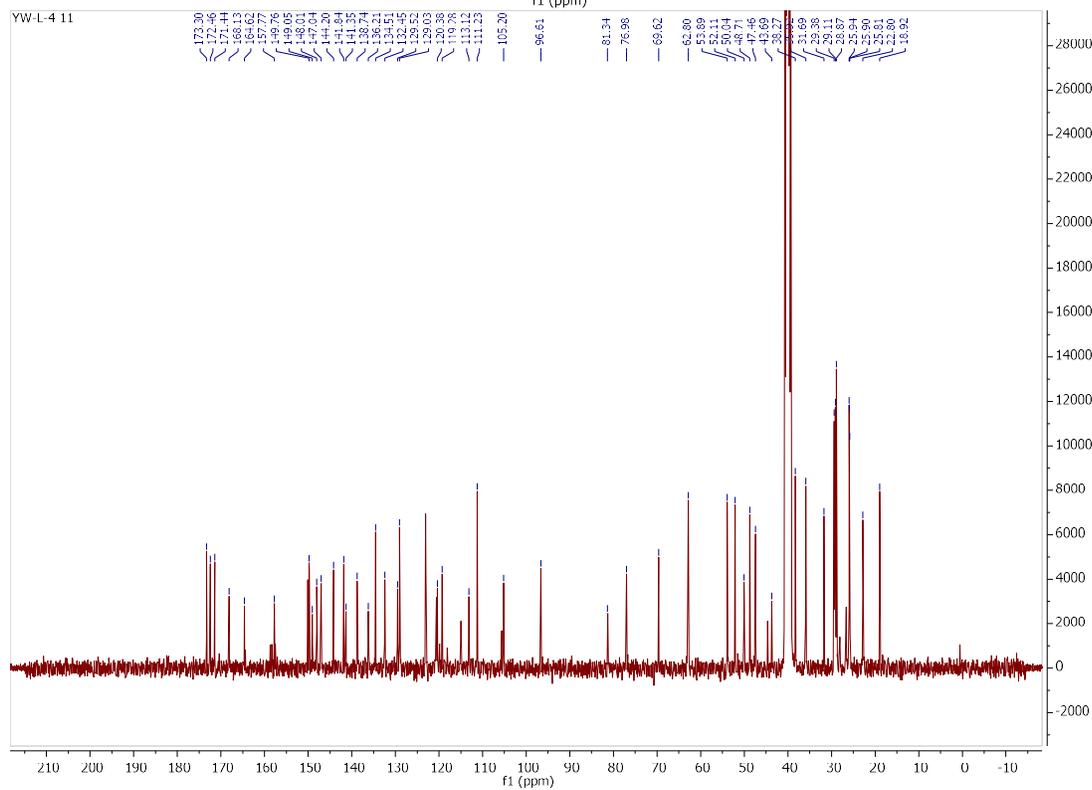
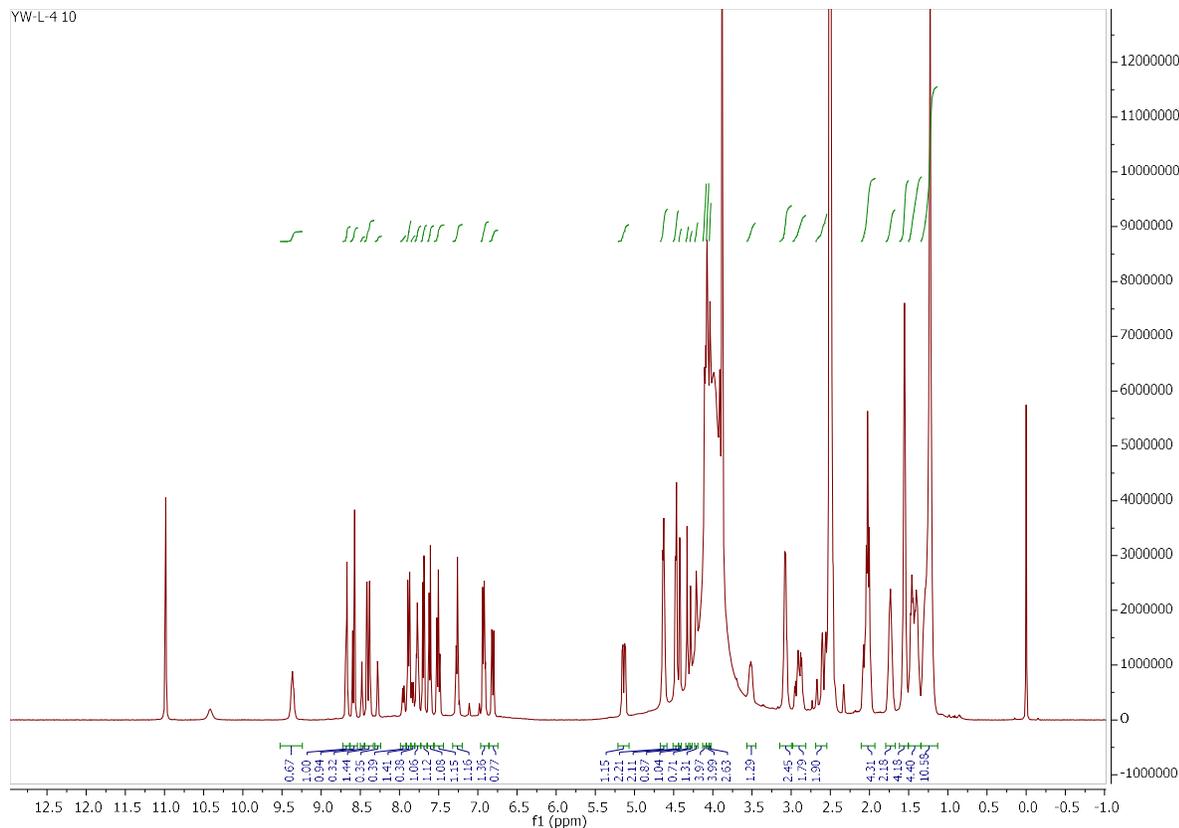
YW-L-2 10



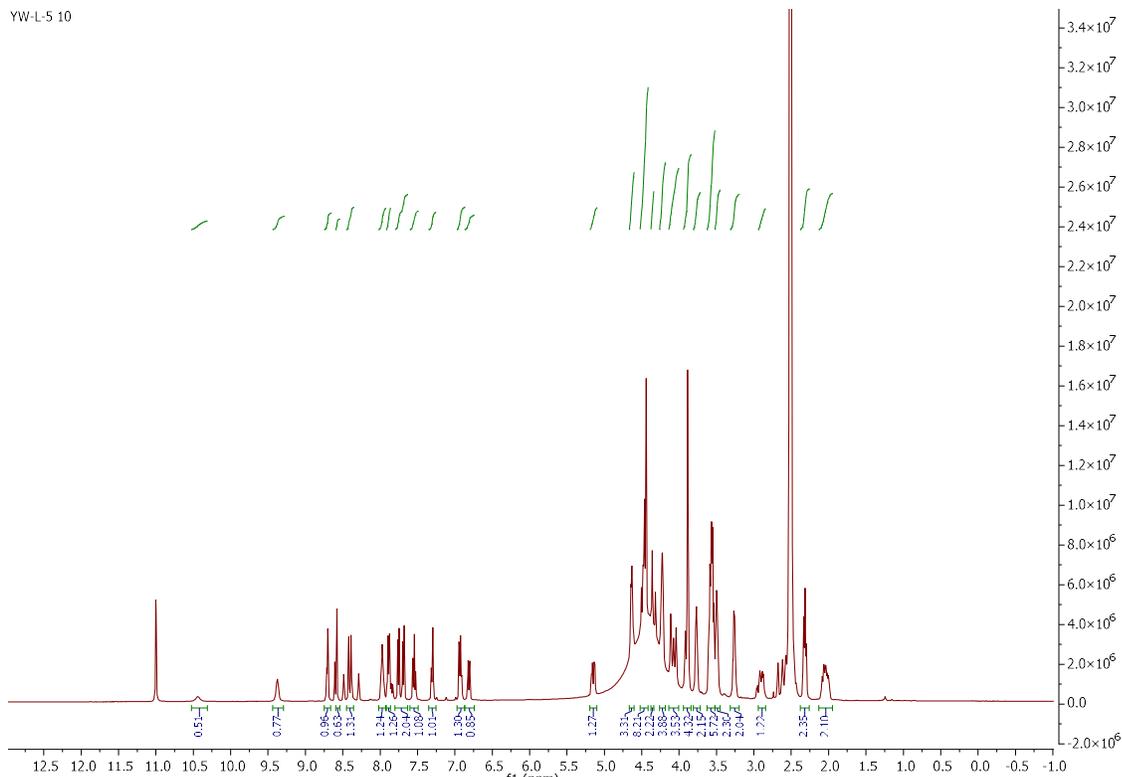
YW-L-2 11



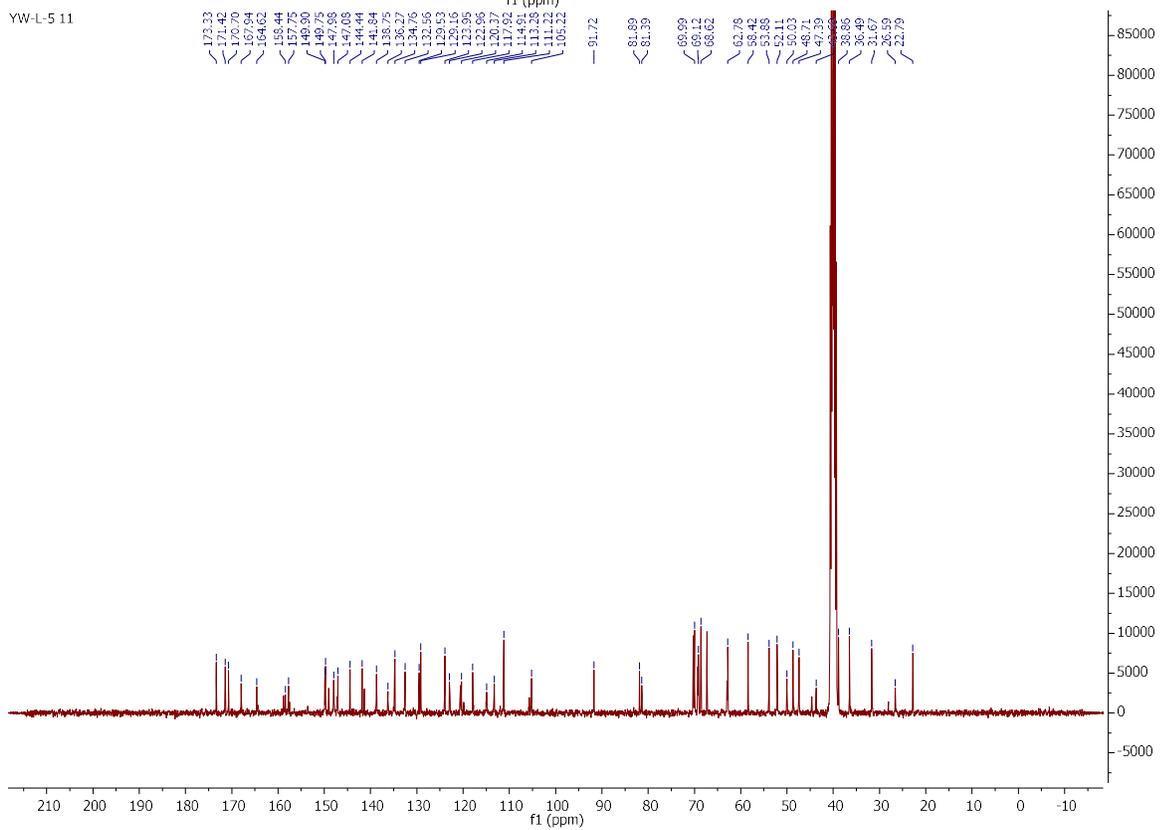




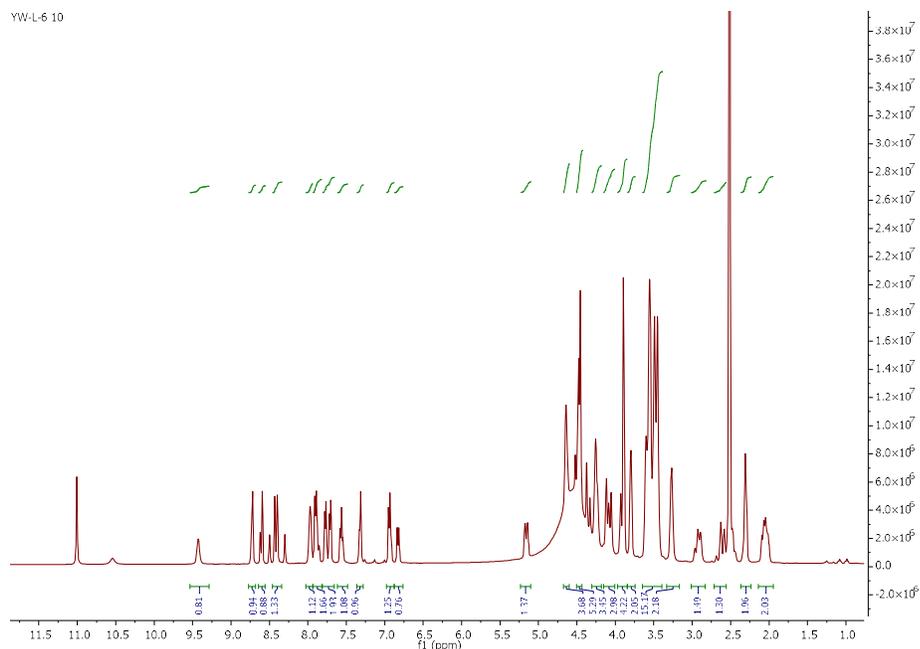
YW-L-5 10



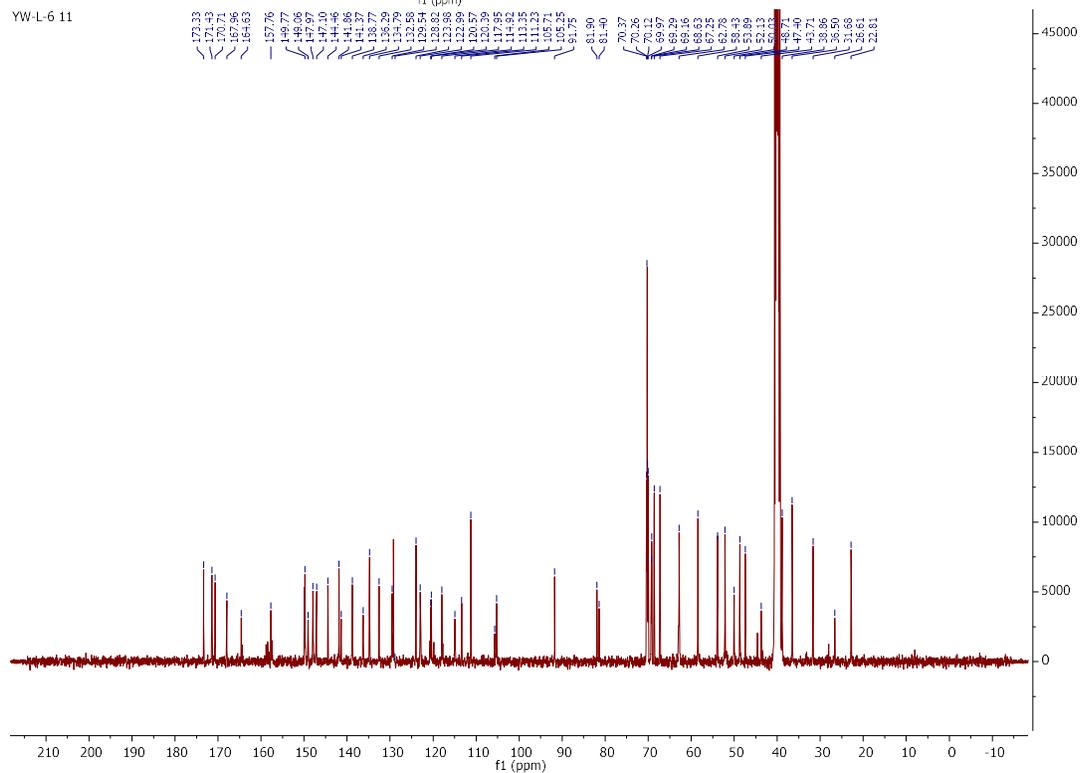
YW-L-5 11



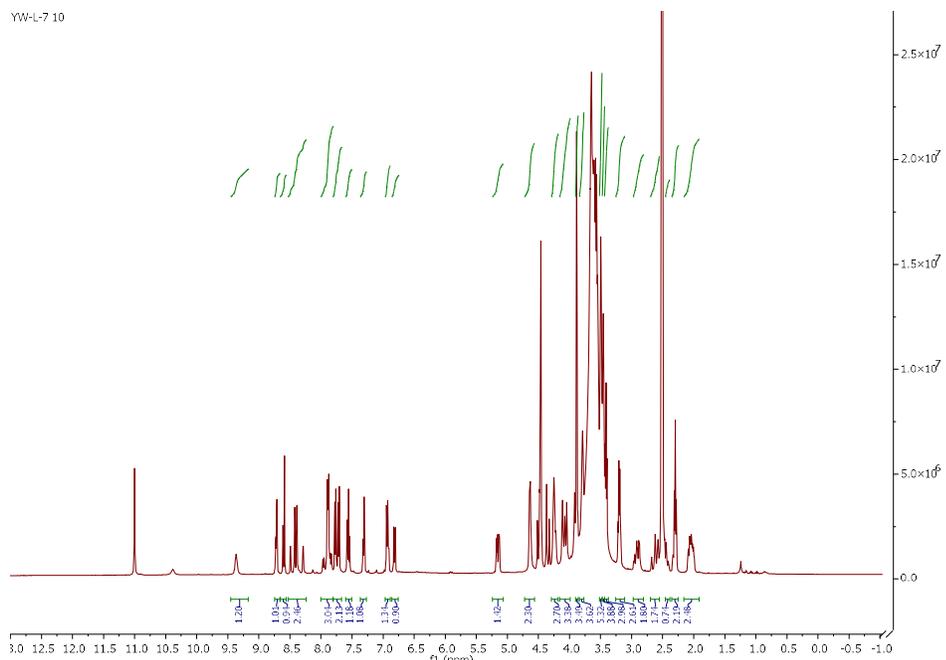
YW-L-6 10



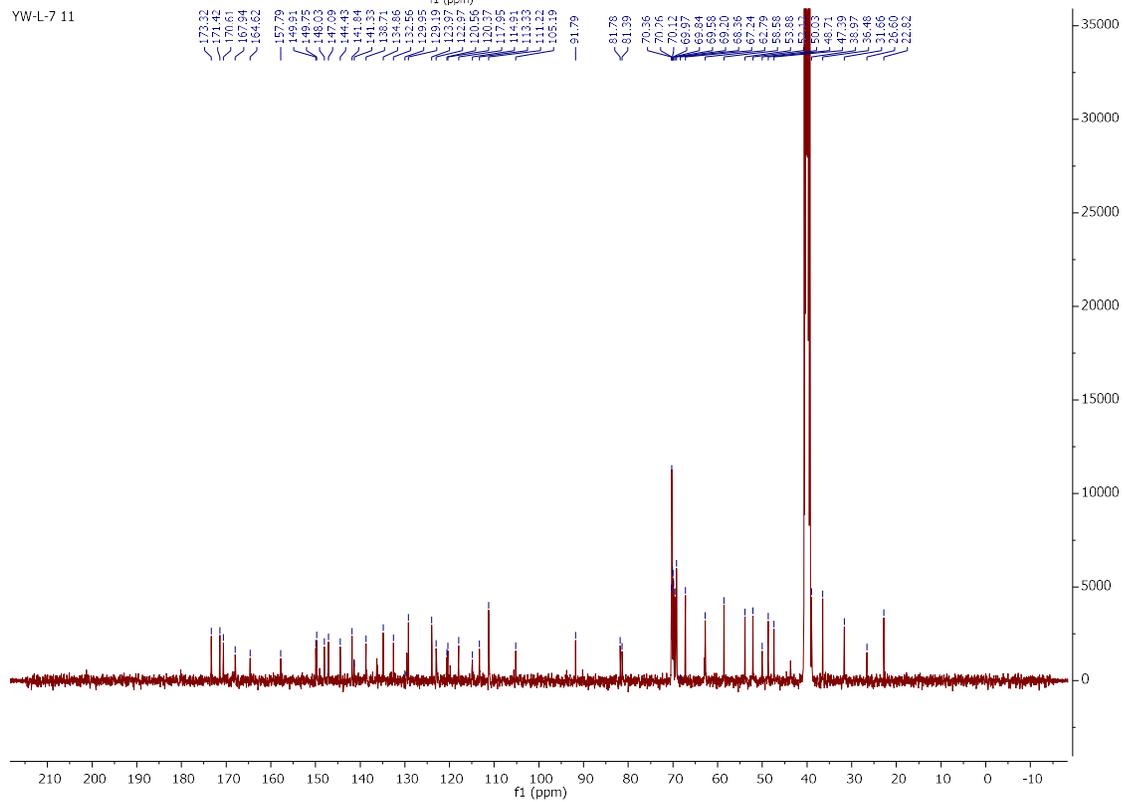
YW-L-6 11



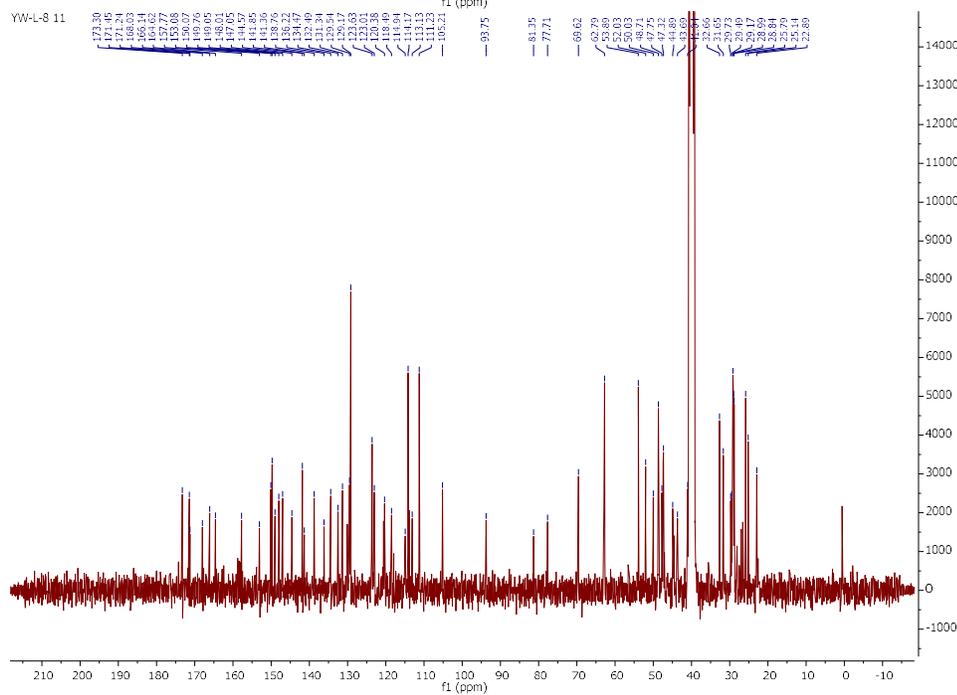
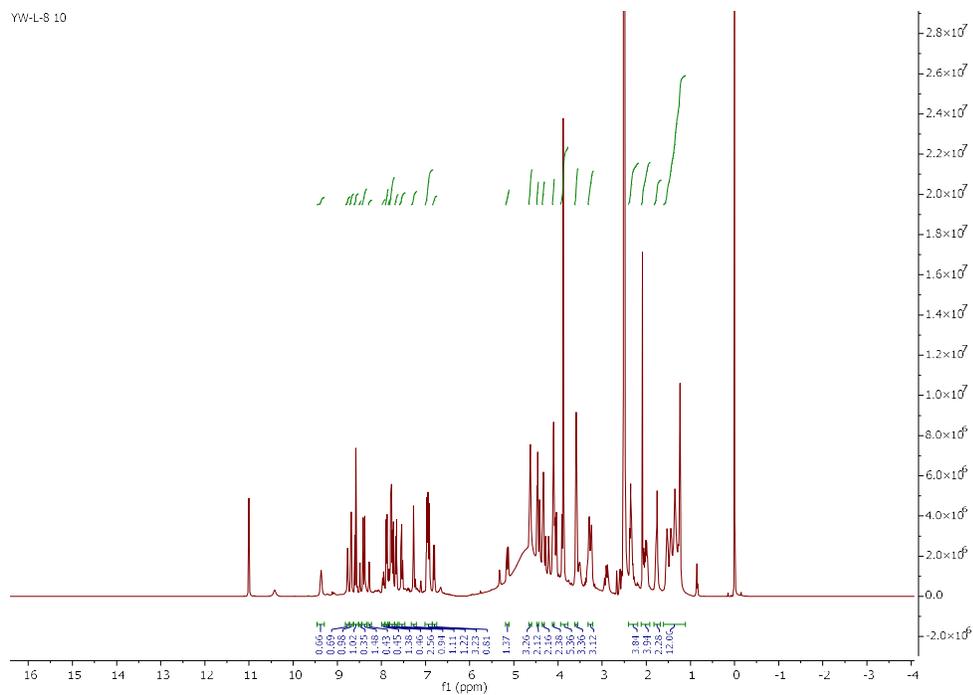
YW-L-7 10



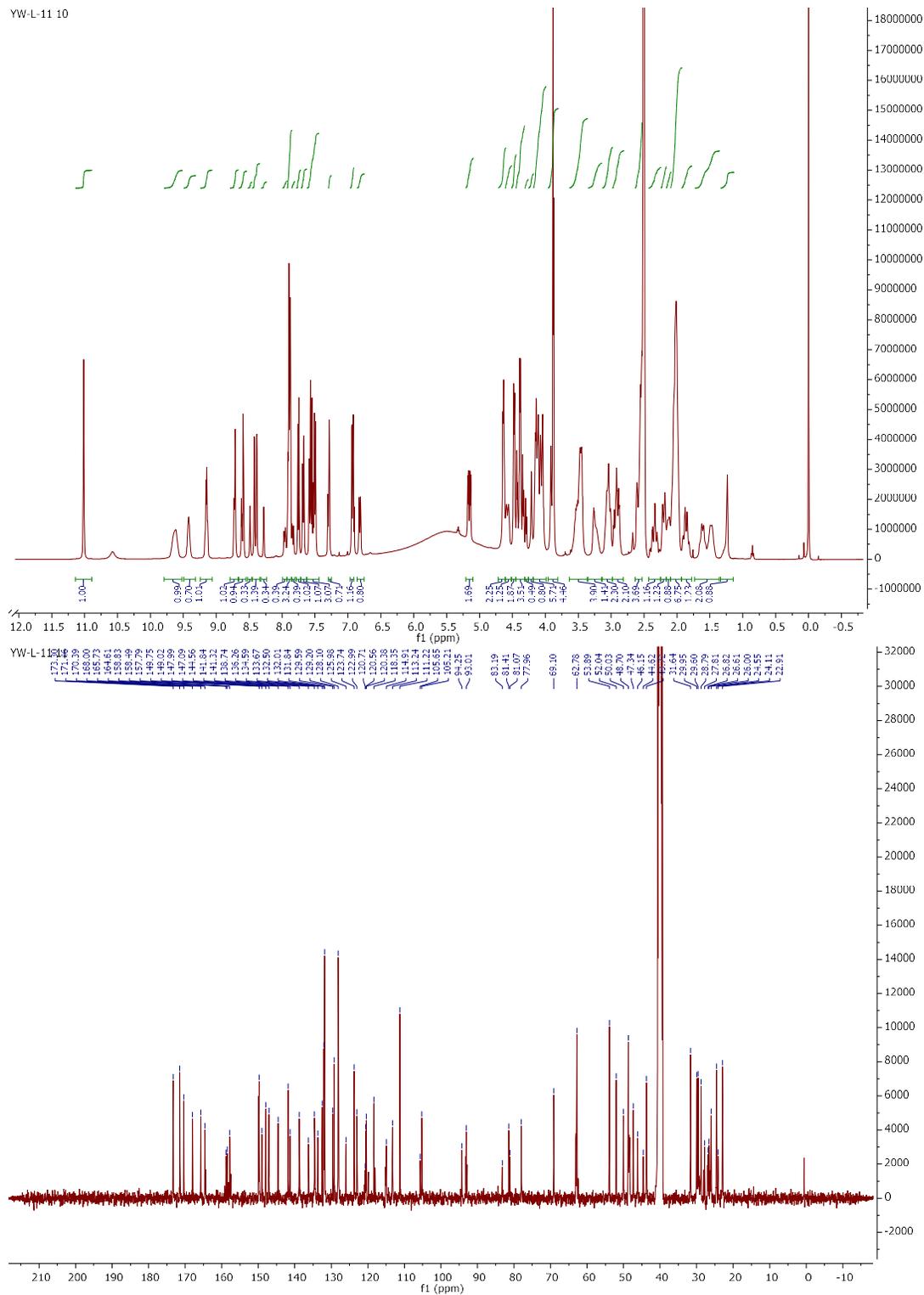
YW-L-7 11



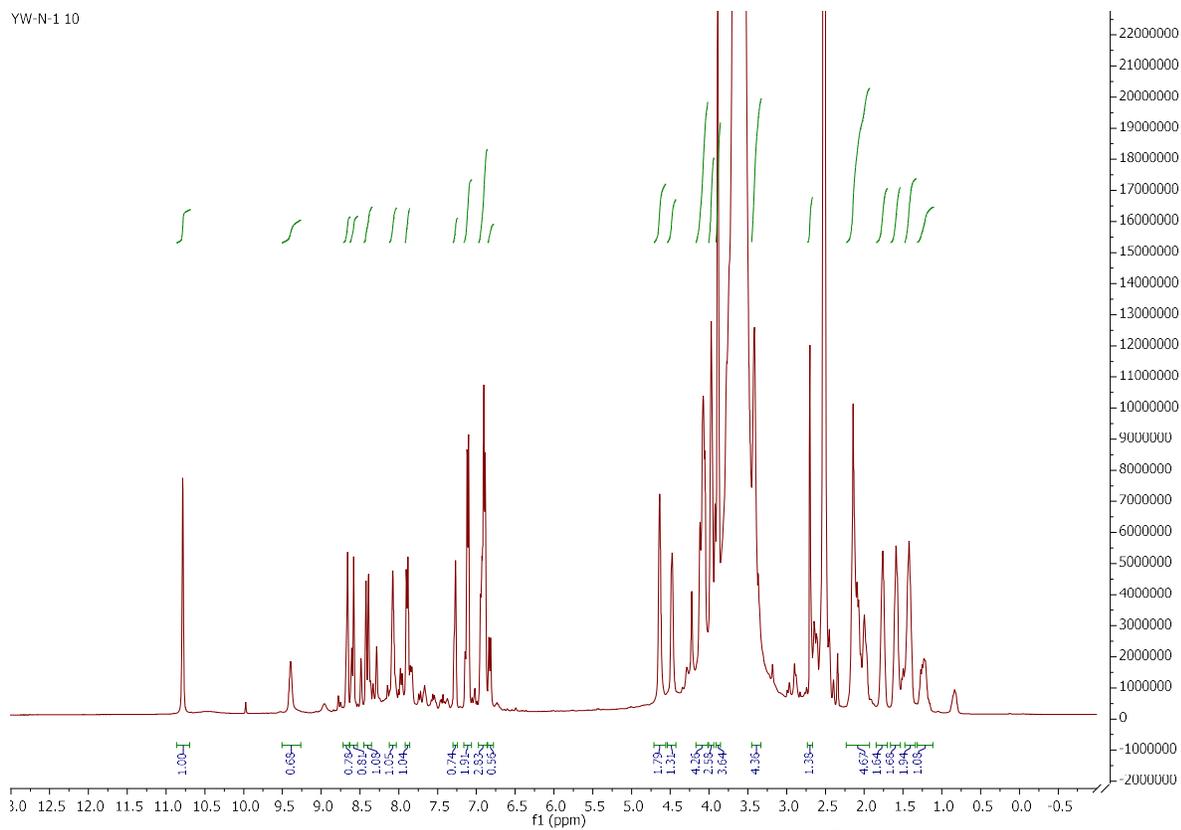
YW-L-8 10



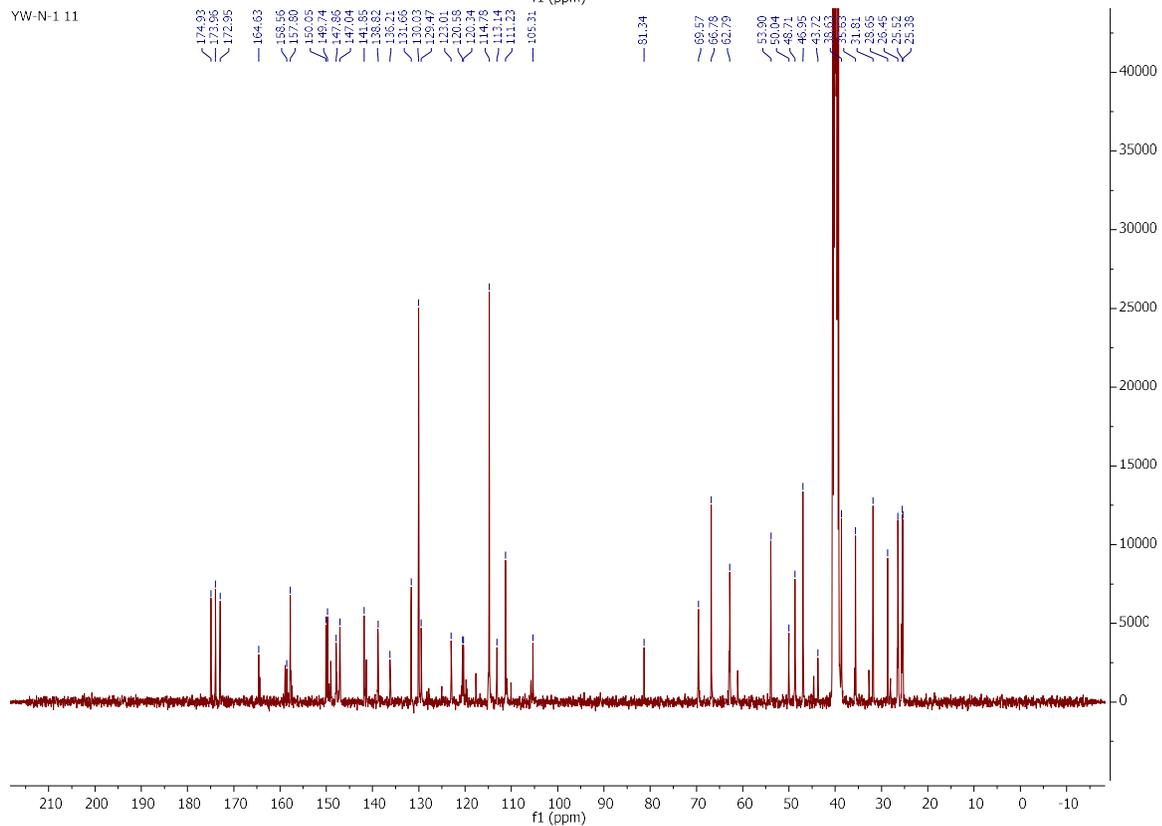
YW-L-11 10



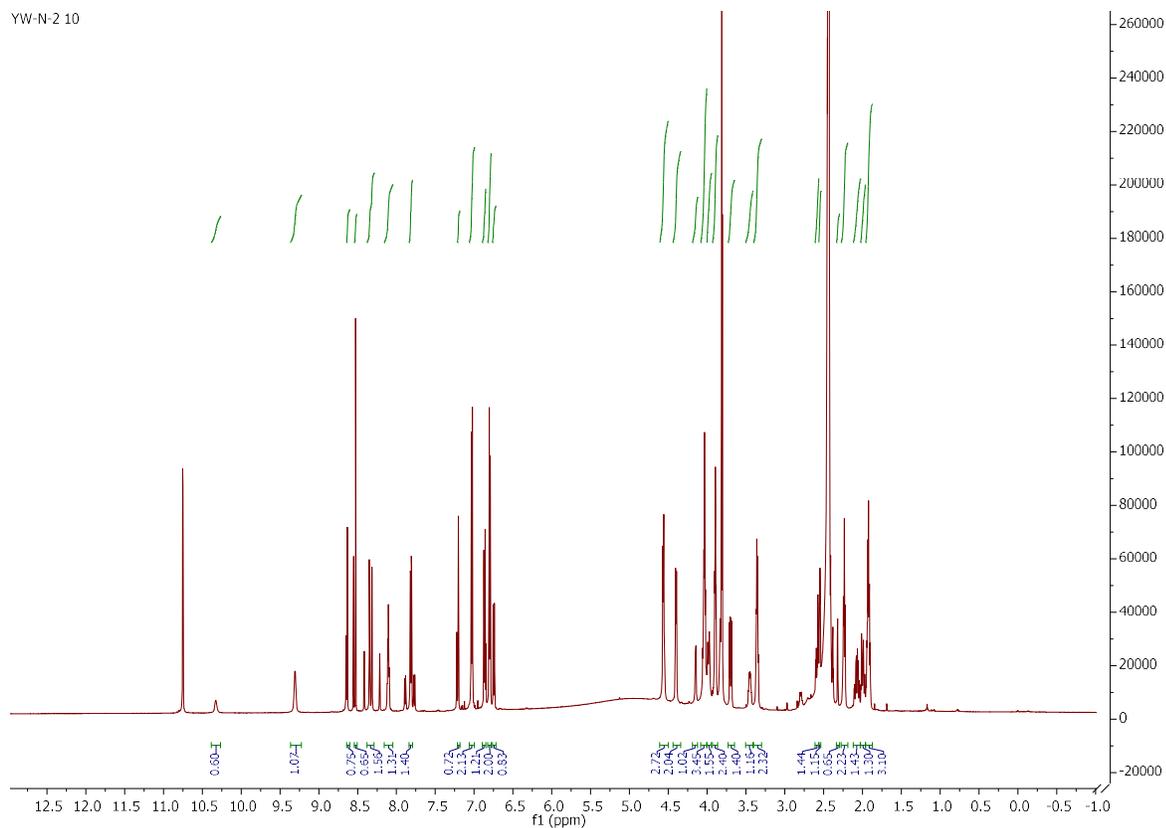
YW-N-1 10



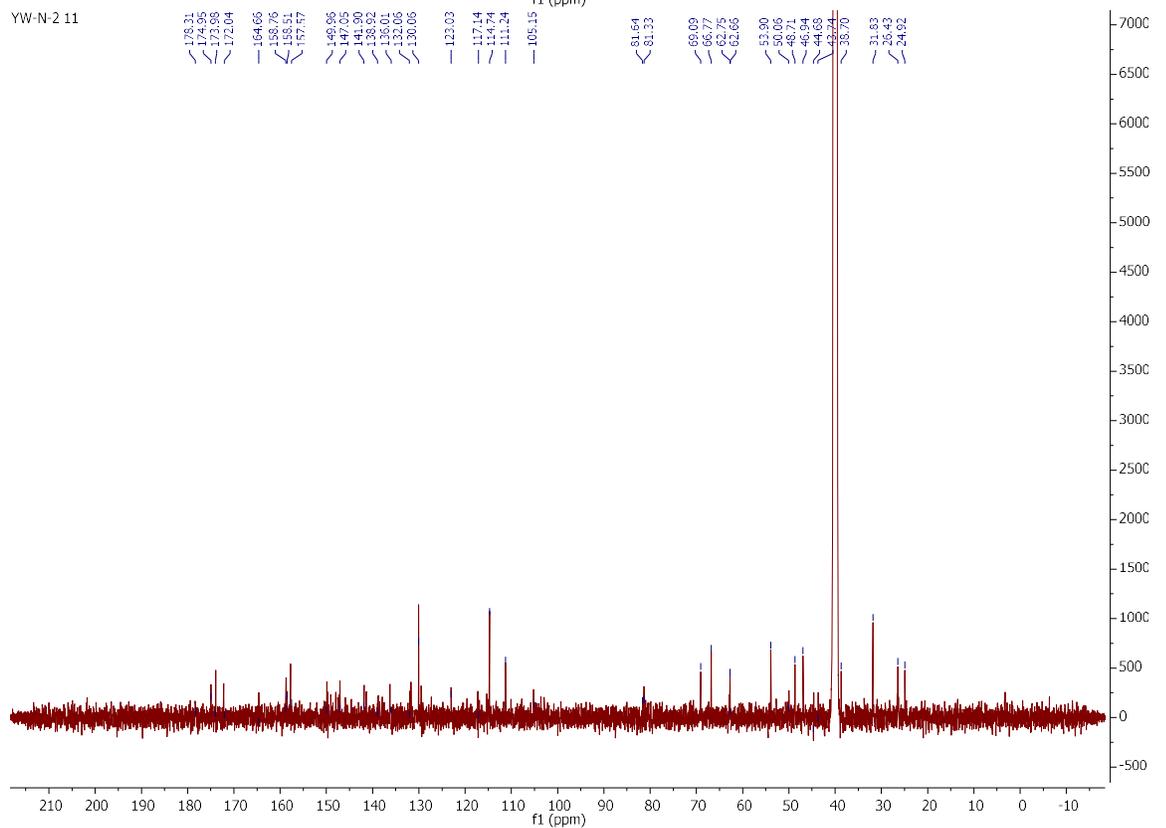
YW-N-1 11



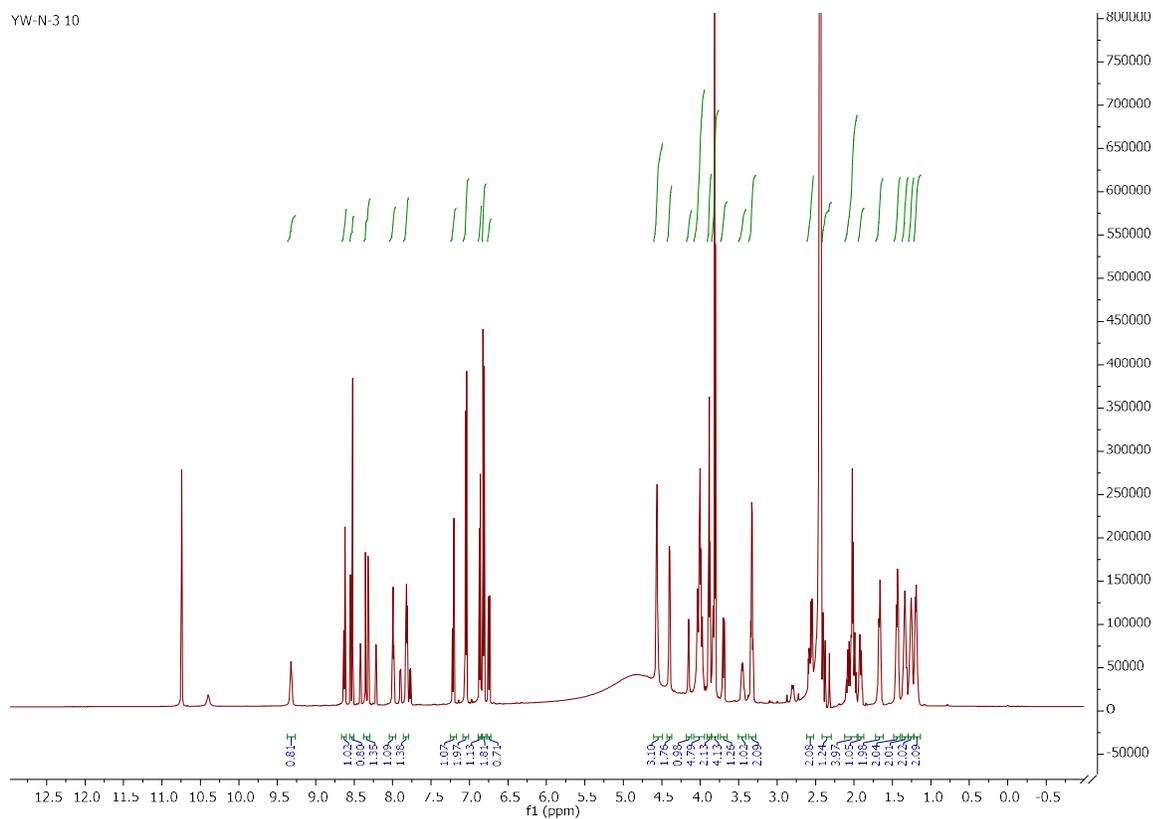
YW-N-2 10



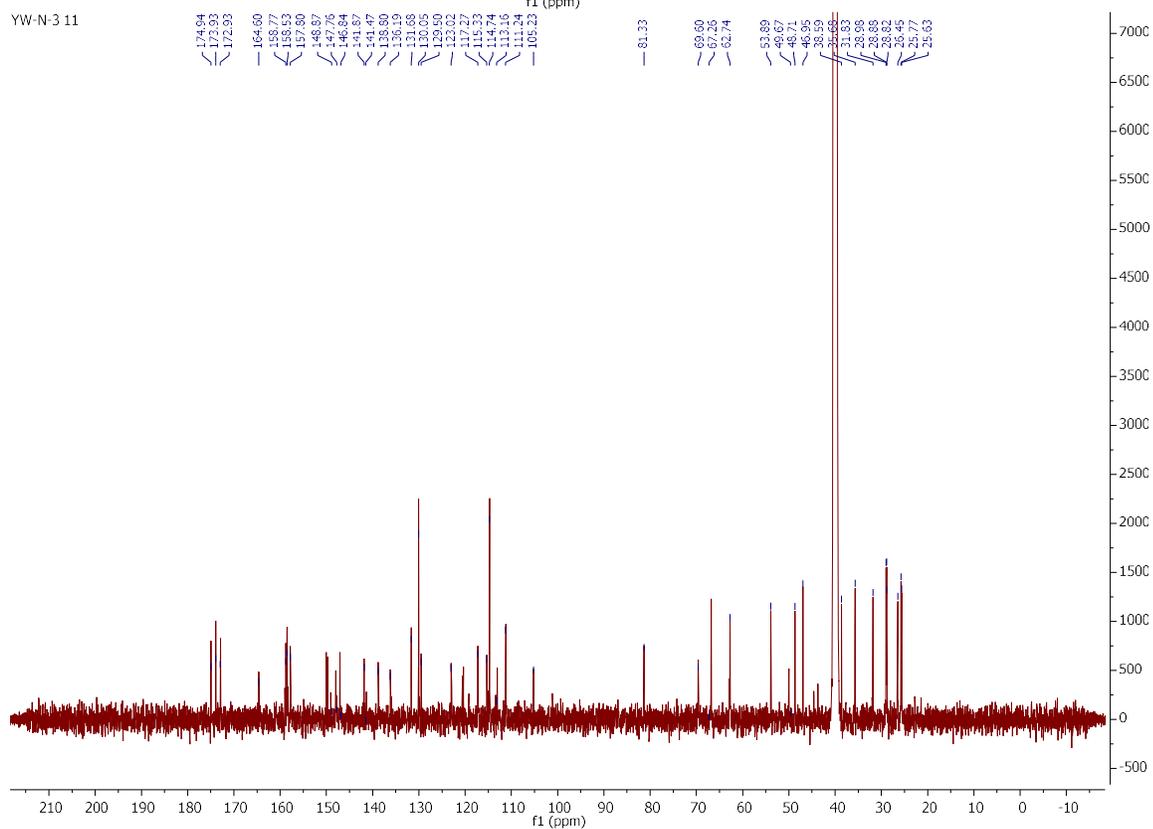
YW-N-2 11



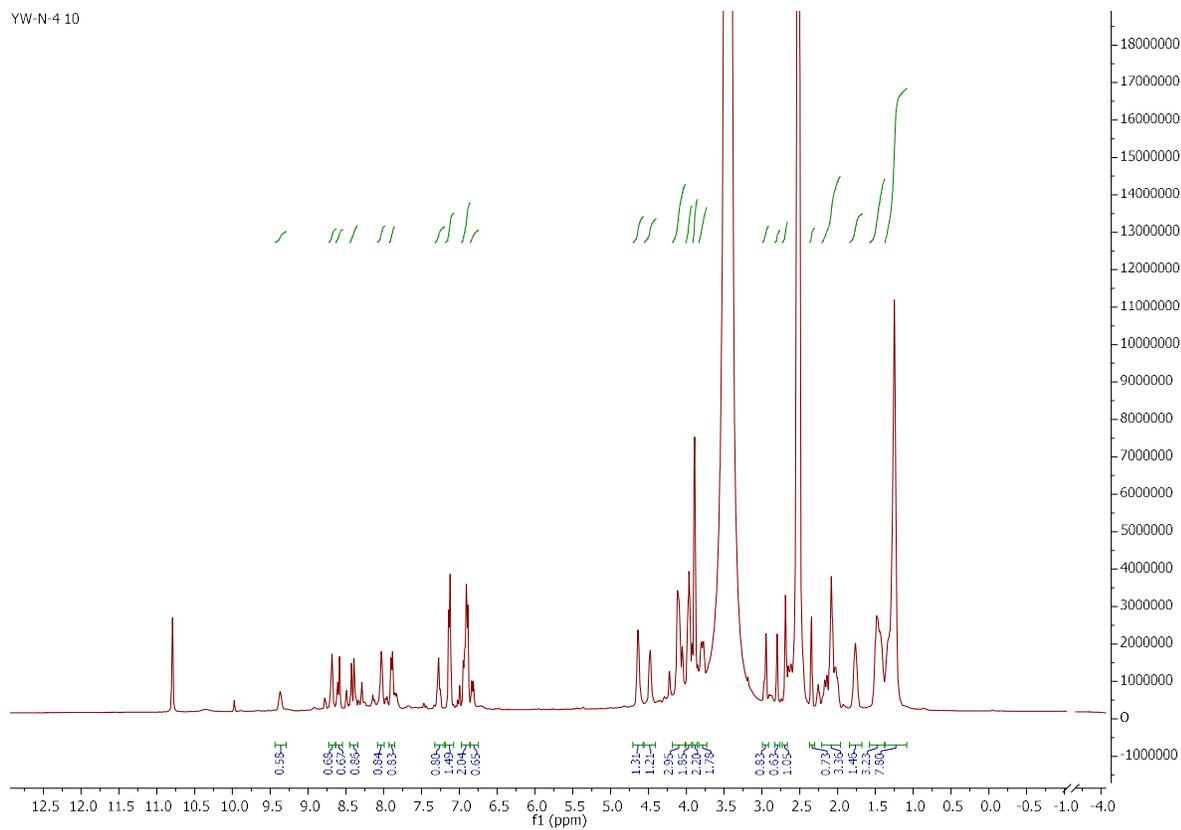
YW-N-3 10



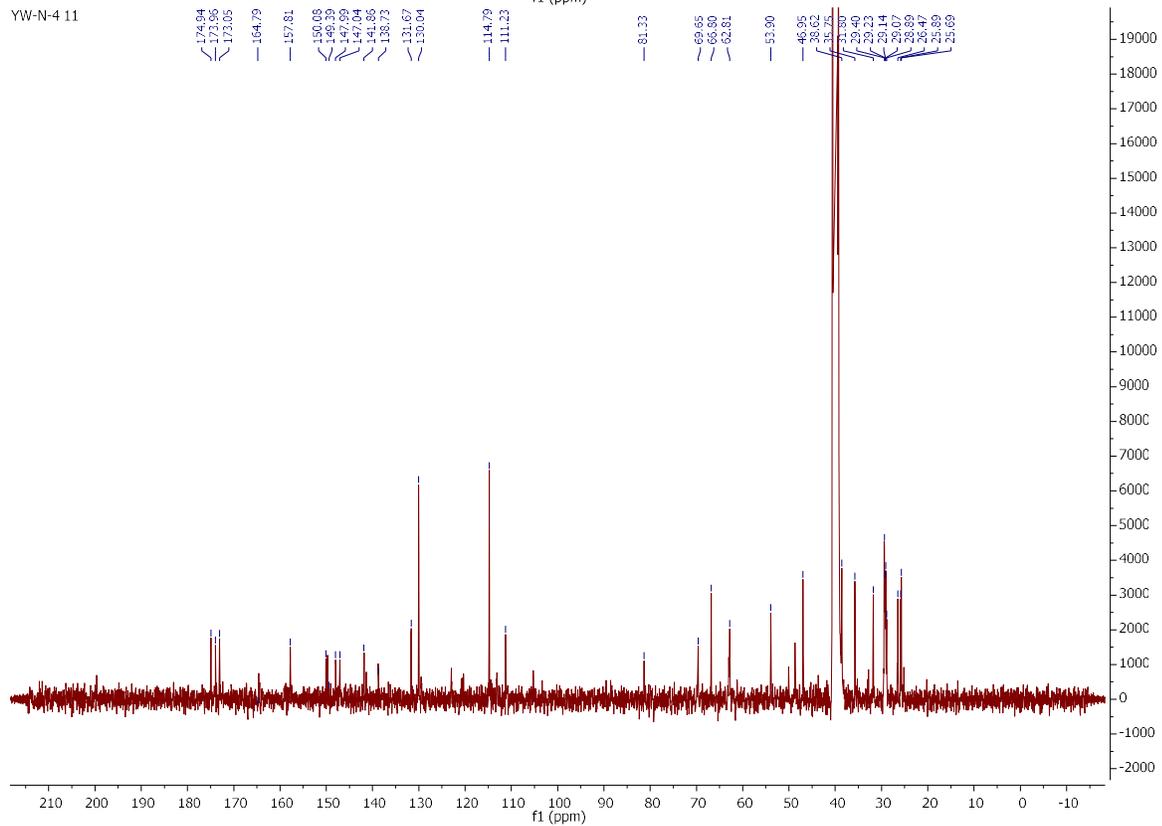
YW-N-3 11



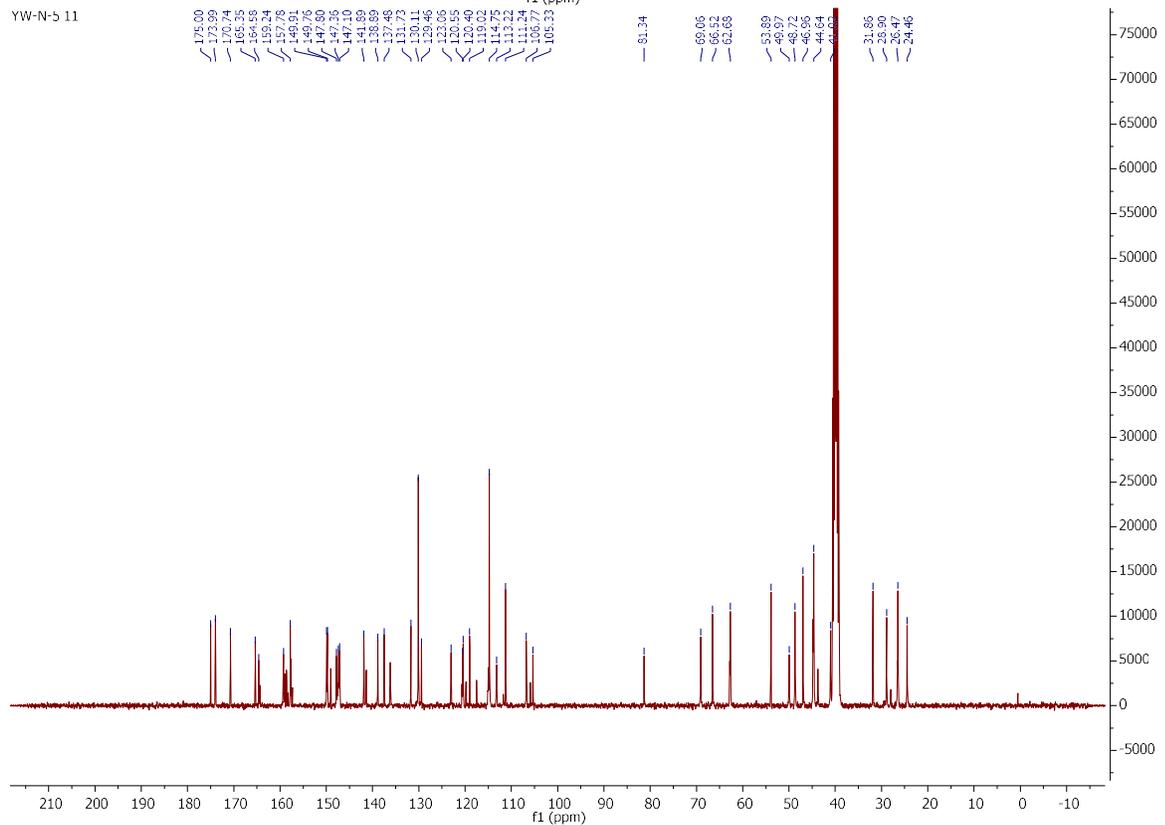
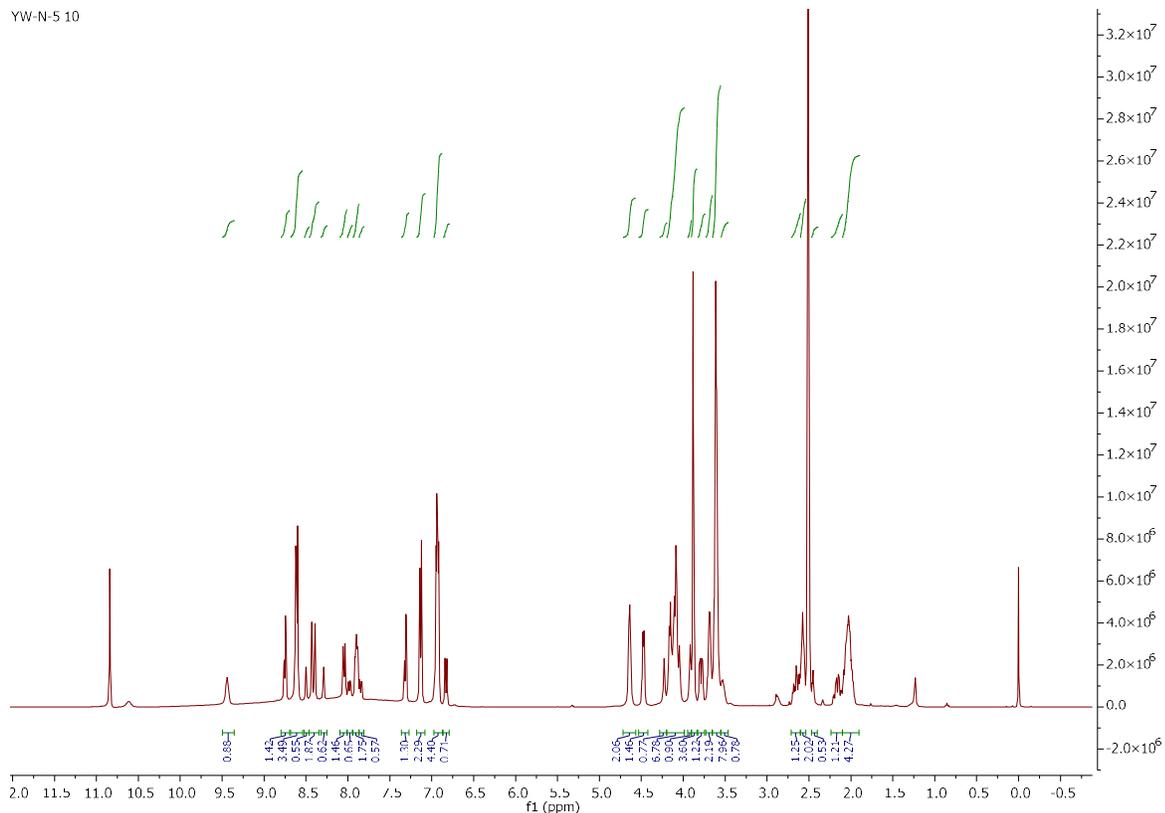
YW-N-4 10



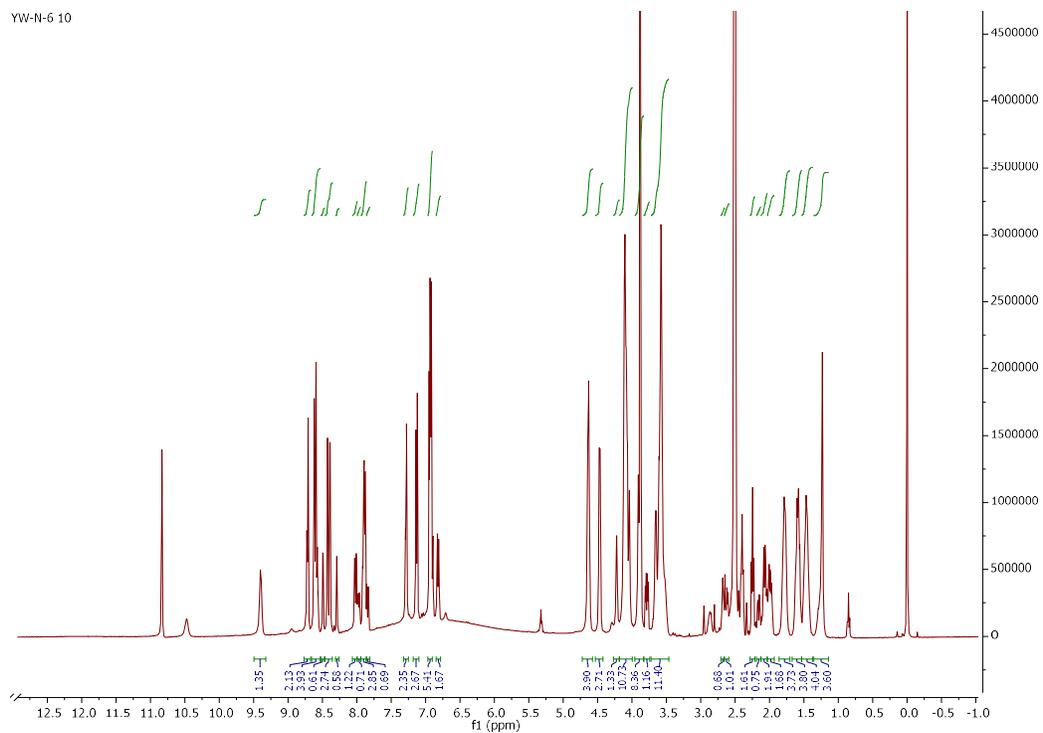
YW-N-4 11



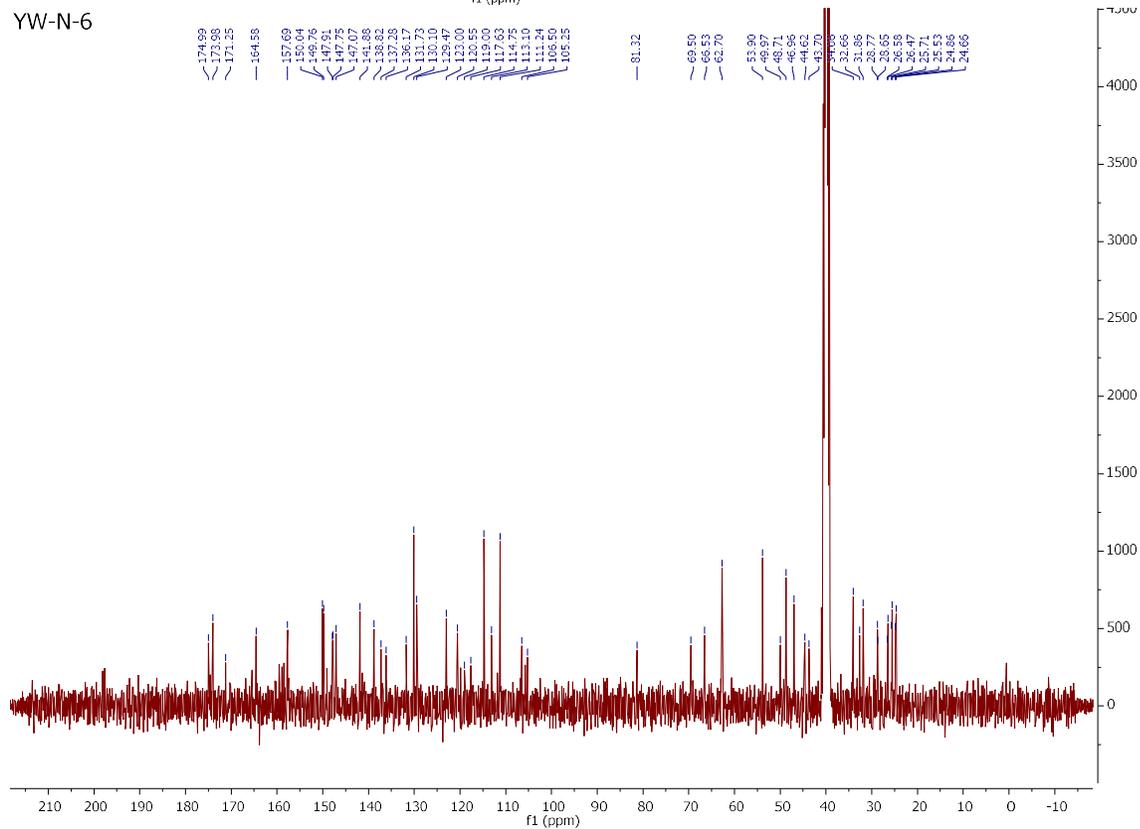
YW-N-5 10



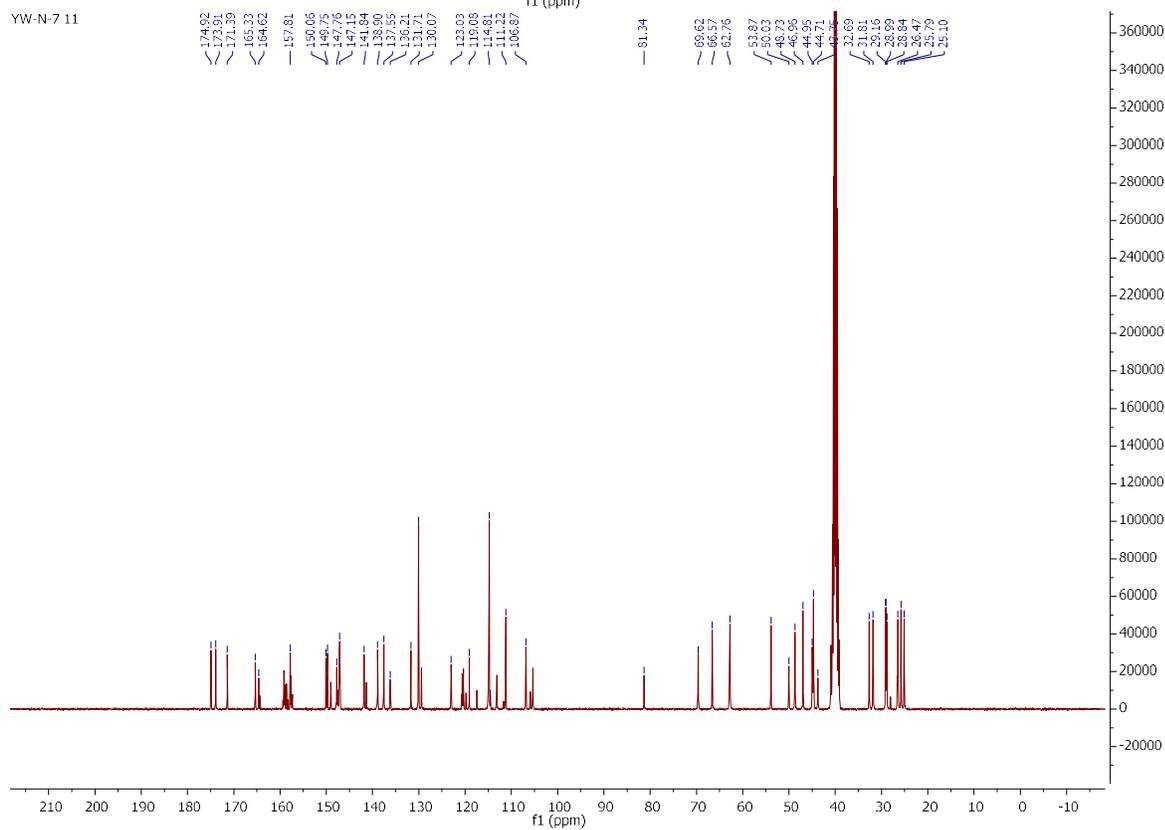
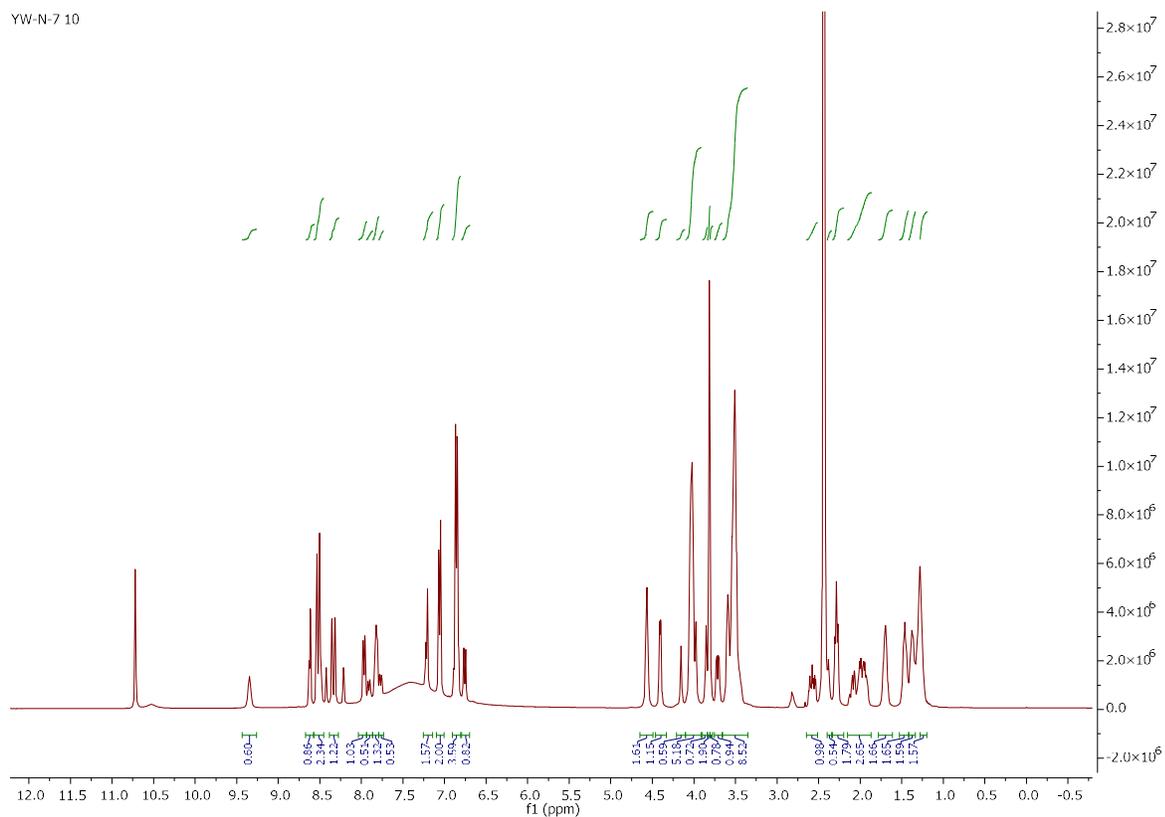
YW-N-6 10



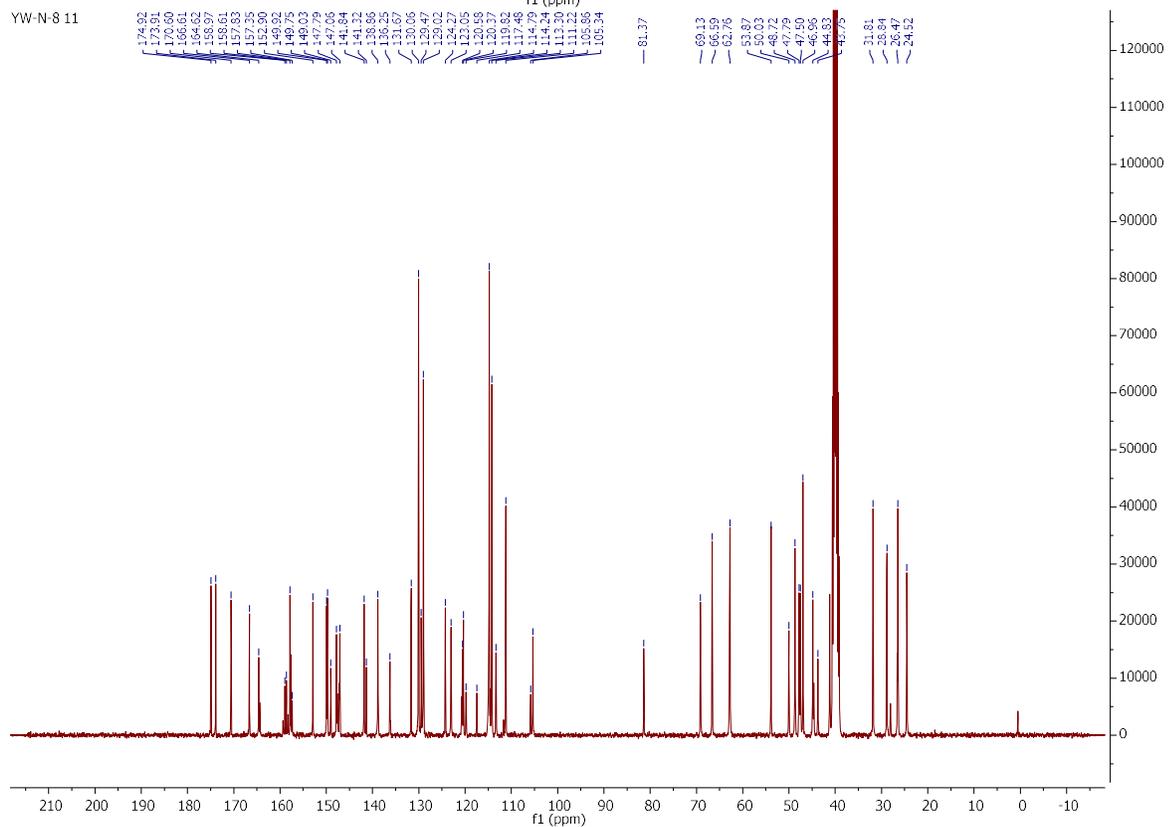
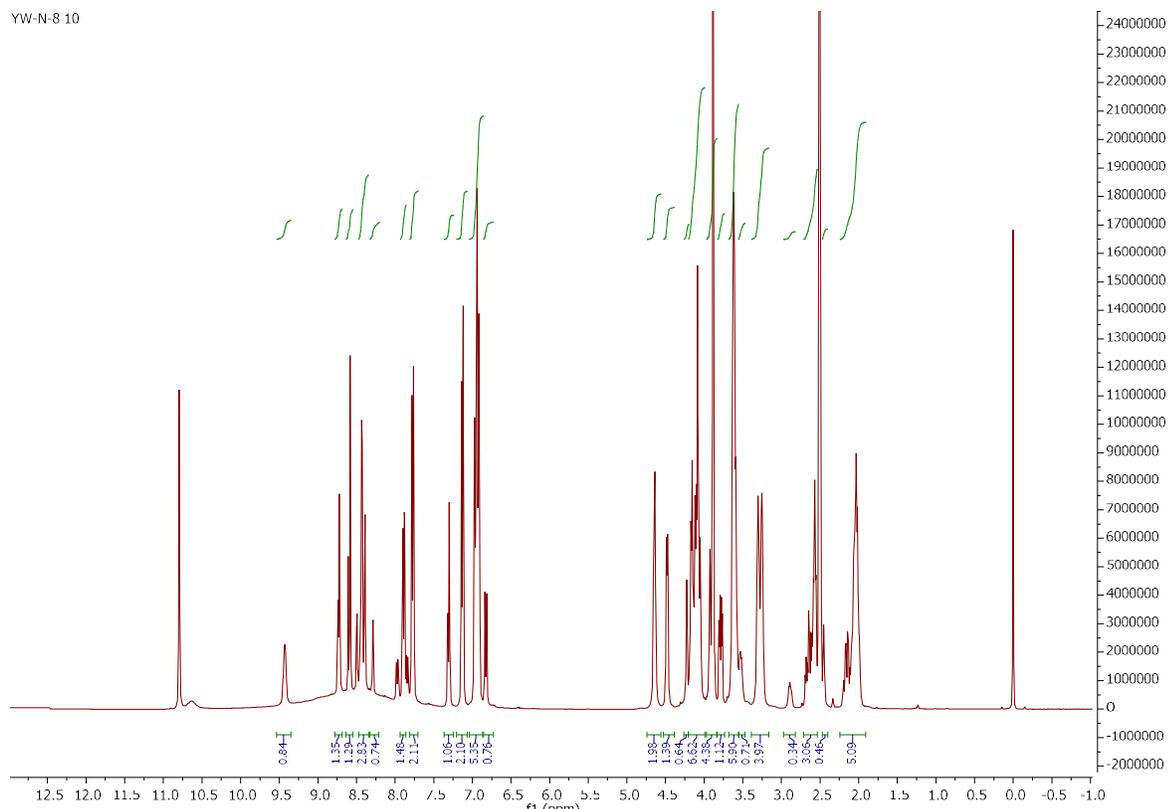
YW-N-6



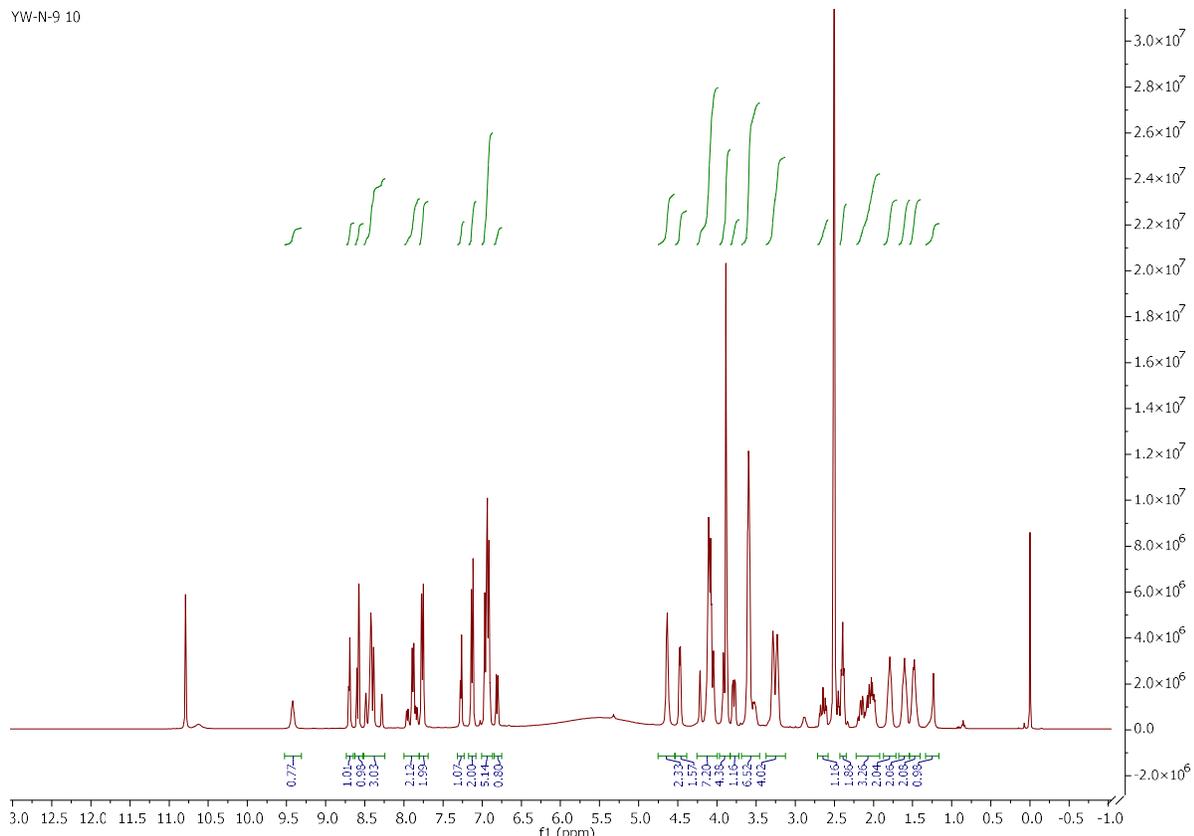
YW-N-7 10



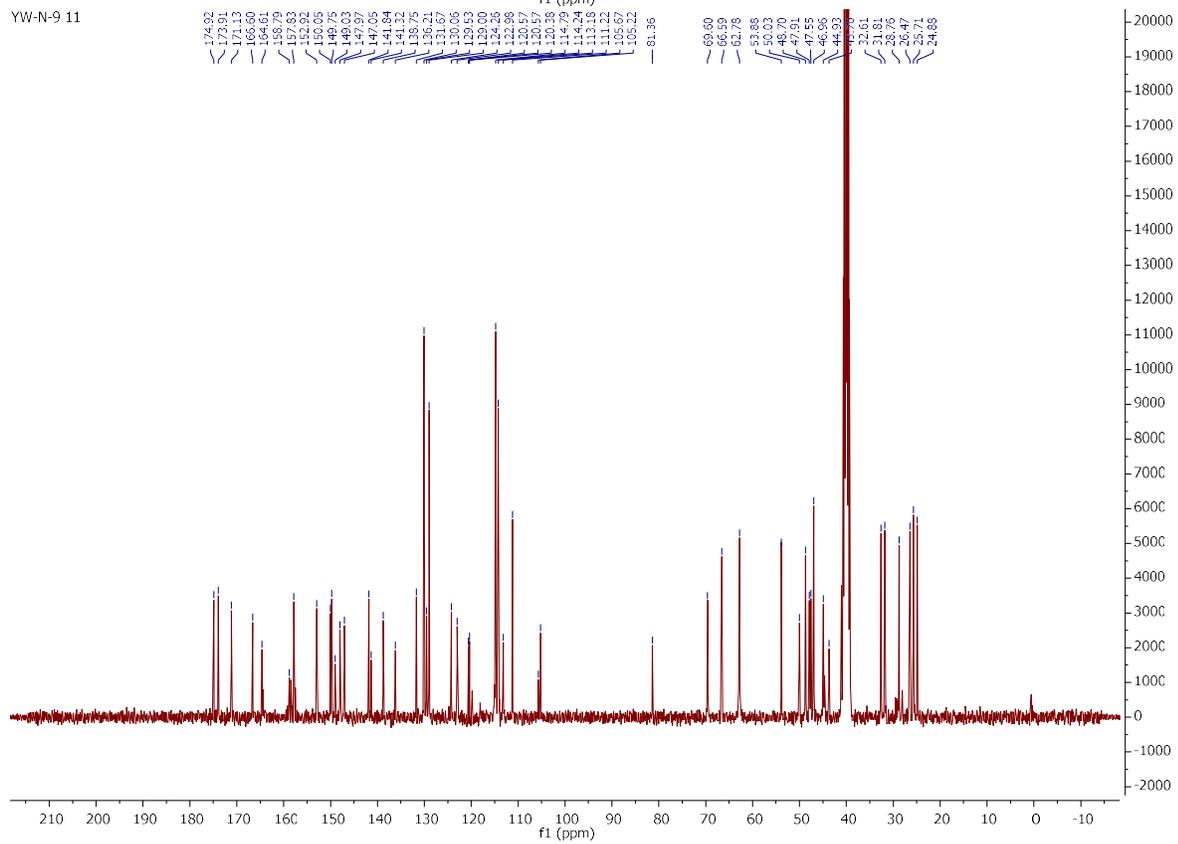
YW-N-8 10



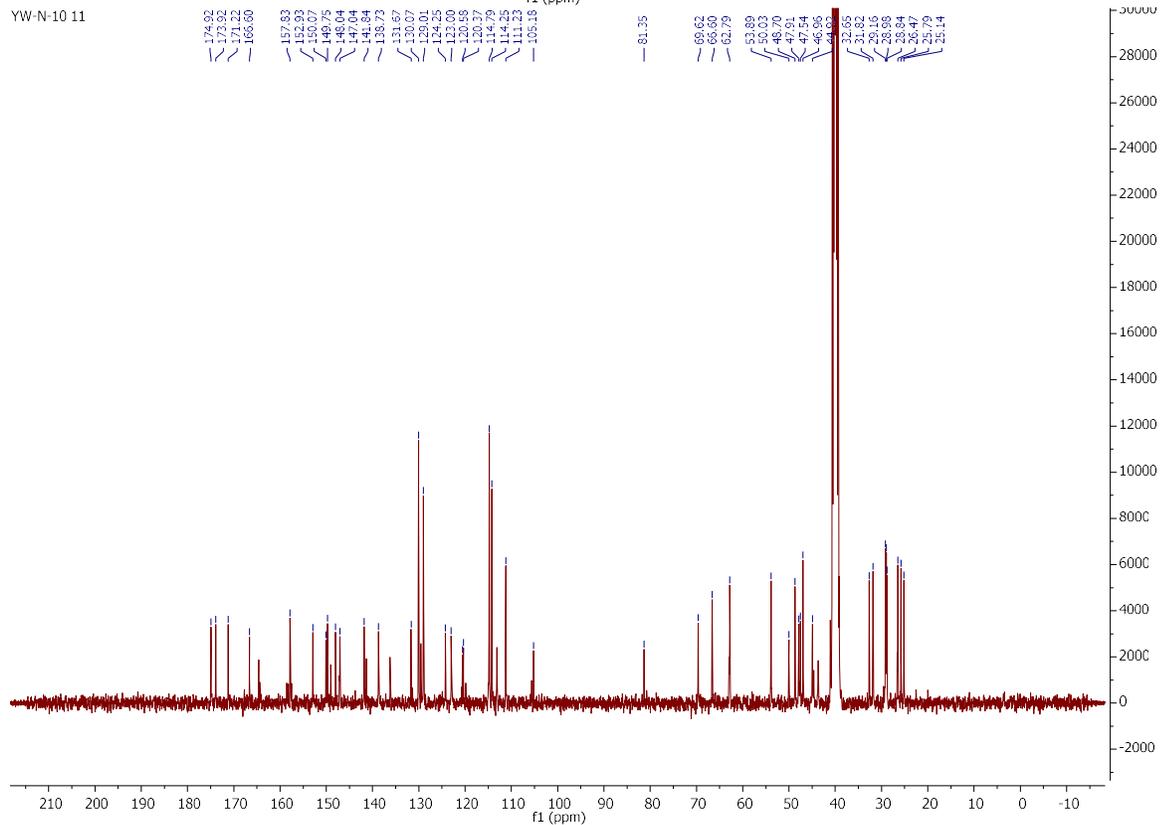
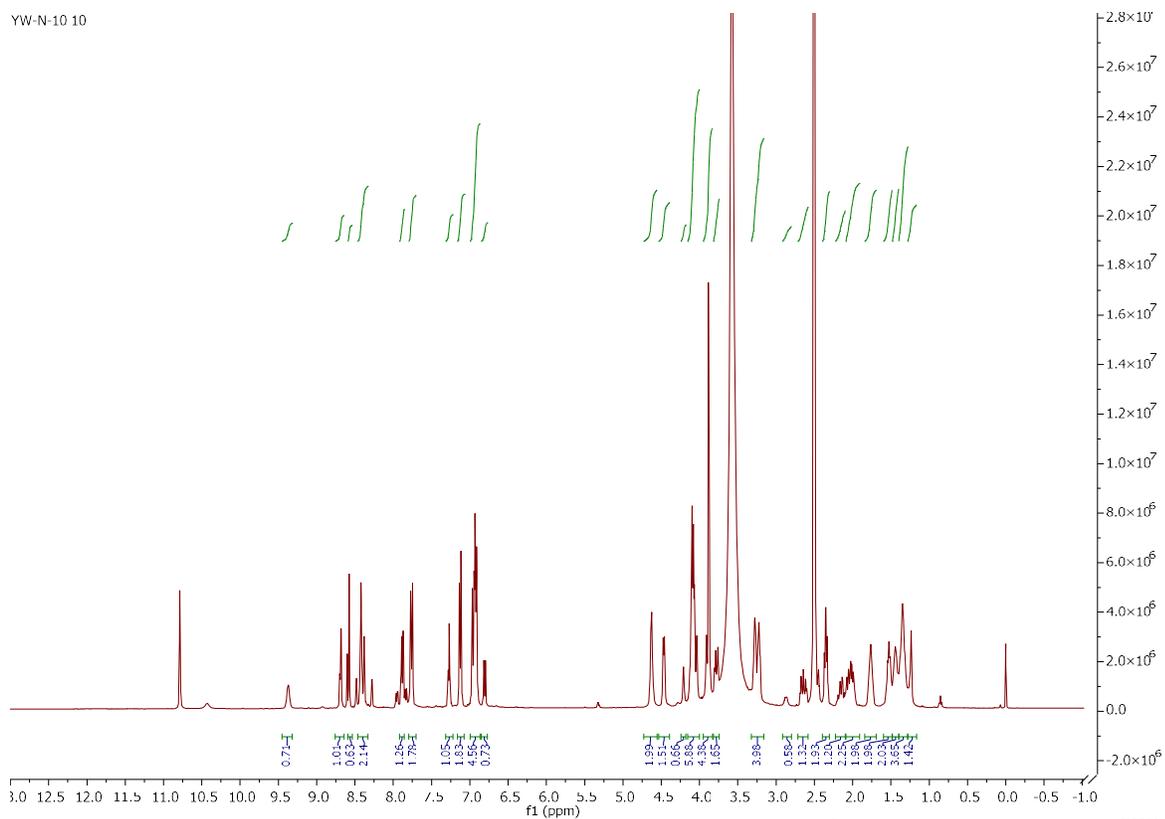
YW-N-9 10



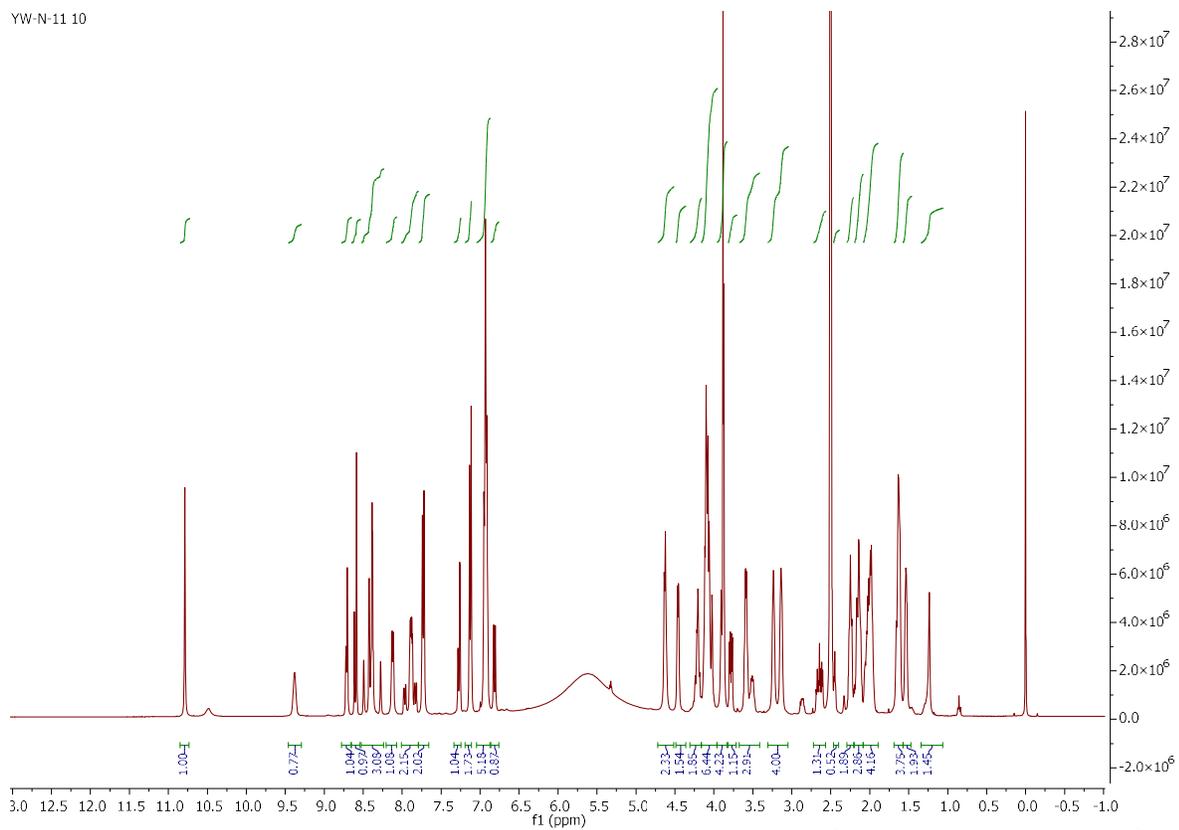
YW-N-9 11



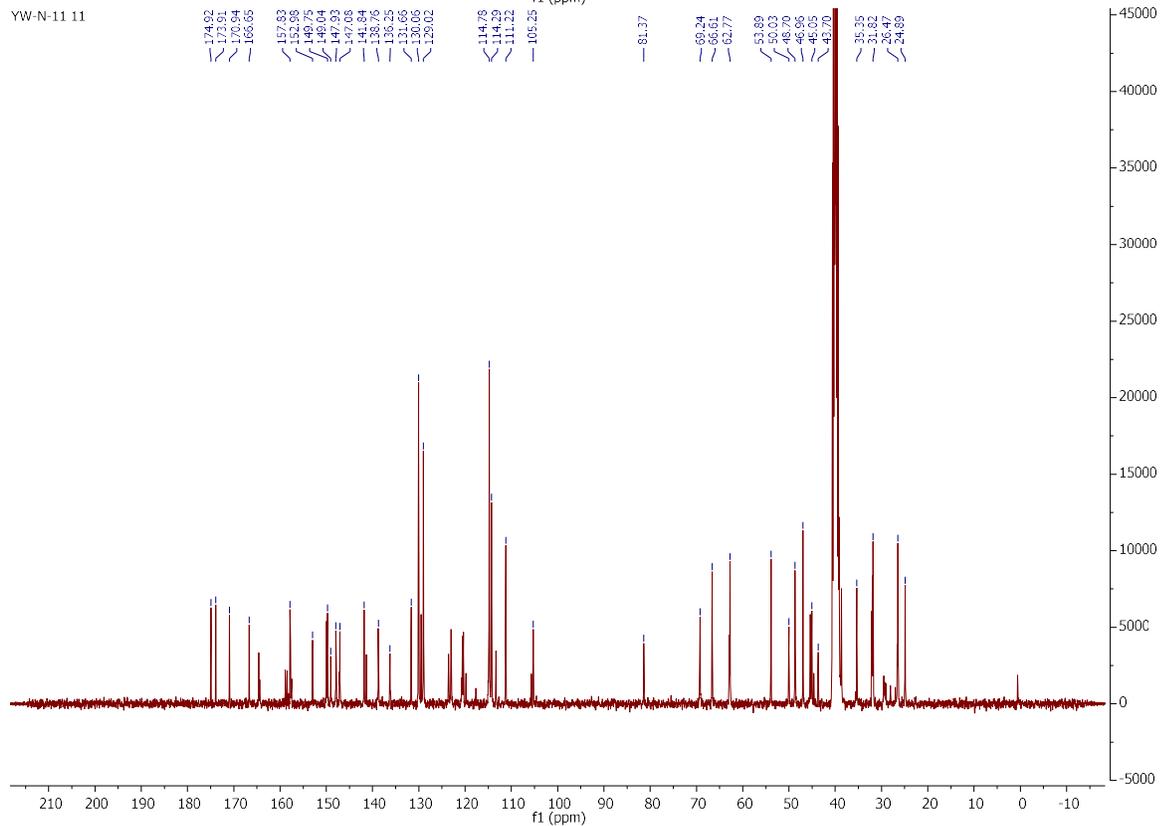
YW-N-10 10



YW-N-11 10



YW-N-11 11

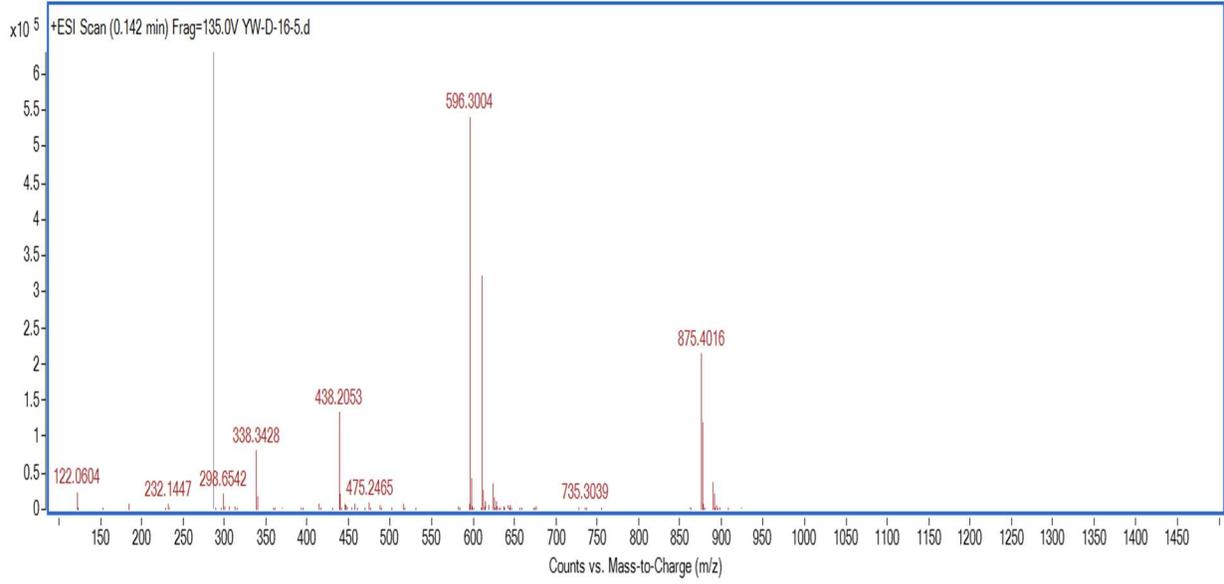


S1.2. HRMS data

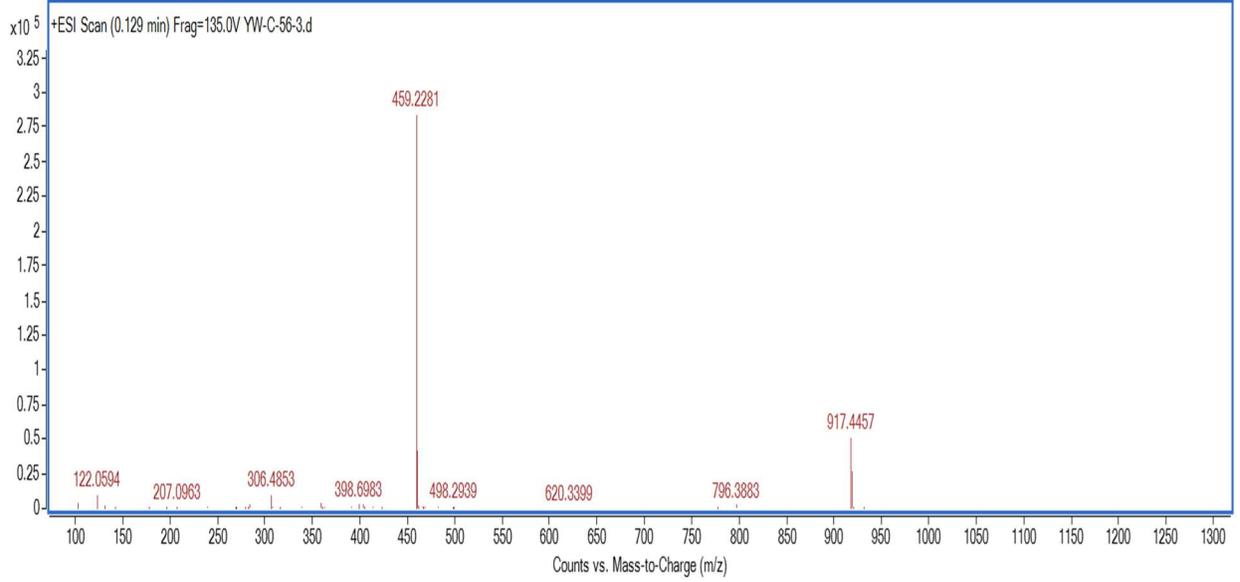
Table 1.2. HRMS data

Compound name	Exact Mass	HRMS
YW-L-1	874.3915	875.4016
YW-L-2	916.4384	917.4439
YW-L-3	958.4854	959.4893
YW-L-4	1004.4909	1005.4914
YW-L-5	936.3919	937.3947
YW-L-6	980.4181	981.4232
YW-L-7	1024.4443	1025.4492
YW-L-8	1034.4551	1035.4603
YW-L-9	1062.4864	1063.4966
YW-L-10	1088.5021	1089.5102
YW-L-11	1112.5021	1113.5224
YW-L-12	1140.5334	1141.4944
YW-N-1	769.3336	770.3434
YW-N-2	797.3649	798.3754
YW-N-3	825.3962	826.4066
YW-N-4	867.4432	868.4533
YW-N-5	958.4238	959.4336
YW-N-6	986.4551	987.4647
YW-N-7	1014.4864	1015.4968
YW-N-8	957.4286	958.4385
YW-N-9	985.4599	986.4700
YW-N-10	1013.4912	1014.5008
YW-N-11	1011.4755	1012.4851

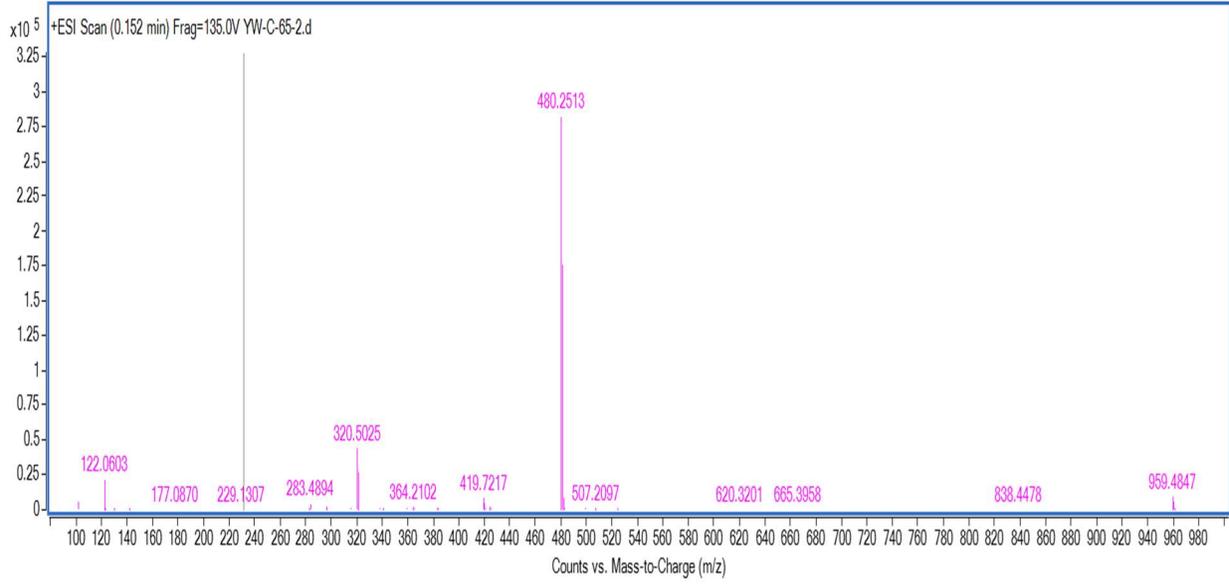
YW-L-1



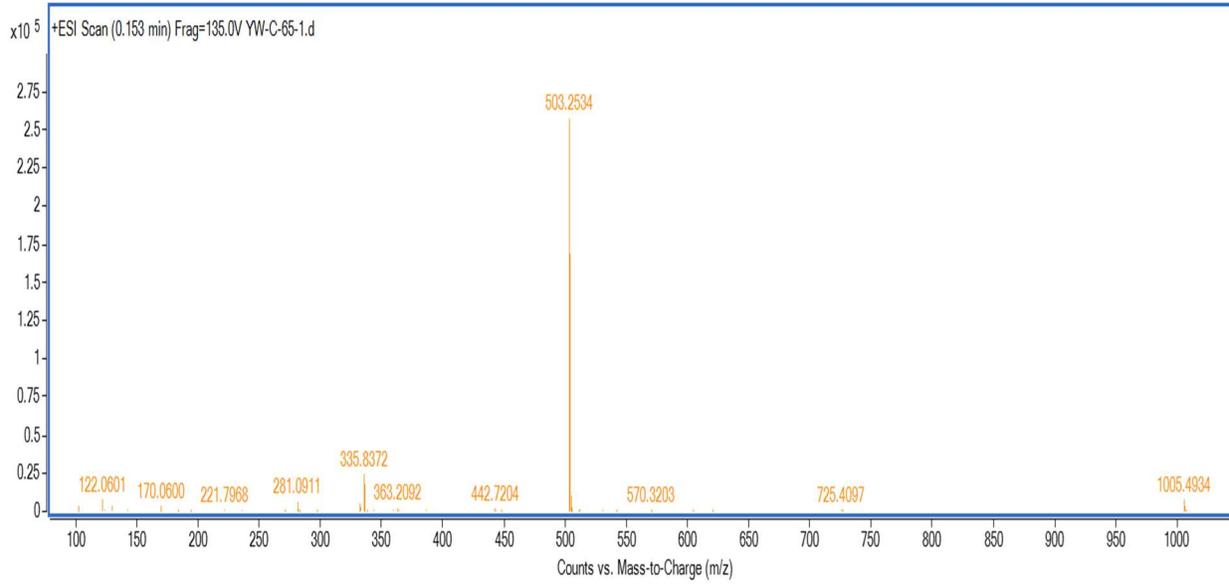
YW-L-2



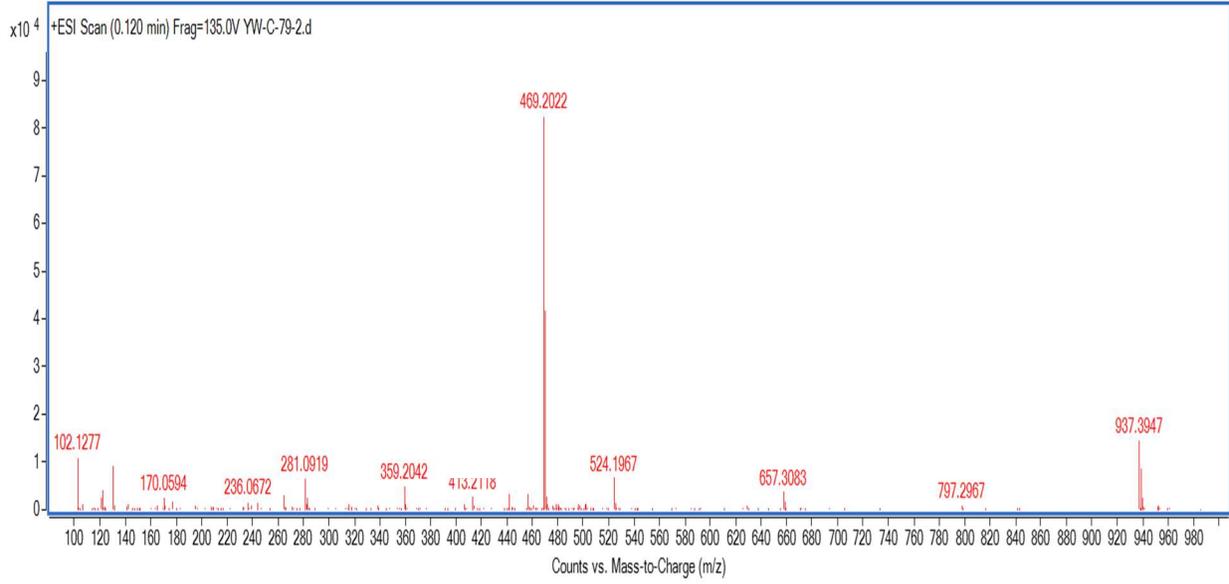
YW-L-3



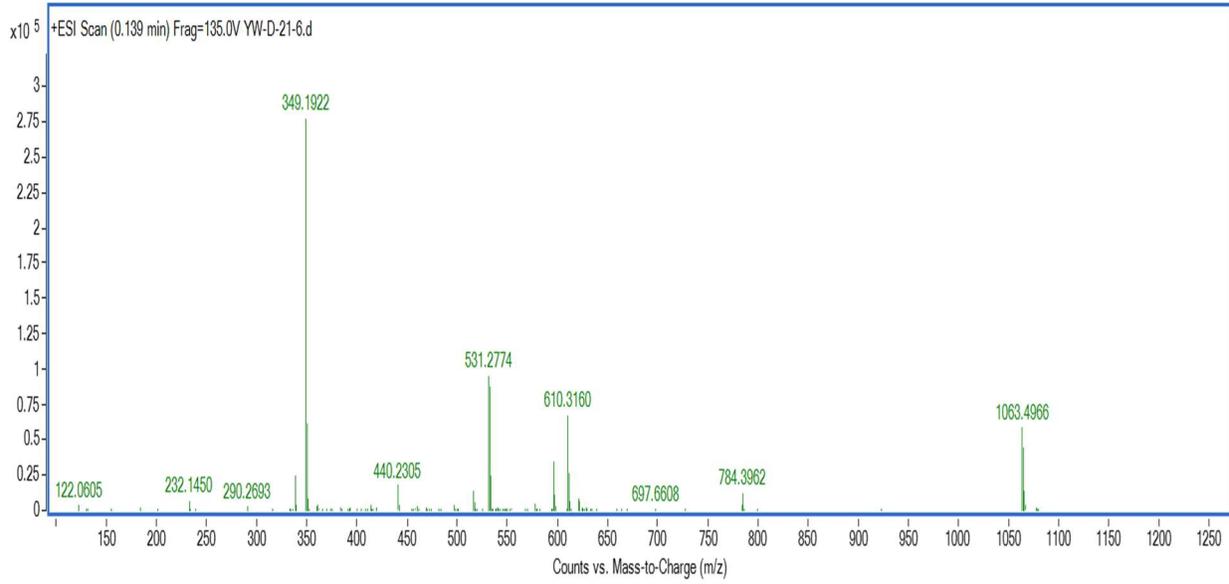
YW-L-4



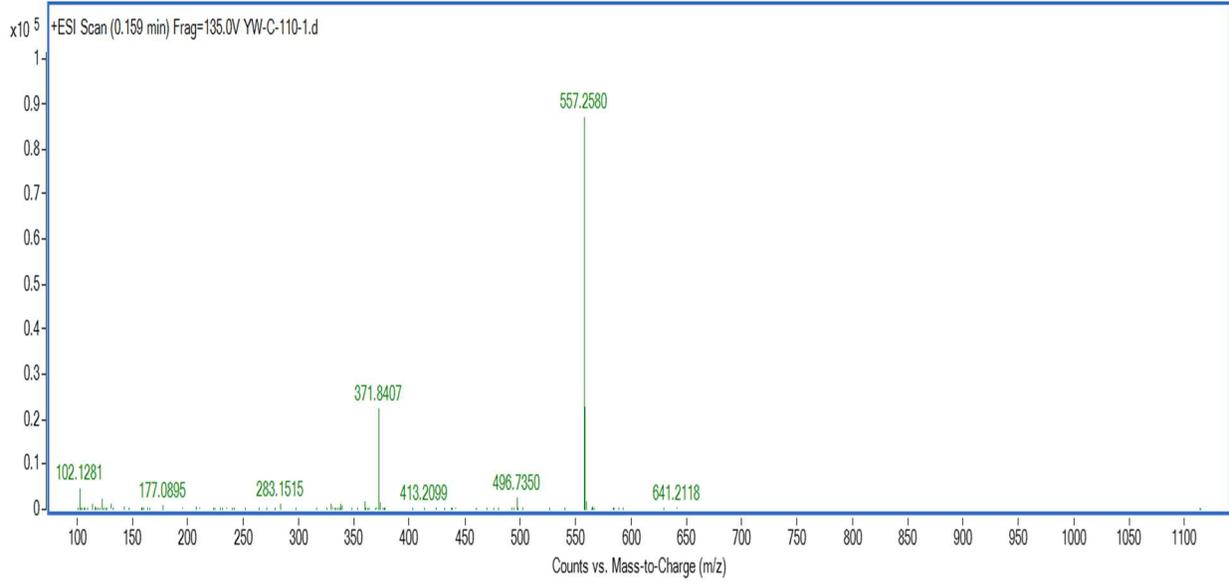
YW-L-5



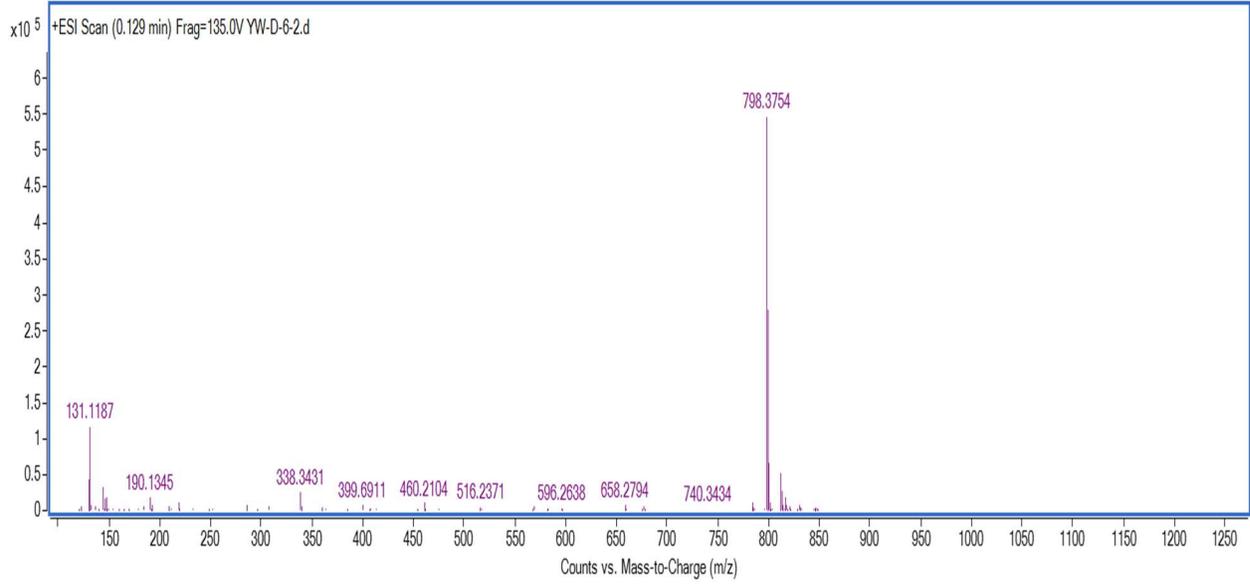
YW-L-9



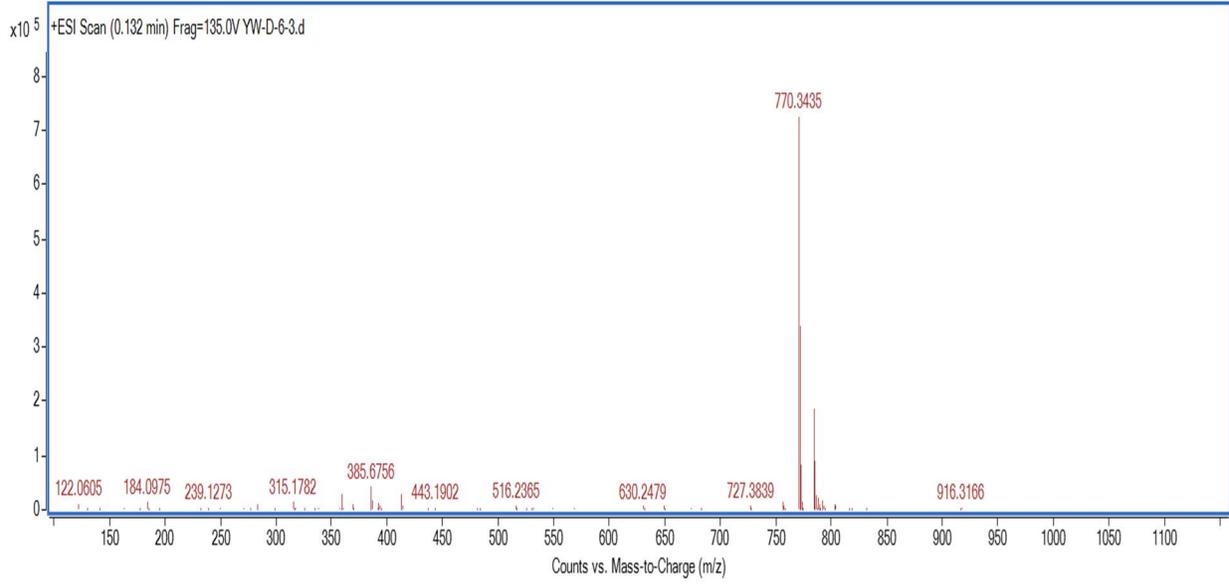
YW-L-11



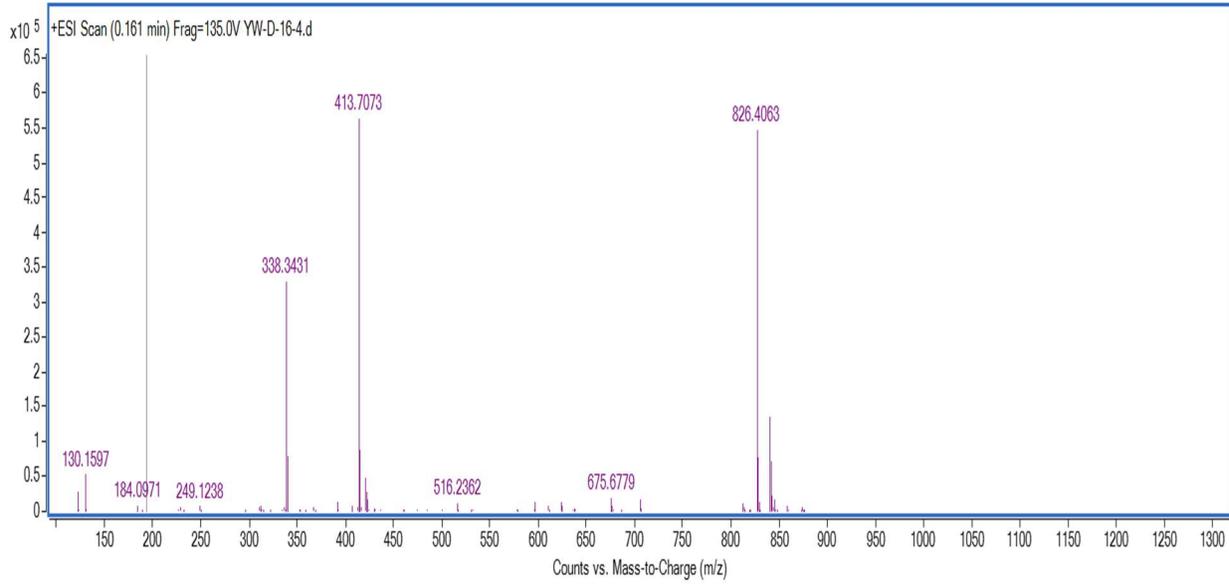
YW-N-1



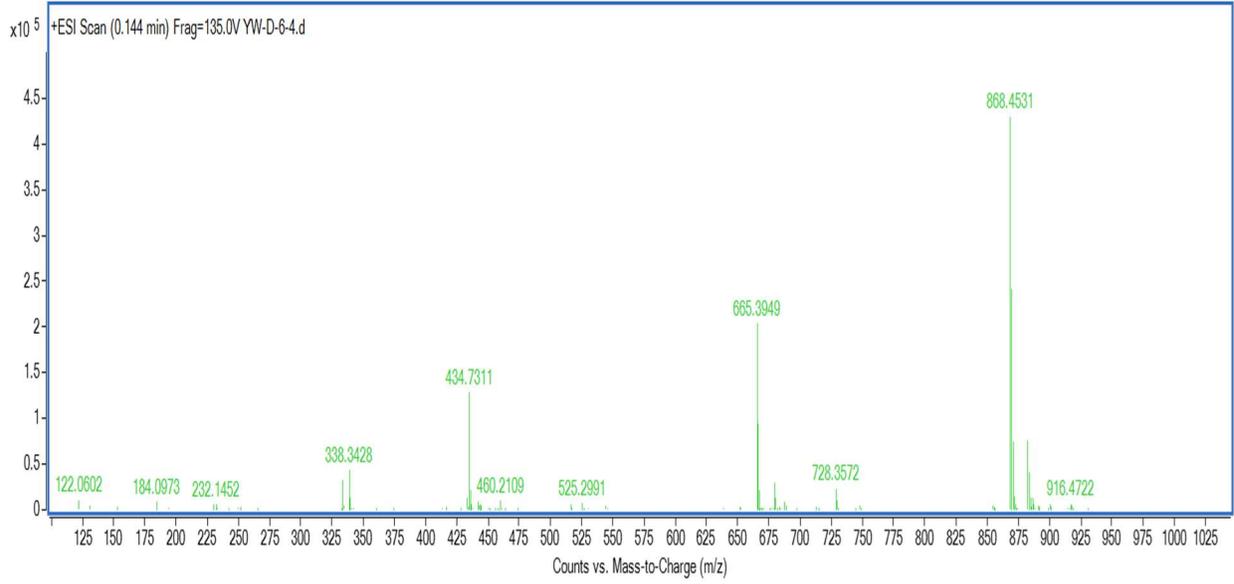
YW-N-2



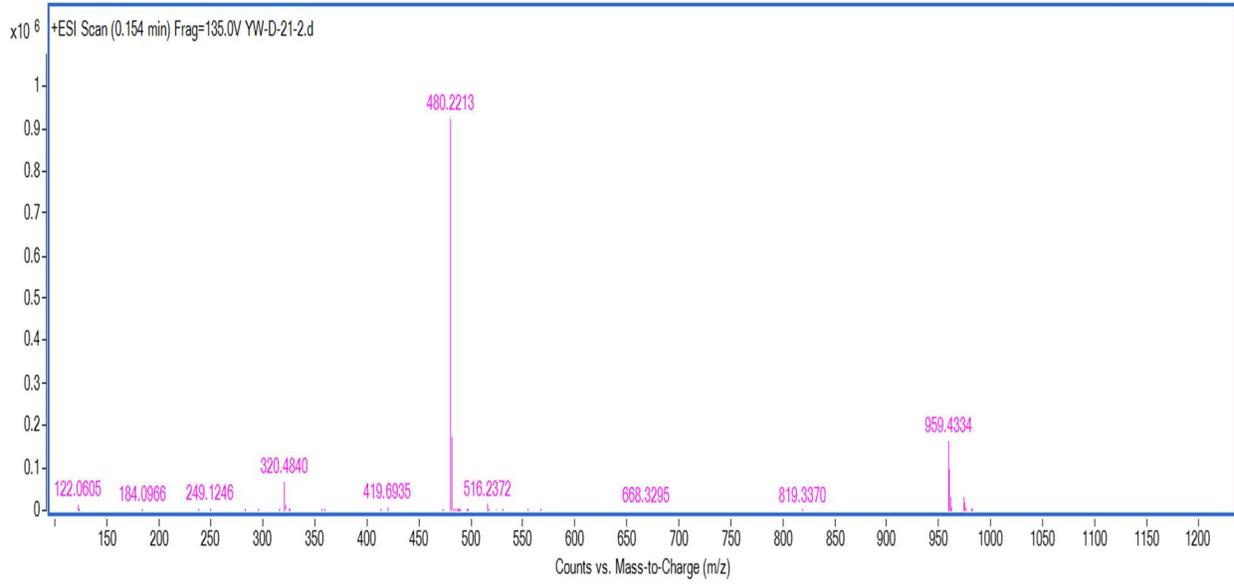
YW-N-3



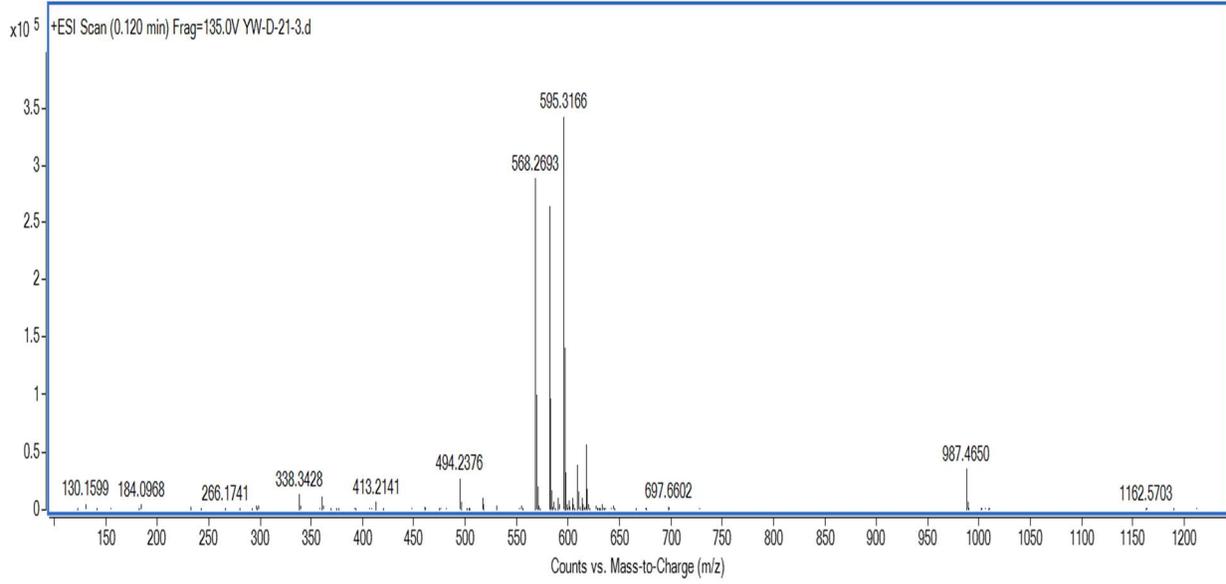
YW-N-4



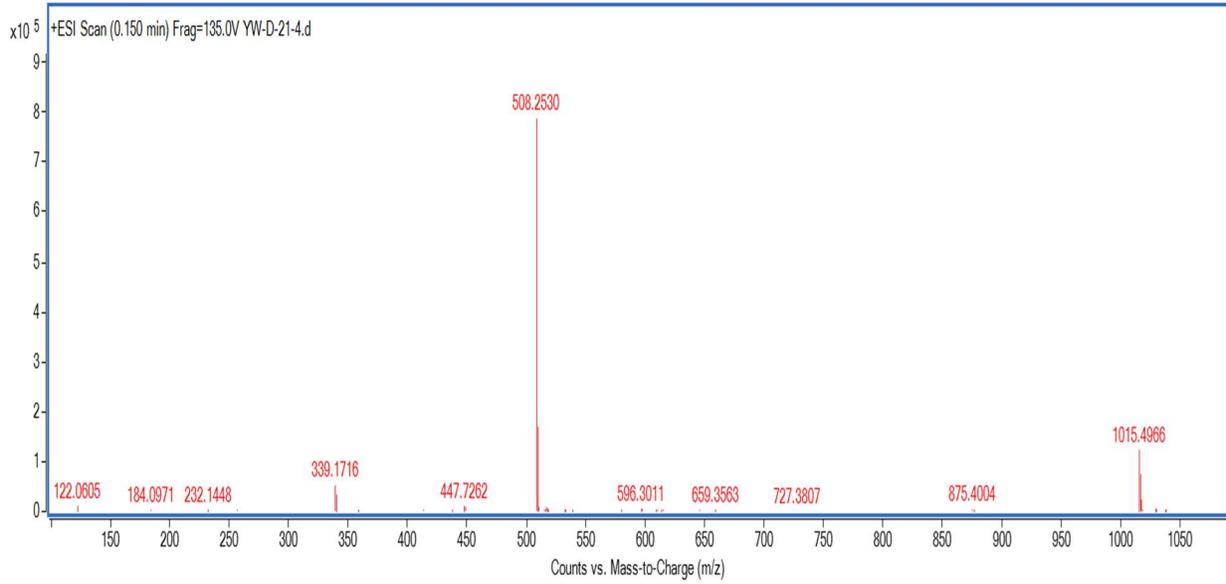
YW-N-5



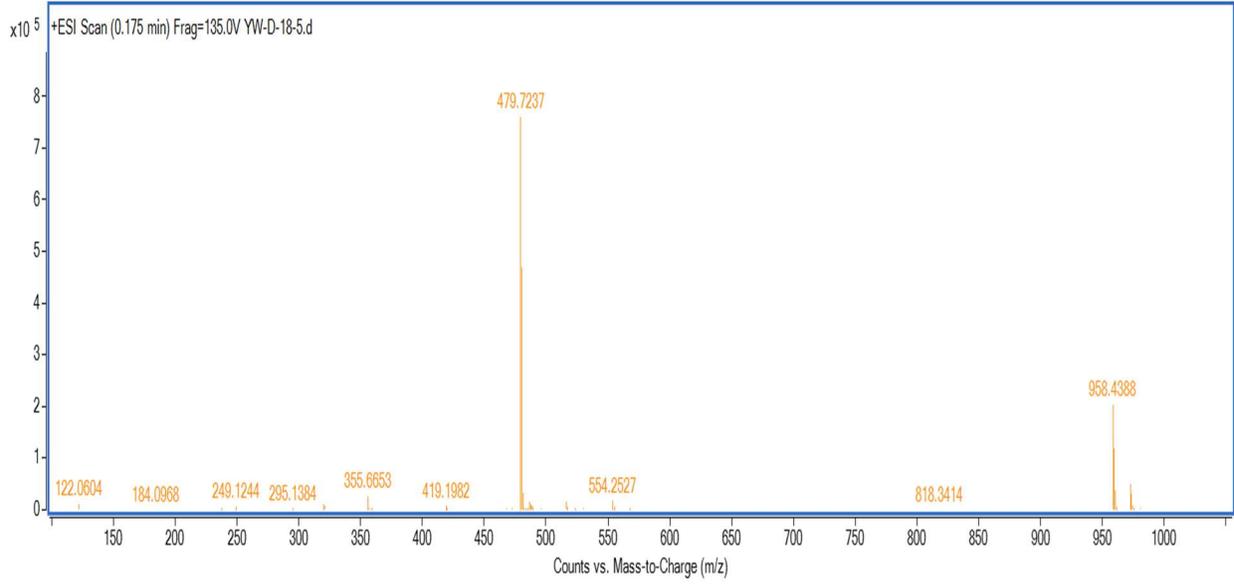
YW-N-6



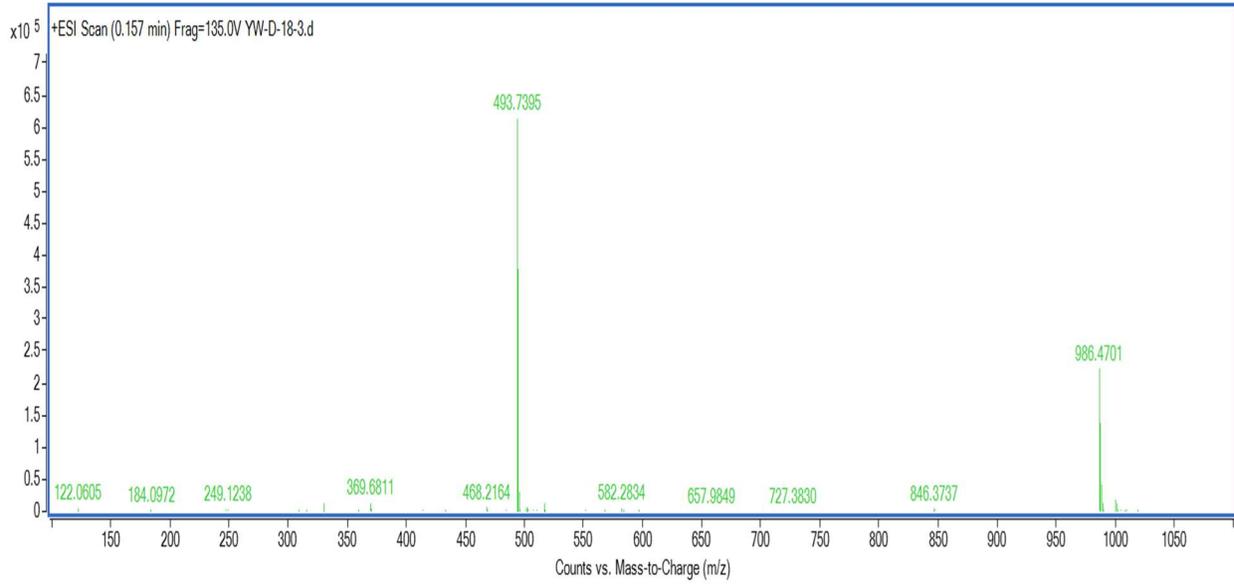
YW-N-7



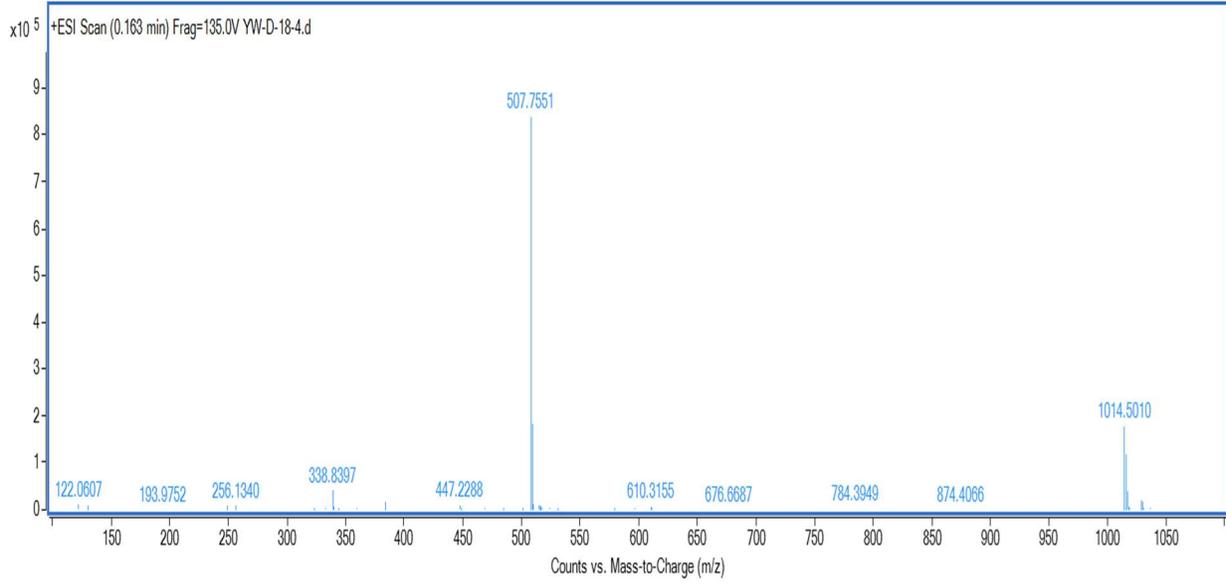
YW-N-8



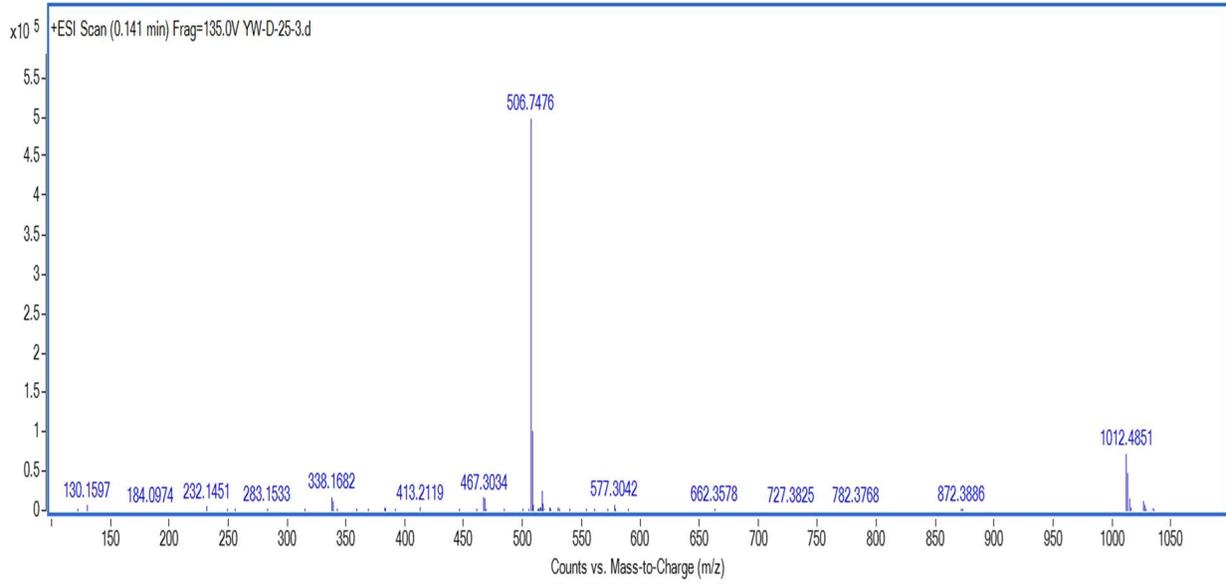
YW-N-9



YW-N-10



YW-N-11



YW-T-1

^1H NMR (400 MHz, DMSO- d_6) δ 9.41 (s, 1H), 8.64 (s, 1H), 8.56 (s, 1H), 8.40 (d, $J = 7.5$ Hz, 1H), 8.13 (s, 1H), 7.88 (t, $J = 7.5$ Hz, 2H), 7.77 (t, $J = 7.9$ Hz, 1H), 7.52 (d, $J = 8.5$ Hz, 1H), 7.42 (d, $J = 7.2$ Hz, 1H), 7.26 (s, 1H), 6.93 (d, $J = 8.4$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 5.08 (d, $J = 7.5$ Hz, 2H), 4.63 (s, 3H), 4.48 (d, $J = 6.1$ Hz, 2H), 4.24 (d, $J = 12.6$ Hz, 3H), 4.07 (d, $J = 23.7$ Hz, 5H), 3.92 (s, 1H), 3.88 (d, $J = 4.3$ Hz, 4H), 3.49 (s, 3H), 2.88 (d, $J = 7.3$ Hz, 2H), 2.60 (s, 2H), 2.30 (s, 2H), 2.00 (s, 4H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 173.25, 172.40, 170.39, 167.23, 165.71, 164.62, 157.73, 156.16, 149.90, 149.75, 149.03, 147.90, 147.03, 141.84, 138.76, 137.43, 136.20, 136.10, 133.69, 129.48, 122.98, 120.49, 116.91, 115.93, 113.24, 111.22, 105.25, 81.36, 69.06, 67.95, 62.77, 53.88, 50.46, 50.03, 49.21, 48.71, 38.44, 31.88, 31.41, 26.60, 24.89, 22.47.

YW-T-2

^1H NMR (400 MHz, Chloroform- d) δ 10.50 (s, 1H), 9.39 (s, 1H), 8.67 (s, 1H), 8.57 (s, 1H), 8.40 (d, $J = 14.9$ Hz, 1H), 8.00 (t, $J = 5.3$ Hz, 1H), 7.88 (d, $J = 8.6$ Hz, 1H), 7.83 – 7.75 (m, 1H), 7.54 (d, $J = 8.5$ Hz, 1H), 7.45 (d, $J = 7.2$ Hz, 1H), 7.26 (s, 1H), 6.93 (d, $J = 8.3$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 5.07 (d, $J = 12.8$ Hz, 1H), 4.63 (s, 4H), 4.46 (s, 3H), 4.26 – 4.18 (m, 3H), 4.15 – 4.00 (m, 5H), 3.91 (s, 1H), 3.87 (d, $J = 4.4$ Hz, 3H), 3.43 (d, $J = 5.6$ Hz, 2H), 2.89 (s, 1H), 2.58 (d, $J = 17.6$ Hz, 1H), 2.06 (t, $J = 7.3$ Hz, 3H), 1.74 (s, 2H), 1.44 (d, $J = 23.5$ Hz, 4H), 1.35 – 1.27 (m, 2H), 1.22 (s, 7H).

^{13}C NMR (101 MHz, Chloroform- d) δ 177.99, 177.89, 175.12, 172.01, 169.37, 162.48, 160.98, 154.81, 154.50, 153.79, 152.68, 151.78, 146.59, 143.54, 142.23, 138.49, 134.25, 127.76, 125.48,

125.33, 121.70, 120.74, 117.88, 115.97, 110.00, 86.09, 74.38, 72.69, 67.67, 58.63, 54.78, 53.97, 53.46, 48.46, 43.14, 40.47, 36.16, 34.16, 34.07, 33.90, 33.83, 33.64, 31.35, 30.63, 30.39, 27.24.

YW-T-3

^1H NMR (400 MHz, DMSO- d_6) δ 9.38 (s, 1H), 8.66 (d, $J = 5.2$ Hz, 1H), 8.59 (d, $J = 10.6$ Hz, 1H), 8.40 (d, $J = 11.6$ Hz, 1H), 8.07 (t, $J = 5.0$ Hz, 1H), 7.88 (d, $J = 8.6$ Hz, 1H), 7.56 (t, $J = 7.8$ Hz, 1H), 7.26 (d, $J = 8.2$ Hz, 1H), 7.16 (d, $J = 8.6$ Hz, 1H), 7.00 (d, $J = 7.0$ Hz, 1H), 6.93 (d, $J = 8.5$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 6.71 (s, 1H), 5.04 (dd, $J = 12.8, 5.3$ Hz, 1H), 4.64 (d, $J = 5.2$ Hz, 3H), 4.47 (d, $J = 6.1$ Hz, 2H), 4.08 (dd, $J = 15.9, 10.2$ Hz, 6H), 3.60 – 3.46 (m, 1H), 3.45 – 3.32 (m, 2H), 3.25 (d, $J = 5.6$ Hz, 2H), 2.87 (t, $J = 12.9$ Hz, 2H), 2.58 (s, 1H), 2.53 (s, 1H), 2.15 (s, 2H), 2.04 (dd, $J = 28.2, 9.6$ Hz, 2H), 1.70 (d, $J = 8.8$ Hz, 5H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 173.26, 173.12 (d, $J = 29.5$ Hz), 170.55, 169.19, 167.75, 164.62, 157.77, 150.00, 149.88 (d, $J = 24.9$ Hz), 149.05, 147.99, 147.05, 146.94 (d, $J = 22.5$ Hz), 138.74, 136.64, 136.21, 132.66, 129.52, 122.99, 117.61, 113.09, 111.11 (d, $J = 23.6$ Hz), 105.21, 81.36, 69.33, 62.95, 62.79, 53.89, 50.03, 48.85 (d, $J = 28.5$ Hz), 41.96, 38.47, 35.25, 31.44, 28.34, 26.60, 22.40 (d, $J = 45.6$ Hz).

YW-T-4

^1H NMR (400 MHz, DMSO- d_6) δ 9.39 (s, 1H), 8.66 (s, 1H), 8.57 (s, 1H), 8.40 (d, $J = 10.0$ Hz, 1H), 7.88 (d, $J = 8.6$ Hz, 1H), 7.78 (t, $J = 4.9$ Hz, 1H), 7.56 (t, $J = 7.8$ Hz, 1H), 7.26 (s, 1H), 7.03 (dd, $J = 25.5, 7.8$ Hz, 2H), 6.93 (d, $J = 8.3$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 6.51 (s, 1H), 5.04 (dd, $J = 12.8, 5.3$ Hz, 1H), 4.63 (s, 2H), 4.47 (d, $J = 6.1$ Hz, 2H), 4.07 (d, $J = 24.0$ Hz, 4H), 3.97 – 3.81 (m, 4H), 3.58 – 3.45 (m, 1H), 3.27 (s, 2H), 3.03 (d, $J = 6.0$ Hz, 2H), 2.88 (t, $J = 12.9$ Hz, 1H), 2.58 (d, $J = 17.6$ Hz, 1H), 2.21 – 1.94 (m, 4H), 1.80 – 1.61 (m, 4H), 1.60 – 1.49 (m, 2H), 1.45 – 1.19 (m, 5H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 173.29, 172.11, 170.57, 169.42, 164.62, 157.75, 150.01, 149.75, 147.96, 146.86, 141.84, 138.74, 136.73, 136.20, 132.65, 129.52, 122.99, 120.37, 117.59, 114.92, 113.12, 111.22, 110.84, 109.47, 105.22, 81.36, 69.36, 62.78, 53.88, 50.03, 49.01, 48.70, 42.24, 35.34, 31.45, 29.58, 29.09, 28.40, 26.60, 26.49, 22.62, 22.33.

YW-T-6

^1H NMR (400 MHz, DMSO- d_6) δ 9.37 (s, 1H), 8.71 (d, $J = 4.4$ Hz, 1H), 8.59 (d, $J = 10.4$ Hz, 1H), 8.40 (d, $J = 13.9$ Hz, 1H), 8.07 (t, $J = 5.2$ Hz, 1H), 7.88 (dd, $J = 8.7, 2.2$ Hz, 1H), 7.63 – 7.51 (m, 1H), 7.31 (d, $J = 7.3$ Hz, 1H), 7.15 (d, $J = 8.6$ Hz, 1H), 7.02 (d, $J = 7.0$ Hz, 1H), 6.93 (d, $J = 8.4$ Hz, 1H), 6.81 (d, $J = 8.8$ Hz, 1H), 6.71 (s, 1H), 5.05 (dd, $J = 12.8, 5.2$ Hz, 1H), 4.63 (s, 2H), 4.47 (d, $J = 6.0$ Hz, 2H), 4.25 (s, 3H), 4.15 – 4.01 (m, 5H), 3.88 (d, $J = 4.5$ Hz, 5H), 3.79 (s, 4H), 3.61 – 3.40 (m, 16H), 3.40 – 3.30 (m, 2H), 3.23 (d, $J = 5.7$ Hz, 2H), 2.88 (t, $J = 13.0$ Hz, 2H), 2.69 – 2.55 (m, 2H), 2.29 (t, $J = 6.4$ Hz, 2H), 2.04 (dt, $J = 13.1, 6.2$ Hz, 2H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 173.28, 171.12, 170.55, 169.18, 167.76, 164.62, 157.79, 149.75, 149.05, 148.04, 147.09, 146.94 (d, $J = 29.7$ Hz), 141.84, 138.72, 136.67, 129.55, 122.97, 120.56, 117.61, 114.91, 113.32, 111.12 (d, $J = 20.5$ Hz), 81.39, 70.70 – 69.72 (m), 69.15, 67.18, 62.79, 53.89, 50.03, 48.85 (d, $J = 28.9$ Hz), 41.89, 38.50, 36.57, 31.45, 26.60, 22.63.

YW-V-1

^1H NMR (400 MHz, DMSO- d_6) δ 10.48 (s, 1H), 9.39 (s, 1H), 9.00 (s, 1H), 8.69 (s, 1H), 8.64 – 8.47 (m, 2H), 8.42 (d, $J = 14.7$ Hz, 1H), 8.02 – 7.81 (m, 3H), 7.41 (d, $J = 8.2$ Hz, 4H), 7.29 (s, 1H), 6.94 (d, $J = 8.2$ Hz, 1H), 6.83 (d, $J = 8.8$ Hz, 1H), 4.65 (s, 4H), 4.58 (d, $J = 8.8$ Hz, 3H), 4.37 (s, 2H), 4.23 (s, 3H), 4.18 – 3.99 (m, 6H), 3.91 (d, $J = 11.5$ Hz, 5H), 3.67 (s, 2H), 3.53 (s, 1H), 2.46 (s, 3H), 2.41 – 2.31 (m, 1H), 2.25 (s, 1H), 2.06 (d, $J = 9.9$ Hz, 2H), 1.92 (s, 1H), 1.73 (d, $J = 22.0$ Hz, 4H), 0.95 (s, 9H). ^{13}C NMR (101 MHz, DMSO- d_6) δ 172.41, 170.15, 164.63, 157.76, 151.95,

149.90 (d, $J = 28.3$ Hz), 147.99, 147.06, 141.85, 139.99, 129.53, 129.11, 127.89, 123.05, 113.17, 111.23, 105.23, 81.36, 69.34, 62.79, 59.17, 56.86, 53.89, 48.72, 42.12, 38.45, 35.28 (d, $J = 90.0$ Hz), 28.41, 26.86, 22.47, 16.41.

YW-V-2

^1H NMR (400 MHz, DMSO- d_6) δ 10.33 (s, 1H), 9.30 (s, 1H), 8.91 (s, 1H), 8.61 (s, 1H), 8.53 (s, 2H), 8.33 (d, $J = 13.5$ Hz, 2H), 8.21 (s, 1H), 7.82 (d, $J = 8.5$ Hz, 1H), 7.76 (d, $J = 9.2$ Hz, 1H), 7.33 (q, $J = 8.0$ Hz, 4H), 7.19 (s, 1H), 6.86 (d, $J = 8.5$ Hz, 1H), 6.74 (d, $J = 8.8$ Hz, 1H), 4.61 – 4.52 (m, 3H), 4.47 (d, $J = 9.3$ Hz, 2H), 4.39 (s, 5H), 4.35 (s, 3H), 4.28 (s, 5H), 4.16 (s, 7H), 4.06 – 3.98 (m, 7H), 3.97 (s, 2H), 3.84 (s, 1H), 3.81 (d, $J = 3.8$ Hz, 4H), 3.57 (s, 2H), 3.50 – 3.38 (m, 1H), 2.60 (s, 1H), 2.37 (s, 3H), 2.26 (s, 1H), 2.21 (s, 1H), 1.99 (s, 4H), 1.83 (s, 1H), 1.69 (s, 3H), 1.36 (s, 5H), 1.19 (s, 11H), 0.86 (s, 9H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 172.55, 172.41, 170.17, 165.01, 158.77, 158.42, 157.75, 151.94, 150.07, 149.75, 147.98, 147.04, 141.84, 141.36, 139.99, 138.76, 136.21, 131.64, 130.09, 129.52, 129.10, 127.89, 123.01, 113.13, 111.23, 105.22, 81.34, 69.63, 69.32, 62.95, 62.79, 59.16, 56.81, 56.73, 53.88, 50.03, 48.71, 43.70, 42.11, 38.43, 35.69, 35.32, 29.40, 29.21, 29.16, 29.12, 28.90, 26.84, 25.90, 16.40.

YW-V-3

^1H NMR (400 MHz, DMSO- d_6) δ 10.65 (s, 1H), 9.46 (s, 1H), 9.00 (d, $J = 2.9$ Hz, 1H), 8.74 (d, $J = 7.1$ Hz, 1H), 8.42 (d, $J = 13.4$ Hz, 1H), 7.91 (d, $J = 9.0$ Hz, 3H), 7.41 (q, $J = 7.9$ Hz, 4H), 7.33 (d, $J = 8.1$ Hz, 1H), 6.94 (q, $J = 7.8$ Hz, 1H), 6.84 (d, $J = 8.8$ Hz, 1H), 4.65 (d, $J = 6.4$ Hz, 2H), 4.56 (d, $J = 9.0$ Hz, 1H), 4.52 – 4.39 (m, 3H), 4.36 (s, 1H), 4.32 – 4.18 (m, 4H), 4.17 – 4.01 (m, 3H), 3.94 (s, 1H), 3.89 (d, $J = 3.9$ Hz, 3H), 3.81 (t, $J = 4.3$ Hz, 2H), 3.73 – 3.39 (m, 16H), 2.90 (q,

$J = 7.7, 6.5$ Hz, 1H), 2.45 (d, $J = 3.1$ Hz, 3H), 2.41 – 2.27 (m, 1H), 2.06 (q, $J = 9.5$ Hz, 2H), 1.91 (ddt, $J = 12.7, 8.1, 3.8$ Hz, 1H), 0.94 (s, 9H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 172.40, 170.42, 170.00, 164.62, 157.86, 151.97, 149.93, 149.76, 148.12, 147.83, 147.09, 141.86, 141.36, 140.00, 138.86, 131.67, 130.08, 129.49, 129.11, 127.90, 123.02, 120.38, 117.45, 114.92, 113.37, 111.22, 105.34, 81.40, 70.37, 70.26, 70.17, 69.94, 69.33, 69.17, 67.41, 62.76, 59.19, 56.77, 53.88, 50.03, 48.72, 43.74, 42.12, 38.42, 36.12, 35.83, 26.79, 16.38.

YW-V-4

^1H NMR (400 MHz, DMSO- d_6) δ 9.40 (s, 1H), 8.99 (d, $J = 2.5$ Hz, 1H), 8.73 (d, $J = 6.9$ Hz, 1H), 8.59 (d, $J = 7.4$ Hz, 1H), 8.41 (d, $J = 14.6$ Hz, 1H), 8.00 – 7.80 (m, 3H), 7.41 (q, $J = 8.0$ Hz, 4H), 7.32 (d, $J = 8.3$ Hz, 1H), 6.94 (d, $J = 8.5$ Hz, 1H), 6.83 (d, $J = 8.8$ Hz, 1H), 4.65 (d, $J = 5.5$ Hz, 2H), 4.56 (d, $J = 9.0$ Hz, 1H), 4.47 (t, $J = 7.1$ Hz, 2H), 4.43 (d, $J = 7.6$ Hz, 1H), 4.36 (s, 1H), 4.24 (d, $J = 6.0$ Hz, 4H), 4.16 – 4.01 (m, 3H), 3.89 (d, $J = 3.5$ Hz, 3H), 3.79 (t, $J = 4.4$ Hz, 2H), 3.72 – 3.42 (m, 9H), 2.45 (d, $J = 2.7$ Hz, 3H), 2.38 (dd, $J = 14.1, 7.0$ Hz, 1H), 2.06 (q, $J = 10.4, 9.4$ Hz, 2H), 1.91 (tt, $J = 9.7, 4.2$ Hz, 1H), 0.93 (s, 9H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 172.40, 170.41, 170.00, 164.62, 158.88, 157.72, 151.95, 149.93, 149.76, 149.05, 148.14, 147.92, 147.08, 141.84, 141.35, 139.98, 138.79, 136.29, 131.65, 130.09, 129.51, 129.10, 127.89, 123.00, 120.57, 120.38, 114.92, 113.31, 111.22, 105.28, 81.40, 70.34, 69.98, 69.33, 69.17, 67.45, 62.92, 62.78, 59.19, 56.85, 56.75, 53.88, 50.03, 48.72, 43.72, 42.12, 38.42, 36.12, 35.84, 26.79, 16.39.

W-2-3 C₄

¹H NMR (400 MHz, DMSO-*d*₆) δ 12.24 (s, 1H), 9.49 (s, 1H), 8.70 (s, 1H), 8.59 (s, 1H), 8.43 (t, *J* = 17.2 Hz, 2H), 7.89 (d, *J* = 7.9 Hz, 2H), 7.29 (s, 1H), 6.91 (s, 1H), 6.82 (d, *J* = 8.4 Hz, 1H), 4.56 (d, *J* = 68.6 Hz, 3H), 4.20 – 4.10 (m, 2H), 4.06 (d, *J* = 11.5 Hz, 2H), 3.90 (d, *J* = 16.5 Hz, 4H), 3.45 (d, *J* = 56.9 Hz, 6H), 3.17 (s, 1H), 2.43 (t, *J* = 7.1 Hz, 2H), 2.12 – 2.04 (m, 1H), 2.04 – 1.94 (m, 2H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 174.48, 164.61, 157.82, 149.90, 147.58 (d, *J* = 96.9 Hz), 141.83, 136.25, 129.61, 122.95, 120.55, 114.95, 113.26, 111.19, 105.17, 81.38, 68.83, 62.77, 53.86, 50.01, 48.67, 43.75, 30.46, 26.60, 24.41.

W-2-3 C₆

¹H NMR (400 MHz, DMSO-*d*₆) δ 12.04 (s, 1H), 9.45 (s, 1H), 8.69 (s, 1H), 8.59 (s, 1H), 8.41 (d, *J* = 18.8 Hz, 2H), 8.19 (s, 1H), 7.89 (d, *J* = 8.8 Hz, 2H), 7.81 (d, *J* = 10.7 Hz, 1H), 7.28 (s, 1H), 6.98 – 6.76 (m, 2H), 4.56 (d, *J* = 68.0 Hz, 3H), 4.10 (t, *J* = 6.4 Hz, 3H), 3.92 (s, 2H), 3.87 (s, 5H), 3.52 (s, 1H), 2.25 (t, *J* = 7.3 Hz, 2H), 2.07 (d, *J* = 10.3 Hz, 1H), 1.77 (p, *J* = 6.9 Hz, 2H), 1.59 (p, *J* = 7.3 Hz, 2H), 1.45 (p, *J* = 7.8 Hz, 3H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 174.90, 157.87, 153.16 – 146.68 (m), 138.66, 136.21, 129.59, 122.91, 121.74 (d, *J* = 236.3 Hz), 114.99, 113.16, 111.16, 105.17, 81.35, 69.55, 62.77, 53.81, 50.08, 48.68, 34.08, 28.64, 25.52, 24.66.

W-2-3 C₈

¹H NMR (400 MHz, DMSO-*d*₆) δ 12.03 (s, 1H), 9.50 (s, 1H), 8.68 (s, 1H), 8.59 (s, 1H), 8.41 (q, *J* = 13.4, 11.7 Hz, 2H), 7.89 (d, *J* = 8.7 Hz, 2H), 7.28 (s, 1H), 6.99 – 6.76 (m, 2H), 4.56 (d, *J* = 69.1 Hz, 3H), 4.18 – 3.98 (m, 5H), 3.97 – 3.78 (m, 6H), 3.53 (s, 1H), 2.21 (t, *J* = 7.3 Hz, 2H), 1.76 (p, *J* = 6.7 Hz, 2H), 1.51 (p, *J* = 7.5 Hz, 2H), 1.43 (q, *J* = 7.2 Hz, 2H), 1.39 – 1.23 (m, *J* = 6.9 Hz, 5H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 174.96, 164.60, 157.83, 150.08, 148.06, 147.07, 141.84, 138.68, 129.60, 122.94, 120.53, 14.99, 113.09 (t, $J = 191.9$ Hz), 105.18, 81.35, 69.61, 62.77, 53.83, 50.01, 48.65, 34.11, 28.90 (d, $J = 9.7$ Hz), 25.76, 24.90.

L-2-2

^1H NMR (400 MHz, DMSO- d_6) δ 12.54 (s, 1H), 8.64 (s, 1H), 7.95 (d, $J = 8.9$ Hz, 1H), 6.86 (d, $J = 9.0$ Hz, 1H), 3.71 – 3.54 (m, 4H), 3.51 – 3.37 (m, 4H), 1.43 (s, 9H).

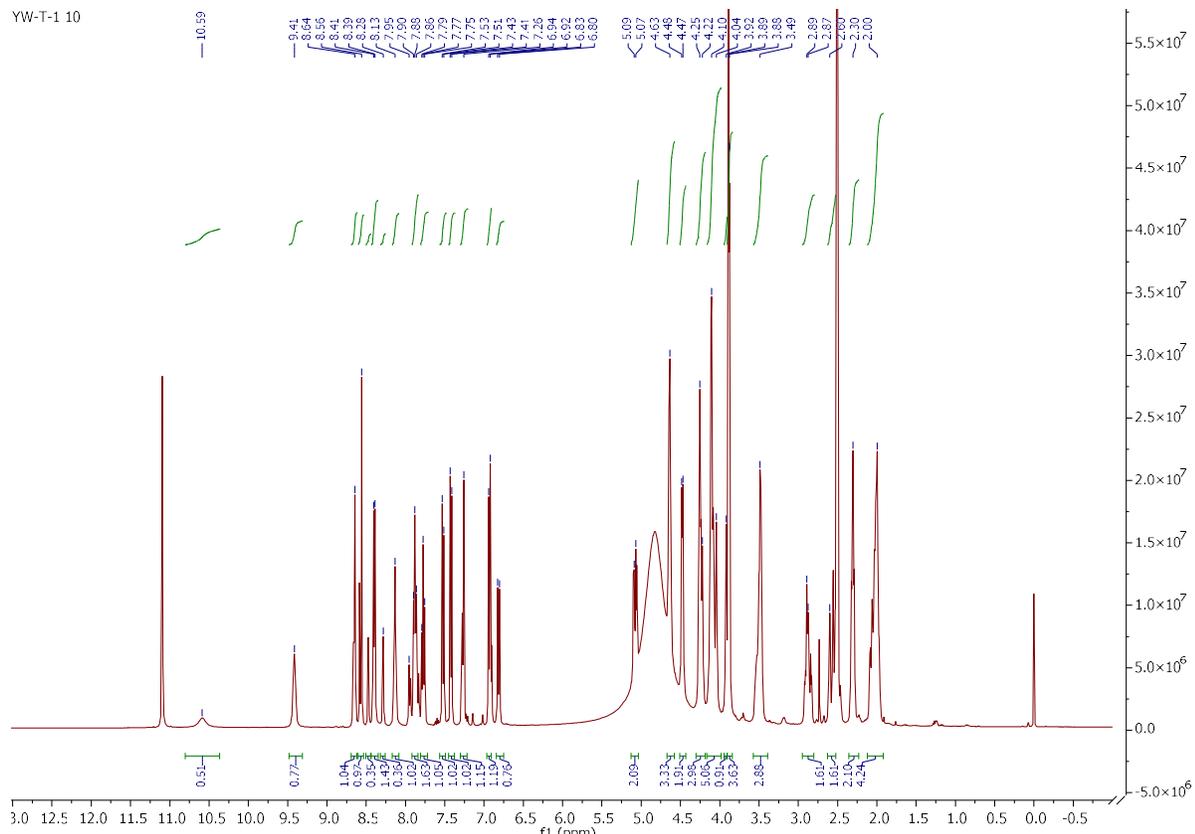
^{13}C NMR (101 MHz, DMSO- d_6) δ 167.03, 160.60, 154.37, 150.79, 138.82, 115.44, 106.22, 79.58, 44.33, 28.53.

L-3-2

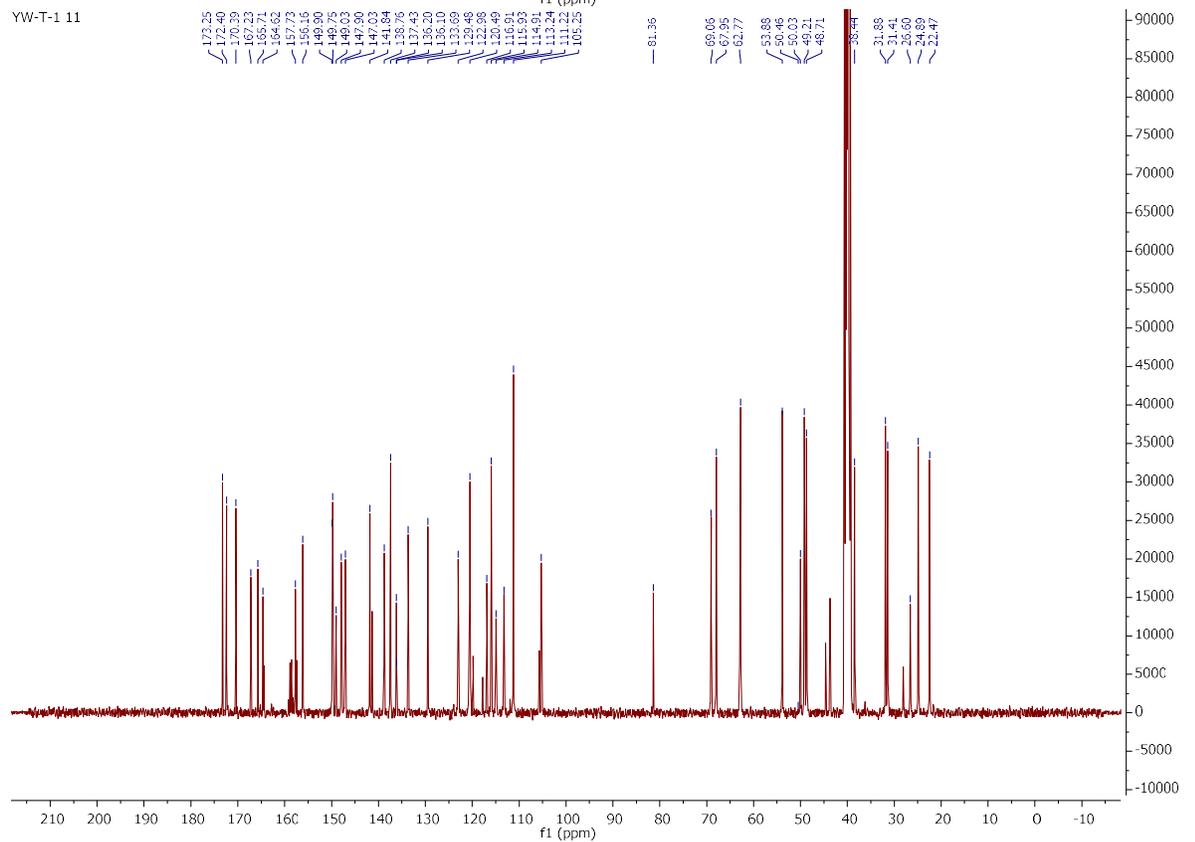
^1H NMR (400 MHz, DMSO- d_6) δ 12.44 (s, 1H), 7.78 (d, $J = 8.4$ Hz, 2H), 6.96 (d, $J = 8.6$ Hz, 2H), 3.45 (t, $J = 5.2$ Hz, 4H), 1.42 (s, 9H).

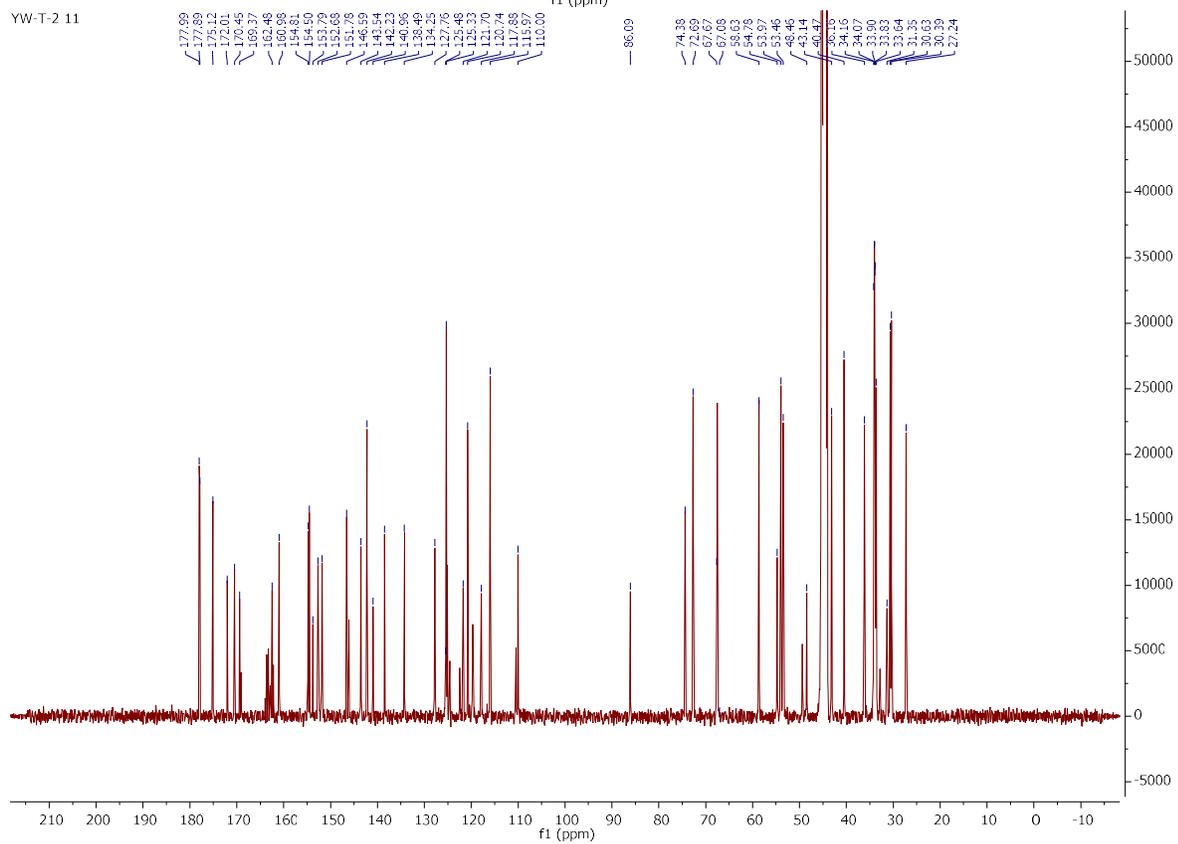
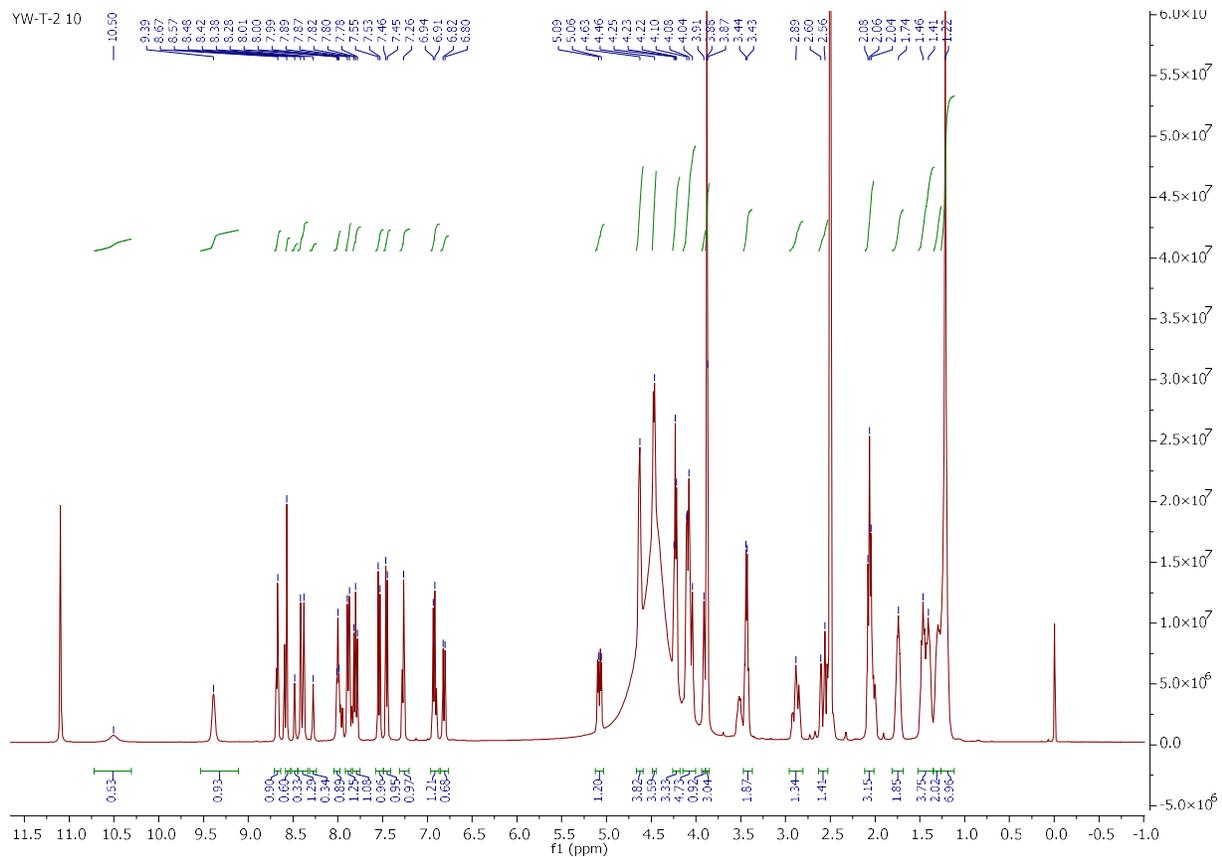
^{13}C NMR (101 MHz, DMSO- d_6) δ 167.70, 154.33, 153.98, 131.33, 120.22, 114.10, 79.54, 47.11, 28.53.

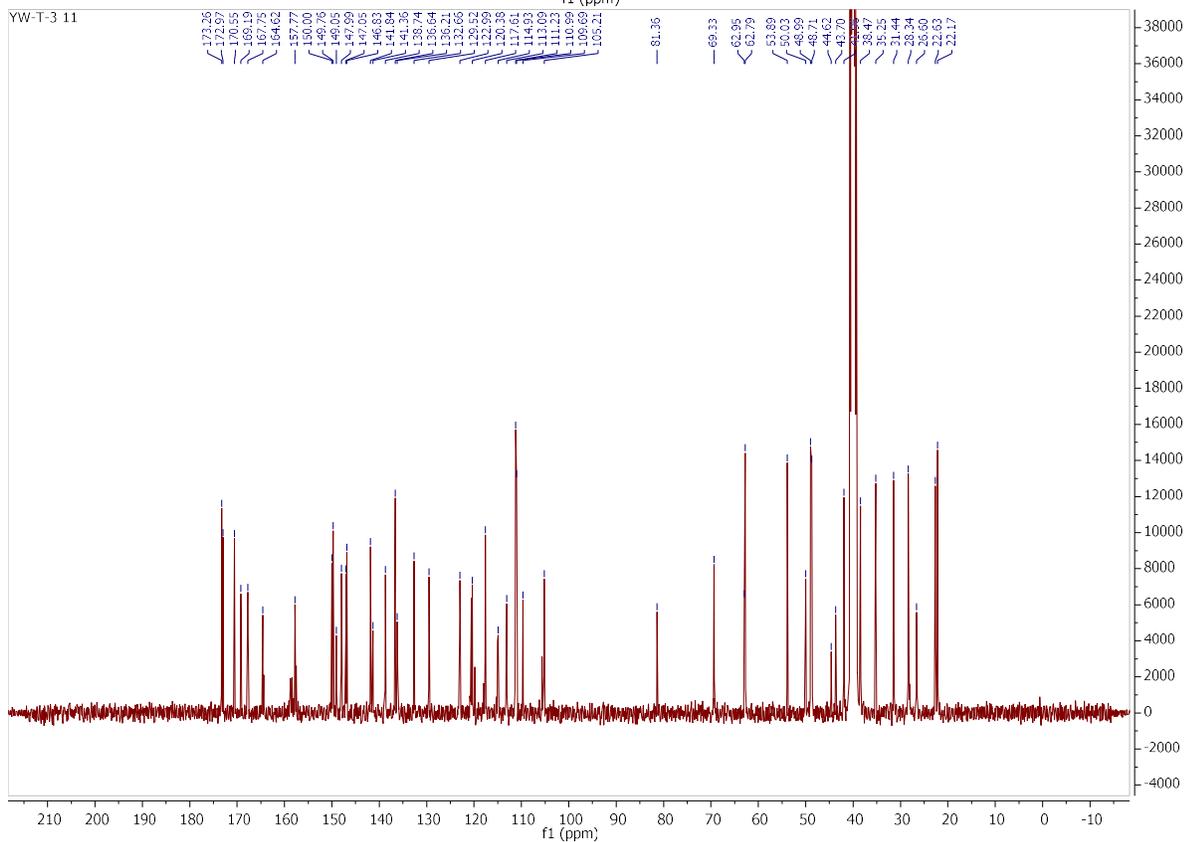
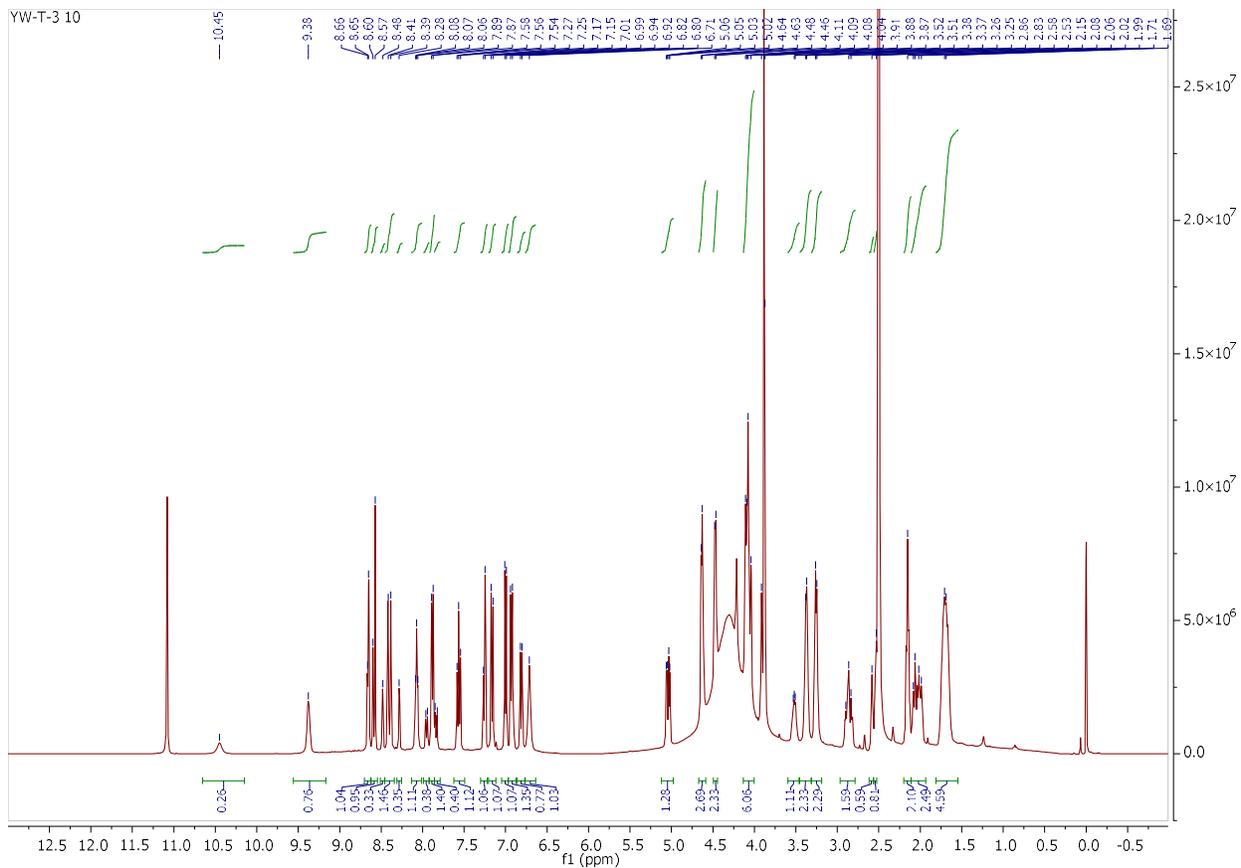
YW-T-1 10



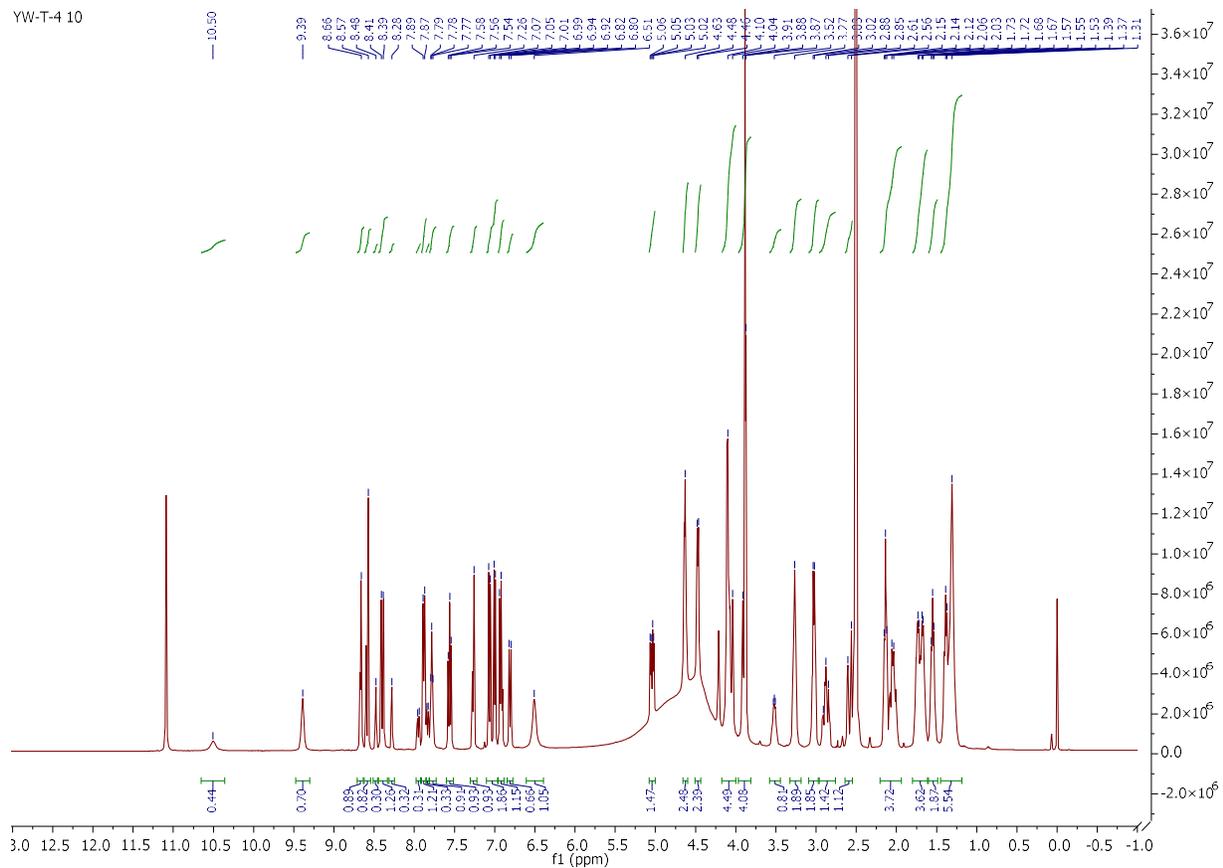
YW-T-1 11



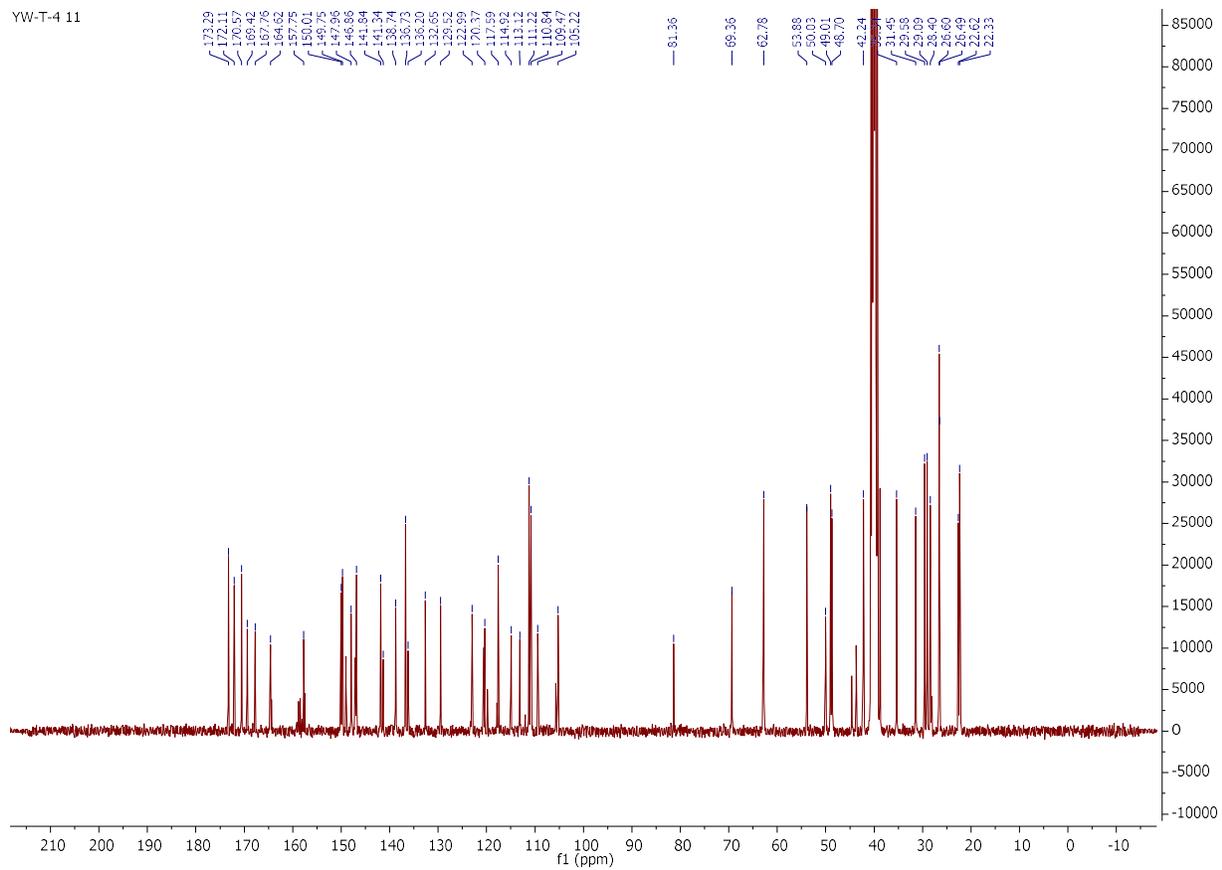




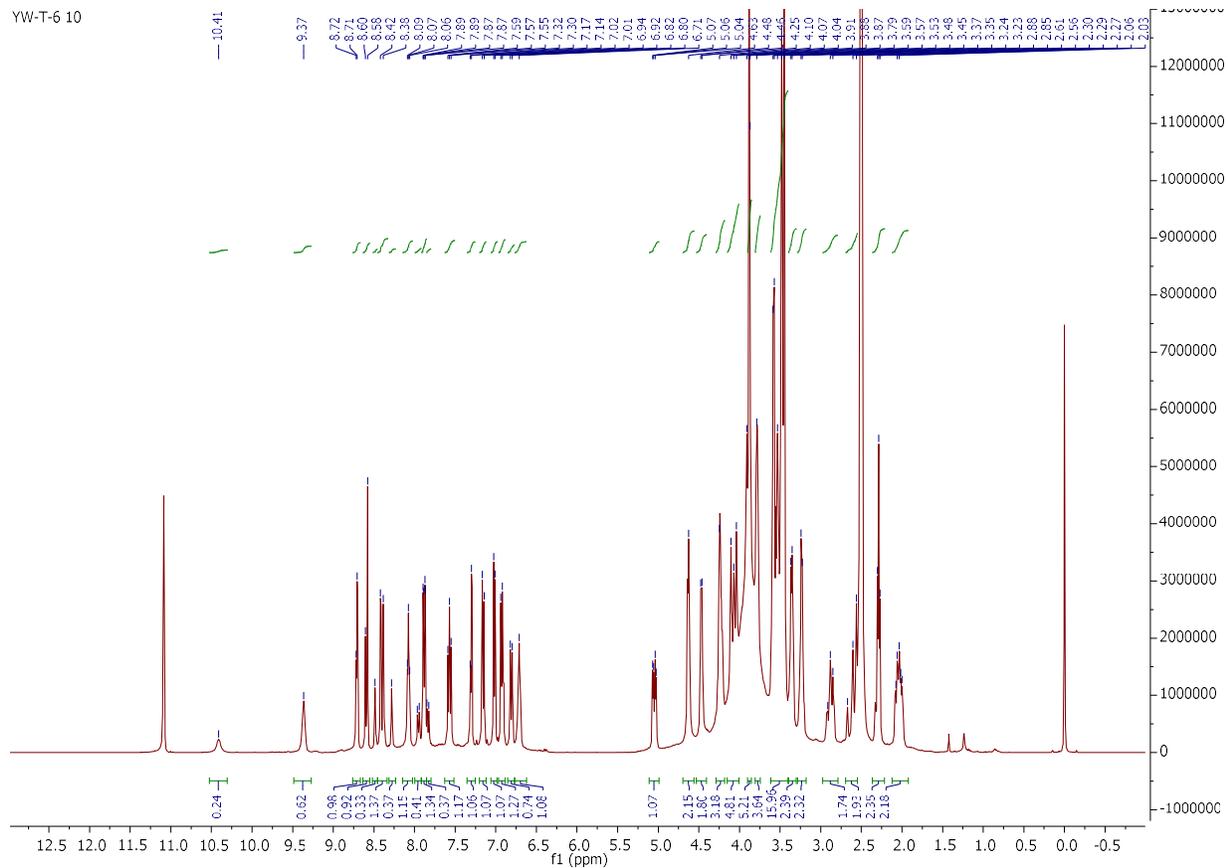
YW-T-4 10



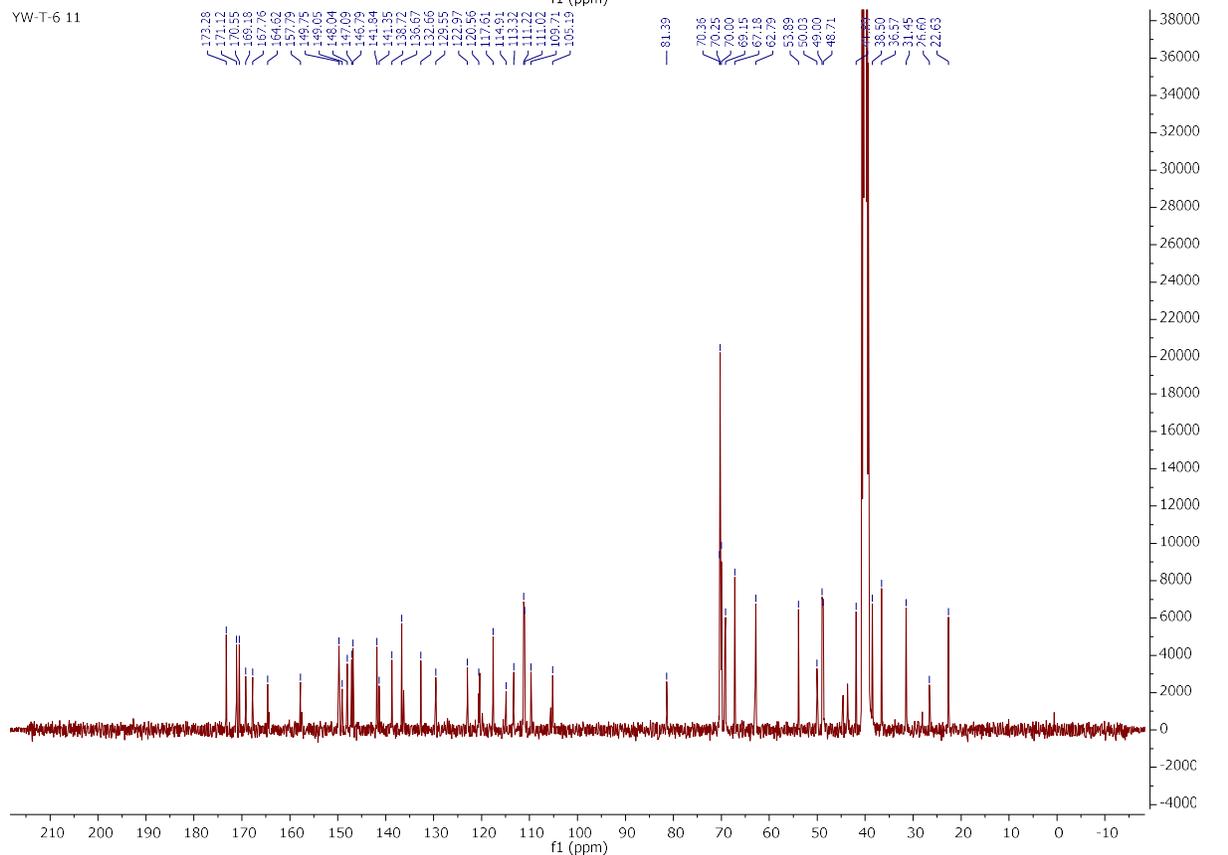
YW-T-4 11



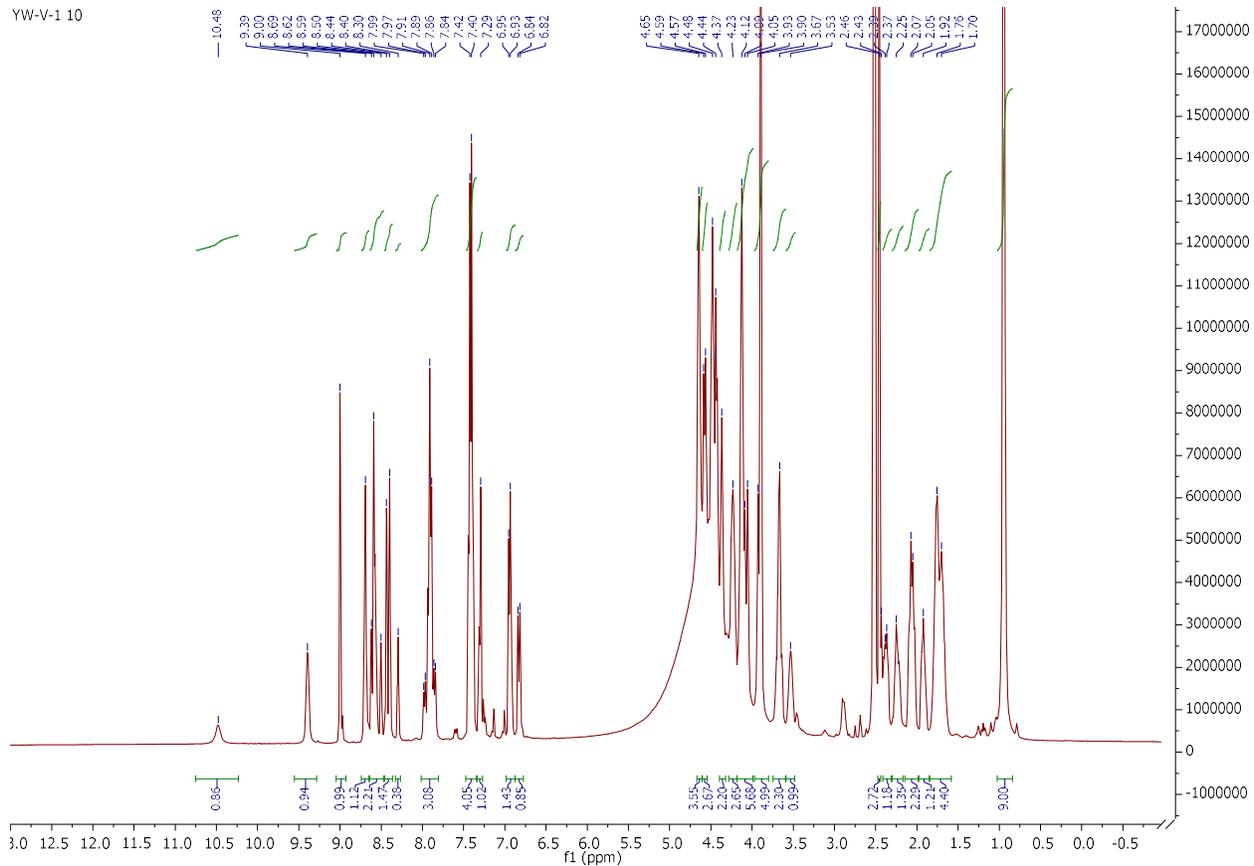
YW-T-6 10



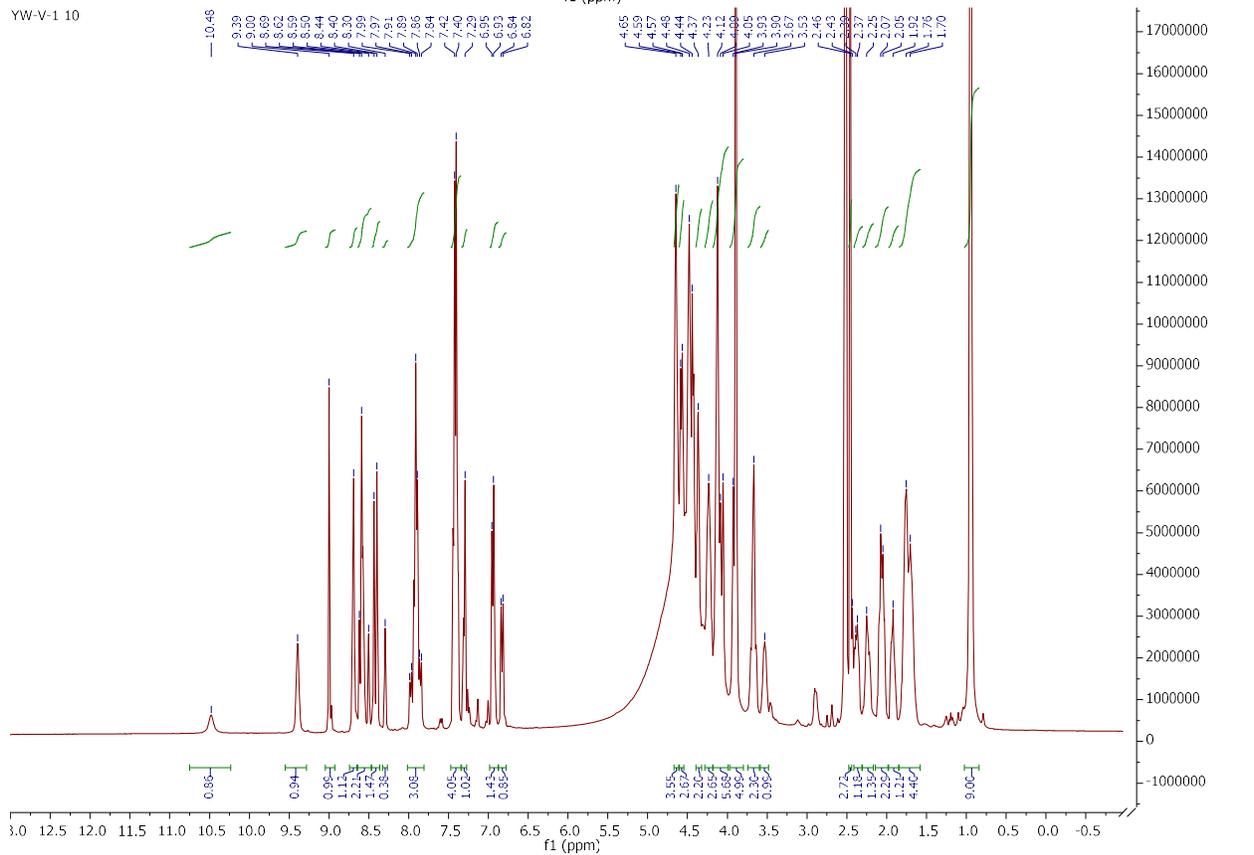
YW-T-6 11



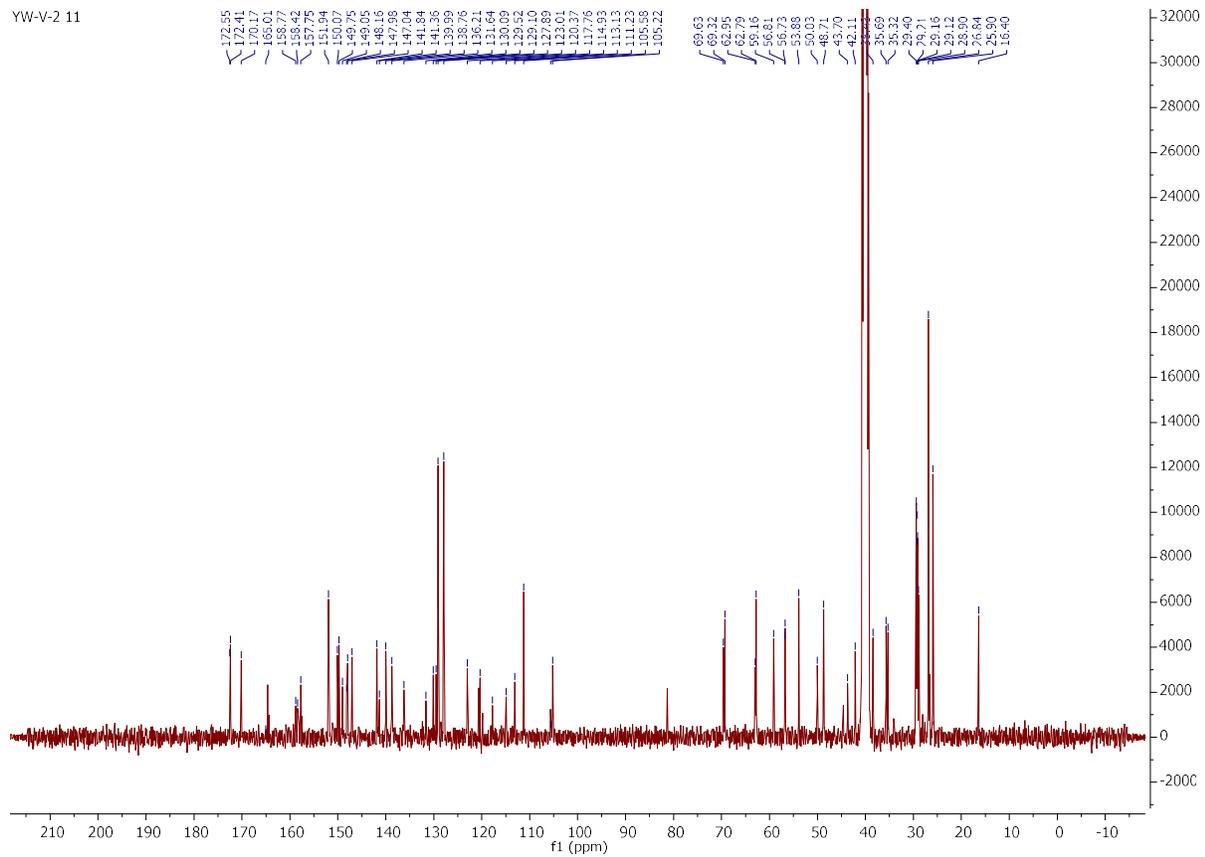
YW-V-1 10



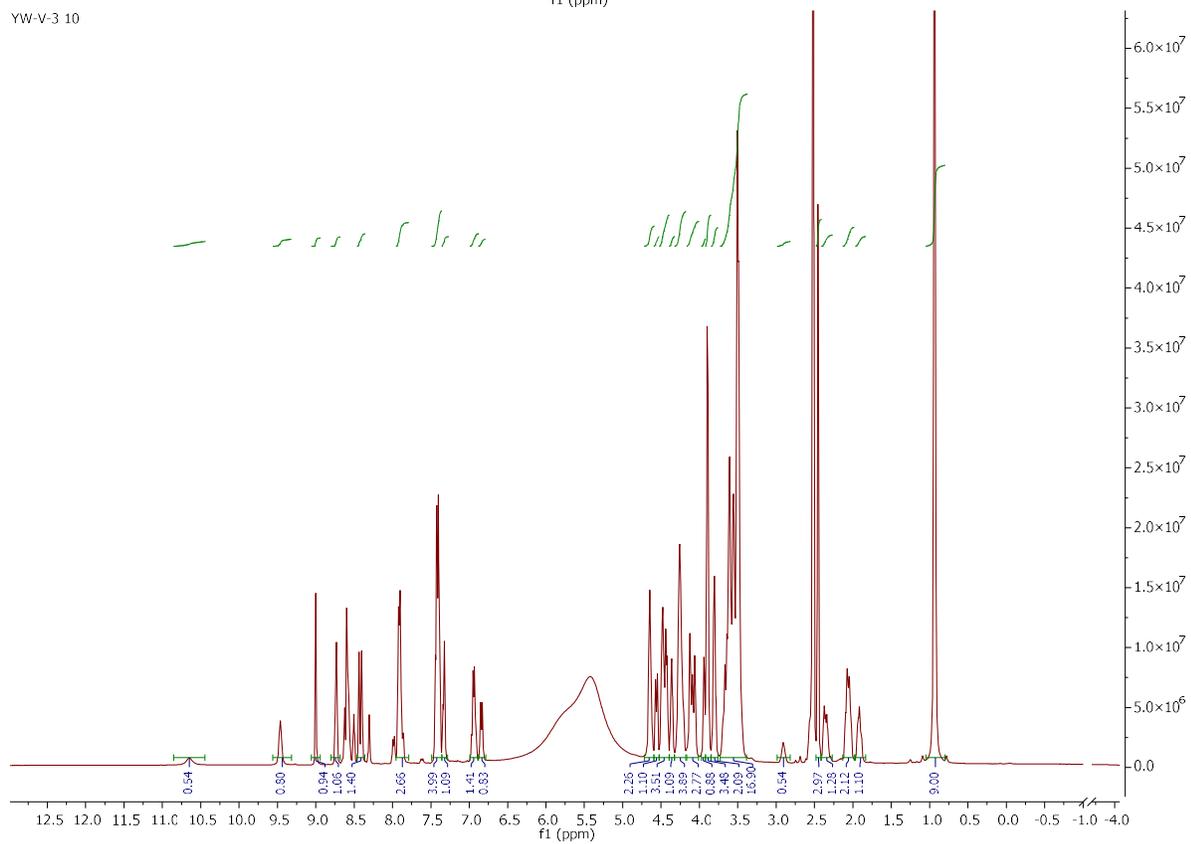
YW-V-1 10



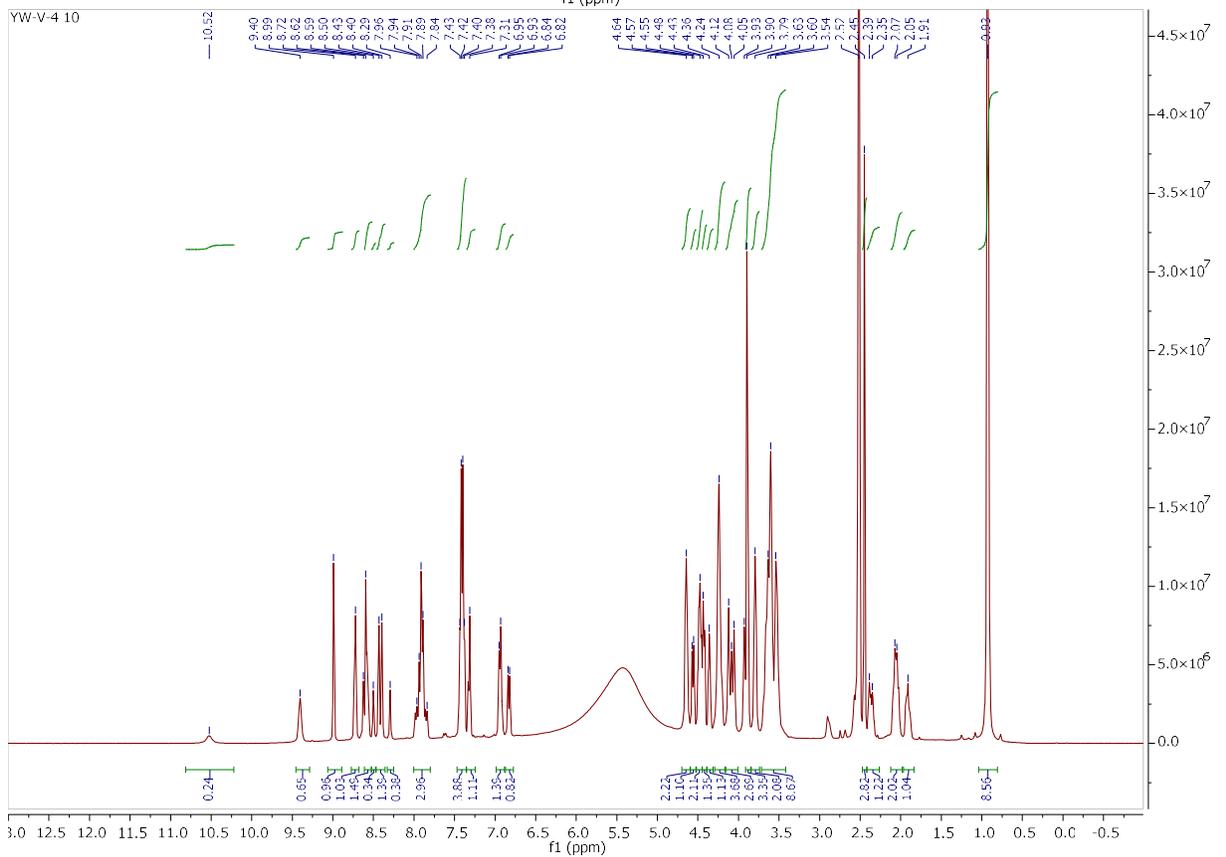
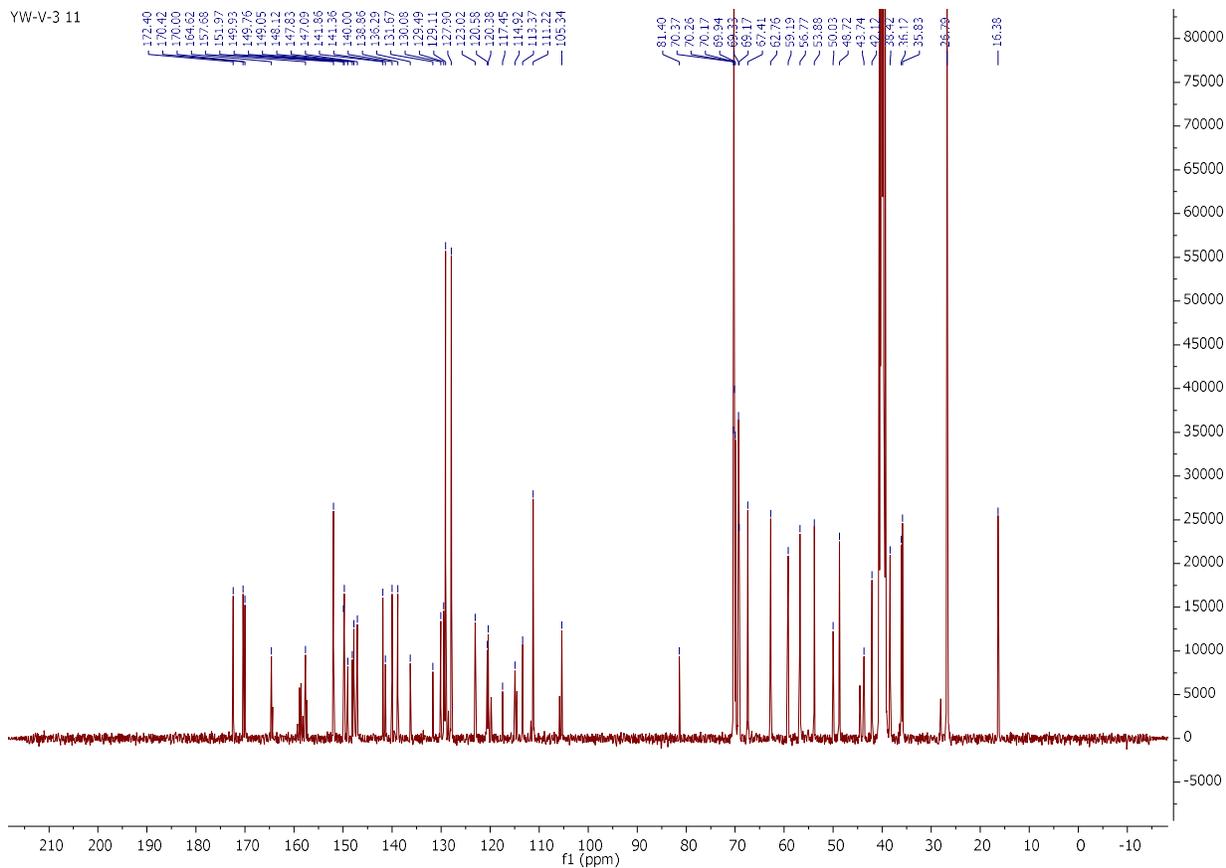
YW-V-2 11

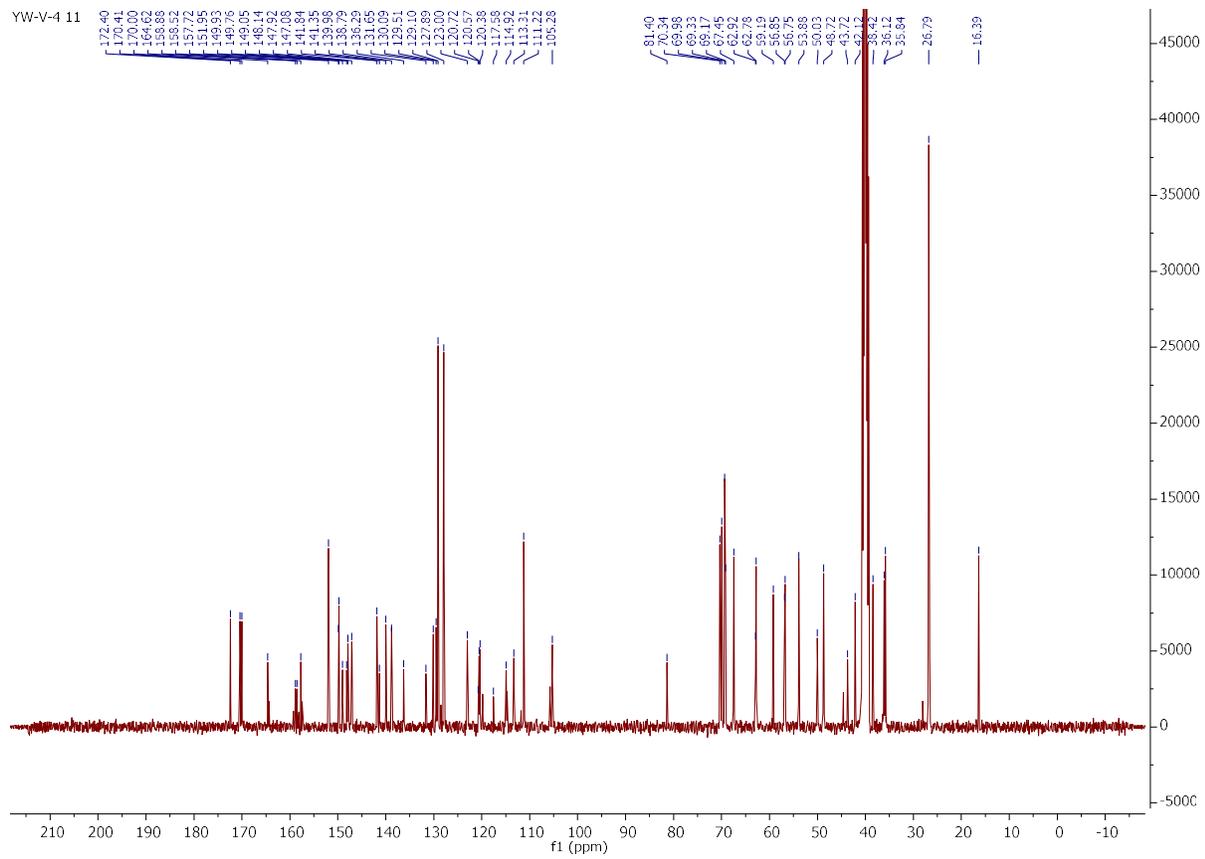


YW-V-3 10



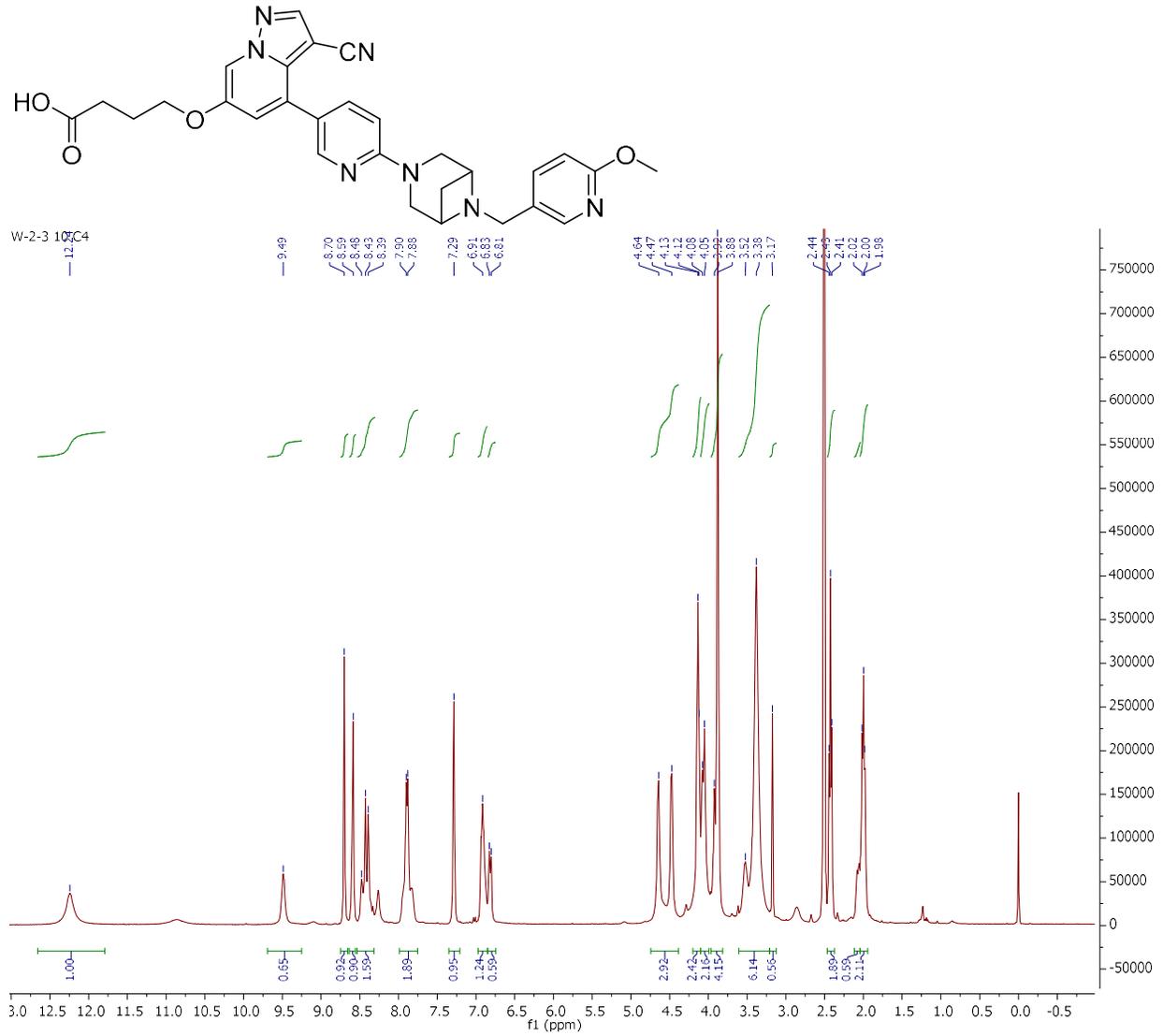
YW-V-3 11



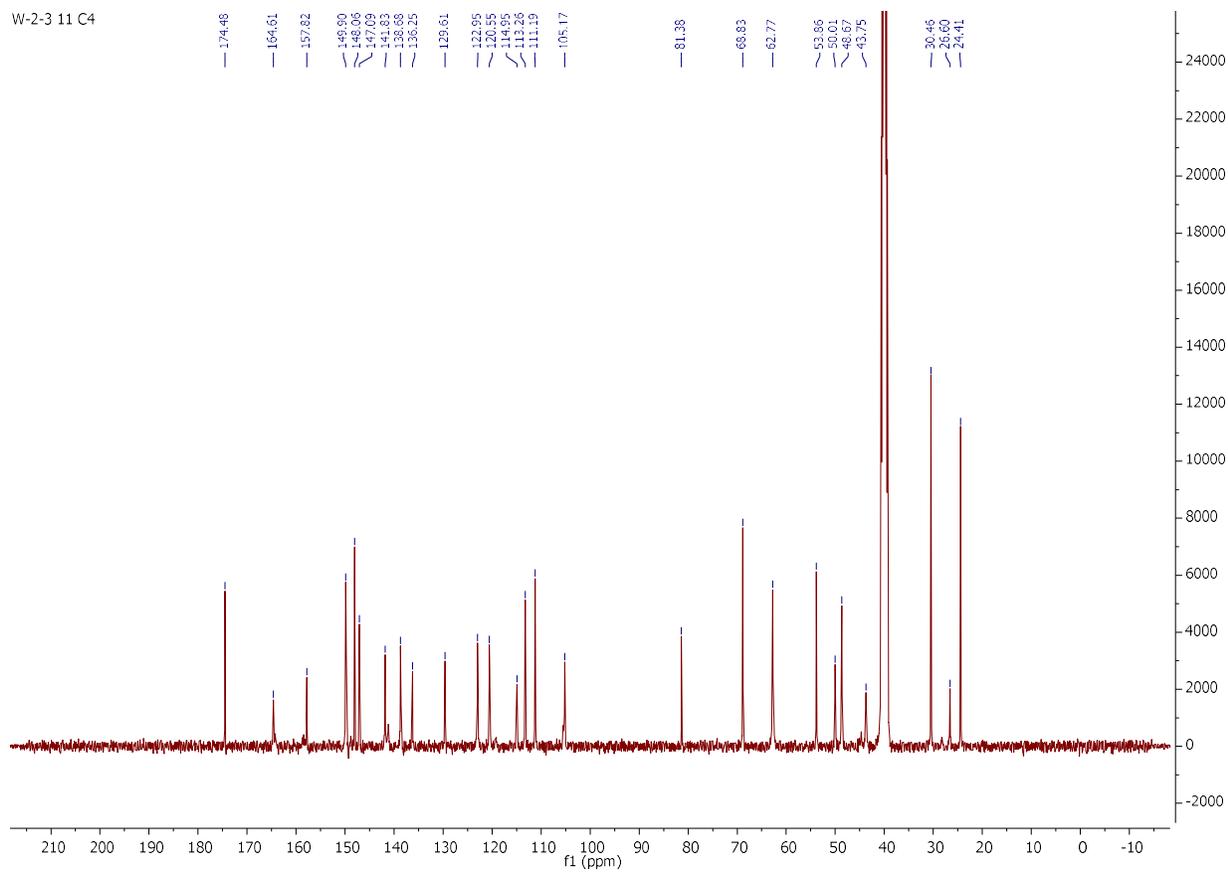


W-2-3 (C₄, C₆, C₈, C₁₀, C₁₁)

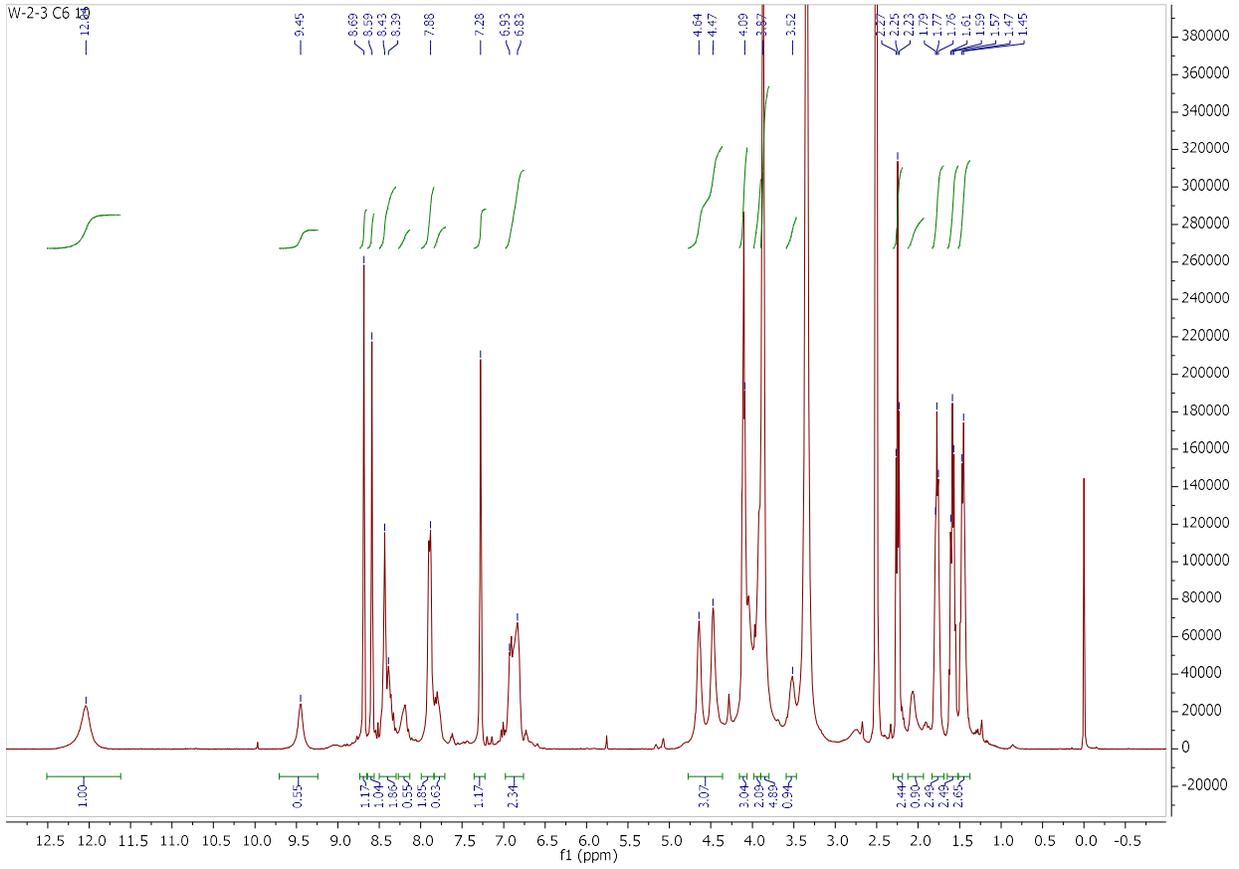
W-2-3 C₄



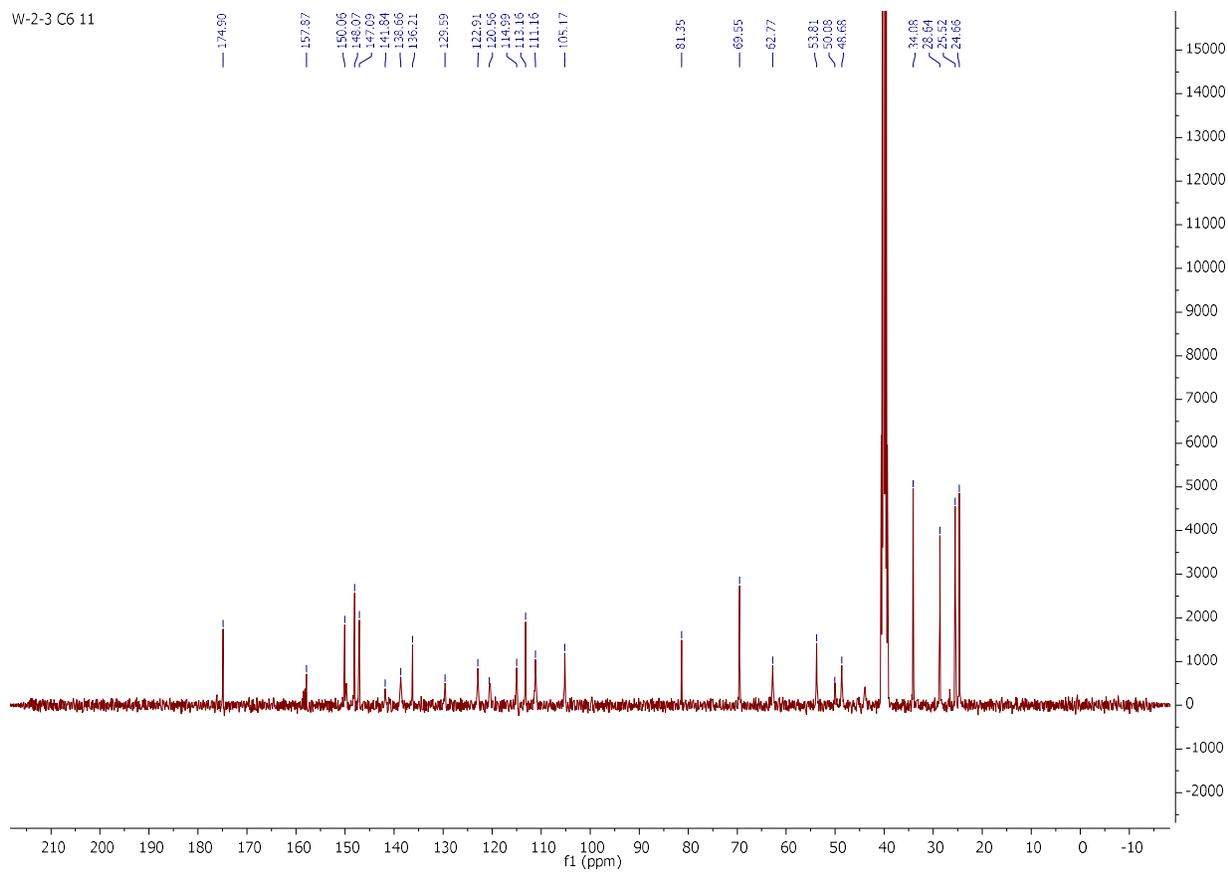
W-2-3 11 C4



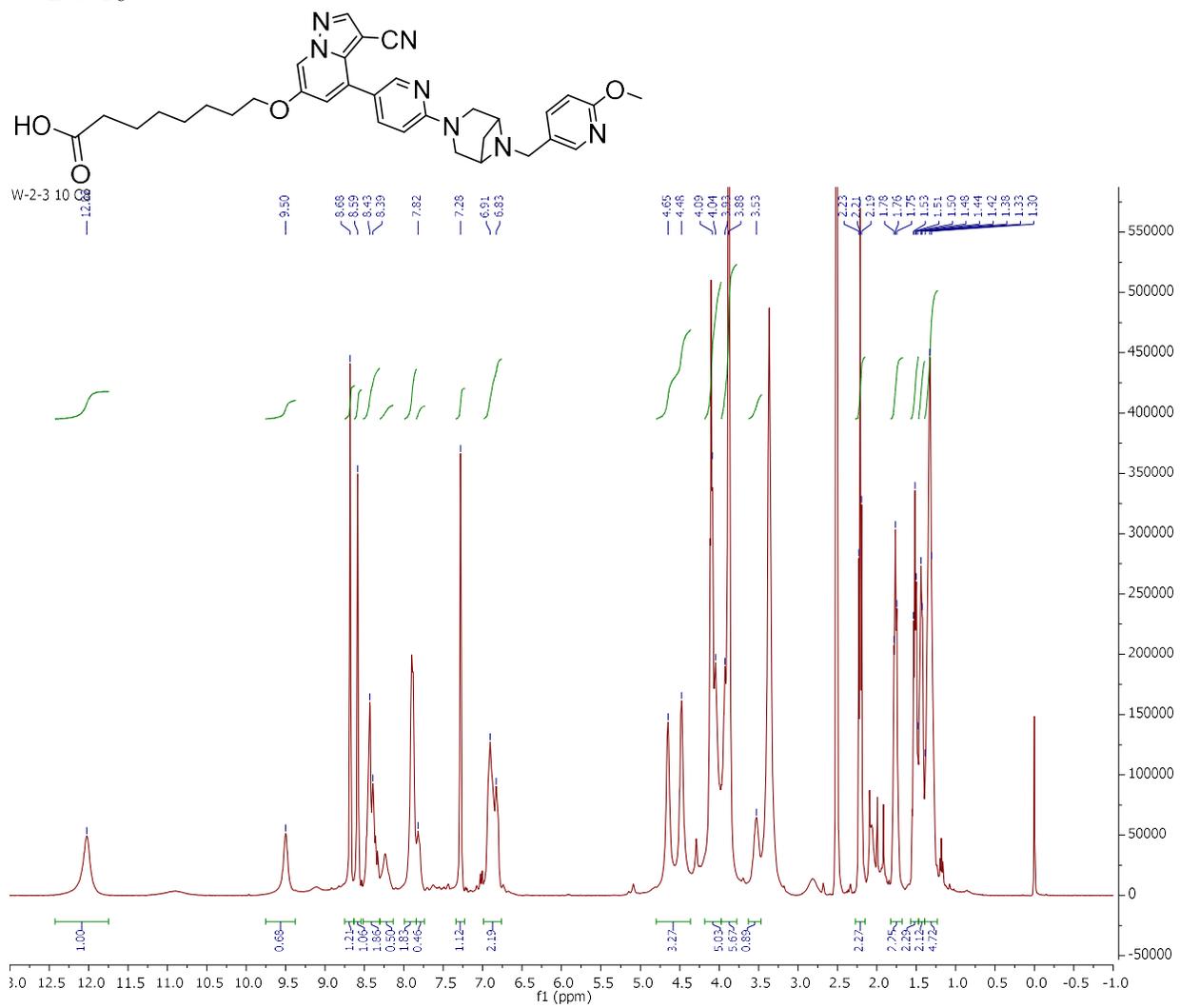
W-2-3 C₆

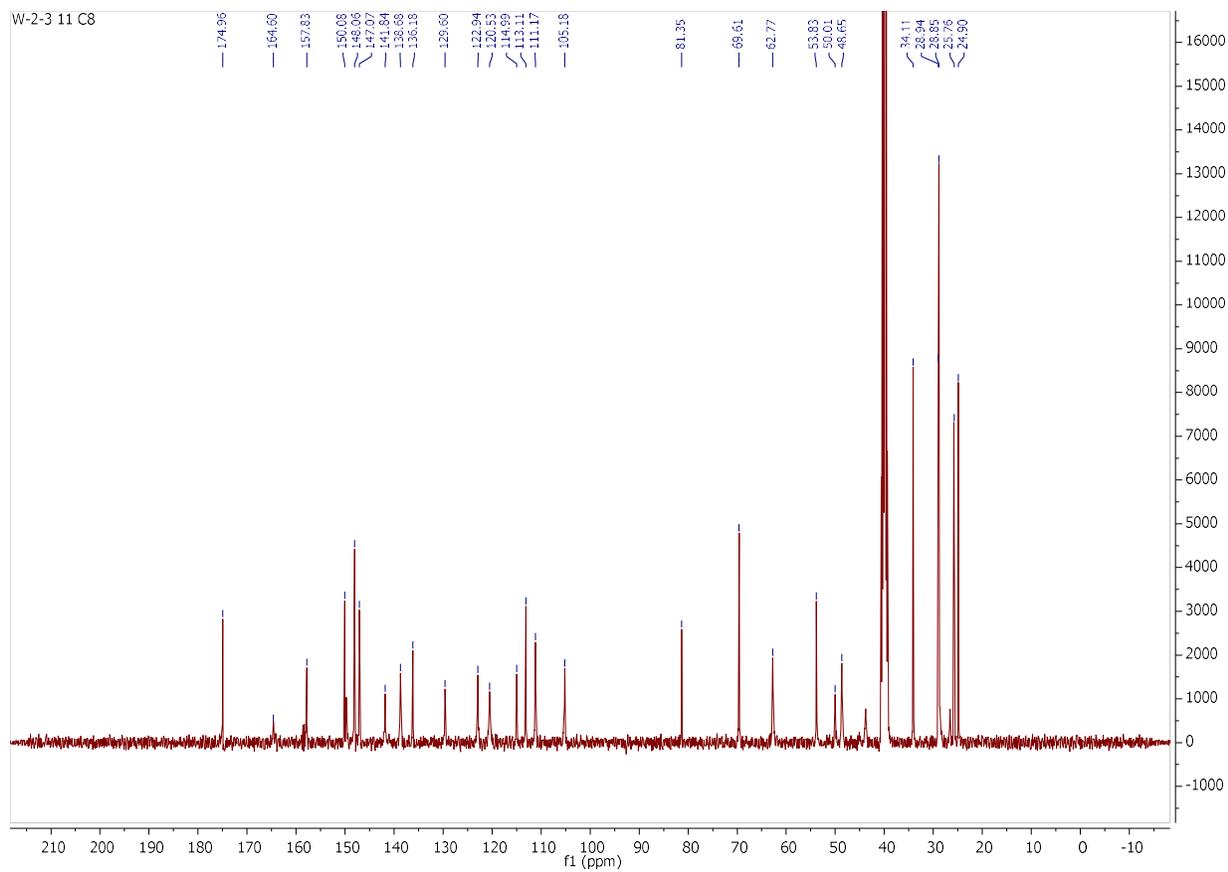


W-2-3 C6 11

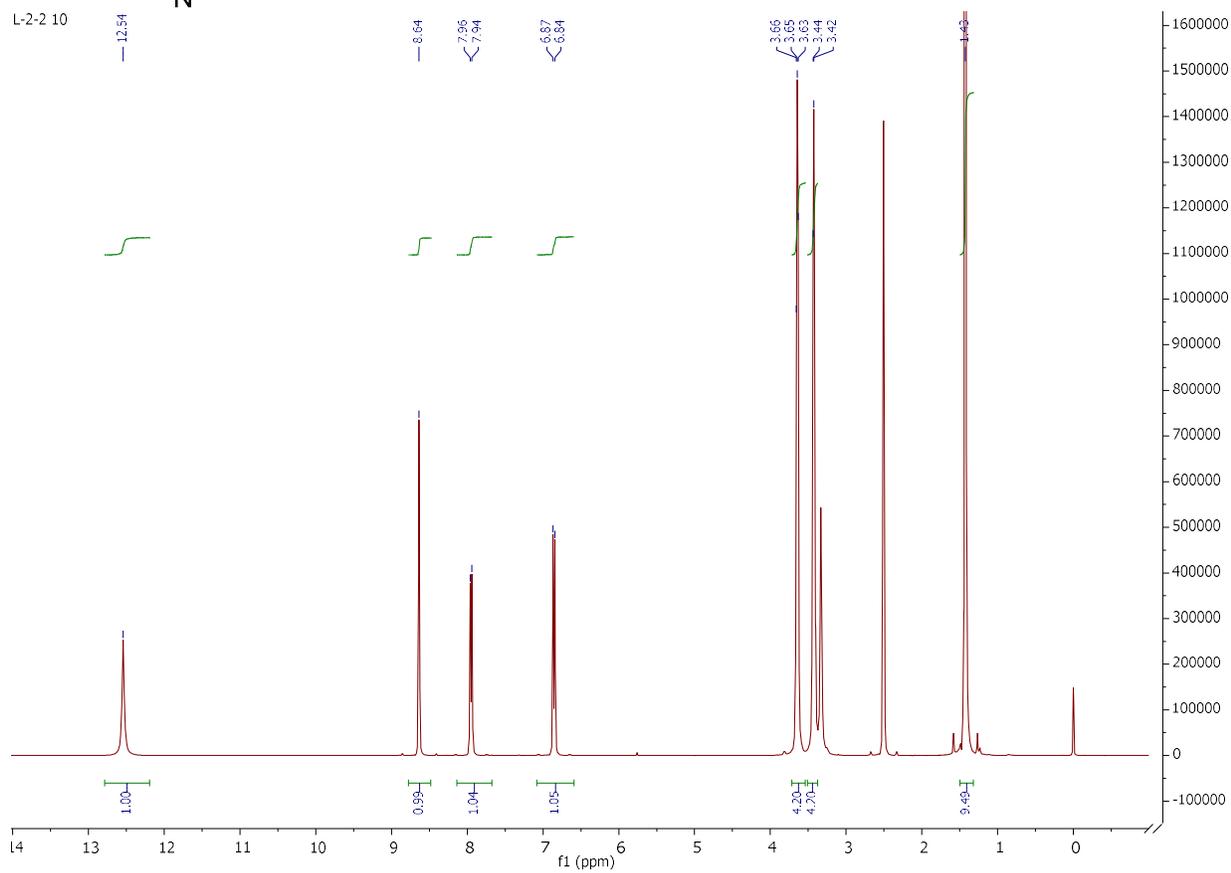
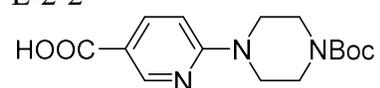


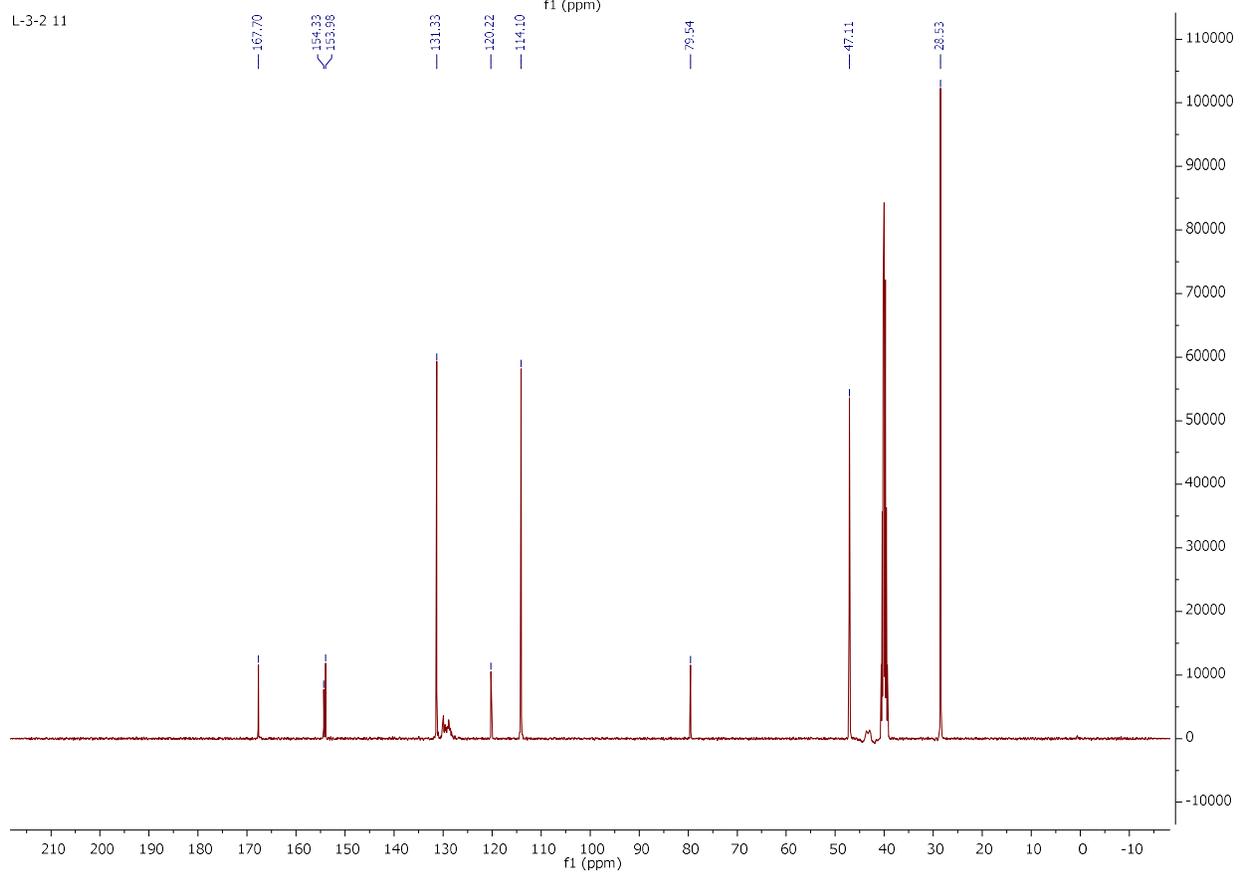
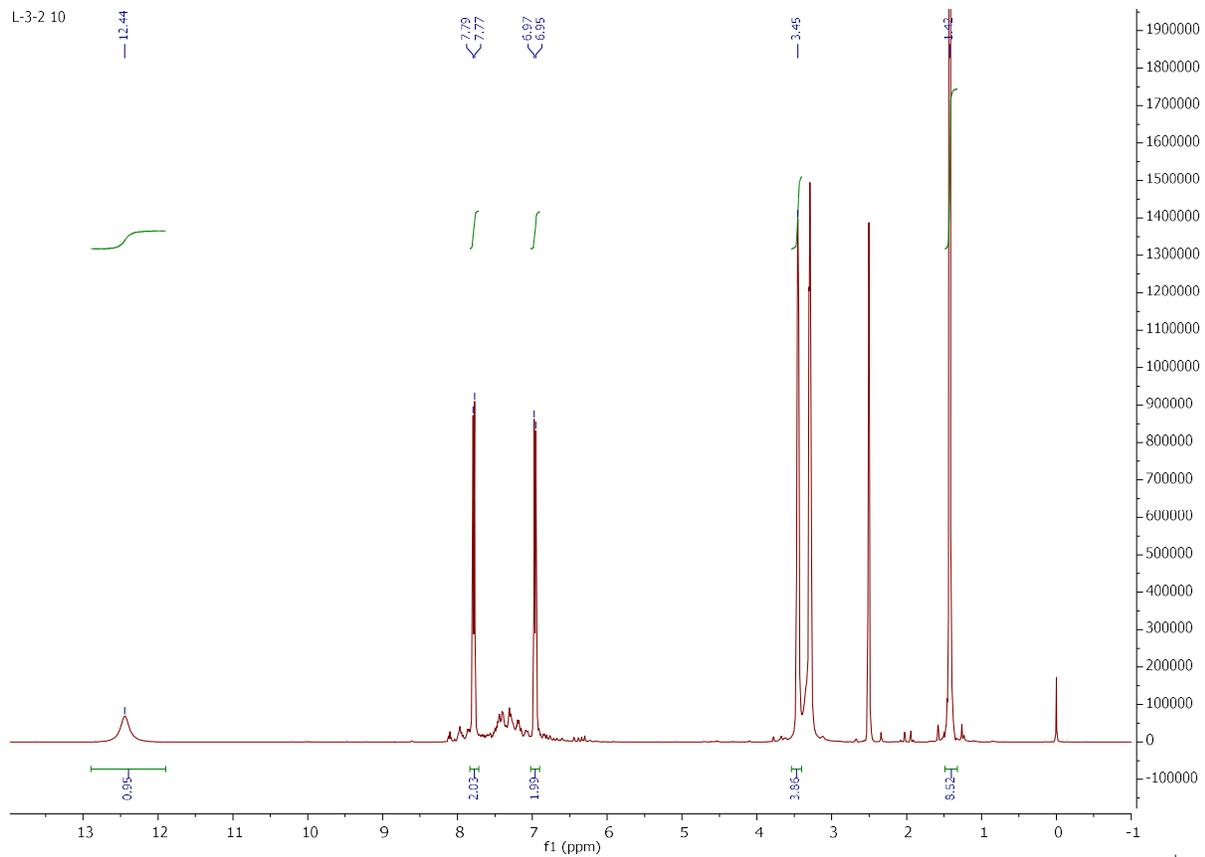
W-2-3 C₈





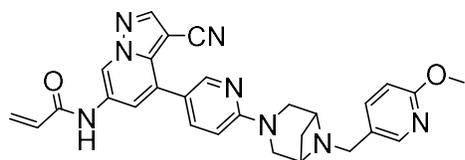
L-2-2





Appendix B: ^1H -NMR and ^{13}C -NMR

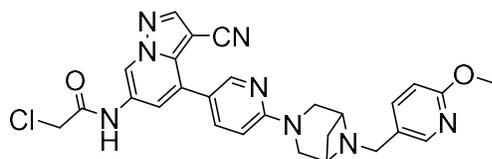
YW-D-50-2



^1H NMR (400 MHz, DMSO- d_6) δ 9.55 (s, 1H), 9.42 (s, 1H), 8.64 (s, 1H), 8.54 – 8.25 (m, 2H), 8.02 – 7.80 (m, 2H), 7.49 (d, J = 17.2 Hz, 1H), 6.90 (dd, J = 29.6, 8.6 Hz, 3H), 6.47 (d, J = 6.7 Hz, 1H), 6.39 – 6.30 (m, 1H), 5.87 (d, J = 9.9 Hz, 1H), 4.64 (s, 2H), 4.48 (d, J = 6.1 Hz, 1H), 4.23 (s, 1H), 4.17 – 4.02 (m, 2H), 3.98 – 3.83 (m, 4H), 3.53 (s, 1H), 2.08 (s, 1H).

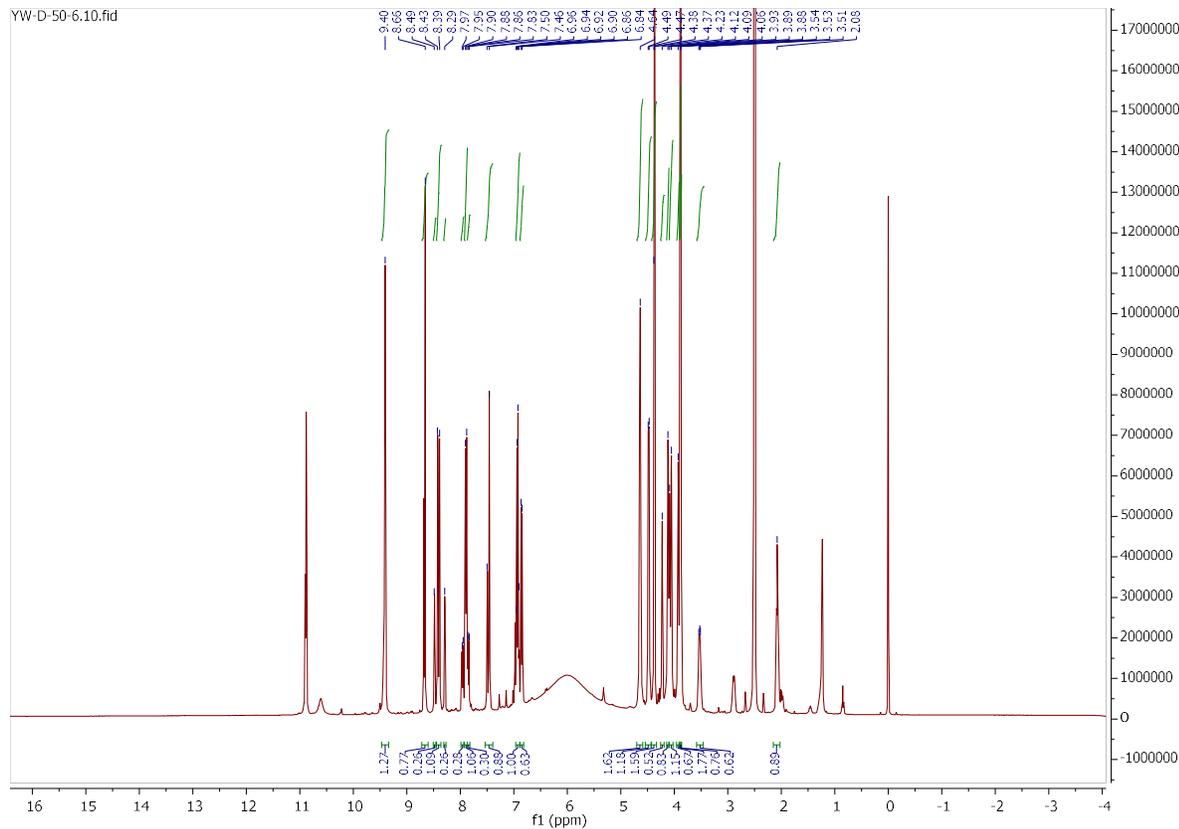
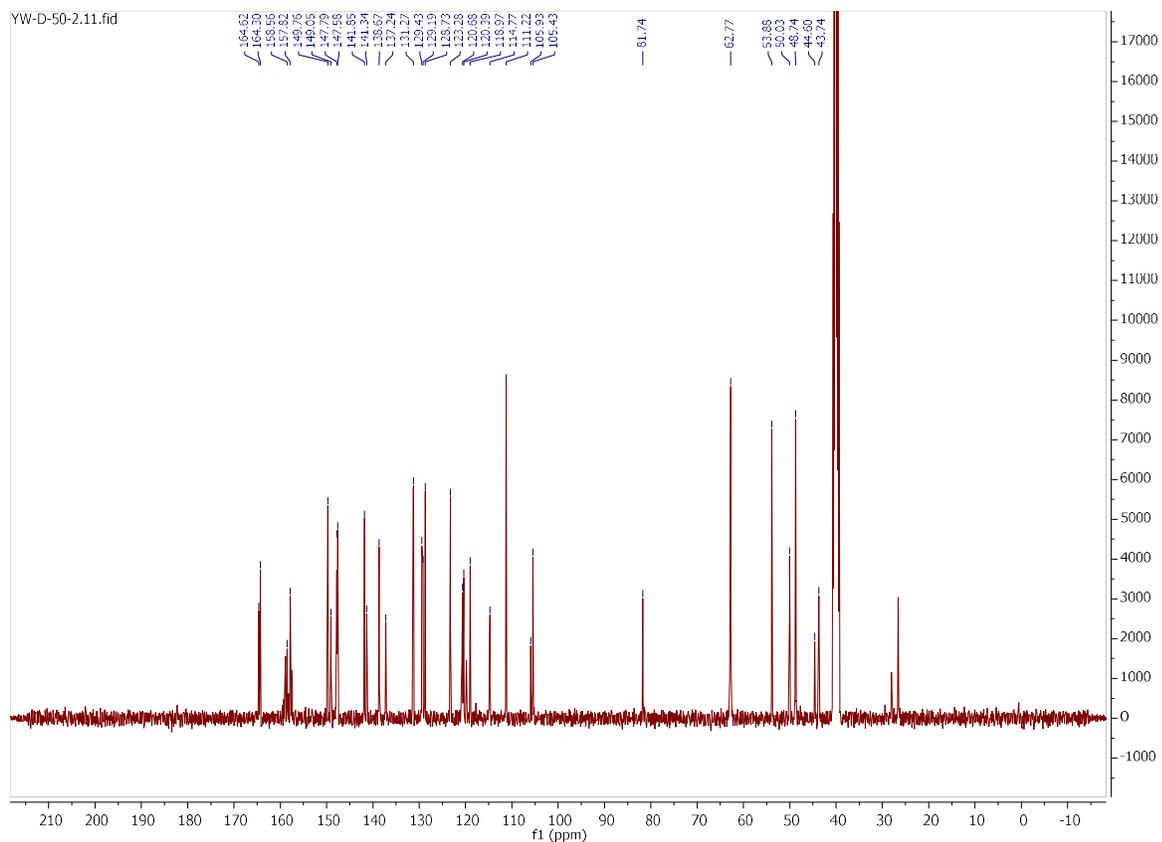
^{13}C NMR (101 MHz, DMSO- d_6) δ 164.62, 164.30, 158.56, 157.82, 149.76, 149.05, 147.79, 147.58, 141.85, 141.34, 138.67, 137.24, 131.27, 129.43, 129.19, 128.73, 123.28, 120.68, 120.39, 118.97, 114.77, 111.22, 105.93, 105.43, 81.74, 62.77, 53.88, 50.03, 48.74, 44.60, 43.74.

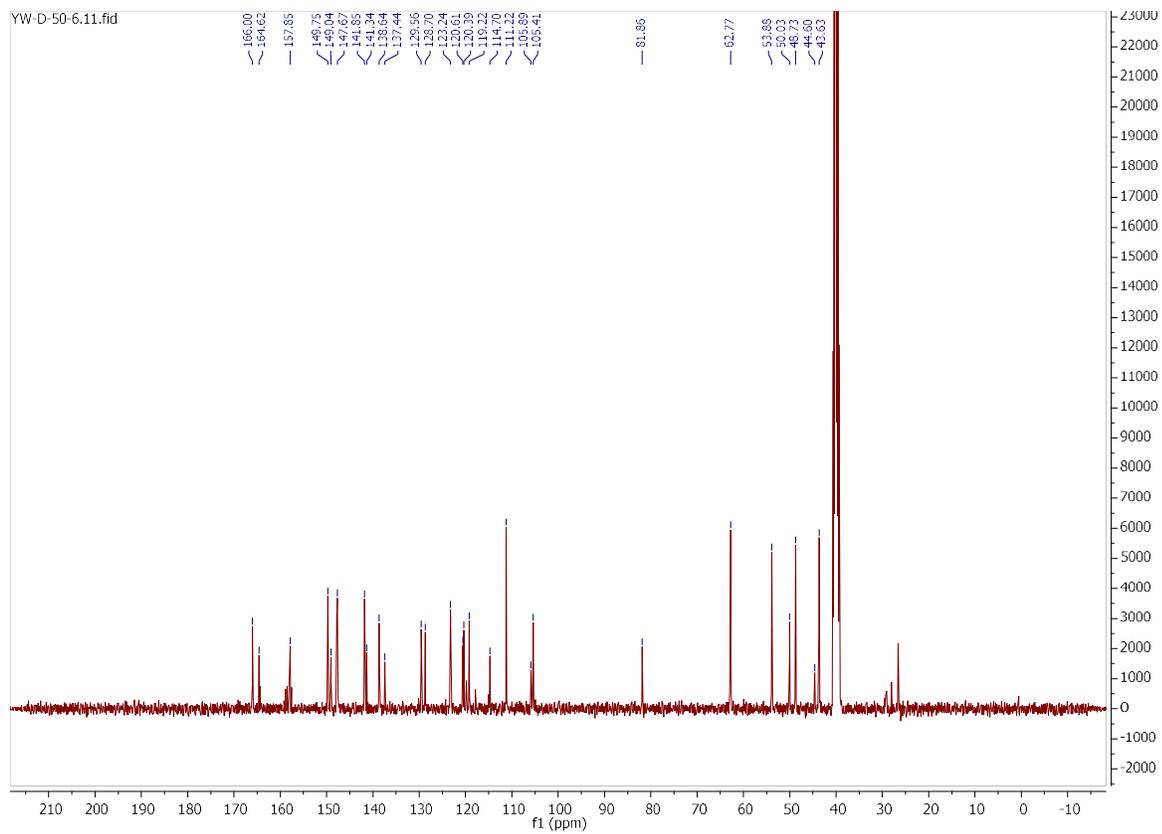
YW-D-50-6



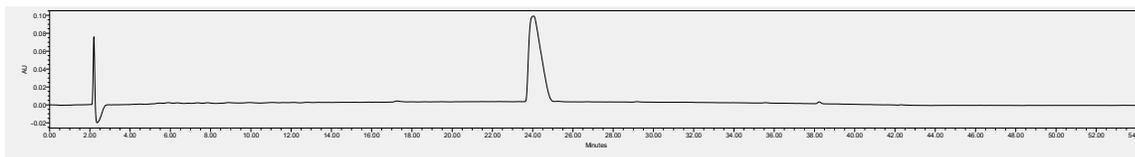
^1H NMR (400 MHz, DMSO- d_6) δ 9.40 (s, 1H), 8.66 (s, 1H), 8.41 (d, J = 15.1 Hz, 1H), 7.89 (d, J = 9.1 Hz, 1H), 7.48 (d, J = 16.9 Hz, 1H), 6.93 (q, J = 7.2, 6.6 Hz, 1H), 6.85 (d, J = 8.8 Hz, 1H), 4.64 (s, 2H), 4.48 (d, J = 6.1 Hz, 1H), 4.38 (d, J = 5.2 Hz, 2H), 4.23 (s, 1H), 4.12 (s, 1H), 4.07 (d, J = 12.7 Hz, 1H), 3.93 (s, 1H), 3.89 (s, 2H), 3.88 (s, 1H), 3.59 – 3.46 (m, 1H), 2.08 (s, 1H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 166.00, 164.62, 157.85, 149.75, 149.04, 147.67, 141.85, 141.34, 138.64, 137.44, 129.56, 128.70, 123.24, 120.61, 120.39, 119.22, 114.70, 111.22, 105.89, 105.41, 81.86, 62.77, 53.88, 50.03, 48.73, 44.60, 43.63.

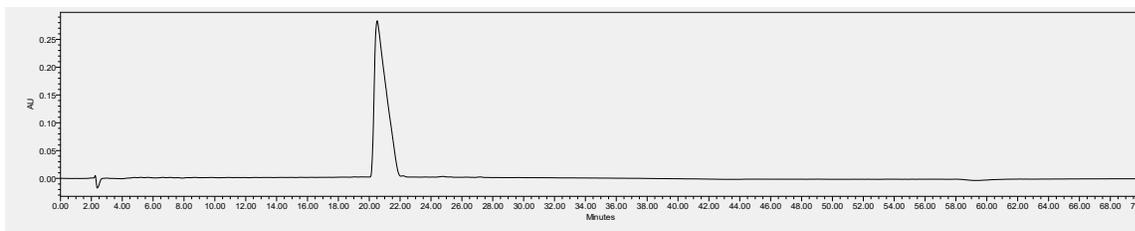




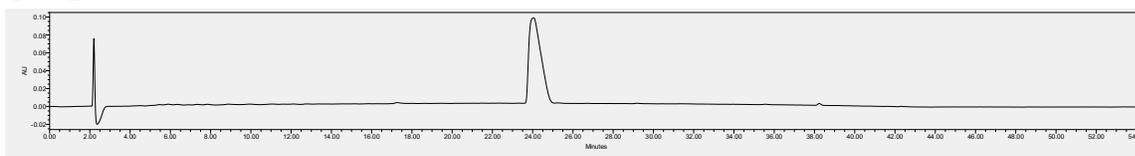
Appendix C: HPLC data , ¹HNMR, and ¹³C NMR spectrum



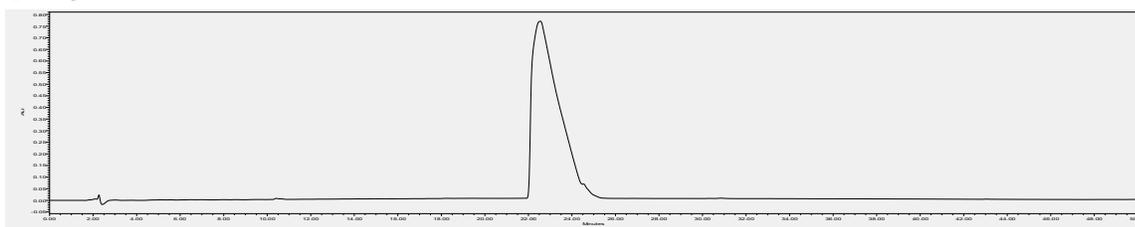
YW-1



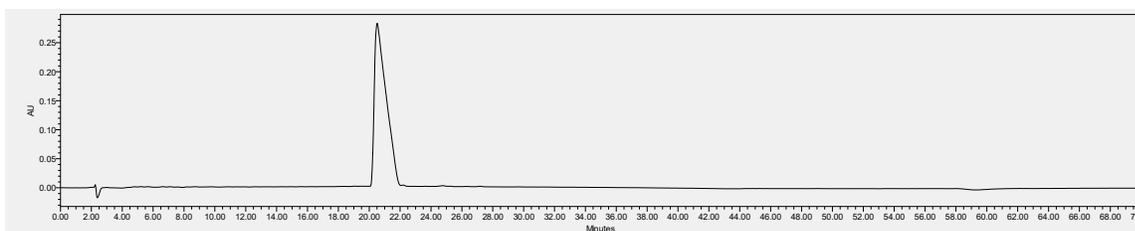
YW-2



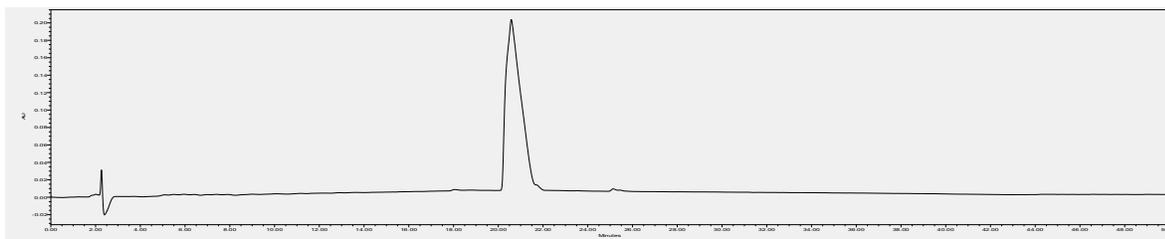
YW-3



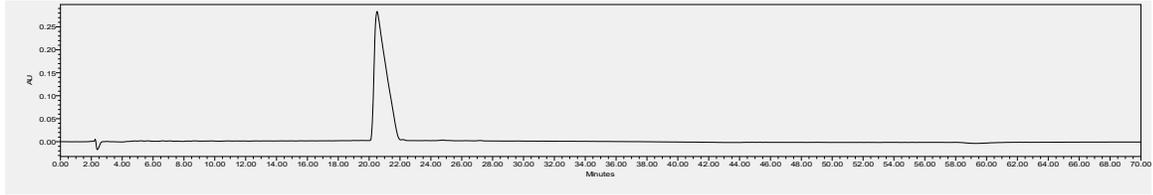
YW-4



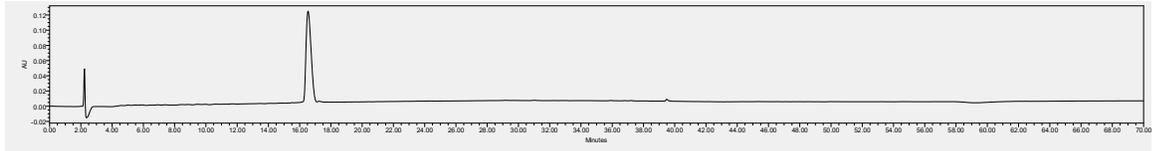
YW-5



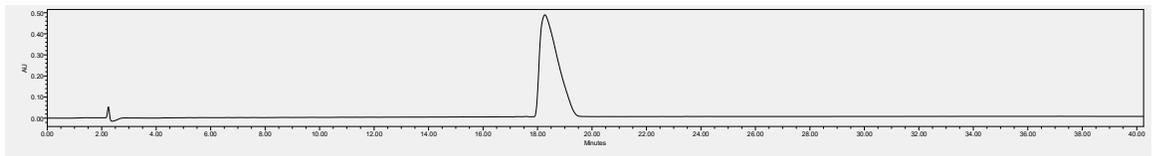
YW-6



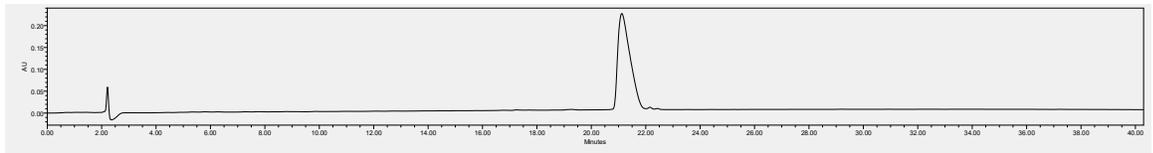
YW-7



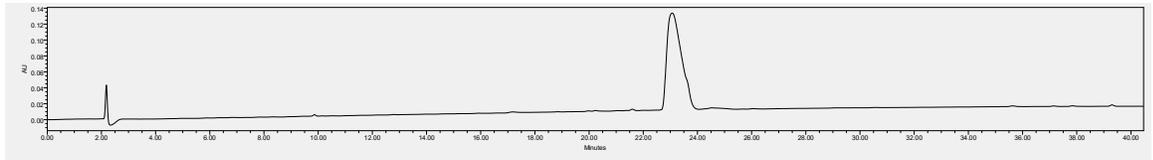
YW-8



YW-9



YW-10



YW-11

YW-1:

^1H NMR (600 MHz, DMSO- d_6) δ 8.09 (s, 1H), 7.96 – 7.71 (m, 9H), 7.57 (d, J = 8.9 Hz, 1H), 7.37 (d, J = 22.1 Hz, 1H), 7.17 (d, J = 12.8 Hz, 1H), 6.99 (d, J = 14.9 Hz, 1H), 4.10 – 3.88 (m, 3H), 3.74 (d, J = 16.5 Hz, 1H), 3.50 (ddd, J = 33.1, 14.7, 7.5 Hz, 4H), 3.26 – 3.16 (m, 1H), 3.02 (qd, J = 6.4, 3.2 Hz, 2H), 2.75 (dddd, J = 28.1, 13.9, 6.9, 3.1 Hz, 5H), 2.11 – 1.95 (m, 2H), 1.64 – 1.30 (m, 10H), 1.24 (s, 26H), 0.89 – 0.83 (m, 6H), 0.80 (dd, J = 10.4, 6.5 Hz, 3H).

^{13}C NMR (151 MHz, DMSO- d_6) δ 172.72, 172.49, 170.78, 158.82, 158.61, 38.94, 36.02, 35.73, 31.74, 30.46, 29.50, 29.15, 28.97, 27.14, 24.78, 23.80, 22.55, 22.01, 21.72, 14.41.

YW-2:

^1H NMR (600 MHz, DMSO- d_6) δ 8.20 – 7.96 (m, 1H), 7.94 – 7.71 (m, 9H), 7.30 – 7.10 (m, 6H), 4.27 – 4.09 (m, 1H), 4.08 – 3.94 (m, 1H), 3.93 – 3.75 (m, 1H), 3.56 (ddd, J = 87.1, 13.0, 5.8 Hz, 4H), 3.02 (p, J = 6.4 Hz, 2H), 2.94 (dd, J = 13.4, 7.4 Hz, 1H), 2.85 – 2.53 (m, 7H), 2.45 (q, J = 7.0 Hz, 1H), 1.93 (td, J = 7.3, 4.0 Hz, 2H), 1.53 (dt, J = 17.2, 10.4, 5.2 Hz, 4H), 1.45 – 1.10 (m, 27H), 1.09 – 0.94 (m, 2H), 0.85 (t, J = 6.9 Hz, 3H).

^{13}C NMR (151 MHz, DMSO- d_6) δ 171.27, 170.78, 158.54, 129.45, 128.50, 128.39, 118.54, 116.71, 49.99, 47.31, 44.32, 38.95, 31.75, 29.52, 29.38, 29.16, 27.17, 25.67, 22.55, 22.02, 14.42.

YW-3:

^1H NMR (600 MHz, DMSO- d_6) δ 7.95 – 7.69 (m, 9H), 4.07 – 3.76 (m, 3H), 3.50 (dd, J = 17.8, 10.9 Hz, 4H), 3.24 (dd, J = 7.3, 3.6 Hz, 1H), 3.01 (h, J = 6.2 Hz, 2H), 2.85 – 2.63 (m, 5H), 2.02 (dq, J = 12.1, 7.0 Hz, 2H), 1.62 – 1.49 (m, J = 6.5 Hz, 4H), 1.49 – 1.35 (m, 4H), 1.24 (s, 24H), 1.05 (d, J = 6.7 Hz, 1H), 0.97 (d, J = 6.7 Hz, 1H), 0.85 (t, J = 6.9 Hz, 3H).

^{13}C NMR (151 MHz, DMSO- d_6) δ 172.19, 171.93, 158.82, 158.62, 117.72, 116.43, 49.91, 47.31, 43.58, 38.95, 35.74, 31.75, 30.39, 29.51, 29.47, 29.16, 27.16, 25.70, 22.55, 22.01, 18.70, 18.46, 14.42.

YW-4

^1H NMR (600 MHz, DMSO- d_6) δ 8.11 (d, $J = 7.8$ Hz, 1H), 7.98 – 7.68 (m, 9H), 7.03 – 6.91 (m, 3H), 6.67 – 6.58 (m, 2H), 4.17 – 3.75 (m, 3H), 3.57 (td, $J = 13.7, 6.2$ Hz, 2H), 3.41 – 3.28 (m, 3H), 3.25 (dd, $J = 15.1, 9.2$ Hz, 1H), 3.01 (h, $J = 6.2$ Hz, 2H), 2.92 (dd, $J = 13.4, 7.7$ Hz, 1H), 2.83 – 2.56 (m, 6H), 2.43 (dd, $J = 13.5, 9.3$ Hz, 1H), 1.95 (tt, $J = 7.6, 3.1$ Hz, 2H), 1.55 (tt, $J = 14.0, 5.2$ Hz, 4H), 1.46 – 1.30 (m, 4H), 1.30 – 1.13 (m, 22H), 1.09 (h, $J = 7.0$ Hz, 2H), 0.85 (t, $J = 7.0$ Hz, 3H).

^{13}C NMR (151 MHz, DMSO- d_6) δ 172.62, 171.79, 170.95, 158.97, 158.76, 156.14, 130.27, 128.92, 118.51, 116.53, 115.34, 115.23, 49.92, 47.40, 38.93, 37.31, 36.07, 35.72, 31.75, 29.47, 29.17, 27.16, 25.68, 22.55, 22.02, 14.41.

YW-5

^1H NMR (600 MHz, DMSO- d_6) δ 8.03 (s, 1H), 7.89 – 7.63 (m, 9H), 4.08 – 3.76 (m, 4H), 3.57 – 3.38 (m, 6H), 3.03 (dp, $J = 17.3, 6.8, 6.0$ Hz, 3H), 2.84 – 2.63 (m, 6H), 2.10 – 1.99 (m, 2H), 1.53 (h, $J = 6.4, 5.4$ Hz, 4H), 1.43 (dt, $J = 30.3, 7.3$ Hz, 4H), 1.24 (s, 24H), 0.86 (t, $J = 6.9$ Hz, 3H).

^{13}C NMR (151 MHz, DMSO- d_6) δ 172.75, 171.00, 159.29, 158.54, 50.10, 49.42, 38.97, 35.75, 31.75, 30.40, 29.53, 29.31, 29.17, 27.18, 25.72, 22.56, 14.42.

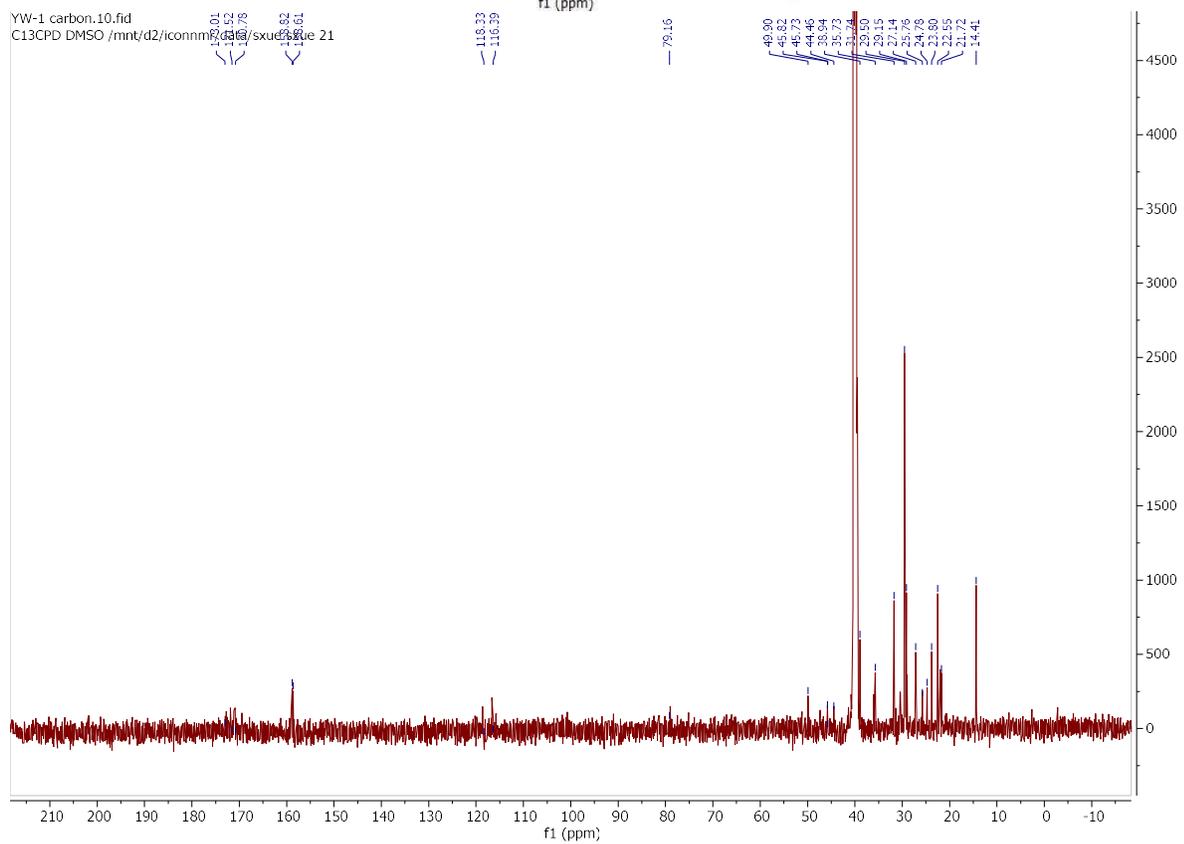
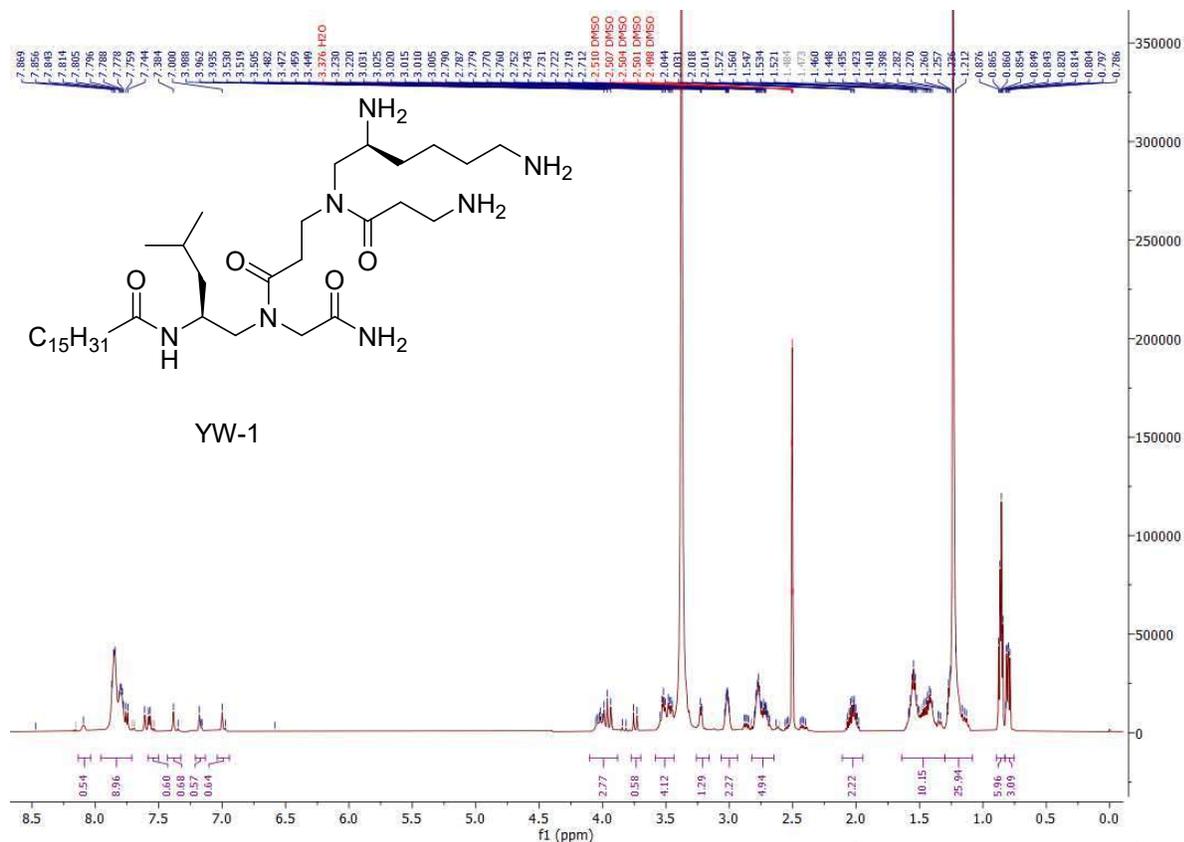
YW-6:

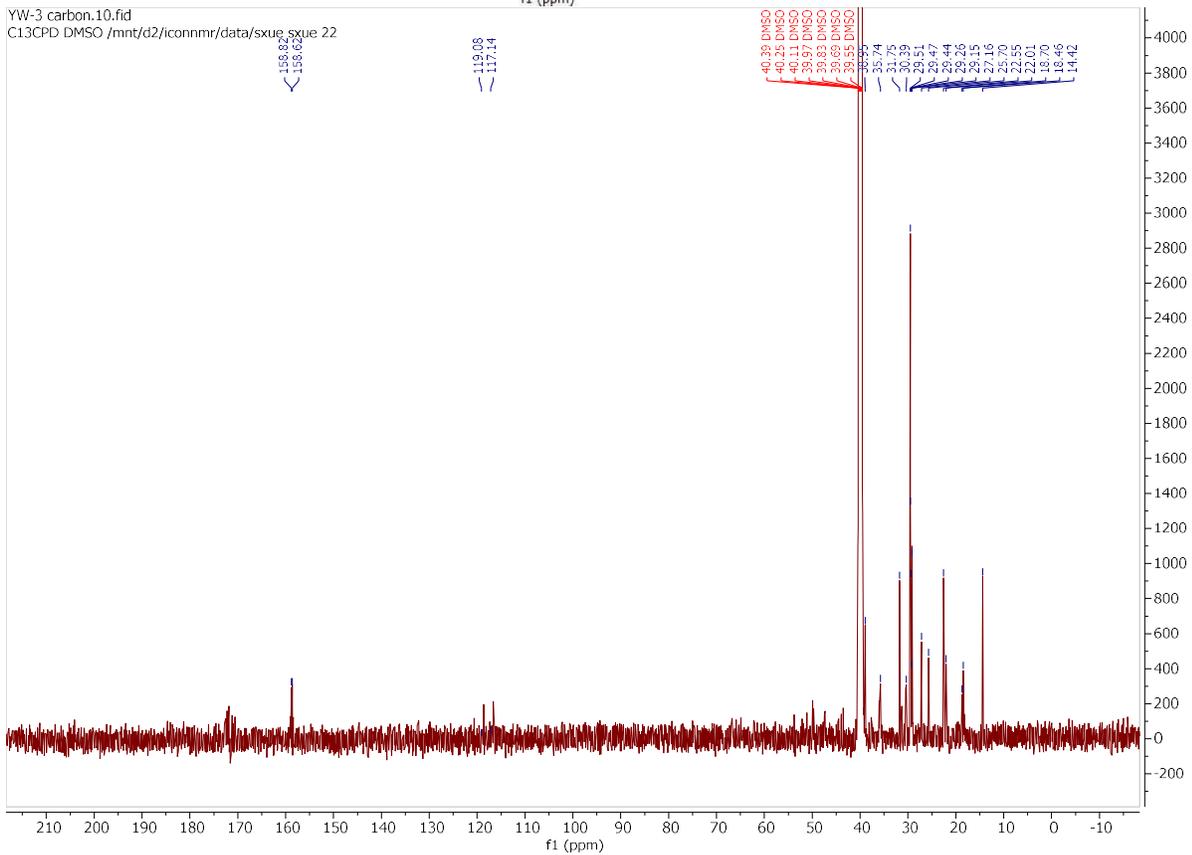
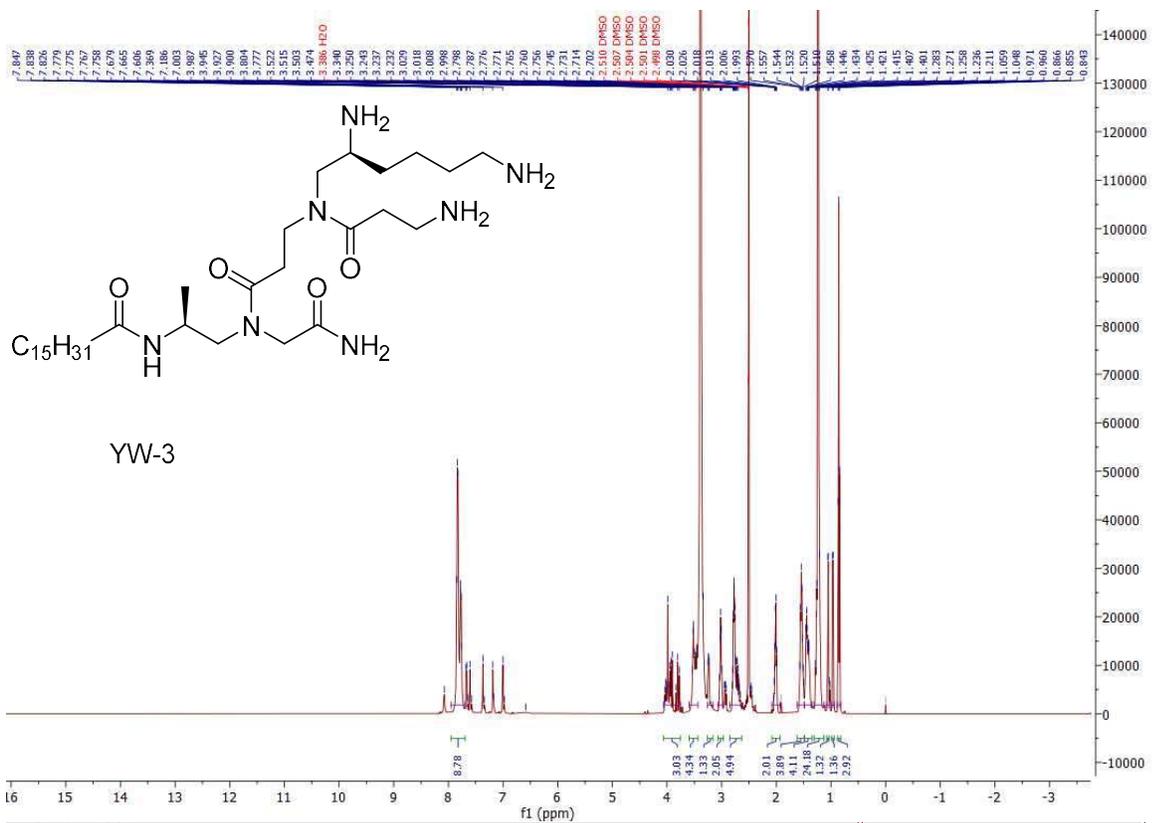
^1H NMR (600 MHz, DMSO- d_6) δ 8.06 (s, 1H), 7.78 (d, $J = 31.6$ Hz, 13H), 3.95 (dd, $J = 20.6, 4.1$ Hz, 2H), 3.88 (d, $J = 14.1$ Hz, 1H), 3.72 (d, $J = 16.5$ Hz, 1H), 3.58 – 3.40 (m, 4H), 3.01 (s, 2H),

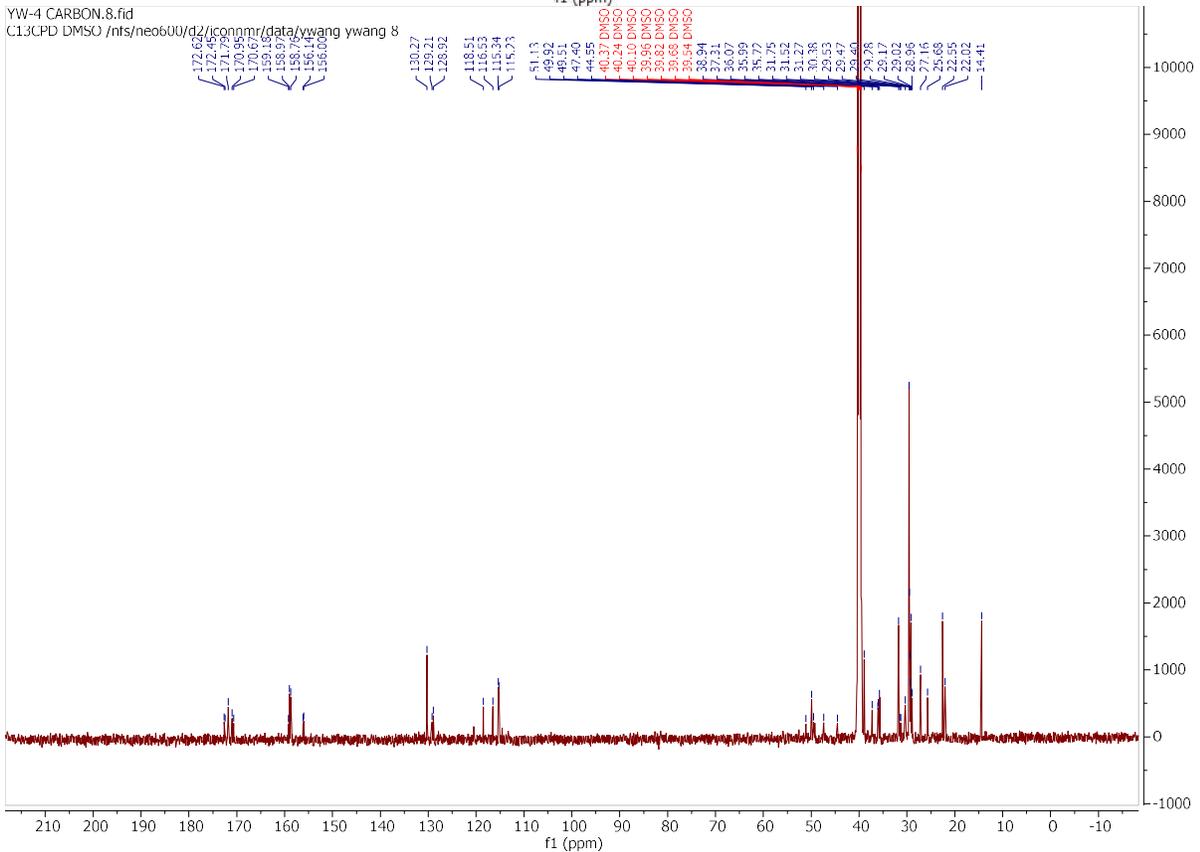
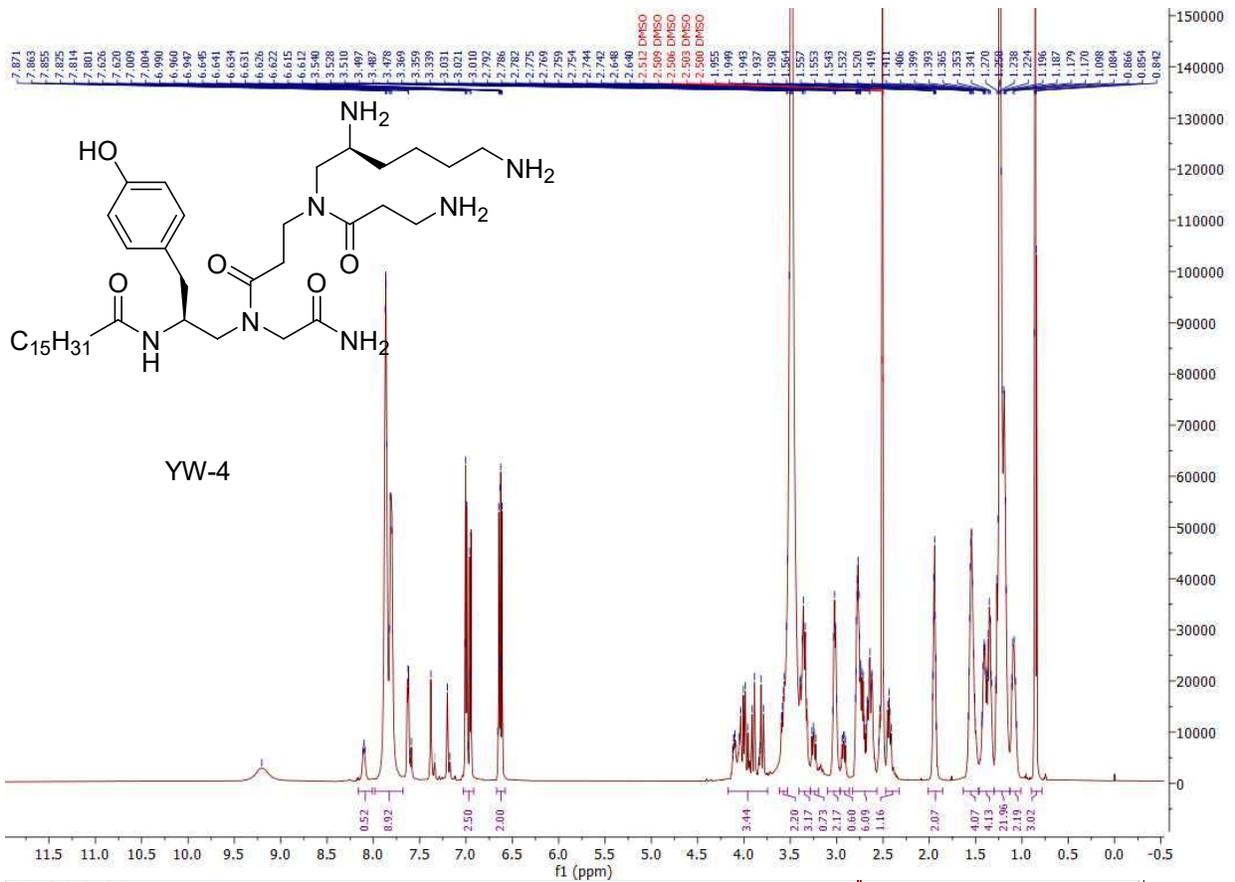
2.95 (dd, J = 13.7, 8.3 Hz, 1H), 2.76 (dtd, J = 31.4, 17.3, 14.7, 7.3 Hz, 8H), 2.45 – 2.35 (m, 1H),
2.03 (dq, J = 9.5, 7.0, 6.6 Hz, 2H), 1.62 – 1.36 (m, 13H), 1.24 (s, 31H), 0.86 (t, J = 6.9 Hz, 3H).
¹³C NMR (151 MHz, DMSO-d₆) δ 171.75, 169.58, 159.06, 158.32, 47.62, 45.31, 43.94, 39.03,
31.75, 29.54, 29.17, 22.55, 14.42.

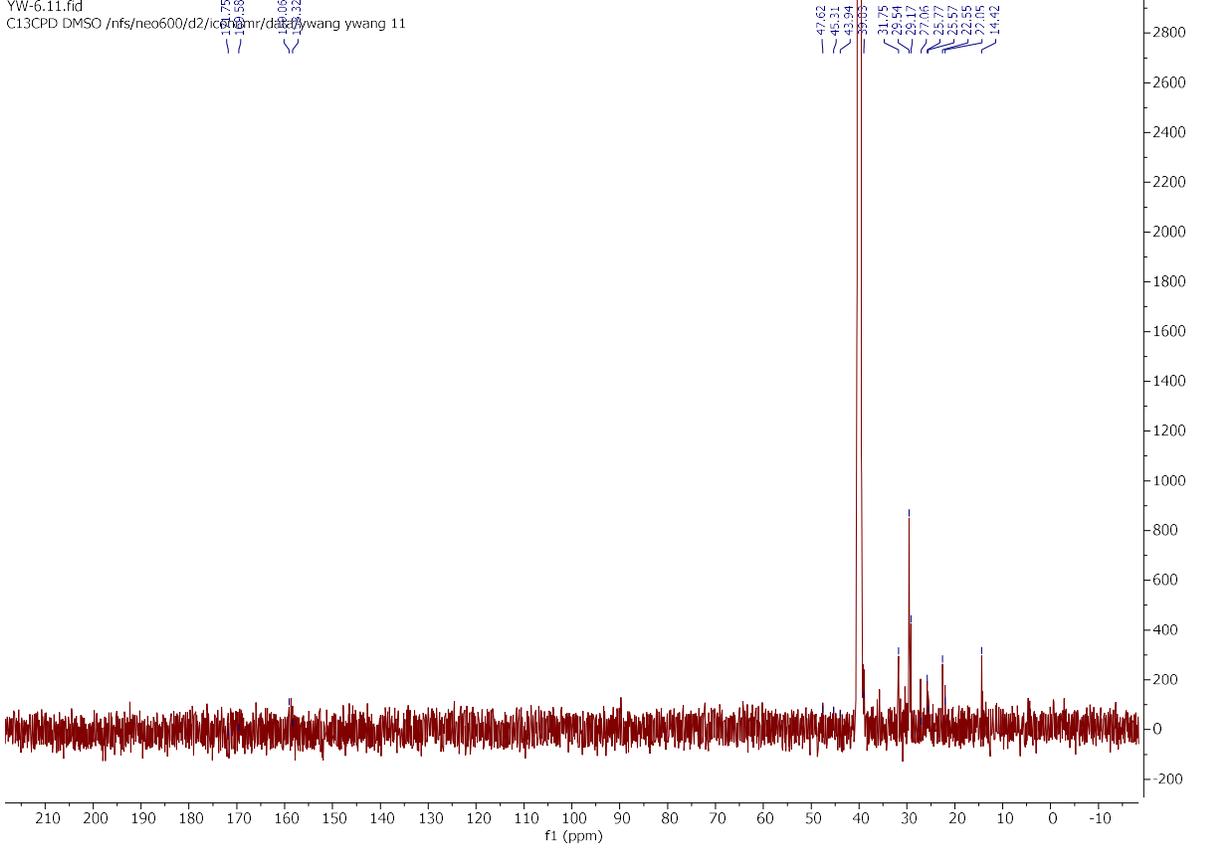
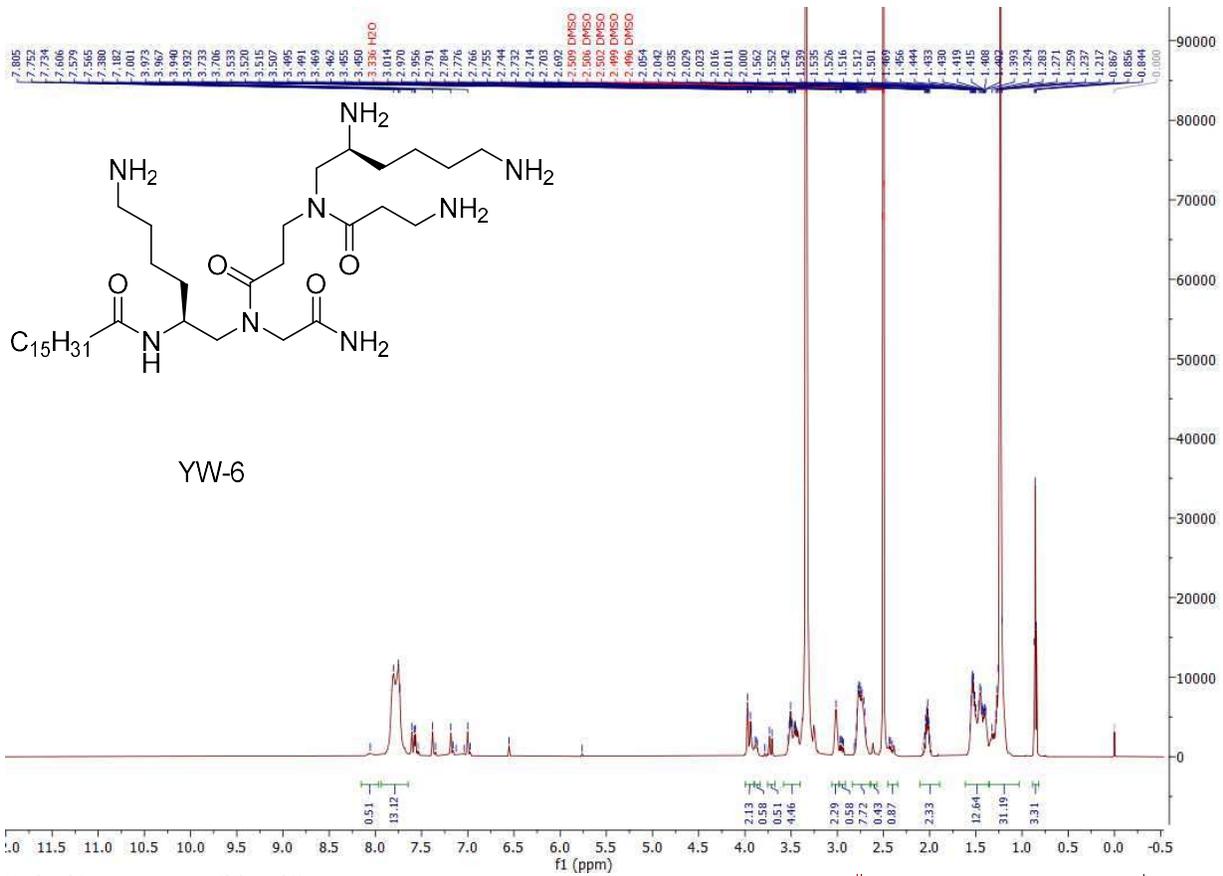
YW-7:

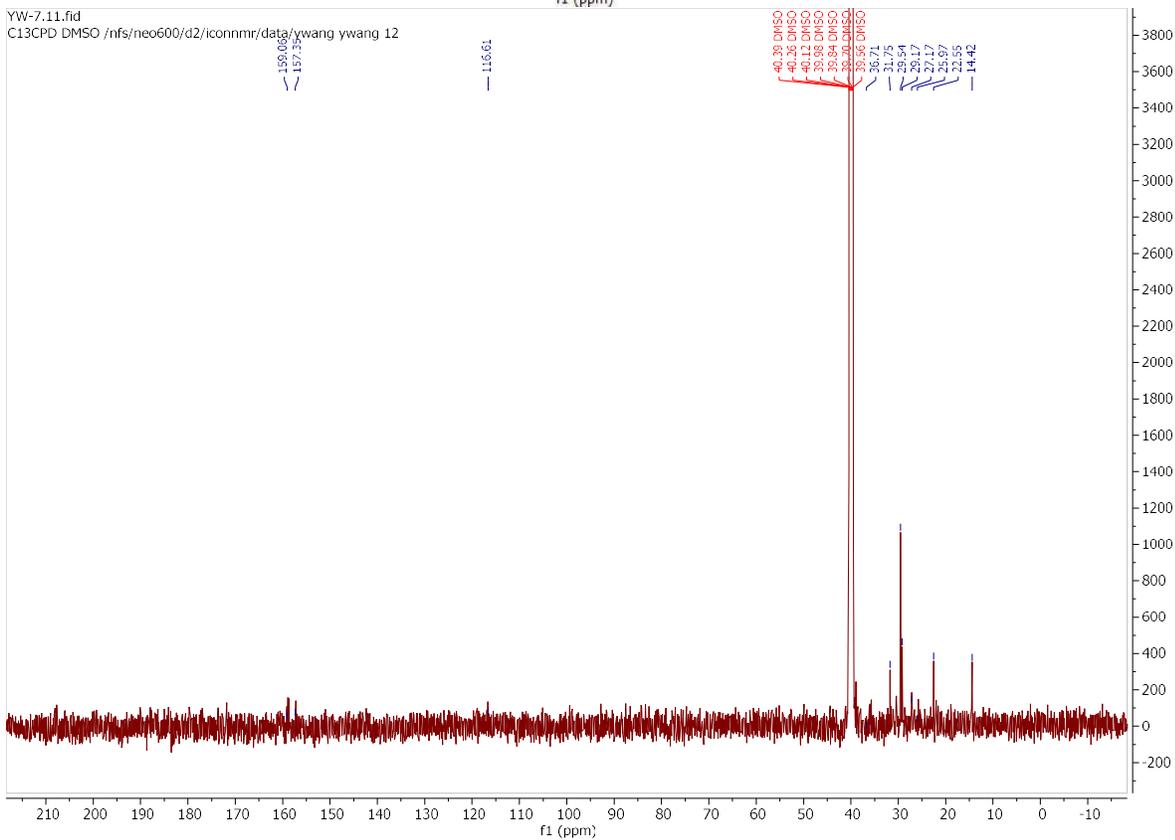
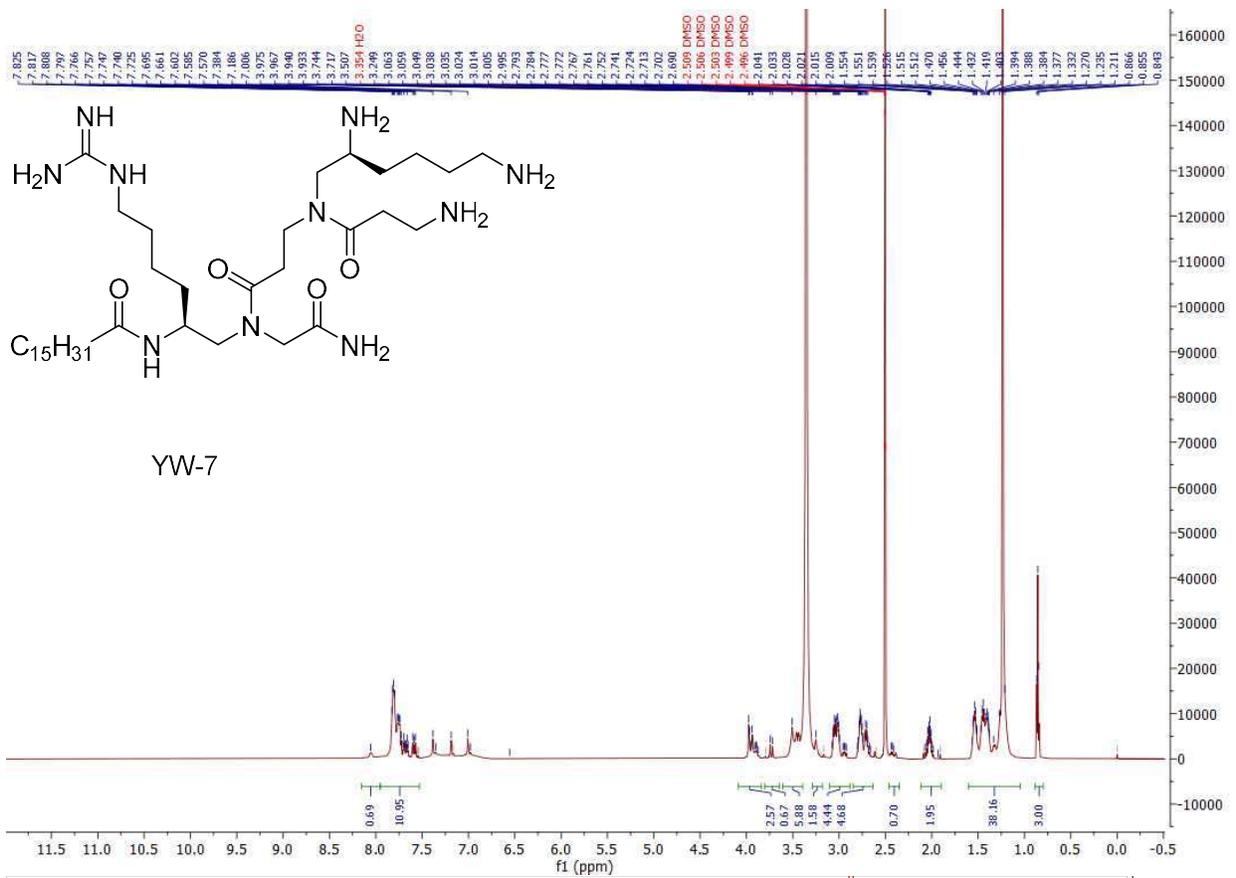
¹H NMR (600 MHz, DMSO-d₆) δ 8.05 (s, 1H), 7.95 – 7.53 (m, 11H), 4.09 – 3.84 (m, 3H), 3.73
(d, J = 16.5 Hz, 1H), 3.51 (s, 6H), 3.25 (s, 2H), 3.10 – 2.88 (m, 4H), 2.85 – 2.63 (m, 5H), 2.46 –
2.35 (m, 1H), 2.02 (dq, J = 10.1, 6.9 Hz, 2H), 1.60 – 1.05 (m, 38H), 0.85 (t, J = 6.9 Hz, 3H).
¹³C NMR (151 MHz, DMSO-d₆) δ 172.04, 171.95, 158.82, 158.66, 157.24, 118.85, 117.36, 50.22,
47.53, 44.10, 39.00, 36.22, 36.07, 35.73, 31.75, 29.54, 29.17, 22.55, 14.42.



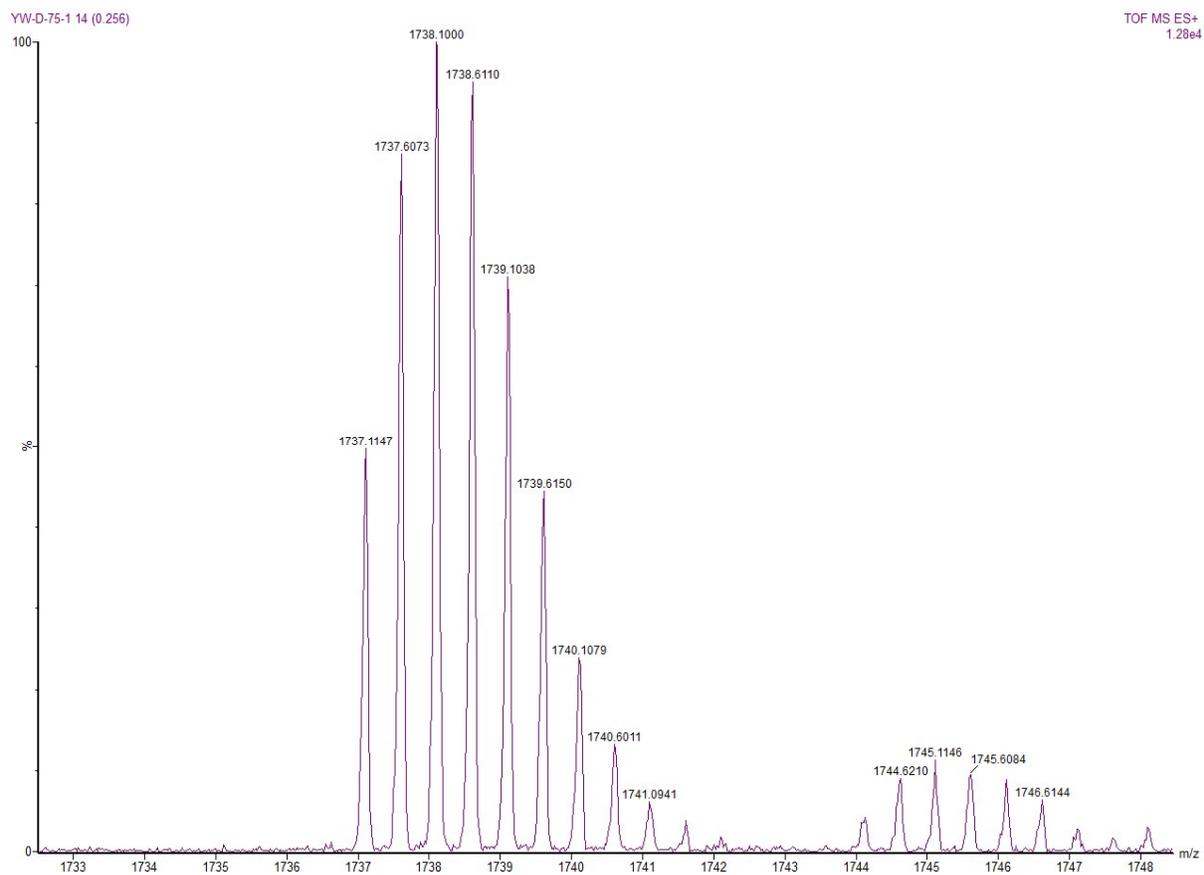








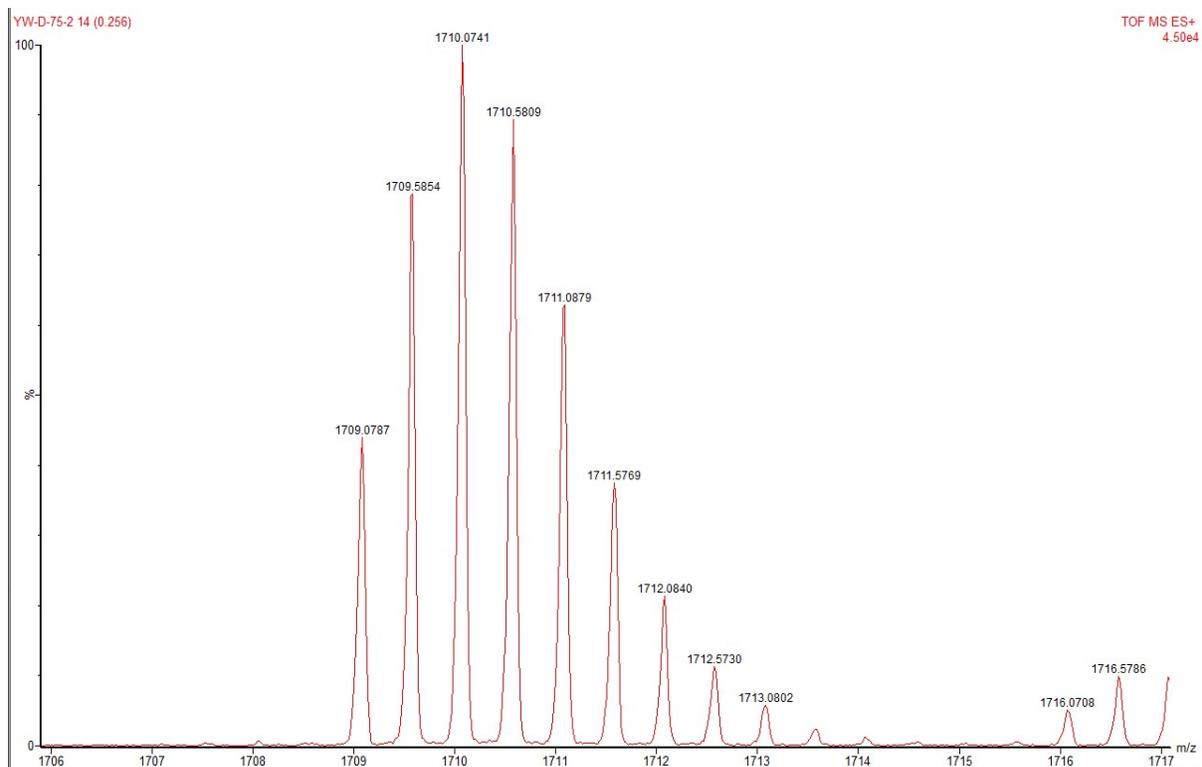
Appendix D: ESI-MS Data



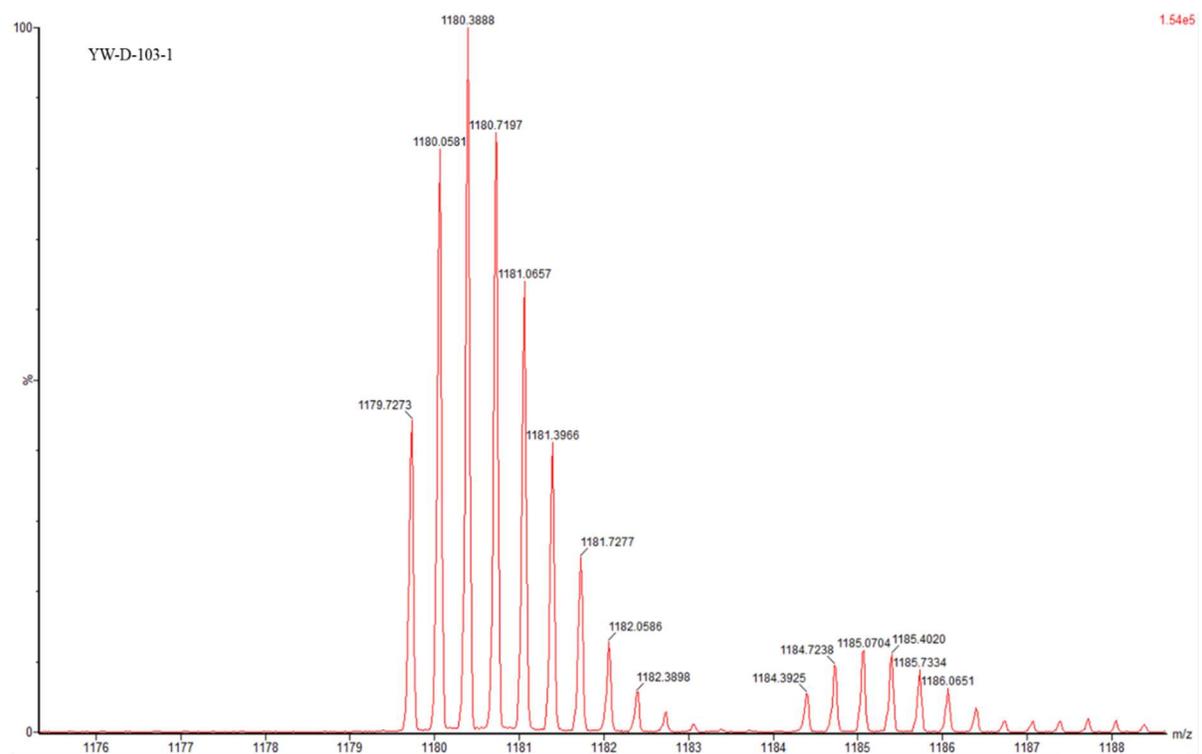
YW-D-75-1 ((E.M.+2)/2)

Chemical Formula: $C_{141}H_{258}N_{40}O_{38}S_{11}$

Exact Mass: 3471.6414



YW-D-75-2 ((E.M.+2)/2)
 Chemical Formula: $C_{137}H_{250}N_{40}O_{38}S_{11}$
 Exact Mass: 3415.5788

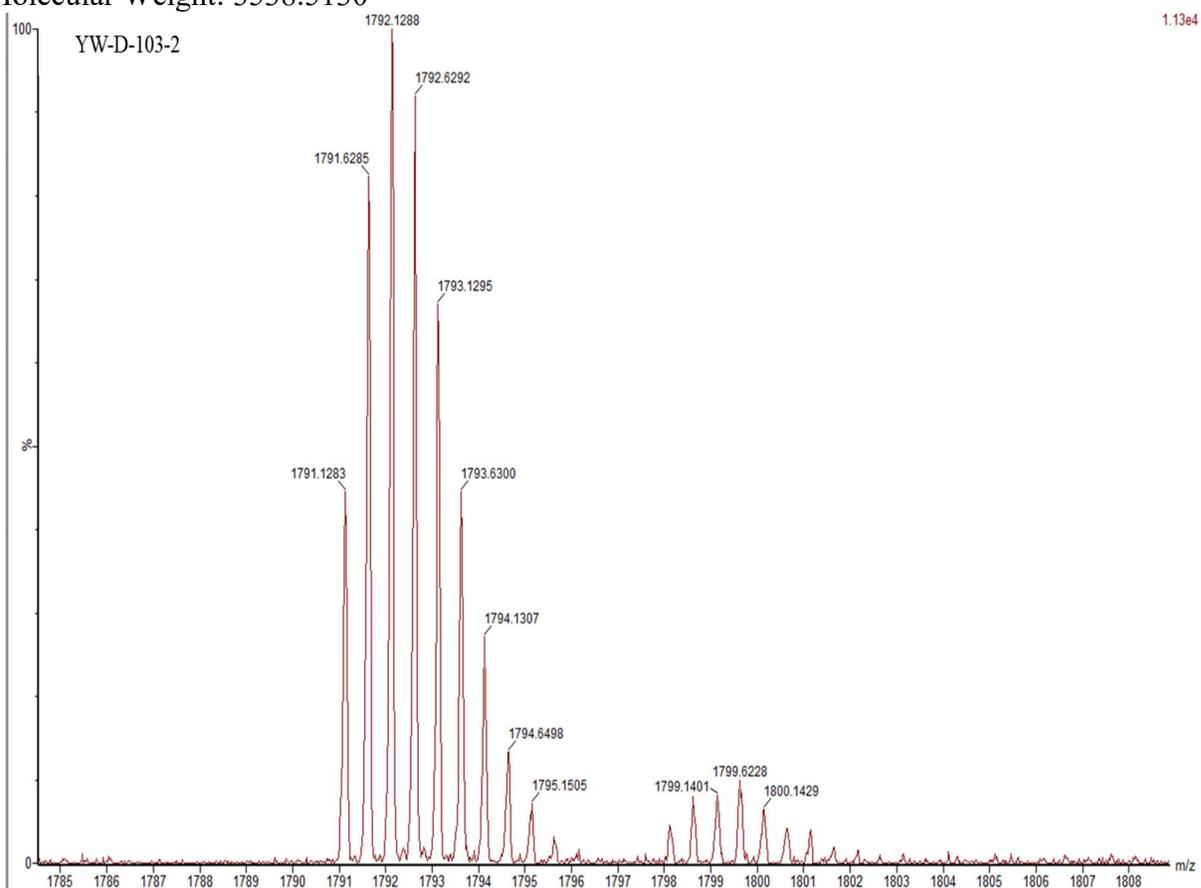


YW-D-103-1 ((E.M.+2)/2)

Chemical Formula: $C_{141}H_{258}N_{40}O_{42}S_{11}$

Exact Mass: 3535.6210

Molecular Weight: 3538.5130

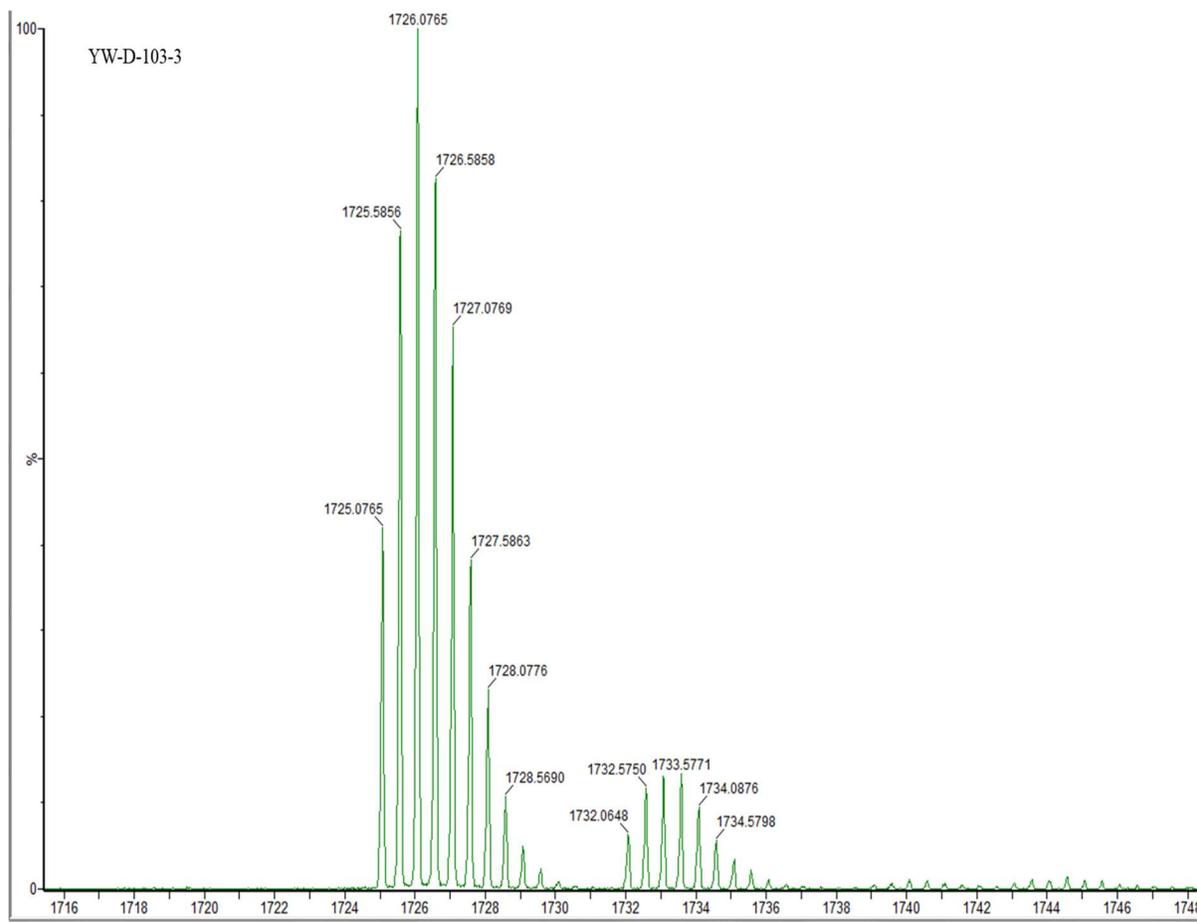


YW-D-103-2 ((E.M.+2)/2)

Chemical Formula: $C_{143}H_{262}N_{40}O_{43}S_{11}$

Exact Mass: 3579.6472

Molecular Weight: 3582.5660

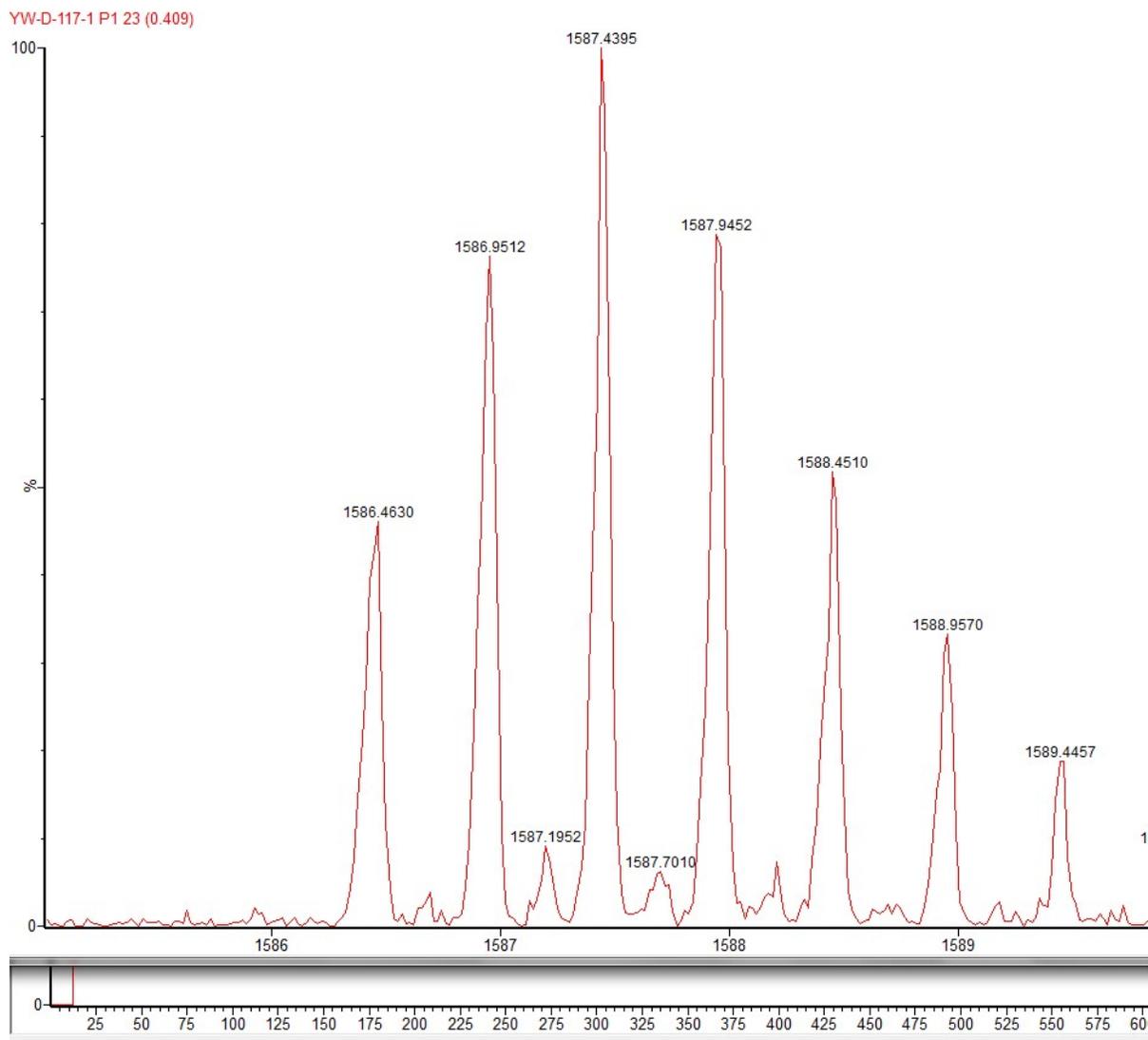


YW-D-103-3 ((E.M.+2)/2)

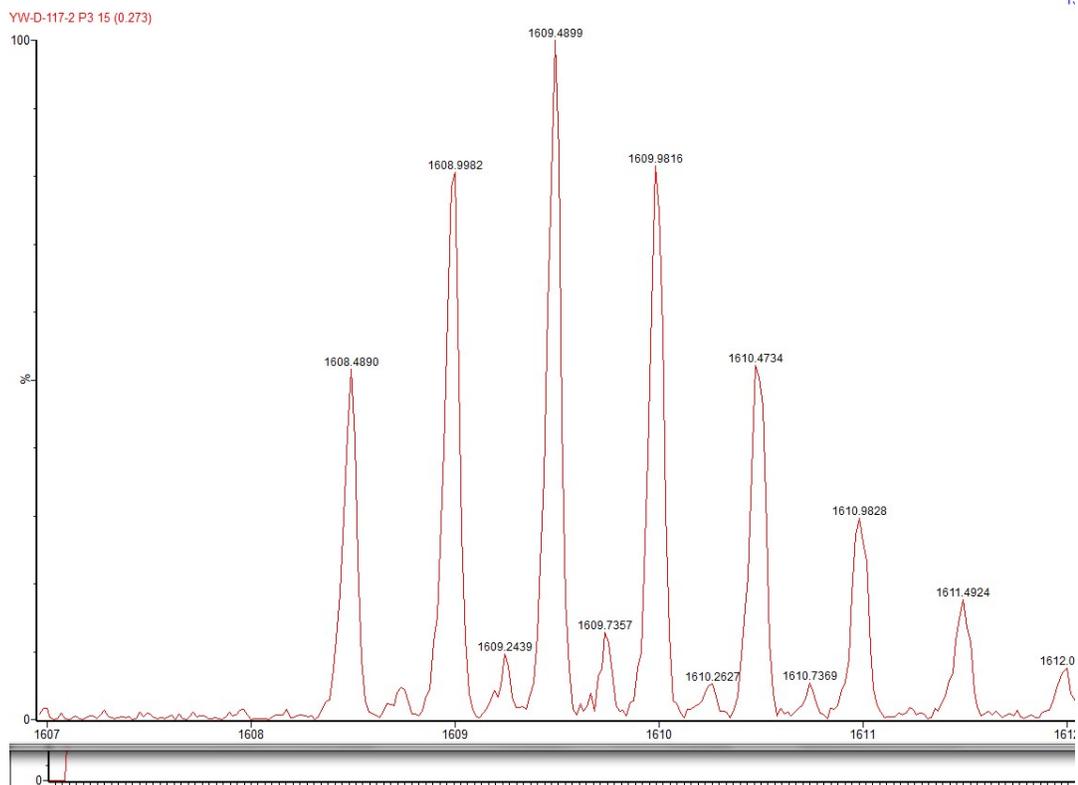
Chemical Formula: $C_{137}H_{250}N_{40}O_{40}S_{11}$

Exact Mass: 3447.5686

Molecular Weight: 3450.4070

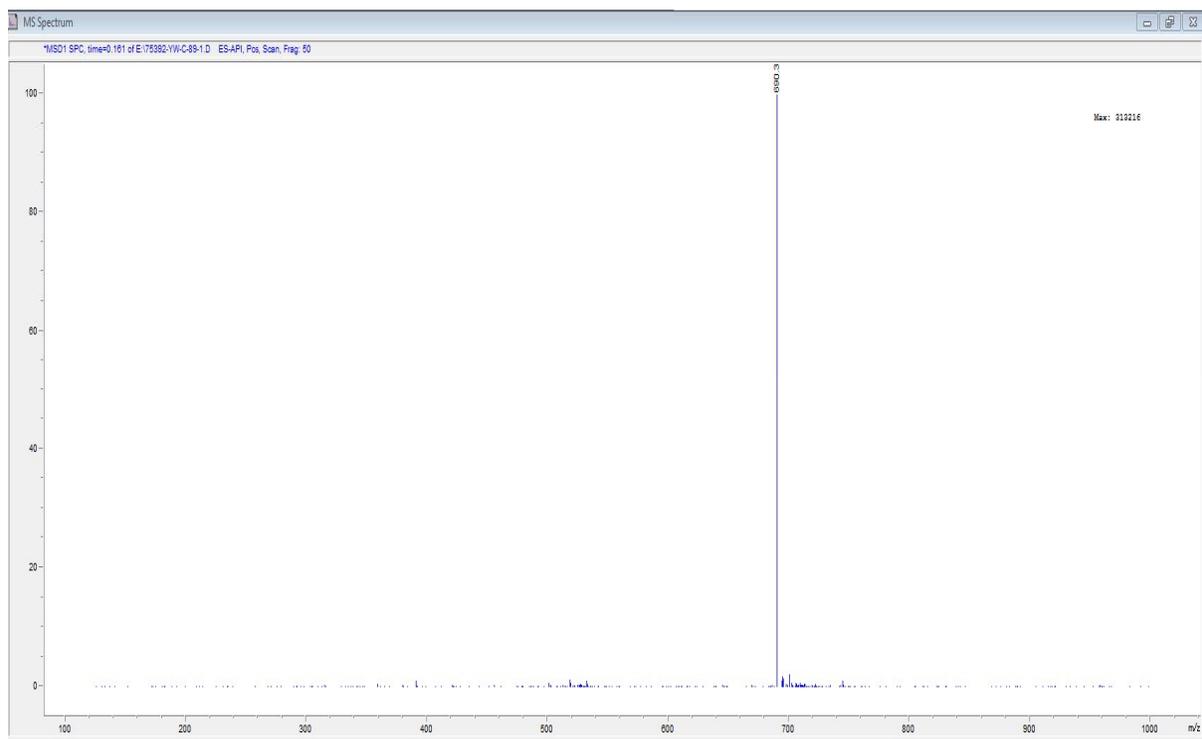


YW-D-117-1 ((E.M.+2)/2)
Chemical Formula: $C_{124}H_{231}N_{40}O_{40}S_{10}$
Exact Mass: 3170.4325

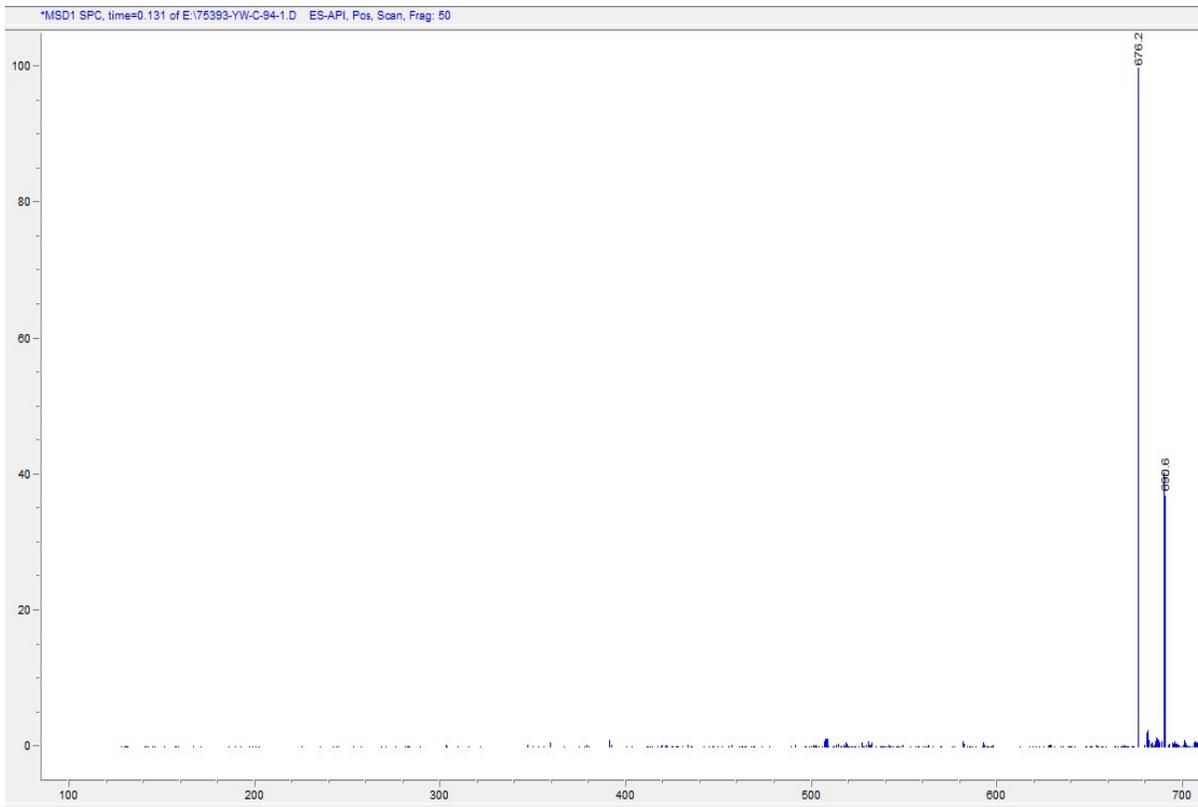


YW-D-117-2 ((E.M.+2)/2)
Chemical Formula: $C_{126}H_{235}N_{35}O_{41}S_{10}$
Exact Mass: 3214.4587

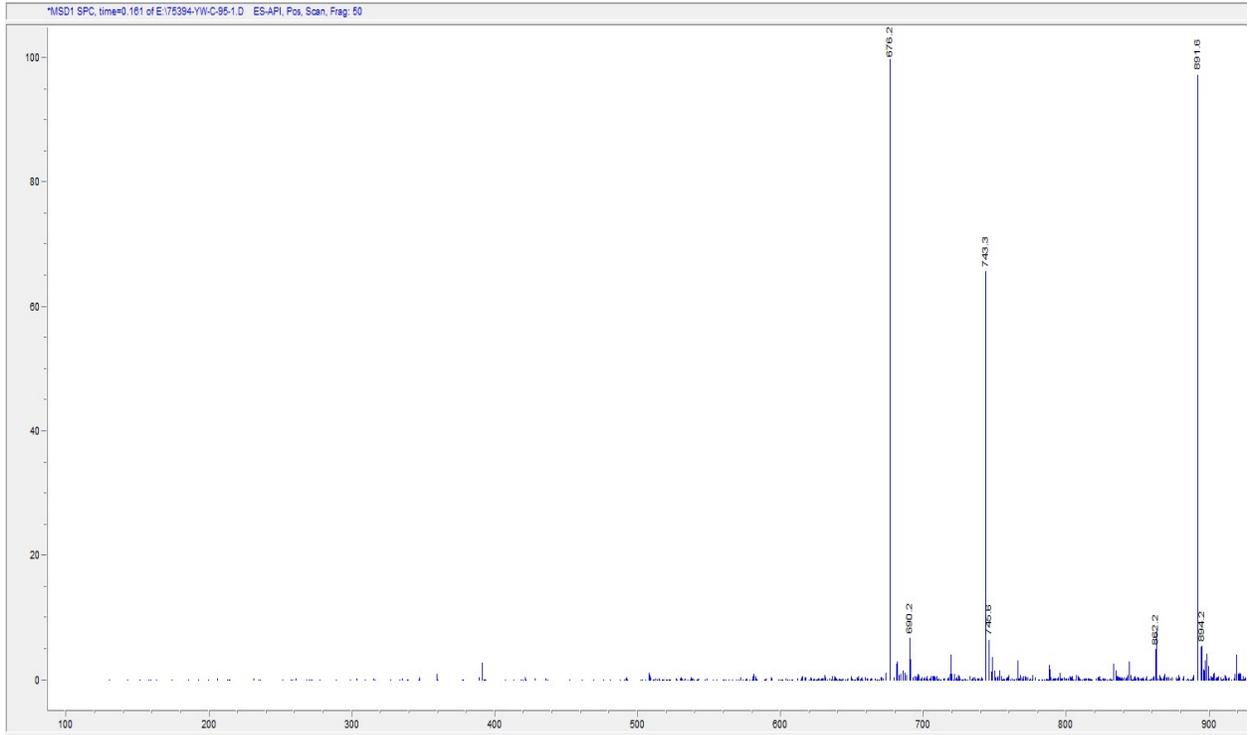
Appendix E: LC-MS and ESI-MS data



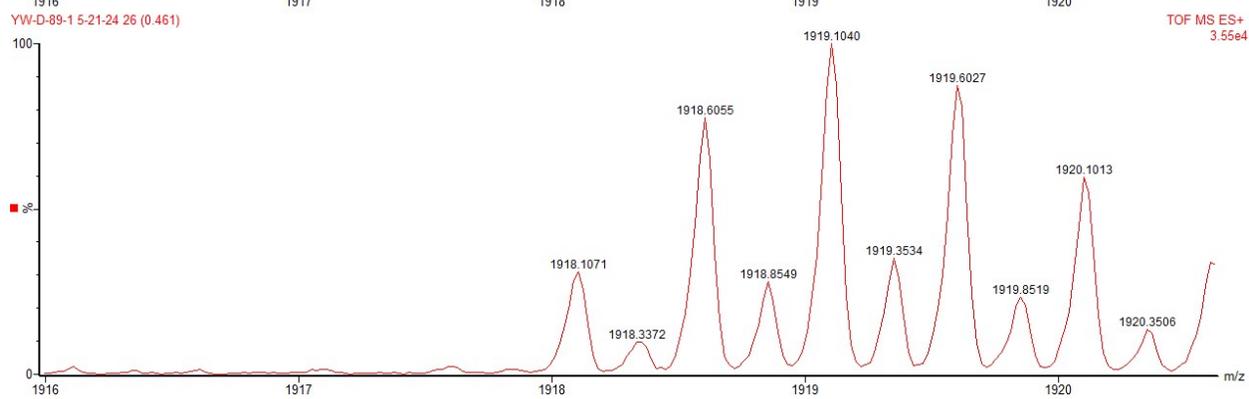
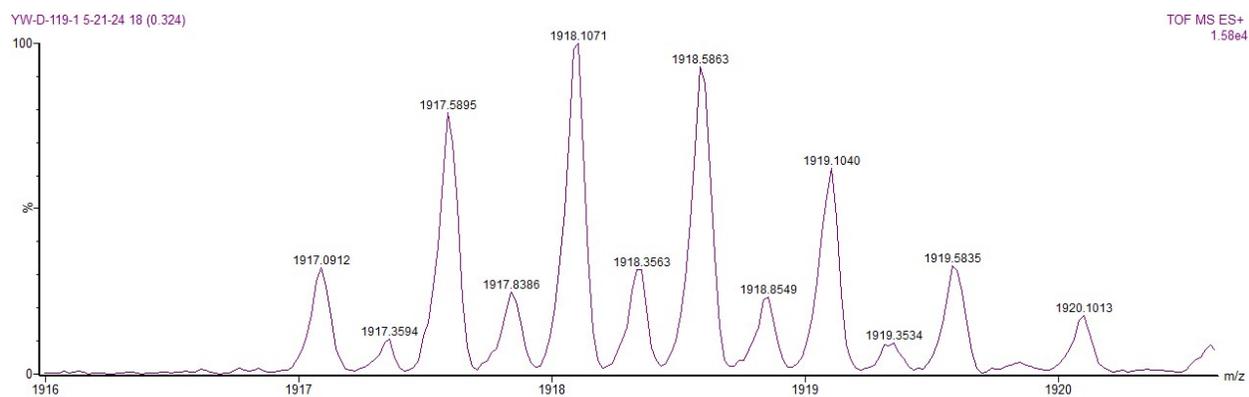
YW-C-89-1
((M.W.+3)/3)



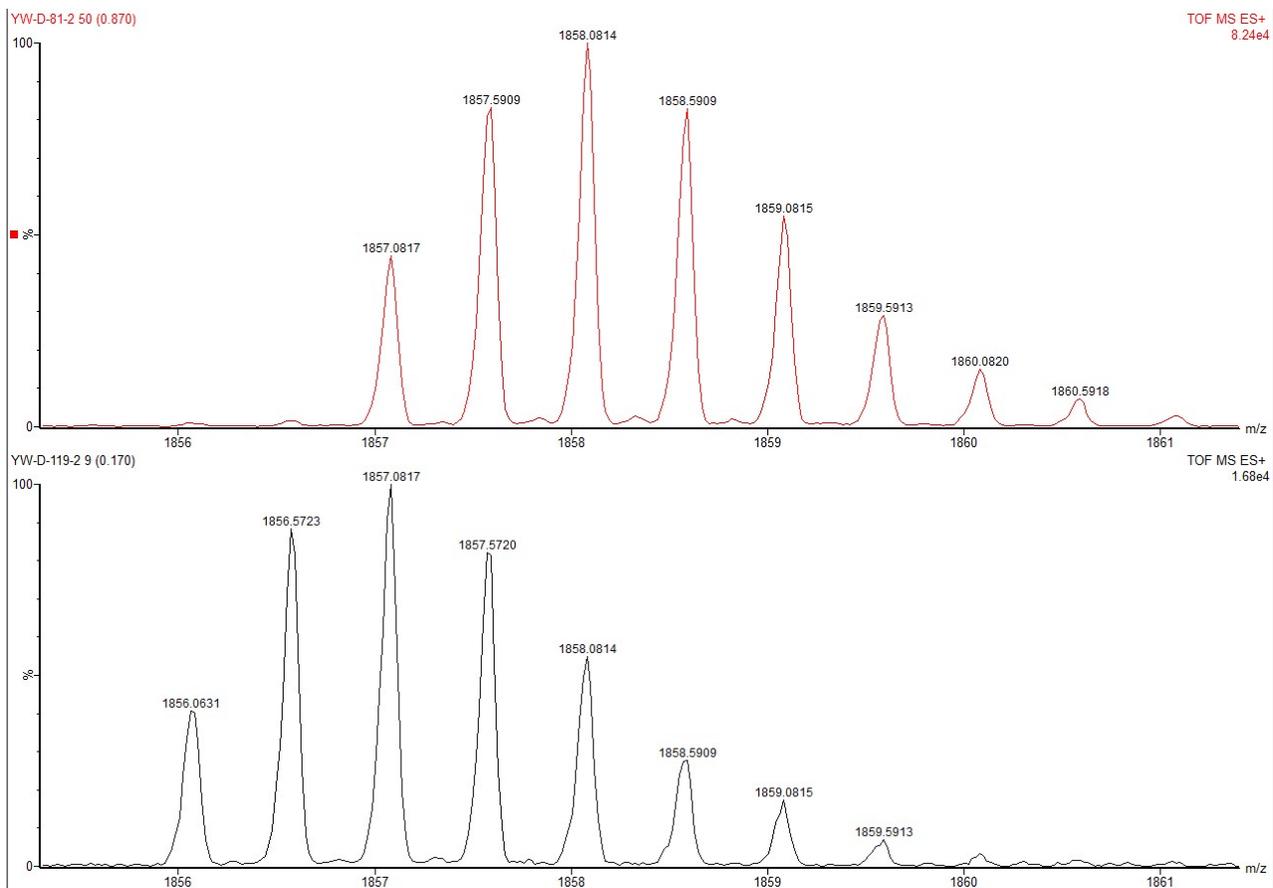
YW-C-94-1
((M.W.+3)/3)



YW-C-95-1
((M.W.+6)/6)



YW-D-119-1
((E.M.+2)/2)



YW-D-119-2
(E.M.+2)